
Barve in laki - Omočljivost - 4. del: Določevanje polarnega in disperznega dela površinske napetosti tekočin prek medfazne napetosti (ISO 19403-4:2017)

Paints and varnishes - Wettability - Part 4: Determination of the polar and dispersive fractions of the surface tension of liquids from an interfacial tension (ISO 19403-4:2017)

Beschichtungsstoffe - Benetzbarkeit - Teil 4: Bestimmung des polaren und dispersen Anteils der Oberflächenspannung von Flüssigkeiten aus einer Grenzflächenspannung (ISO 19403-4:2017)

Peintures et vernis - Mouillabilité - Partie 4: Détermination des fractions polaires et disperses de la tension de surface des liquides à partir de la tension interfaciale (ISO 19403-4:2017)

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Peintures et vernis — Mouillabilité —

*Partie 4: Détermination des fractions polaires et disperses de la
tension de surface des liquides à partir de la tension interfaciale*

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Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

The procedures used to develop this document and those intended for its further maintenance are described in the ISO/IEC Directives, Part 1. In particular the different approval criteria needed for the different types of ISO documents should be noted. This document was drafted in accordance with the editorial rules of the ISO/IEC Directives, Part 2 (see www.iso.org/directives).

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For an explanation on the voluntary nature of standards, the meaning of ISO specific terms and expressions related to conformity assessment, as well as information about ISO's adherence to the World Trade Organization (WTO) principles in the Technical Barriers to Trade (TBT) see the following URL: www.iso.org/iso/foreword.html.

This document was prepared by Technical Committee ISO/TC 35, *Paints and varnishes*, Subcommittee SC 9, *General test methods for paints and varnishes*.

A list of all parts in the ISO 19403 series can be found on the ISO website.

Paints and varnishes — Wettability —

Part 4:

Determination of the polar and dispersive fractions of the surface tension of liquids from an interfacial tension

1 Scope

This document specifies a test method to determine the polar and dispersive fraction of the surface tension of liquids with optical methods. The method can be applied for the characterization of liquid coating materials, especially when drying effects occur during measurement. The applicability can be restricted for liquids with non-Newtonian rheology¹⁾.

This document assumes that the information of surface tension of the liquid to be tested, as well as at least one suitable reference liquid, is known.

2 Normative references

The following documents are referred to in the text in such a way that some or all of their content constitutes requirements of this document. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

ISO 1409, *Plastics/rubber — Polymer dispersions and rubber latices (natural and synthetic) — Determination of surface tension by the ring method* 403-4:2020

ISO 4618, *Paints and varnishes — Terms and definitions*

ISO 15528, *Paints, varnishes and raw materials for paints and varnishes — Sampling*

ISO 19403-1, *Paints and varnishes — Wettability — Part 1: Terminology and general principles*

ISO 19403-3, *Paints and varnishes — Wettability — Part 3: Determination of the surface tension of liquids using the pendant drop method*

EN 14370, *Surface active agents — Determination of surface tension*

3 Terms and definitions

For the purposes of this document, the terms and definitions given in ISO 4618 and ISO 19403-1 apply.

ISO and IEC maintain terminological databases for use in standardization at the following addresses:

- IEC Electropedia: available at <http://www.electropedia.org/>
- ISO Online browsing platform: available at <http://www.iso.org/obp>

4 Principle

One drop of the respective liquid to be tested is reproduced within an optical cell, which is completely filled with a reference liquid, hanging from or ascending from a needle. The reproduced drop shall deviate significantly from the spherical shape due to its mass difference from the reference liquid.

1) This term is defined in DIN 1342-1.

