

Designation: D5227 – 01(Reapproved 2008)^{ε1}

Standard Test Method for Measurement of Hexane Extractable Content of Polyolefins¹

This standard is issued under the fixed designation D5227; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon (ε) indicates an editorial change since the last revision or reapproval.

 ε^1 NOTE—Reapproved with editorial changes in November 2008.

1. Scope

1.1 This test method describes an extraction/gravimetric procedure for determination of the amount of hexane soluble low molecular weight material present in polyethylene, polypropylene, ethylene-propylene copolymers, and ethylene-vinyl acetate copolymers. This test method is a modification of the Food and Drug Administration (FDA) procedure for determining hexane extractables of polyolefins. This test method is based upon the presumption that the weight of the residue extract present in the solvent is equal to the amount extracted from the film sample and could therefore be quantified by measuring the weight loss of the extracted film, eliminating the complex and time-consuming evaporation process described in 21 CFR 177.1520.

1.2 This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.

1.3 The values stated in SI units are to be regarded as the standard. Units used in 21 CFR 177.1520 are also used in this test method. Units are in conformance with Federal Code 21 CFR 177.1520, from which this test method is derived.

NOTE 1-There is no known ISO equivalent ISO to this test method.

2. Referenced Documents

2.1 ASTM Standards:²

D883 Terminology Relating to Plastics

- D1239 Test Method for Resistance of Plastic Films to Extraction by Chemicals
- D1600 Terminology for Abbreviated Terms Relating to Plastics

E131 Terminology Relating to Molecular SpectroscopyE691 Practice for Conducting an Interlaboratory Study to Determine the Precision of a Test Method

- 2.2 Federal Document:³
- 21 CFR 177.1520

3. Terminology

3.1 The definitions given in Terminology D883, D1600, and E131 are applicable to this test method.

- 3.2 Abbreviations:
- 3.2.1 EVA-ethylene-vinyl acetate copolymer.
- 3.2.2 *LDPE*—low-density polyethylene.
- 3.2.3 HDPE—high-density polyethylene.
- 3.2.4 *LLDPE*—linear low-density polyethylene.
- 3.2.5 FDA—Food and Drug Administration.
- 3.2.6 *PP*—polypropylene.

4. Summary of Test Method

4.1 Film samples are extracted with hexane for 2 h at 49.5 \pm 0.5°C, dried, and weighed.

4.2 The loss in weight of the film is presumed to be equal to the extractable content determined by solvent evaporation in the FDA protocol.

5. Significance and Use

5.1 FDA requirements for maximum extractables are specified for resin and uses. This test method provides a means to determine the amount of hexane-soluble low molecular weight material present in polyolefins. It is applicable to resins containing greater than 0.20 % extractables.

6. Apparatus

- 6.1 Water Bath, maintained at 49.5 \pm 0.5°C.
- 6.2 Resin Kettle, 1500-mL.⁴

¹ This test method is under the jurisdiction of ASTM Committee D20 on Plastics and is the direct responsibility of Subcommittee D20.70 on Analytical Methods.

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² For referenced ASTM standards, visit the ASTM website, www.astm.org, or contact ASTM Customer Service at service@astm.org. For *Annual Book of ASTM Standards* volume information, refer to the standard's Document Summary page on the ASTM website.

³ Available from Standardization Documents Order Desk, DODSSP, Bldg. 4, Section D, 700 Robbins Ave., Philadelphia, PA 19111-5098, http://www.dodssp.daps.mil.

⁴ Ace Glass, Inc., Cat. No. 6476–15, or its equivalent, has been found satisfactory.

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6.3 Kettle Head, 3-neck, with one 45/50 and two 24/40 female joints, and appropriate stoppers.⁵

6.4 Clamp.⁶

6.5 Allihn Condenser, Size C, with 45/50 male joint.⁷

6.6 Plastic Sleeves, tetrafluoroethylene (TFE), to fit Allihn condenser 45/50 male joint.

6.7 Vacuum Oven, capable of maintaining $80 \pm 5^{\circ}$ C and a minimum of 25-in. Hg pressure.

6.8 Magnetic Stirring Bar, egg-shaped, TFE-coated, 11/2 by 5/8 in.

6.9 Submersible Magnetic Stirring Motor, with power supply.8

6.10 Analytical Balance, capable of weighing to 0.1 mg.

7. Reagents and Materials

7.1 n-Hexane, aromatic free (<1 mg/L), minimum 85 % *n*-Hexane-reagent grade or equivalent.⁹ The solvent must be free of aromatic compounds that would significantly increase the solubility of the resin. The solvent grade specified represents the minimum required purity.

