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Fine ceramics (advanced ceramics, advanced technical ceramics) — Test method for determination of photocatalytic activity on semiconducting photocatalytic materials by dissolved oxygen iTeh STCONSUMPTION EVIEW

(standards, iteh.a) Méthode d'essai relative à la détermination de l'activité photocatalytique sur matériaux photocatalytiques semiconducteurs par la consommation d'oxygène dissous https://standards.iteh.ai/catalog/standards/sist/773c4147-316e-4e6f-bbac-54cc0c892c31/iso-19722-2017



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ISO copyright office Ch. de Blandonnet 8 • CP 401 CH-1214 Vernier, Geneva, Switzerland Tel. +41 22 749 01 11 Fax +41 22 749 09 47 copyright@iso.org www.iso.org

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Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

The procedures used to develop this document and those intended for its further maintenance are described in the ISO/IEC Directives, Part 1. In particular the different approval criteria needed for the different types of ISO documents should be noted. This document was drafted in accordance with the editorial rules of the ISO/IEC Directives, Part 2 (see www.iso.org/directives).

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For an explanation on the meaning of ISO specific terms and expressions related to conformity assessment, as well as information about ISO's adherence to the World Trade Organization (WTO) principles in the Technical Barriers to Trade (TBT) see the following URL: www.iso.org/iso/foreword.html.

This document was prepared by Technical Committee ISO/TC 206, *Fine ceramics*.

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Introduction

International Standards covering test methods for determination of photocatalytic activity have been published. A wide variety of photocatalytic functions, such as water and air purification, antibacterial effect, and self-cleaning, require different evaluation methods. However, much easier methods to evaluate a common semiconducting photocatalytic activity are strongly demanded, in particular in research and development activities for testing of performance of semiconducting photocatalyst and photocatalytic materials under development.

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Fine ceramics (advanced ceramics, advanced technical ceramics) — Test method for determination of photocatalytic activity on semiconducting photocatalytic materials by dissolved oxygen consumption

1 Scope

This document specifies the test method for determination of concentration of dissolved oxygen consumed due to photocatalytic oxidation of phenol in aqueous phase by semiconducting photocatalytic substances. The method is applicable to powder test sample or film test piece of semiconducting photocatalystic material targeting water contaminants. This test method is not applicable for evaluating the materials conjugated with other base material, such as organic binder which can also be decomposed by the photocatalytic activity.

This document is applicable to the test method for the activity of powder test sample or film test piece of semiconducting photocatalystic material targeting water contaminants.

2 Normative references STANDARD PREVIEW

The following documents are referred to in the text in such a way that some or all of their content constitutes requirements of this document. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

ISO 5814, Water quality \sim Determination of dissolved oxygen $_{47}$ Electrochemical probe method

ISO 10677, Fine ceramics (advanced ceramics, advanced technical ceramics) — Ultraviolet light source for testing semiconducting photocatalytic materials

ISO/IEC 17025, General requirements for the competence of testing and calibration laboratories

3 Terms and definitions

For the purposes of this document, the terms and definitions given in ISO 20507 and the following apply.

ISO and IEC maintain terminological databases for use in standardization at the following addresses:

- IEC Electropedia: available at http://www.electropedia.org/
- ISO Online browsing platform: available at http://www.iso.org/obp

3.1

semiconducting photocatalyst

substance that displays photocatalytic action based on its electronic band structure

Note 1 to entry: This applies to metal oxides like titanium dioxide and sulfides. Photocatalysts which are not semiconducting includes metal complexes.

3.2

photocatalytic materials

material in which or on which the photocatalyst is added by coating, impregnation, mixing, etc.

Note 1 to entry: Materials include ceramic, metal, plastic, cloth, etc. for general purpose.

3.3

DO

dissolved molecular oxygen in aqueous phase

3.4

DO analyser

measuring instrument for continuous measurement of DO(3.3) in aqueous using DO electrode (3.5)

3.5

DO electrode

electrode to measure DO(3.3) in aqueous phase

3.6

photocatalytic oxygen demand

POD

quantity of molecular oxygen in aqueous phase consumed in photocatalysis

3.7

blank POD(%)

percentage of concentration of *DO* (3.3) consumed under a test condition without phenol addition

4 Symbols

Designation	Symbol	Unit
room temperature en STANDARD		₩°C
water temperature (standards i	teh W.T.	°C
concentration	C	mol/l
concentration of DO ISO 19722:20	₁₇ c _{D0}	mg/l
photocatalyticoxygenademandi/catalog/standards/sis	t/773c4 P0D 316e-4e	6f-b mg/ l
c _{DO} before UV light irradiation the dark /iso-19	$722-201\overline{c}_{i\rm DO}$	mg/l
<i>c</i> _{DO} after UV light irradiation in the dark	c_{fDO}	mg/l
volume of test solution	V	ml
wavelength	λ	nm
UV light irradiation intensity	Ι	mW/cm ²

