

---

---

**Infant formula and adult  
nutritionals — Determination  
of pantothenic acid by ultra high  
performance liquid chromatography  
and tandem mass spectrometry  
method (UHPLC-MS/MS)**

iTeh STANDARD PREVIEW

(standards.iteh.ai)

*Formules infantiles et produits nutritionnels pour adultes —  
Détermination de la teneur en acide pantothénique par  
chromatographie liquide à ultra haute performance et spectrométrie  
de masse en tandem (CLUHP-SM/SM)*

ISO 20639:2015

<https://standards.iteh.ai/catalog/standards/sist/deb66ace-37fb-48a4-a678-a463b0e15ff2/iso-20639-2015>



**iTeh STANDARD PREVIEW**  
**(standards.iteh.ai)**

ISO 20639:2015

<https://standards.iteh.ai/catalog/standards/sist/deb66ace-37fb-48a4-a678-a463b0e15ff2/iso-20639-2015>



**COPYRIGHT PROTECTED DOCUMENT**

© ISO 2015, Published in Switzerland

All rights reserved. Unless otherwise specified, no part of this publication may be reproduced or utilized otherwise in any form or by any means, electronic or mechanical, including photocopying, or posting on the internet or an intranet, without prior written permission. Permission can be requested from either ISO at the address below or ISO's member body in the country of the requester.

ISO copyright office  
Ch. de Blandonnet 8 • CP 401  
CH-1214 Vernier, Geneva, Switzerland  
Tel. +41 22 749 01 11  
Fax +41 22 749 09 47  
copyright@iso.org  
www.iso.org

# Contents

Page

Foreword .....	iv
<b>1 Scope .....</b>	<b>1</b>
<b>2 Terms and definitions .....</b>	<b>1</b>
<b>3 Principle .....</b>	<b>1</b>
<b>4 Reagents and materials .....</b>	<b>1</b>
<b>5 Apparatus .....</b>	<b>3</b>
<b>6 Procedure .....</b>	<b>3</b>
6.1 Sample preparation .....	3
6.1.1 General .....	3
6.1.2 Dry blended powder samples .....	3
6.1.3 Wet blended powder samples .....	4
6.1.4 Liquid samples .....	4
6.2 Extraction .....	4
6.3 Analysis .....	4
6.3.1 Chromatographic analysis .....	4
6.3.2 UHPLC conditions .....	4
6.3.3 MS/MS conditions .....	5
6.3.4 Identification .....	5
<b>7 Calculations .....</b>	<b>5</b>
<b>Annex A (informative) Examples of chromatograms .....</b>	<b>6</b>
<b>Annex B (informative) Precision data .....</b>	<b>7</b>
<b>Bibliography .....</b>	<b>8</b>

iTech STANDARD PREVIEW  
(standards.iteh.ai)  
ISO 20639:2015  
<https://standards.iteh.ai/catalog/standards/sist/deb66ace-37fb-48a4-a678-a463b0e15ff2/iso-20639-2015>

## Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

The procedures used to develop this document and those intended for its further maintenance are described in the ISO/IEC Directives, Part 1. In particular the different approval criteria needed for the different types of ISO documents should be noted. This document was drafted in accordance with the editorial rules of the ISO/IEC Directives, Part 2. [www.iso.org/directives](http://www.iso.org/directives)

Attention is drawn to the possibility that some of the elements of this document may be the subject of patent rights. ISO shall not be held responsible for identifying any or all such patent rights. Details of any patent rights identified during the development of the document will be in the Introduction and/or on the ISO list of patent declarations received. [www.iso.org/patents](http://www.iso.org/patents)

Any trade name used in this document is information given for the convenience of users and does not constitute an endorsement.

For an explanation on the meaning of ISO specific terms and expressions related to conformity assessment, as well as information about ISO's adherence to the WTO principles in the Technical Barriers to Trade (TBT) see the following URL: [Foreword - Supplementary information](#)

The committee responsible for this document is ISO/TC 34, *Food products* in collaboration with AOAC INTERNATIONAL. It is being published by ISO and separately by AOAC INTERNATIONAL. The method described in this International Standard is equivalent to the AOAC Official Method 2012.16: *Pantothenic acid (vitamin B<sub>5</sub>) in infant formula and adult/pediatric nutritional formula ultra high pressure liquid chromatography — tandem mass spectrometry method*.