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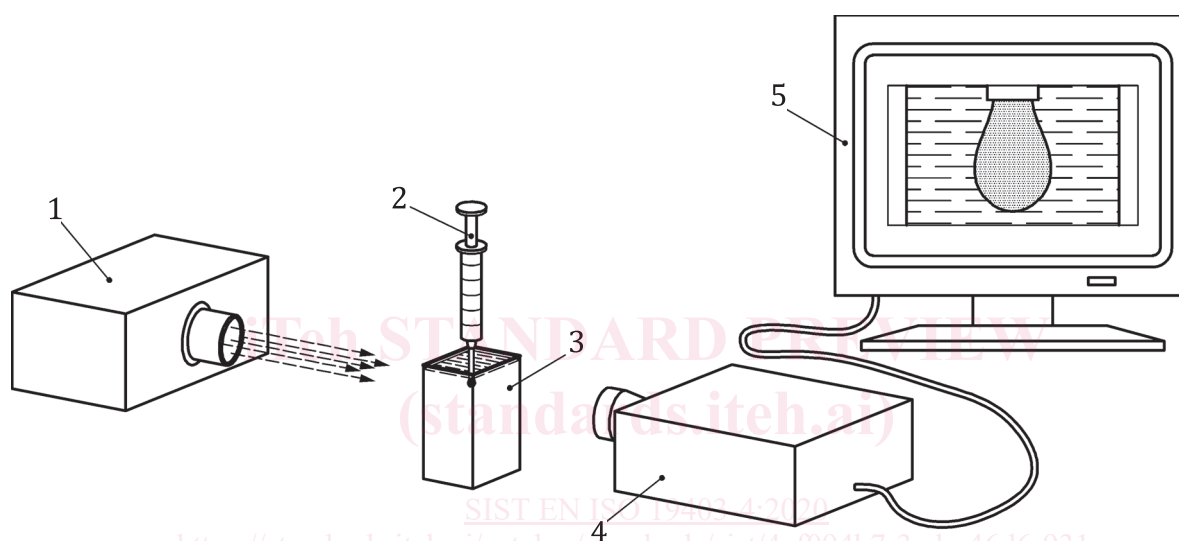
The interfacial tension is calculated from the shape of the reproduced pendant or ascending drop in accordance with the Young-Laplace equation. The polar and dispersive fraction of the surface tension of the liquid to be tested can be determined from the obtained interfacial tension and the known surface tensions of the liquid to be tested and the reference liquid.

5 Apparatus and materials

Ordinary laboratory apparatus, together with the following.

5.1 Drop contour analysis system, for measurement of the surface tension of pendant drops.

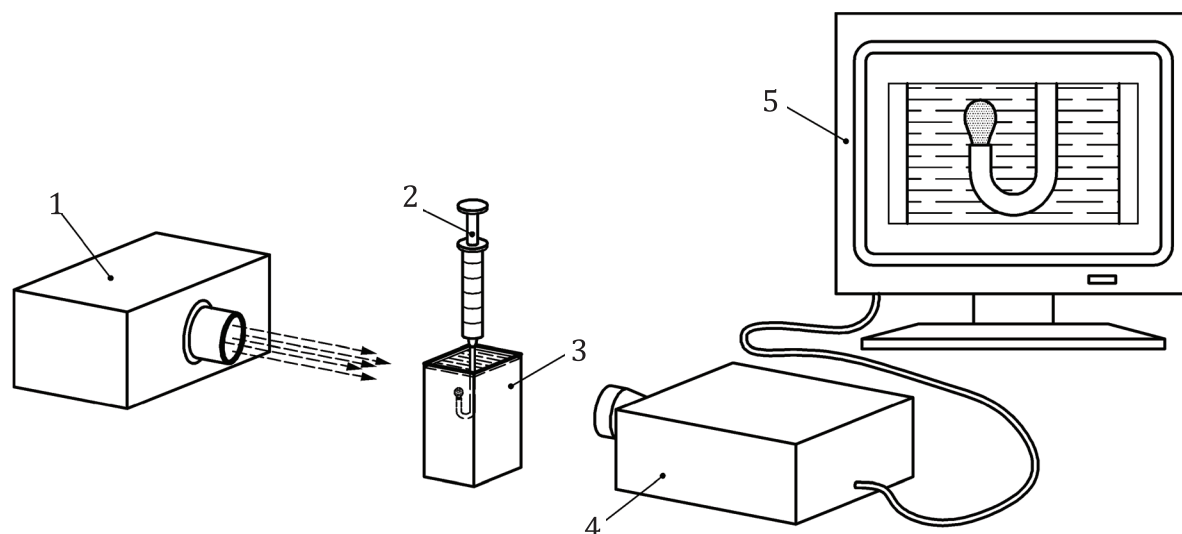
Any state-of-the-art drop contour analysis system with digital image capture and analysis. [Figure 1](#) and [Figure 2](#) show a schematic example of a drop contour analysis system.