5 Principle

Photocatalysis in pure water generally produces molecular $oxygen(O_2)$ from water molecule (H₂O) oxidation^[3]. In the water polluted by some of the organic compounds, major photocatalysis oxidizes the organic compounds^[4]. Photocatalyst needs O_2 to oxidize organic compounds to CO_2 and water in the environment^[4]. Then the quantity of O_2 that photocatalysis needs is much larger than O_2 production from H₂O oxidation because the organic compound is photocatalytically oxidized much easier than water molecule. O₂ then has major three functions in semiconducting photocatalyst. The first function is to improve charge separation by accepting conduction band electron. The second one is to produce active oxygen species that have the ability to oxidize organic compounds. The final one is oxidation. O₂ combines with organic radicals (intermediates) produced by semiconducting photocatalytic oxidation; O_2 is an indispensable species in the semiconducting photocatalysis^[4]. Therefore, the photocatalysis to oxidize organic compound means O₂ consumption. In the photocatalysis to oxidize and mineralize the organic compounds, partially oxygenated by-products are produced. Under the progress in the continuous oxidation of partially oxygenated by-products, the photocatalysis consume O_2 to mineralisation. On the basis of the photocatalytic mineralization and functions of O_2 , the semiconducting photocatalytic activity can be evaluated by determining O₂ consumption. This test method is especially effective in the photocatalysis in aqueous phase. Target photocatalytic materials are either powder test samples or a film test pieces.

6 Materials

6.1 Reagent

Reagent is phenol and the assay is >99 wt%.

6.2 Purified water

Water used for the preparation of all solutions shall be distilled or deionised water.

6.3 Purified air

Air in the atmosphere aerated through 1 000 ml purified water.

6.4 Purified water saturated with dissolved oxygen

Purified water at R.T. ± 1 °C saturated with DO.

6.5 Test solution

Suspension with the powder test sample in phenol solution or phenol solution for the film test piece.

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7 Test apparatus Teh STANDARD PREVIEW

7.1 General

Apparatus shall be used to evaluate the semiconducting photocatalytic materials with a suitable test method. The powder test sample is suspended and the film test piece is immersed in the test solution. The following apparatus is required $\frac{4}{4cc0c892c31/iso-19722-2017}$

7.2 Ultraviolet (UV) — Irradiation light source

Use black light fluorescence lamps as black light lamp (BL) and black light blue lamps (BLB). The black light fluorescence lamps shall have a peak wavelength $\lambda = 351$ nm as specified in ISO 10677.

7.3 UV radiometer

A radiometer with a detector whose sensitivity peak is at $\lambda = 351$ nm shall be used to measure the UV-light intensity. The radiometer shall be calibrated to closely match the characteristic of the UV light irradiation light source as specified in ISO 10677 or be corrected to ascertain sensitivity within the wavelength range to be adsorbed by the powder test sample or the film test piece with suitable approaches.

7.4 UV light intensity

I is adjusted to be 1,5 mW/cm² at the centre of a test vessel for the powder test sample or at the centre of the film test piece surface for the film test piece (see <u>Annex C</u>).

7.5 DO analyser

To measure c_{DO} , an electrode of DO analyser has an oxygen permeable membrane and the performance of electrode and equipment is specified in ISO 5814. Sensitivity correction and operation of the electrode and the DO analyser shall be performed following their manuals of suppliers and manufactures. Notice that portable equipment does not have stability for voltage, if indicated values are unstable, it is necessary to use stabilised power supplies.

7.6 Magnetic stirrer and magnetic stirring bar

A rotational number for stirring is 1 400 r/min to 2 100 r/min. The size of a magnetic stirring bar is approximately o.d. 7 mm \times L 20 mm which is adjusted to be the size of the test vessel. Digital laser tachometers (non-contact system) are suitable for measurement of a rotational number. The rotational number is measured in the air.

8 Arrangement of test method

8.1 Measuring device setup

The apparatus shall be used to evaluate the activity of photocatalytic materials by measuring a decrease in c_{DO} in the test solution with photo irradiation necessary for the photocatalytic reaction after suspending the powder test sample or immersing the film test piece in the test solution, consisting of the test vessel, the light source, the DO analyser, a thermostatic bath, and a pump. This test is in a closed system to measure the decrease in c_{DO} in the test solution. An example of the measuring device setup is shown in Annex A.

8.2 Test vessel and implement

Schematic diagrams are shown in <u>Annex B</u>. The test vessel and glass implement shall be made of borosilicate glass which can resist near-UV light irradiation of $\lambda > 300$ nm and absorbs less UV. The test vessel for the test solution shall be cylindrical and the volume shall be 200 ml.

