## INTERNATIONAL STANDARD

ISO 13344

Third edition 2015-12-15

## Estimation of the lethal toxic potency of fire effluents

Détermination du pouvoir toxique létal des effluents du feu

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<u>SIST ISO 13344:2018</u> https://standards.iteh.ai/catalog/standards/sist/9975673b-9334-4997-81f3-f9d87034d69a/sist-iso-13344-2018



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## **Foreword**

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

The procedures used to develop this document and those intended for its further maintenance are described in the ISO/IEC Directives, Part 1. In particular the different approval criteria needed for the different types of ISO documents should be noted. This document was drafted in accordance with the editorial rules of the ISO/IEC Directives, Part 2 (see <a href="https://www.iso.org/directives">www.iso.org/directives</a>).

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For an explanation on the meaning of ISO specific terms and expressions related to conformity assessment, as well as information about ISO's adherence to the WTO principles in the Technical Barriers to Trade (TBT) see the following URL: Foreword - Supplementary information

The committee responsible for this document is ISO/TC 92, Fire safety, Subcommittee SC 3, Fire threat to people and environment.

SIST ISO 13344:2018

This third edition cancels and replaces the second edition (ISO 13344:2004), which has been technically revised. The following changes have been made:  $\frac{1}{4}$  d69a/sist-iso-13344-2018

- ISO 19702 has been added as a normative reference and citations added in 6.2.3 and 9.2.2;
- the first paragraph in 4.3 has been deleted;
- the note in 13.2 has been deleted.

## Introduction

The pyrolysis or combustion of every combustible material produces a fire effluent atmosphere, which, in sufficiently high concentration, is toxic. It is, therefore, desirable to establish a standard test method for the estimation of the toxic potency of such fire effluents.

It is further desirable, in view of worldwide resistance to the exposure of animals in standard tests, that this method should not make mandatory the use of such animals in its procedures. The mandatory portion of this standard test does not, therefore, specify the use of animal exposures. It only refers to animal exposure data already reported in the literature, with calculations being employed to express test results as they would have been obtained had animals actually been employed.

For those cases in which confirmation of test results using animal exposures can be justifiably permitted, an optional procedure to do so is presented in  $\underline{\text{Annex } A}$ .

The two parameters calculated using this standard are the FED (Fractional Effective Dose) and the  $LC_{50}$ . When either of these is used in performing a hazard analysis, certain information must accompany the term to avoid confusion. In the case of the FED, that is the toxicological effect on which the FED is based and the animal species for which the FED has been determined. In the case of the  $LC_{50}$ , that information is the length of the exposure and the animal species for which the  $LC_{50}$  has been determined.

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## Estimation of the lethal toxic potency of fire effluents

## 1 Scope

This International Standard provides a means for estimating the lethal toxic potency of the fire effluent produced from a material while exposed to the specific combustion conditions of a physical fire model. The lethal toxic potency values are specifically related to the fire model selected, the exposure scenario and the material evaluated.

Lethal toxic potency values associated with 30-min exposures of rats are predicted using calculations which employ combustion atmosphere analytical data for carbon monoxide (CO), carbon dioxide (CO<sub>2</sub>), oxygen (O<sub>2</sub>) (vitiation) and, if present, hydrogen cyanide (HCN), hydrogen chloride (HCl), hydrogen bromide (HBr), hydrogen fluoride (HF), sulfur dioxide (SO<sub>2</sub>), nitrogen dioxide (NO<sub>2</sub>), acrolein and formaldehyde. The chemical composition of the test specimen may suggest additional combustion products to be quantified and included. If the fire effluent toxic potency cannot be attributed to the toxicants analysed (Annex A), this is an indication that other toxicants or factors must be considered.

This International Standard is applicable to the estimation of the lethal toxic potency of fire effluent atmospheres produced from materials, products or assemblies under controlled laboratory conditions and should not be used in isolation to describe or appraise the toxic hazard or risk of materials, products or assemblies under actual fire conditions. However, results of this test may be used as elements of a fire hazard assessment that takes into account all of the factors which are pertinent to an assessment of the fire hazard of a particular end use; see ISO 19706.