Key

- 1 light source
- 2 dosing unit with graduated microsyringe
- 3 optical cell
- 4 image taking and analysing unit
- 5 screen

Figure 1 — Drop contour analysis system with pendant drop

**Key**

- 1 light source
- 2 dosing unit with graduated microsyringe
- 3 optical cell
- 4 image taking and analysing unit
- 5 screen

Figure 2 — Drop contour analysis system with ascending drop

The image taking system should be oriented in a way that the optimal image resolution ratio (ratio of width and height) can be used.

NOTE The device used can differ from the schematic diagram in regard to light path and the set-up of the components.

5.2 Dosing unit.

The dosing unit makes it possible to dose a pendant liquid drop, which deviates significantly from the spherical shape due to its own mass, on a circular-cylindrical needle with constant wall thickness within the detection area of the camera.

The needle used can be straight or bent in a J-shape.

J-shaped needles, whose free end points upwards are necessary when the density of the liquid to be tested, is lower than the density of the reference liquid.

For the measurement of the surface tension on the pendant or ascending drop, usually a larger outside diameter of the needle is needed than for the measurement of the contact angle on the horizontal drop. The outside diameters of the needles used shall be in the range between 0,5 mm and 2,5 mm. The ideal outside diameter of the needle depends on the relationship between the interfacial tension, σ_{LR} , and the density difference, $\Delta\rho_{LR} = \rho_L - \rho_R$, of the liquid to be tested, ρ_L , and the reference liquid, ρ_R . The higher the quotient, $\sigma_{LR}/\Delta\rho_{LR}$, the larger should be the outside diameter of the needle.

5.3 Optical cell.

The optical cell used shall have plane and side walls running parallel to each other. Through these sides made of optically impeccable clear-transparent materials (e.g. made of glass or insoluble, optically useable plastic), the observation of the formed drop is carried out in a way so that a representation

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without optical errors is possible. In addition, the inner and outer walls of the optical cell shall be free of contaminations (e.g. finger prints, adhering particles and surfactants).

5.4 Reference liquids.

The reference liquid to be used and the liquid to be tested shall not be miscible and shall be able to form a meniscus. Reference liquids shall be chemically homogenous, strictly dispersive and colourless with a melting point lower than 20 °C. Preferably, liquid hydrocarbons or perfluorohydrocarbons free from polar contaminations can be used. Among the hydrocarbons, especially *n*-decane, *n*-dodecane, *n*-tetradecane and *n*-hexadecane are suitable as reference liquids. As an alternative, also perfluoroalkanes, such as *n*-perfluorohexane and *n*-perfluorooctane, can be used as test liquids. The perfluoroalkanes can be used as reference liquids if the solubility of the liquid to be tested is too high compared to the hydrocarbons.

NOTE The viscosity of the mentioned reference liquids at 23 °C is in the range of that of water. This is of advantage for the setting of the equilibrium.

6 Sampling

Take a representative sample of the liquid to be tested in accordance with ISO 15528.

7 Procedure

7.1 General

7.1.1 Setting up the drop contour analysis device

Choose the location of the drop contour analysis device, so that it is not exposed to

- vibrations,
- intense air flows (e.g. caused by air conditioning), and
- intense exposure to light from outside (e.g. windows, bright lighting).

Align the drop contour analysis device horizontally.

Obtain the value of the local acceleration of gravity of the installation location and enter in the respective position of the manufacturer software.

7.1.2 Test conditions

Carry out the test at (23 ± 2) °C (see ISO 3270) and make sure that all test media have this temperature.

7.1.3 Cleaning and conditioning of the reference liquid

The measurable interfacial tension compared to water shall be at least 52 mN/m at 23 °C after the cleaning of the recommended *n*-alkanes (see 5.4)[5].

Especially, the residual content of ketones in the *n*-alkanes can be removed by the following cleaning procedure. The cleaning of the hydrocarbons used as reference liquids can be carried out in chromatography columns with a minimum length of 60 cm, which are filled with silica gel (e.g. particle size mesh 60). The silica gel used shall have polar surfaces in order to sufficiently absorb the polar contaminations that originate from the production process during passing through the hydrocarbons. Cleaning by distillation is also possible.

The cleaned *n*-alkanes shall be stored in light-proof glass bottles in the refrigerator at a maximum temperature of 4 °C.