# Infant formula and adult nutritionals — Determination of pantothenic acid by ultra high performance liquid chromatography and tandem mass spectrometry method (UHPLC-MS/MS)

**WARNING** — The use of this International Standard can involve hazardous materials, operations and equipment. This International Standard does not purport to address all the safety problems associated with its use. It is the responsibility of the user of this International Standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.

## 1 Scope

This International Standard specifies a method for the quantitative determination of pantothenic acid, excluding bound forms, in infant formula and adult nutritionals (i.e. powders) using ultra high performance liquid chromatography and tandem mass spectrometry method (UHPLC-MS/MS).

## 2 Terms and definitions

For the purposes of this document, the following terms and definitions apply.

### 2.1

#### **adult nutritional**

nutritionally complete, specially formulated food, consumed in liquid form, which may constitute the sole source of nourishment, made from any combination of milk, soy, rice, whey, hydrolysed protein, starch and amino acids, with and without intact protein

### 2.2

#### **infant formula**

breast-milk substitute specially manufactured to satisfy, by itself, the nutritional requirements of infants during the first months of life up to the introduction of appropriate complementary feeding

[SOURCE: Codex Standard 72-1981]

## 3 Principle

Pantothenic acid is extracted using a 0,4 mol/l ammonium acetate buffer solution. After filtration, the final solution is subjected to ultra high performance liquid chromatography tandem mass spectrometry (UHPLC-MS/MS).

## 4 Reagents and materials

During the analysis, unless otherwise stated, use only reagents of recognized analytical grade and distilled or demineralized water or water of equivalent purity.

#### 4.1 Standards

4.1.1 **Calcium D-pantothenate**, Sigma<sup>1)</sup> or equivalent CAS 137-08-6.

4.1.2 **Calcium pantothenate-[<sup>13</sup>C<sub>6</sub>, <sup>15</sup>N<sub>2</sub>]**, IsoSciences<sup>1)</sup> or equivalent CAS 356786-94-2.

4.2  **$\alpha$ -Amylase**, Sigma A3176<sup>1)</sup>, from porcine pancreas, about 25 U/mg or equivalent.

#### 4.3 Solvents

4.3.1 **Acetonitrile**, LC grade or equivalent.

4.4 **Ammonium acetate**, ACS grade, > 98 % (Fluka 9690)<sup>1)</sup>.

4.5 **Acetic acid**, ACS grade.

4.6 **Formic acid**, ACS grade.

4.7 **1 % Formic acid in water**, ACS grade.

#### 4.8 Preparation of standard solutions

4.8.1 **Pantothenic acid (PA) stock solution**,  $\rho = 250 \mu\text{g/ml}$ . Weigh 54,5 mg of calcium pantothenate (4.1.1) into a 200 ml volumetric flask (take into account the moisture content given in the supplier's certificate or dry to constant mass at 105 °C) and dilute to volume with water. Store aliquots at -20 °C.

4.8.2 **Pantothenic acid intermediate solution**,  $\rho = 10 \mu\text{g/ml}$ . Transfer 1 ml of PA solution (4.8.1) into a 25 ml volumetric flask and dilute to volume with water. Store aliquots at -20 °C.

4.8.3 **Calcium pantothenate-[<sup>13</sup>C<sub>6</sub>, <sup>15</sup>N<sub>2</sub>] solution [IS (Internal Standard)] stock solution**,  $\rho = 20 \mu\text{g/ml}$ . Weigh 5,0 mg of calcium pantothenate-[<sup>13</sup>C<sub>6</sub>, <sup>15</sup>N<sub>2</sub>] (4.1.2) into a 250 ml volumetric flask and dilute to volume with water. Store aliquots at -20 °C.

