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Food products — Determination of the total nitrogen content by combustion according to the Dumas principle and calculation of the crude protein content —

Part 2:
Cereals, pulses and milled cereal products

Produits alimentaires — Détermination de la teneur en azote total par combustion selon le principe Dumas et calcul de la teneur en protéines brutes —

Partie 2: Céréales, légumineuses et produits céréaliers de mouture

ICS: 67.060;67.050

ISO/CEN PARALLEL PROCESSING

This draft has been developed within the International Organization for Standardization (ISO), and processed under the **ISO lead** mode of collaboration as defined in the Vienna Agreement.

This draft is hereby submitted to the ISO member bodies and to the CEN member bodies for a parallel five month enquiry.

Should this draft be accepted, a final draft, established on the basis of comments received, will be submitted to a parallel two-month approval vote in ISO and formal vote in CEN.

To expedite distribution, this document is circulated as received from the committee secretariat. ISO Central Secretariat work of editing and text composition will be undertaken at publication stage.

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Foreword

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The procedures used to develop this document and those intended for its further maintenance are described in the ISO/IEC Directives, Part 1. In particular the different approval criteria needed for the different types of ISO documents should be noted. This document was drafted in accordance with the editorial rules of the ISO/IEC Directives, Part 2. www.iso.org/directives

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For an explanation on the meaning of ISO specific terms and expressions related to conformity assessment, as well as information about ISO's adherence to the WTO principles in the Technical Barriers to Trade (TBT) see the following URL: Foreword - Supplementary information

The committee responsible for this document is ISO/TC 34, *Food products*, Subcommittee SC 4, *cereals and pulse*.

This second edition cancels and replaces the first edition, which has been technically revised.

Introduction

For a long time, the Kjeldahl method has been the most frequently used method for the determination of the protein content of food products. However, in recent years, the Kjeldahl method has increasingly been replaced by the Dumas method, which is faster and does not use dangerous chemicals. Although the principles of the two methods are different, both measure the nitrogen content of the product. Nitrogen content can be converted into protein content by using an appropriate factor. The value of this factor varies depending on the relative amounts of different proteins and their amino-acid composition in a given product.

Neither the Dumas nor the Kjeldahl method distinguishes between protein and non-protein nitrogen. In most cases, results obtained by the Dumas method are slightly higher than those of the Kjeldahl method. This is due to the fact that the Dumas method measures almost all of the non-protein nitrogen, whereas the Kjeldahl method measures only a part of it.

Taking into consideration the fact that the protein content of a product calculated by both methods only approximates to the true value, it is a matter of discretion which one is accepted. The most appropriate solution should be the use of a second factor for the elimination of the systematic error caused by the non-protein nitrogen content of the different products. However, this second factor has to be determined for each product, like the existing factors which indicate the ratio of the protein content to the nitrogen content.

Food products — Determination of the total nitrogen content by combustion according to the Dumas principle and calculation of the crude protein content —

Part 2:
Cereals, pulses and milled cereal products

1 Scope

This part of ISO 16634 specifies a method for the determination of the total nitrogen content and the calculation of the crude protein content of cereals, pulses and milled cereal products.

This method, like the Kjeldahl method (see References [1] and [6]), does not distinguish between protein nitrogen and non-protein nitrogen. For the calculation of the protein content, various conversion factors are used (see Annex D).

2 Normative references

The following referenced documents are indispensable for the application of this document. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

ISO 712, *Cereals and cereal products — Determination of moisture content — Reference method*

ISO 6540, *Maize — Determination of moisture content (on milled grains and on whole grains)*

ISO 24557, *Pulses — Determination of moisture content — Air-oven method*

3 Terms and definitions

For the purposes of this document, the following terms and definitions apply.

3.1
nitrogen content
mass fraction of the total nitrogen determined by the procedure specified in this part of ISO 16634

NOTE The mass fraction is expressed as a percentage.

3.2
crude protein content
nitrogen content (3.1) multiplied by a factor, usually 5,7 for wheat, rye and their milled products and 6,25 for others products falling within the scope of this part of ISO 16634

NOTE The factors for calculation of the crude protein content from the total nitrogen content are derived from the Kjeldahl method, which is the reference method for the determination of total nitrogen content. As the

method specified in this part of ISO 16634 uses the same factors as the Kjeldahl method, the validity of these factors has to be verified due to the slight difference in the results obtained with the Kjeldahl and Dumas methods.

