

# SLOVENSKI STANDARD oSIST prEN 459-2:2019

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Gradbeno apno - 2. del: Metode preskušanja

Building lime - Part 2: Test methods

Baukalk - Teil 2: Prüfverfahren

Chaux de construction - Partie 2: Méthodes d'essai PREVIEW

Ta slovenski standard je istoveten z: (standards iteh ai)

<u>kSIST FprEN 459-2:2020</u>

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ICS:

91.100.10 Cement. Mavec. Apno. Malta Cement. Gypsum. Lime. Mortar

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# EUROPEAN STANDARD NORME EUROPÉENNE EUROPÄISCHE NORM

# DRAFT prEN 459-2

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Will supersede EN 459-2:2010

#### **English Version**

# Building lime - Part 2: Test methods

Chaux de construction - Partie 2: Méthodes d'essai

Baukalk - Teil 2: Prüfverfahren

This draft European Standard is submitted to CEN members for enquiry. It has been drawn up by the Technical Committee CEN/TC 51.

If this draft becomes a European Standard, CEN members are bound to comply with the CEN/CENELEC Internal Regulations which stipulate the conditions for giving this European Standard the status of a national standard without any alteration.

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EUROPEAN COMMITTEE FOR STANDARDIZATION COMITÉ EUROPÉEN DE NORMALISATION EUROPÄISCHES KOMITEE FÜR NORMUNG

CEN-CENELEC Management Centre: Rue de la Science 23, B-1040 Brussels

Com	tents	Page
Europ	oean foreword	5
Intro	duction	6
1	Scope	7
2	Normative references	7
3	Terms and definitions	
4	Sampling	
4.1	General	7
4.2	Sampling of powdered material	
4.3	Sampling of granular material	
4.4	Sampling of lime putty and milk of lime	
4.5	Preparation of the test portion	
5	General requirements for testing	
5.1	Methods of chemical analysis	
5.2	Number of tests	10
5.3	Repeatability and reproducibility	
5.4	Expression of masses, volumes, factors and results PREVIEW  Blank determinations	11
5.5	Blank determinations ell STANDARD FREVIEW	11
5.6	Reagents (standards.itch.ai) Evaluation of test results	11
5.7	Evaluation of test results	11
5.7.1	General KSIST FprEN 459-2:2020 Test results for quickline RSIST FprEN 459-2:2020 Test results for quickline RSIST FprEN 459-2:2020	11
5.7.2	Test results for quicklime Note that appropriately the state of the st	11
5.7.3	Test results for all other types (4444)71024ee/ksix-ibren-459-2-2020	12
5.7.4	Test results for available lime	12
6	Chemical analysis	
6.1	Fusion with lithium tetraborate	
6.1.1	General	12
6.1.2	Principle	
6.1.3	Reagents	
6.1.4	Apparatus	
6.1.5		_
6.2	Extraction with hydrochloric acid	
6.2.1	General	
6.2.2	Principle	
6.2.3	Reagents	
6.2.4	Apparatus	
6.2.5	Procedure	
6.3	Calcium oxide (CaO) and magnesium oxide (MgO)	
6.3.1	General	
6.3.2	Principle	
6.3.3	Reagents	
6.3.4	Apparatus	
6.3.5	Procedure	
6.3.6	Evaluation and expression of results	
6.4	Sulfate (expressed as SO <sub>3</sub> )	
6.4.1	General	17