The intended use of fire safety-engineering calculations is for life-safety prediction for people and is most frequently for time intervals somewhat shorted than 30 min. This extrapolation across species and exposure intervals is outside the scope of this International Standard! B-69d87034d69a/sist-iso-13344-2018

This International Standard does not purport to address all of the safety problems associated with its use. It is the responsibility of the user of this International Standard to establish appropriate safety and health practices.

### 2 Normative references

The following referenced documents are indispensable for the application of this document. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

ISO 13571, Life-threatening components of fire — Guidelines for the estimation of time to compromised tenability in fires

ISO 13943:2008, Fire safety — Vocabulary

ISO 19701, Methods for sampling and analysis of fire effluents

ISO 19702, Guidance for sampling and analysis of toxic gases and vapours in fire effluents using Fourier transform infrared (FTIR) spectroscopy

ISO 19706, Guidelines for assessing the fire threat to people

#### 3 Terms and definitions

For the purposes of this document, the terms and definitions given in ISO 13943:2008 apply.

## 4 Principle

**4.1** This method subjects a test sample to the combustion conditions of a specific physical fire model.

Concentrations of the major gaseous toxicants in the fire effluent atmosphere are monitored over a 30-min period, with ( $C \cdot t$ ) products for each interval being determined from integration of the areas under the respective concentration vs time plots. The ( $C \cdot t$ ) product data, along with either the mass charge or the mass loss of the test sample during the test, are then used in calculations to predict the 30-min LC<sub>50</sub> of the test sample.

**4.2** Since there can be toxicants present other than those measured, this value of the  $LC_{50}$  is a maximum.

If the chemical formulation and professional experience suggest that additional toxicants might contribute significantly to the  $LC_{50}$  value, the accuracy of the predicted  $LC_{50}$  may then be experimentally determined using a bioassay (see Annex A). Agreement within the experimental uncertainty supports attributing the lethality of the smoke to the monitored toxicants.

**4.3** Toxic potencies are estimated from combustion product analytical data without the exposure of experimental animals. Such a methodology is based on extensive experimentation using exposure of rats to the common fire gases, both singly and in combinations; see Reference[1]. The principle can be expressed mathematically, as shown in Formula (1); see Reference[2]:

$$L_{\text{FED}} = \sum_{i=1}^{n} \int_{0}^{t} \frac{C_{i}}{(C \cdot t)_{i}} dt \text{ iTeh STANDARD PREVIEW}$$
where

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 $C_i$  is the concentration, expressed in microlitres per litre, of the toxic component, i;

 $(C \cdot t)_i$  is the concentration-time product, expressed in microlitres per litre times minutes, for the specific exposure doses required to produce the toxicological effect.

When, as in this test method, the time values of 30 min numerically cancel, the FED becomes simply the ratio of the average concentration of a gaseous toxicant to its  $LC_{50}$  value for the same exposure time. When the FED is equal to 1, the mixture of gaseous toxicants should be lethal to 50 % of exposed animals.

## 5 Significance and use

**5.1** This test method has been designed to provide data for use in the estimation of lethal toxic fire hazard as a means for the evaluation of materials and products and to assist in their research and development.

The data are not, in themselves, an indication of toxic hazard or relative toxic hazard of a commercial product.

5.2 The method is used to predict the  $LC_{50}$  of fire effluents produced upon exposure of a material or product to fire.

Experimental confirmation might be needed to determine whether the major gaseous toxicants can account for the observed toxic effects as well as for the lethal toxic potency (see Annex A).

- 5.3 Predicted LC<sub>50</sub> values determined in this test method are associated only with the physical fire model used.
- **5.4** This test method does not attempt to address the toxicological significance of changes in particulate/aerosol size, fire effluent transport, distribution or deposition, or changes in the concentration of any fire effluent constituent as a function of time as may occur in a real fire.

