### ISO

#### INTERNATIONAL ORGANIZATION FOR STANDARDIZATION

# ISO RECOMMENDATION R 1908

iTeh STANDARD PREVIEW
CRESYLIC ACID AND XYLENOLS
(standards iteh.ai)
FOR INDUSTRIAL USE

ISO/R 1908:1971

TEST'SFORGABSENCESOFGHYDROGEN 3002PHIDE 000c-

1st EDITION

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#### **BRIEF HISTORY**

The ISO Recommendation R 1908, Cresylic acid and xylenols for industrial use — Test for absence of hydrogen sulphide, was drawn up by Technical Committee ISO/TC 47, Chemistry, the Secretariat of which is held by the Ente Nazionale Italiano di Unificazione (UNI).

Work on this question led to the adoption of Draft ISO Recommendation No. 1908, which was circulated to all the ISO Member Bodies for enquiry in November 1969. It was approved, subject to a few modifications of an editorial nature, by the following Member Bodies:

Australia Israel Spain Belgium Italy Switzerland Chile Japan Thailand Netherlands PRE New Zealand Poland ds.iteh.ai) Portugal Czechoslovakia Turkey France U.A.R. Germany United Kingdom Greece U.S.S.R.

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No Member Body opposed the approval of the Draft/iso-r-1908-1971

This Draft ISO Recommendation was then submitted by correspondence to the ISO Council, which decided to accept it as an ISO RECOMMENDATION.

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R 1908

May 1971

### CRESYLIC ACID AND XYLENOLS FOR INDUSTRIAL USE

#### TEST FOR ABSENCE OF HYDROGEN SULPHIDE

WARNING. These materials burn the skin and can be absorbed into the system through the skin. It is essential for the sampler to wear protective gloves, for example of polyvinyl chloride, and also a face shield. Inhalation of the vapours from hot material is to be avoided.

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#### ISO/R 1908:1971

#### 1. SCOPE AND FIELD OF APPLICATION iteh.ai/catalog/standards/sist/bc5c22c3-0862-43eb-800c-

This ISO Recommendation describes a method of testing for the absence of hydrogen sulphide and is applicable to cresylic acid of high *m*-cresol content, cresylic acid of high *o*-cresol content and xylenols, for industrial use.

NOTE. - This is a simple qualitative test for the absence of hydrogen sulphide and is in no way quantitative.

#### 2. SAMPLING

Apply the principles given in ISO Recommendation . . .\*. The following principles should also be observed :

Place the laboratory sample representative of the material taken from the bulk in a clean, dry, dark-coloured, glass-stoppered bottle of such a size that it is nearly filled by the sample. If it is necessary to seal this bottle, care should be taken to avoid contaminating the contents.

#### 3. PRINCIPLE

Formation of a brown colour on lead acetate paper due to reaction with any hydrogen sulphide present.

<sup>\*</sup> Sampling of chemical products will form the subject of a future ISO Recommendation.

#### 4. REAGENT

Distilled water or water of equivalent purity should be used in the test.

#### 4.1 Lead acetate paper

Immerse strips of double acid washed filter paper, approximately 50 mm  $\times$  100 mm, for 1 minute in a solution prepared by dissolving 10 g of lead acetate in 90 ml of water and adding 5 ml of glacial acetic acid and 10 ml of pure glycerol. Drain off the liquid from the strips, suspend them in a hydrogen sulphide free atmosphere and allow to dry as completely as possible at ambient temperature. Then trim off and discard 25 mm from the top and bottom of each strip. Cut the strips into 25 mm squares and store in a glass-stoppered, airtight container.

#### 5. APPARATUS

Ordinary laboratory apparatus.

#### 6. PROCEDURE

Measure 20 ml of the laboratory sample into a 100 ml conical flask.

Place a lead acetate paper (4.1) over the mouth of the flask and heat the latter in a boiling water bath for 5 minutes.

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#### 7. EXPRESSION OF RESULTS

Report whether any brown as distinct from yellow colour has developed indicating the presence of hydrogen suphide.

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#### 8. TEST REPORT

The test report should give the following particulars:

- (a) the reference of the method used;
- (b) the results and the method of expression used;
- (c) any unusual features noted during the determination;
- (d) any operation not included in this ISO Recommendation or regarded as optional.

#### **ANNEX**

This document forms one of a series of ISO Recommendations on methods of test for phenol, cresols, cresylic acid and xylenols for industrial use.

The complete list of the Recommendations already prepared or in course of preparation is as follows:

#### PHENOL, o-CRESOL, m-CRESOL, p-CRESOL, CRESYLIC ACID, XYLENOLS

ISO/R 1897, Determination of water by the Karl Fischer method. ISO/R 1898, Determination of water by the Dean and Stark method. ISO/R 1899, Determination of neutral oils and pyridine bases.

#### PHENOL, o-CRESOL, m-CRESOL, p-CRESOL

ISO/R 1900, Determination of residue on evaporation.

ISO/R 1901, Determination of crystallizing point.

ISO/R 2208, Determination of crystallizing point after drying with a molecular sieve.\*

ISO/R 1902, Test for impurities insoluble in sodium hydroxide solution - Visual test.

ISO/R 2273, Determination, after combustion, of total sulphur (conductimetric method) and chlorine content (potentiometric or spectrophotometric method.\*

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#### LIQUEFIED PHENOL, m-CRESOL, CRESYLIC ACID, XYLENOLS

ISO/R 1903, Determination of density at 20 °C. ISO/R 1908:1971 https://standards.iteh.ai/catalog/standards/sist/bc5c22c3-0862-43eb-800c-7234b9c7f51d/iso-r-1908-1971

#### **PHENOL**

ISO/R 1904, Determination of phenol content – Bromination method.\*

#### LIQUEFIED PHENOL

ISO/R 1905, Test for impurities insoluble in water – Visual test.

#### CRESYLIC ACID AND XYLENOLS

ISO/R 1906, Determination of distillation range. ISO/R 1907, Determination of residue on distillation. ISO/R 1908, Test for absence of hydrogen sulphide. ISO/R 1909, Measurement of colour.

ISO/R 1910, Determination of o-cresol content.

#### CRESYLIC ACID

ISO/R 1911, Determination of m-cresol content.

NOTE.-A laboratory sample of not less than 500 ml (for phenol and cresols) or 1000 ml (for cresylic acid and xylenols) is necessary to carry out the whole series of tests described in these documents.

At present at the stage of Draft ISO Recommendation.

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