4 Principle

Samples are converted to gases by heating in a combustion tube. Interfering components are removed from the resulting gas mixture. The nitrogen compounds in the gas mixture, or a representative part of them, are converted to molecular nitrogen which is quantitatively determined by a thermal-conductivity detector. The nitrogen content is then calculated by a microprocessor.

5 Reagents

Use only reagents of recognized analytical grade or reagents of equivalent purity as specified by instrument manufacturers. Except for the reference materials (see 5.12), all reagents shall be free from nitrogen.

5.1 Carrier gas(es): use either 5.1.1 or 5.1.2.

5.1.1 Carbon dioxide, as pure as possible, but with a minimum CO_2 volume fraction of 99,99 %.

5.1.2 Helium, as pure as possible, but with a minimum He volume fraction of 99,99 %.

5.2 Oxygen, as pure as possible, but with a minimum O_2 volume fraction of 99,99 %.

5.3 Sulfur dioxide and halogen absorbent, to eliminate any sulfur from the sample [e.g. lead chromate ($PbCrO_4$) or steel wool].

5.4 Copper oxide/platinum catalyst, for the post-combustion tube.

Platinum catalyst [5 % of Pt on alumina (Al_2O_3)] is blended with CuO in the ratio 1 part:7 parts or 1 part:8 parts in accordance with the manufacturer's recommendations.

To prevent separation as a result of the different bulk densities of the two materials, it is recommended not to prepare the mixture before filling the tube but to pour the platinum catalyst and copper oxide simultaneously into the post-combustion tube using a suitable funnel.

5.5 Silver or copper wool.

This shall be disaggregated before being inserted into the post-combustion or reduction tube.

5.6 Silica (quartz) or **glass wool** or **cotton wool**, as recommended by the instrument manufacturer.

5.7 Copper or **tungsten (wire, cuttings, turnings** or **powder)**, for the reduction tube.

The use of copper or tungsten in one of these forms can improve the precision of analytical results for samples with low nitrogen contents (about 1 % mass fraction).

5.8 Diphosphorus pentoxide (P_2O_5) or **granulated magnesium perchlorate** [$Mg(ClO_4)_2$], or another suitable drying agent, to fill the drying tubes.

5.9 Hollow corundum spheres or **aluminium oxide pellets**, for the combustion tube.

5.10 Copper oxide (CuO), as filling material for the combustion tube.

5.11 Sodium hydroxide (NaOH), on a support material.

5.12 Aspartic acid ($C_4H_7NO_4$) or **ethylenediaminetetraacetic acid** ($C_{10}H_{16}N_2O_8$) or **glutamic acid** ($C_5H_9NO_4$) or **hippuric acid** ($C_9H_9NO_3$) **standard**, or other suitable reference materials with a known, constant, certified nitrogen content.

The minimum recovery should preferably be 99 % mass fraction.

5.13 Light petroleum, with a boiling range between 30 °C and 60 °C, or **acetone** or **ethanol**.

6 Apparatus

Usual laboratory equipment and, in particular, the following:

6.1 Analytical balance, capable of weighing to the nearest 0,000 1 g.

6.2 Grinding device, appropriate to the nature of the sample.

6.3 Sieve, of nominal opening size 800 µm or 1 mm, made of non-ferrous material.

6.4 Crucibles (e.g. made of stainless steel, quartz, ceramic material or platinum) or **tin capsules** or **nitrogen-free filter paper**, suitable for the Dumas apparatus used.

NOTE 1 Several instruments provided with an automatic sampler are commercially available.

NOTE 2 Some solid samples (e.g. powders) can be pressed to form pellets.

6.5 Dumas apparatus[1], fitted with a furnace able to maintain a given temperature greater than or equal to 850 °C, with a thermal-conductivity detector and suitable device for signal integration.

Suitable Dumas apparatus operates according to the general flowchart given in Annex A, although different arrangements and components may be used.

NOTE Schematic diagrams of three commercially available instruments are shown as examples in Figures B.1, B.2 and B.3.

To avoid leaks, the sealing O-rings shall be slightly lubricated with high-vacuum grease prior to installation.

Experience has shown that it is important to clean all pieces of silicaware and glassware carefully and to remove fingerprints from tubes, using a suitable solvent (e.g. acetone), before inserting them in the furnace.