- **5.5** The propensity for fire effluents from any material to have the same effects on humans as on rats in fire situations can only be inferred to the extent that the biological system of the rat is correlated with the human system.
- **5.6** This test method does not address any other acute sublethal effects of smoke, e.g. sensory and upper-respiratory-tract irritation, reduced motor capability, heat or thermal radiation injury, etc.
- **5.7** This test method does not address the long-term lethal effects of smoke exposure or the lethal effects of chronic exposures to smoke.
- **5.8** The FED values,  $L_{\text{FED}}$ , estimated from this method differ from those obtained using the equations in ISO 13571. The values obtained here are derived from rat lethality data. The FED values from ISO 13571 are derived from consensus estimates of the incapacitating effects of fire gases on people.

## 6 Apparatus

## 6.1 Physical fire model

- **6.1.1** The physical fire model, or laboratory combustion device, and the conditions under which it is operated shall be chosen so as to have demonstrated relevance to one or more of the specific classes or stages of fires identified and characterized in ISO 19706.
- **6.1.2** When obtaining data on the effluent from the combustion of a commercial product or assembly, i.e. other than a homogeneous material; the configuration and condition of the test specimen in the physical fire model shall be relevant to the appropriate fire exposure of the commercial product or assembly.
- 6.1.3 Repeatability and interlaboratory reproducibility of the physical fire model shall be demonstrated to be within the uncertainty range for the FED calculations for irritant and asphyxiant gases in ISO 13571.
- **6.1.4** The physical fire model shall be adaptable to analytical requirements.

## 6.2 Gas sampling

- **6.2.1** Continuous gas sampling shall be used to measure CO, CO<sub>2</sub> and O<sub>2</sub> levels.
- **6.2.2** The gas analysers shall have the following ranges, as a minimum:
- carbon monoxide, 0% by volume to 1% by volume ( $0\mu$ /l to  $10000\mu$ /l);
- carbon dioxide, 0 % by volume to 10 % by volume (0  $\mu$ l/l to 100 000  $\mu$ l/l);
- oxygen, 0 % by volume to 21 % by volume (0  $\mu$ l/l to 210 000  $\mu$ l/l).
- **6.2.3** Other gas analyses (for example, for HCN, HCl, HBr,  $NO_x$ ,  $SO_2$ , acrolein, formaldehyde and other chemical species) shall be performed, as appropriate to the chemical composition of the test sample and/or expectation of potential combustion products, by a method of choice with guidance from ISO 19701 and ISO 19702.

#### 7 Hazards

**7.1** This test procedure involves combustion processes.

Therefore, hazards to operating personnel exist from inhalation of combustion products. To avoid accidental leakage of toxic combustion products into the surrounding atmosphere, the entire exposure system shall be placed in a laboratory fume hood or under a canopy hood.

- **7.2** The venting system shall be checked for proper operation before testing and shall discharge into an exhaust system with adequate capacity.
- **7.3** Operating personnel have the responsibility to ensure that they are in compliance with all pertinent regulations regarding release and/or disposal of combustion products or gases.

## 8 Test specimens

- **8.1** Test specimens shall be prepared in accordance with the operating restrictions and conditions applicable to the physical fire model used and with consideration of the end use of the finished product being examined.
- **8.2** Test specimens shall be conditioned at an ambient temperature of 23 °C  $\pm$  3 °C (73 °F  $\pm$  5 °F) and relative humidity of (50  $\pm$  10) % for at least 24 h prior to testing or until constant mass is attained.

## 9 Calibration of the apparatus (standards.iteh.ai)

- **9.1** Physical fire model calibrations shall be conducted in accordance with the applicable operating methodology of the physical fire model, itch ai/catalog/standards/sist/9975673b-9334-4997-81f3
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- **9.2** Gas analyser calibrations shall be conducted at the beginning of each series of tests.
- **9.2.1** The gas analysers (for  $O_2$ ,  $CO_2$ , and CO) shall be calibrated using nitrogen gas for "zeroing" and an appropriate gas mixture close to, but less than, the analyser full-scale reading.