4.8.4 **Solutions for the five-level standard curve**. Transfer appropriate volumes of the PA intermediate solution (10  $\mu\text{g/ml}$ ) (4.8.2) into 10 ml volumetric flasks to obtain five different concentrations of PA (0,08  $\mu\text{g/ml}$ , 0,16  $\mu\text{g/ml}$ , 0,32  $\mu\text{g/ml}$ , 0,64  $\mu\text{g/ml}$  and 1,2  $\mu\text{g/ml}$ ). Add 500  $\mu\text{l}$  of the IS stock solution (20  $\mu\text{g/ml}$ ) (4.8.3) and dilute to volume with water. The concentration of IS in each standard solution is 1  $\mu\text{g/ml}$ . Store aliquots of these solutions at -20 °C for no longer than one month before use.

4.8.5 **Ammonium acetate solution**,  $c = 400 \text{ mmol/l}$ , pH = 3,8 (used for sample extraction). Into a 500 ml beaker, add (30,8  $\pm$  0,10) g ammonium acetate. Add about 300 ml water and stir to dissolve with a magnetic stirrer. Adjust to pH = 3,8  $\pm$  0,1, carefully adding glacial acetic acid (about 150 ml is needed). Transfer into a 1 000 ml volumetric flask and make up to volume with water. This solution is stable for one month at 4 °C.

---

1) This is an example of a suitable product available commercially. This information is given for the convenience of users of this document and does not constitute an endorsement by ISO of the product named. Equivalent products may be used if they can be shown to lead to the same results.

## 5 Apparatus

Usual laboratory glassware and equipment and, in particular, the following.

**5.1 Balances**, with readability of 0,1 mg, capacity 210 g; with readability of 0,1 g, capacity 4 100 g.

**5.2 pH-meter**, with readability of 0,01 pH unit.

**5.3 Homogenizer**<sup>2)</sup>.

**5.4 Stir plate with magnetic stirrers.**

**5.5 Filters.** Syringe filters, 0,22 µm pore size, 33 mm internal diameter, Millex-GV PVDF (Millipore)<sup>3)</sup>. Membrane disc filters, 0,45 µm pore size (Millipore)<sup>3)</sup> or equivalent.

**5.6 UHPLC-MS/MS system**, UPLC column, e.g. ACQUITY UPLC<sup>®3)</sup> coupled with triple quadrupole detector equipped with electrospray ionization (ESI) source and T3 column (1,8 µm, 100 mm × 2,1 mm internal diameter; Waters Corp.)<sup>3)</sup> or equivalent.

## 6 Procedure

### 6.1 Sample preparation

#### 6.1.1 General

If the product contains starch, add 50 mg α-amylase to the suspensions and incubate for 15 min at 40 °C to decrease viscosity and facilitate handling. Mix liquid samples well to ensure homogeneity and continue directly to extraction. If the powder sample homogeneity is unknown, assume that it is non-homogenous and proceed with [6.1.2](#).

#### 6.1.2 Dry blended powder samples

For dry blended/non-homogenous powder samples, accurately weigh approximately 25,0 g ( $m_1$ ). Add 200,0 g ( $m_2$ ) water at 40 °C before mixing until a homogeneous suspension is obtained. A homogenizer ([5.3](#)) can be used when necessary. Accurately weigh approximately 15,0 g ( $m_3$ ) aliquot of homogenized sample suspension into a 50 ml volumetric flask. Calculate the sample mass ( $m_s$  is the powder equivalent) using Formula (1):

$$m_s = \frac{m_1 \times m_3}{m_2} \quad (1)$$

where

$m_1$  is the mass of sample weighed, in g;

$m_2$  is the mass of water added before mixing, in g;

2) Polytron PT3000 (drive unit), Aggregate PT-DA 3012 (Kinematics, Lucerne, Switzerland) are examples of suitable products available commercially. This information is given for the convenience of users of this document and does not constitute an endorsement by ISO of these products. Equivalent products may be used if they can be shown to lead to the same results.

3) This is an example of a suitable product available commercially. This information is given for the convenience of users of this document and does not constitute an endorsement by ISO of the product named. Equivalent products may be used if they can be shown to lead to the same results.

$m_3$  is the mass of homogenized sample suspension, in g.