[1] Elementar Analysensysteme, Sumika Chemical Analysis Service and LECO Instruments produce suitable equipment available commercially. This information is given for the convenience of users of this Technical Specification and does not constitute an endorsement by ISO of this equipment. Equivalent products may be used if they can be shown to lead to the same results.

7 Sampling

A representative sample should have been sent to the laboratory. It should not have been damaged or changed during transport or storage.

Sampling is not part of the method specified in this part of ISO 16634. Recommended sampling methods are given in ISO 24333 [7] for cereals and cereal products.

8 Preparation of test sample

The test sample shall be prepared from the laboratory sample in such a way that a homogeneous test sample is obtained.

Using a suitable grinding device (6.2), grind the laboratory sample. Generally, pass the ground material through a sieve (6.3) of nominal opening size 800 µm for small sample sizes (under 300 mg) or a sieve of nominal opening size 1 mm for larger sample sizes (300 mg or more). Mills that produce particle sizes meeting the specifications given in Table 1 will give acceptable results.

Table 1 — Required particle size

Nominal size of sieve openings µm	**Amount passing through sieve** % mass fraction
710	100
500	95 to 100
200	85 or less

Grinding may result in moisture loss and therefore the moisture content of the ground sample should preferably also be determined when reporting nitrogen or protein contents on a dry-matter or constant-moisture basis. Determination of the moisture content shall be carried out in accordance with ISO 712 for cereals other than maize, ISO 6540 for maize and ISO 24557 for pulses.

The grinder efficiency can be checked by replicate preparation of ground samples of a 2+1 mixture of maize and soya seeds. The expected coefficient of variation should be less than 2 % mass fraction.

9 Procedure

9.1 General

Carefully follow the manufacturer's instructions for instrument set-up, optimization, calibration and operation. Switch the instrument on and allow it to stabilize as defined in local procedures.

An instrument performance test should be carried out daily, using the reference material (5.12). The recovery of nitrogen should be > 99,0 % mass fraction.

9.2 Test portion

Weigh, to the nearest 0,000 1 g, at least 0,1 g of the test sample into a crucible or tin capsule or nitrogen-free filter paper (6.4). For samples low in protein (< 1 % mass fraction), the amount of the test

portion can be increased up to 3,5 g, depending on the type of Dumas equipment being used and on the nature of the sample.

Depending on the type of equipment used, if the sample contains over 17 % mass fraction of moisture, it may be necessary to dry it before analysis.

Lower masses may be necessary for very high protein content samples or when only very small amounts of sample are available. In the case of masses less than 0,1 g, a second (validation) determination shall be performed.

9.3 Control of oxygen supply

Control the oxygen supply, in particular the flow, in accordance with the instructions of the material supplier.

With each series of nitrogen content determinations, conduct as many blank runs as necessary to stabilize the equipment, using for each run an equivalent mass of sucrose in place of the test portion. The sucrose blank provides the amount of nitrogen that is introduced in the form of atmospheric air trapped within a powdered organic material. Use the mean value of the blank determinations as an error correction in the calculation of the nitrogen content of each test sample.

9.4 Calibration

For long-term instrument calibration, use pure compounds with a known, constant nitrogen content, e.g. aspartic acid (see 5.12), as standards. Analyse in duplicate three pure compounds, each in three different amounts chosen as a function of the measurement range for the actual samples.

To prepare a calibration curve, carry out at least five determinations with different amounts of the same compound, choosing the compound and the amounts used in such a way that the curve obtained will cover the range of nitrogen contents in the samples to be analysed.

If the test portion contains more than 200 mg of nitrogen, the calibration curve is likely to be non-linear. In the non-linear section, short segments can nevertheless be used for calibration purposes. To ensure the reliability of the curve in these segments, the amount of standard used shall be increased in steps corresponding to 1 mg to 5 mg of nitrogen over the segments.

Calibration can also be performed using standard aqueous solutions.

Check the calibration at least three times at the beginning of a series of analyses and then after every 15 to 25 samples, analysing either one of the standards (see 5.12) or a sample of known value. The value obtained for the nitrogen mass fraction shall differ by less than 0,05 % from the expected value. If it does not, repeat the calibration check after checking instrument performance.

9.5 Determination

With the instrument operating in the stable state, introduce the test portion in accordance with the manufacturer's instructions.

During the analysis, the following processes take place in the instrument (see Figure B.1, B.2 or B.3).

The test portion is quantitatively combusted under standard conditions at a temperature of at least 850 °C, depending on the instrument and the material being analysed.