For all calibrations, the gas shall be set to flow at the same rate and pressure as during a test. For calibration of the  $O_2$  analyser, ambient air (20,9 %  $O_2$  by volume if the air is dry) shall be used, while for the  $CO_2$  and  $CO_3$  analysers, bottled gases containing  $CO_3$  or  $CO_3$  at known concentration are required. A single mixture containing both  $CO_3$  and  $CO_3$  may be used. During the calibration procedure, the gas return lines shall be diverted into an exhaust duct in order to prevent inadvertent accumulation of  $CO_3$  and  $CO_3$  in the exposure chamber.

**9.2.2** Calibration of devices used for analysis of other gases (for example, HCN, HCl and HBr) shall be performed using the guidance provided in ISO 19701 or ISO 19702.

### 10 Procedures

#### 10.1 General

- **10.1.1** The test conditions in the physical fire model shall replicate the combustion conditions in the intended fire stage.
- **10.1.2** The choice of specimen size for initial tests is made with consideration of anticipated toxicant yields such that  $L_{\text{FED}}$ s from 0,7 to 1,3 are obtained (see <u>Clause 11</u>) over the 30-min test period. Analytical

data from at least three tests are used for the calculation of a predicted  $LC_{50}$  for the test sample (Clause 12) to test for possible sensitivity to sample size of combustion conditions in the test apparatus.

## 10.2 Preparation for tests

Test preparation shall be conducted in accordance with the operating procedures for the physical fire model.

## 10.3 Test procedure for obtaining data

- **10.3.1** Weigh the conditioned test specimen and subject it to the operating conditions of the physical fire model.
- **10.3.2** As specified in <u>Clause 12</u>, collect analytical data for a total of 30 min from the initiation of the test or from when the combustion conditions replicating the desired fire stage (<u>6.1.1</u>) are established within the apparatus.
- **10.3.3** Quench the test specimen residue, remove it from the sample holder, and cool it to ambient temperature in an exhaust hood.

Weigh the specimen residue after it has cooled. Use reasonable means to obtain an accurate measure of the mass of the test specimen that has not been combusted, recognizing that some specimens can lose material from the specimen holder, for example, by explosion or spitting.

### 11 Calculations

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### 11.1 General

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The predicted lethal toxic potency ( $\mathbb{E}(\mathfrak{T}_0)$ ) of the effluent from the test specimen is calculated from the combustion atmosphere analytical data for CO, CO<sub>2</sub>, O<sub>2</sub>, and, if present, HCN, HCl and other toxicants. This is done for a given sample mass by first calculating the FED for the test. The LC<sub>50</sub> is then calculated as that sample mass which would yield an FED equal to 1 within a volume of 1 m<sup>3</sup>.

## 11.2 Calculation of FED

- **11.2.1** Two equations have been developed for the estimation of the 30-min lethality FED from the chemical composition of the environment in the physical fire model. Each begins with the precept that the fractional lethal doses of most gases are additive, as developed by Tsuchiya and coworkers<sup>[3]</sup>.
- **11.2.2** Formula (2) was developed empirically by Levin and coworkers (summarized in Reference<sup>[4]</sup> with citations of the original research) from exposure of laboratory rats to individual and mixed gases.

$$L_{\text{FED}} = \frac{m[\text{CO}]}{[\text{CO}_2] - b} + \frac{21 - [\text{O}_2]}{21 - \text{LC}_{50,\text{O}_2}} + \frac{[\text{HCN}]}{\text{LC}_{50,\text{HCN}}} + \frac{[\text{HCI}]}{\text{LC}_{50,\text{HCI}}} + \frac{[\text{HBr}]}{\text{LC}_{50,\text{HBr}}}$$
(2)

which reduces to

$$L_{\text{FED}} = \frac{m[\text{CO}]}{[\text{CO}_2] - b} + \frac{21 - [\text{O}_2]}{(21 - 5, 4)} + \frac{[\text{HCN}]}{150} + \frac{[\text{HCl}]}{3700} + \frac{[\text{HBr}]}{3000}$$

where

m is the slope of the CO-vs-CO<sub>2</sub> curve, which depicts the increasing toxicity of CO as CO<sub>2</sub> concentration increases;