The test solution temperature has to be kept within a certain definite range during the test. A water jacket in which water can circulate at a constant temperature shall be used and a test vessel holder (upper and lower) to fix the test vessel inside the water jacket shown in <u>Annex C</u> shall be prepared. A holder is held has O-rings to keep airtightness of the closed system and with a through-hole as well to overflow extra suspension/test solution because the amount of suspension/test solution equal to the DO electrode volume is overflowed when the DO electrode is inserted into the holder. A plug shall be used to close the through-hole following the step. The holder has a space to hold the film test piece. A spacer adjusted to a thickness of the film test piece shall be used to fill a gap in a holding part for the film test piece is held in the holder. In the case of the suspension with the powder test sample, the spacer to fill an orifice of the holding part for the film test piece shall be used. This test wessel shall be made of airtight, chemical resistant, and non-photoresponsive material and be able to keep the W.T. constant and be what an adequate amount of light for the photocatalytic activity is gained under stirring. The O-rings and other implements shall be near-UV resistant and chemical resistant, and it is preferable to use fluoro-carbon polymer for the O-rings and polytetrafluoroethylene for the others.

9 Test material

9.1 Powder test sample

The powder test sample is the photoctalytic material synthesized for treatment of environmental pollution and does not contain so many organic compounds as plastic cement. The quantity of powder test sample shall be 0,110 g \pm 0,005 g.

9.2 Film test piece

Sample size of the film test piece shall be as follows:

- width: 29,5 mm ± 0,5 mm;
- length: 59,5 mm ± 0,5 mm;
- thickness: $5 \text{ mm} \pm 0,5 \text{ mm}$.

The film test piece is held vertically (lengthwise) using a holder and a minimum effective area of $49,5 \text{ mm} \times 29,5 \text{ mm}$ of the film is subjected to the test.

NOTE 1 If the film test piece cannot maintain the shape in the solution, it is not applicable to an adequate evaluation in this test method.

NOTE 2 A film test piece of less than 5 mm thickness is also available by using adjustable spacer.

10 Procedure of the measurement

10.1 Test temperature

All tests shall be carried out at R.T. of 23 °C ± 5 °C and W.T. shall be at R.T. ± 1 °C. W.T. should be controlled to a certain temperature by thermostatic bath/cryostat.

10.2 Preparation of water

Purified water quality shall be distilled or deionised water. The DO in purified water of 1 000 ml is saturated by aeration of 1 500 ml/min for 60 min and the DO saturation is determined by the DO analyser. Then the c_{DO} obeys the concentration specified in ISO 5814. The air for the aeration should be passed through another 1 000 ml of distilled or deionized water to clean itself.

10.3 Powder test sample

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10.3.1 Preparation of suspension from powder test sample (standards.iteh.ai)

Powder test sample is added into 1 000 ml of the purified water and agitated by a sonicator (within 23 kHz to 43 kHz) for 15 min. After sonication, the suspension is irradiated by a UV lamp at a UV light irradiation intensity of 1 mW/cm² under aeration by purified air in vigorous stirring for 180 min. Then the aeration flow rate shall be above 1 500 ml/min A cylindrical glass vessel (a size of o.d. 80 mm × H 300 mm, the volume is above 1 000 ml) is made of borosilicate glass. After UV light irradiation, the pH value of the suspension shall be within 5 to 7. If it is not such a range, the powder test sample shall be washed to remove materials which change the pH. The temperature of the suspension is adjusted to R.T. Then the saturated c_{D0} shall be the DO value ±0,3 mg/l at the test temperature, which is specified in ISO 5814.

10.3.2 Procedure of the measurement

A volume of 0,5 ml of phenol stock solution is added into 195 ml of the suspension and the concentration of phenol shall be finally 0,33 mmol/l. The magnetic stirring bar is then put into the suspension. Here, the saturated c_{DO} shall be the DO value ±0,3 mg/l at the test temperature, which is specified in ISO 5814. Set the holder into the test vessel with the suspension overflowed. To avoid forming air bubbles in the suspension, insert the DO electrode into the holder with the suspension overflowed and set the plug into the through-hole of the holder. The test vessel is then put into the water jacket. Start stirring and recording the c_{DO} every minute at least for 5 min in the dark. Start and carry on the UV light irradiation for 60 min. After stopping the UV light irradiation, keep recording the c_{DO} every minute for 5 min in the dark. If the c_{DO} becomes 0 before regulated examination time, the time and the POD should be included in the test report.

NOTE The flowing water in the water jacket is purified water and the W.T. is adjusted to R.T.

10.4 Film test piece

10.4.1 Preparation of film test piece

Set the film test piece into the holder and then set the holder in the purified water-filled test vessel. The film test piece is irradiated by a UV lamp at a UV light irradiation intensity of 1 mW/cm^2 under