### 6.1.3 Wet blended powder samples

For wet blended homogenous powder samples, accurately weigh approximately 2,0 g of sample ( $m_s$ ) into a 50 ml volumetric flask. Add 14 g of water at 40 °C. Mix until a homogeneous suspension is obtained.

### 6.1.4 Liquid samples

For liquid sample samples, accurately weigh approximately 20,0 g ( $m_s$ ) into a 50 ml volumetric flask.

## 6.2 Extraction

Using the prepared sample (6.1), add a 25 ml volume of a 0,4 mol/l ammonium acetate solution, pH = 3,8. Dilute the sample extract to volume with water. Add a stir bar and stir for 10 min. Filter a 20 ml portion through folded paper (Grade 597½). Run chromatographic analysis.

## 6.3 Analysis

### 6.3.1 Chromatographic analysis

Transfer a 1,0 ml aliquot of the filtrate obtained in 6.2 into a 15 ml polypropylene tube containing 500 µl of the IS stock solution (4.8.3). It is critical to use the same IS solution as used in the preparation of the standard curve (4.8.4). Dilute the solution to 10 ml with water, cap and mix. Filter through a 0,22 µm syringe filter (5.5). Inject into the UHPLC-MS/MS system.

Examples of typical chromatograms are given in Annex A.

### 6.3.2 UHPLC conditions

Injection volume:	2 µl
Column temperature:	30 °C
Flow rate:	0,45 ml/min
Mobile phase A:	0,1 % (v/v) formic acid in water
Mobile phase B:	Acetonitrile

The gradient programme for the column is given in Table 1.

Table 1 — Gradient for column

Time min	Mobile phase A %	Mobile phase B %
0	92	8
2,2	80	20
2,4	50	50
4,0	50	50
4,1	92	8
7,0	92	8

Direct the liquid chromatography flow into the MS detector only between 0 min and 2 min to prevent source fouling as much as possible.



### 6.3.3 MS/MS conditions

- Positive ESI
- Capillary voltage, 2,2 kV
- Cone voltage, 25 V
- Extractor voltage, 3,0 V
- Source temperature, 140 °C
- Desolvation temperature, 350 °C
- Cone gas flow, 40 l/h
- Desolvation gas flow, 700 l/h

Set the collision energy at 14 V with a dwell time for each monitored transition of 0,1 s. These values are indicative and need to be optimized for each instrument used. Monitor between 0 min and 2,1 min the transitions  $m/z$  220,2  $\rightarrow$  90,1 for PA and  $m/z$  224,2  $\rightarrow$  94,1 for the isotope-labelled IS.

### 6.3.4 Identification

MS detection in the single-reaction monitoring mode includes simultaneous detection of molecular ions corresponding to PA and isotopically labelled PA. The selected mass transitions are  $m/z$  220,2  $\rightarrow$  90,1 and  $m/z$  224,2  $\rightarrow$  94,1, respectively.

## 7 Calculations

Calculate for each standard the peak area ratio between PA and IS. Establish a 5-point calibration curve (ranging from 0,16 ng to 2,4 ng on column) by plotting peak area ratio (y-axis) versus PA concentration (x-axis). Calculate the linear regression. It is recommended to use a weighed regression curve (1/x).

Calculate the slope ( $S$ ) and the intercept ( $I$ ) of the calibration curve.

Calculate the PA mass fraction,  $w$ , in mg/100 g, using Formula (2):

$$w = \frac{(A - I) \times V_1 \times V_3 \times 100}{S \times m \times V_2 \times 1\,000} \quad (2)$$

where

- $A$  is the peak area ratio PA/IS in the test solution;
- $I$  is the intercept of the calibration curve;
- $S$  is the slope of the calibration curve;
- $V_1$  is the volume of the of sample extract, in ml (= 50);
- $V_2$  is the volume of the filtrate pipetted, in ml (= 1);
- $V_3$  is the final volume of the of the test solution, in ml (= 10);
- $m$  is the mass of the test portion, in g;
- 100 is the conversion to 100 g basis;
- 1 000 is the conversion from  $\mu\text{g}$  to mg.