8. Materials

8.1 Blown Film, compression molded films, or cast films are suitable for testing.

8.2 Film, approximately 2.5 g, with a thickness not exceeding 4 mil is required for a single determination.

9. Procedure

9.1 Assemble the resin kettle setup with glass stopper, clamp, and magnetic stirring bar. (See Fig. 1.)

9.2 Add 1000 mL of *n*-Hexane to the kettle assembly. 9.3 Stopper the kettle and clamp the assembly into the water bath set at 49.5 \pm 0.5°C.

NOTE 2-Temperature is a critical factor in this analysis and must not vary more than 1°C. If the temperature exceeds these limits, the test must be discontinued and restarted. The FDA protocol also states the temperature of the contents must be brought to 49.5 ± 0.5 °C within 20 to 25 min.

Note 3-The level of water in the bath must be kept at least 2 cm above the level of the solvent in the kettle to ensure the temperature equilibrium. Position the kettle so that the center bottle of the kettle is sitting on a submersible stirrer. Start stirring and allow the hexane to heat for 1 h.

9.4 Using gloves and metal tweezers to avoid sample contamination, cut about 2.7 g of the prepared film sample (4 mil or less in thickness) into about 1-in. squares using clean sharp scissors.

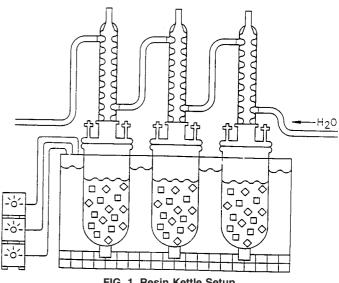


FIG. 1 Resin Kettle Setup

NOTE 4-Care must be exercised when cutting the samples to avoid ragged edges on the specimen. Small shards of film or contamination present at initial weighing can easily be lost during the test, adversely affecting the test results.

9.5 Weigh 2.5 \pm 0.05 g of film squares and record the initial film weight to the nearest 0.1 mg. Also record the number of film pieces.

Note 5-Forty or more squares will be obtained depending on the film thickness. Some laboratories have found that a basket assembly, as shown in Appendix X1, eliminates the need to count the film pieces before and after the solvent extraction step.

9.6 Add the film sample to the hexane making sure all squares become immersed in the solvent. (Use tweezers.) Replace the kettle head with condenser column. Extract for 2 h.

9.7 After the extraction period:

9.7.1 Filter the contents of the resin kettle through the fritted porcelain funnel.

9.7.2 Transfer the film squares, using tweezers, to a 200-mL Berzelius beaker and recount the film pieces to verify that none were lost during transfer.

9.7.3 Cover the beaker with a watchglass and place it in a vacuum oven at 80 \pm 5°C for 2 h.

9.7.4 After 2 h, remove the covered beaker from the vacuum oven and place it in a desiccator to cool to room temperature (about 1 h).

9.8 Remove the film squares using tweezers and weigh them to the nearest 0.1 mg.

9.9 Repeat 9.7.3 and 9.7.4 until a constant weight is obtained.

10. Calculation

10.1 Calculate the weight percent of extractables present in the original sample as follows:

hexane extractables,
$$\% = \frac{(A-B) \times 100 \times 0.935}{A}$$
 (1)

⁵ Ace Glass, Inc., Cat. No. 6486-40, or its equivalent, has been found satisfactory.

⁶ Ace Glass, Inc., Cat. No. 6496-10, or its equivalent, has been found satisfactory.

⁷ Ace Glass, Inc., Cat. No. 6740-06, or its equivalent, has been found satisfactory.

⁸ VWR Scientific Co., Cat. No. 58947-409, or its equivalent, has been found satisfactory.

⁹ EM Science, Omnisolv reagent grade, or its equivalent, has been found satisfactory.