INTERNATIONAL STANDARD

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Copper, lead and zinc sulfide concentrates — Determination of cadmium —

Part 2:

Acid digestion and inductively coupled plasma atomic emission spectrometric method

Concentrés de sulfure de cuivre, de plomb et de zinc — Dosage du cadmium —

Partie 2: Méthode par digestion acide et spectroscopie d'émission atomique à plasma à couplage inductif

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Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

The procedures used to develop this document and those intended for its further maintenance are described in the ISO/IEC Directives, Part 1. In particular, the different approval criteria needed for the different types of ISO documents should be noted. This document was drafted in accordance with the editorial rules of the ISO/IEC Directives, Part 2 (see www.iso.org/directives).

Attention is drawn to the possibility that some of the elements of this document may be the subject of patent rights. ISO shall not be held responsible for identifying any or all such patent rights. Details of any patent rights identified during the development of the document will be in the Introduction and/or on the ISO list of patent declarations received (see www.iso.org/patents).

Any trade name used in this document is information given for the convenience of users and does not constitute an endorsement.

For an explanation of the voluntary nature of standards, the meaning of ISO specific terms and expressions related to conformity assessment, as well as information about ISO's adherence to the World Trade Organization (WTO) principles in the Technical Barriers to Trade (TBT) see www.iso.org/iso/foreword.html.

This document was prepared by Technical Committee ISO/TC 183, *Copper, lead, zinc and nickel ores and concentrates*.

A list of all parts in the ISO 19976 series can be found on the ISO website.

Any feedback or questions on this document should be directed to the user's national standards body. A complete listing of these bodies can be found at www.iso.org/members.html.

Copper, lead and zinc sulfide concentrates — Determination of cadmium —

Part 2:

Acid digestion and inductively coupled plasma atomic emission spectrometric method

WARNING — The use of this document might involve hazardous materials, operations and equipment. It is the responsibility of the user of this document to establish appropriate health and safety practices.

1 Scope

This document specifies an acid digestion and inductively coupled plasma atomic emission spectrometric (ICP-AES) method for the determination of the mass fraction of cadmium in copper, lead and zinc sulfide concentrates as follows:

- a) for copper sulfide concentrates, the method is applicable to the determination of mass fractions of cadmium from 0.01% to 0.30%;
- b) for lead sulfide concentrates, the method is applicable to the determination of mass fractions of cadmium from 0,01 % to 0,30 %;
- c) for zinc sulfide concentrates, the method is applicable to the determination of mass fractions of cadmium from 0,05 % to 1,00 %.

2_{STz} **Normative references** ards/iso/c3a35c11-c6a9-4f40-b504-805ef1a1d8a1/iso-19976-2-2019

The following documents are referred to in the text in such a way that some or all of their content constitutes requirements of this document. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

ISO 385, Laboratory glassware — Burettes

ISO 648, Laboratory glassware — Single-volume pipettes

ISO 1042, Laboratory glassware — One-mark volumetric flasks

ISO 4787, Laboratory glassware — Volumetric instruments — Methods for testing of capacity and for use

ISO 8466-2, Water quality — Calibration and evaluation of analytical methods and estimation of performance characteristics — Part 2: Calibration strategy for non-linear second-order calibration functions

ISO 9599, Copper, lead, zinc and nickel sulfide concentrates — Determination of hygroscopic moisture content of the analysis sample — Gravimetric method

ISO 12743, Copper, lead, zinc and nickel concentrates — Sampling procedures for determination of metal and moisture content

ISO Guide 35, Reference materials — Guidance for characterization and assessment of homogeneity and stability

3 Terms and definitions

No terms and definitions are listed in this document.

ISO and IEC maintain terminological databases for use in standardization at the following addresses:

- ISO Online browsing platform: available at https://www.iso.org/obp
- IEC Electropedia: available at http://www.electropedia.org/

4 Principle

The test sample is dissolved in hydrochloric acid, nitric acid and hydrofluoric acid with a final stage perchloric acid digestion. Yttrium is added as an internal standard to solutions of dissolved samples, reference materials and calibration solutions and cadmium determined by ICP-AES.

5 Reagents

During the analysis, use only reagents of recognized analytical grade and distilled water or water of equivalent purity.

- **5.1 Hydrochloric acid**, concentrated (ρ_{20} 1,19 g/ml).
- **5.2 Hydrochloric acid**, dilute 1+1. Slowly add 500 ml of concentrated hydrochloric acid (5.1) to 500 ml of water while stirring.
- **5.3 Hydrochloric acid**, dilute 1+9. Slowly add 50 ml of concentrated hydrochloric acid (<u>5.1</u>) to 450 ml of water while stirring.
- **5.4** Nitric acid, concentrated (ρ_{20} 1,42 g/ml).
- **5.5** P **Hydrofluoric acid**, concentrated (ρ_{20} 1,15 g/ml). 11-c6a9-4f40-b504-805ef1a1d8a1/iso-19976-2-2019
- **5.6 Perchloric acid**, concentrated (ρ_{20} 1,70 g/ml).
- **5.7 Cadmium metal**, minimum purity 99,99 %.
- **5.8 Cadmium**, standard solution, 1 ml contains 1 mg of Cd.

Weigh, to the nearest 0,1 mg, 1,000 0 g of cadmium metal (5.7) into a 250 ml beaker. Add 10 ml of water and 5 ml of nitric acid (5.4). Cover and warm gently (if necessary) until the cadmium metal is dissolved. Heat to remove nitrogen oxides, then allow to cool, and add about 50 ml of water. Transfer the solution to a 1 000 ml one-mark volumetric flask, add 40 ml of hydrochloric acid (5.2), fill up nearly to the mark with water, mix and cool to room temperature. Then fill up exactly to the mark with water and mix again.

5.9 Cadmium, standard solution, 1 ml contains 100 μg of Cd.

Pipette 10,00 ml of cadmium standard solution (5.8) into a 100 ml one-mark volumetric flask. Add 20 ml of hydrochloric acid (5.2), fill up nearly to the mark with water, mix and cool to room temperature. Then fill up exactly to the mark with water and mix again.