6.4.2	Principle	17
6.4.3	Reagents	
6.4.4	Apparatus	
6.4.5	Procedure	
6.4.6	Evaluation and expression of results	18
6.5	Free water	
6.5.1	General	
6.5.2	Principle	
6.5.3	Apparatus	
6.5.4	Procedure	
6.5.5	Evaluation and expression of results	
6.6	Volumetric determination of carbon dioxide (CO <sub>2</sub> ) (reference method)	
6.6.1	Principle	
6.6.2	Reagents	
6.6.3	Apparatus	
6.6.4	Procedure	
6.6.5	Calibration of the apparatus	
6.6.6	Evaluation and expression of results	
6.7	Gravimetric determination of carbon dioxide (CO <sub>2</sub> ) (alternative method)	
6.7.1	Principle	
6.7.1 6.7.2	Reagents	
6.7.2 6.7.3	Apparatus	
6.7.3 6.7.4	Procedure Teh STANDARD PREVIEW	24
6.7.4 6.7.5	Evaluation and symposium of posults	20
6.7.5 6.8	Evaluation and expression of results	2 / 27
6.8.1	General	
6.8.2	Principle LSIST ForFN 459, 22020	
6.8.3	Apparatus https://standards.itab.ai/catalog/standards/sist/02449556-6644-4abc-95d9-	
6.8.4	Procedure	
6.8.5	Evaluation and expression of results	
6.9	Available lime	
6.9.1	General	
6.9.2	Principle	
6.9.3	Reagents	
6.9.4	Apparatus	
6.9.5	Procedure	
6.9.6	Sugar extraction	
6.9.7	Determination for calcium lime	
6.9.8	Determination for lime with hydraulic properties	
6.9.9	Evaluation and expression of results	
	•	
7	Physical tests	
7.1	Particle size by dry sieving	
7.1.1	General	
7.1.2	Principle	
7.1.3	Apparatus	
7.1.4	Preparation of test portions	
7.1.5	Procedure	
7.1.6	Evaluation and expression of results	
7.2	Particle size by air-jet sieving	
7.2.1	General	
7.2.2	Apparatus	
7.2.3	Procedure	33

/.Z. <del>4</del>	Evaluation and expression of results	34
7.3	Bulk density	34
7.3.1	Apparatus	34
7.3.2	Procedure	36
7.4	Soundness	36
7.4.1	General	36
7.4.2	For hydrated calcium lime and all types of lime with hydraulic properties	36
7.4.3	For hydrated calcium lime, lime putty and hydrated dolomitic lime which include	
	grains larger than 0,2 mm	30
7.4.4	For quicklime, lime putty, dolomitic lime and hydrated dolomitic lime	
7.5	Setting times	
7.5.1	Principle	
7.5.2	Laboratory, apparatus and materials	
7.5.2 7.5.3	Standard consistence test	
7.5.4	Setting time test	
7.3. <del>4</del> 7.6	Reactivity	
7.6 7.6.1	General	
7.6.2	Apparatus	
7.6.2 7.6.3	* *	
	Quality check of the test equipment	
7.6.4	Sample preparation  Procedure	
7.6.5		
7.6.6	Evaluation and expression of test results	
7.7	Yield Slaking vessel IIEN STANDARD PREVIEW	53
7.7.1		
7.7.2	Procedure(standards.itch.ai)	
7.8	Standard mortar by mass and water demand for values of flow and penetration	54
7.8.1	General KSIST ForEN 459-2:2020	54
7.8.2	Composition and preparation of standard mortar https://standards.teh.av/catalog/standards/sist/02449556-6644-4abc-95d9-	55
7.8.3	General  Composition and preparation of standard mortar  Water demand for values of flow and penetration  Water notantian	60
7.9	water retention	ot
7.9.1	Principle	
7.9.2	Apparatus	
7.9.3	Preparation of specimen material	
7.9.4	Procedure	
7.9.5	Evaluation	
	Determination of air content	
	Apparatus	
	Calibration of apparatus	
	Preparation of test mortar	
	Procedure	
	Reporting of results	
	Compressive strength	
	General	
7.11.2	Modifications to EN 196-1	<b> 6</b> 4
Annex	A (informative) Example for the calculation of the results on sieving	68
	B (informative) Precision data for the test methods	
Biblio	graphy	72

## **European foreword**

This document (prEN 459-2:2019) has been prepared by Technical Committee CEN/TC 51 "Cement and building limes", the secretariat of which is held by NBN.

This document is currently submitted to the CEN Enquiry.

This document will supersede EN 492-2:2010.

In comparison with EN 459-2:2010, the following modifications have been:

- description of fusion with lithium tetraborate;
- a new Clause 4.1 included;
- amendments for the determinations of CaO and MgO;
- change of the order of methods for determination of CO<sub>2</sub>;
- Phenolphtalein as indicator replaced by thymolphtalein;
- Amendments for the determination of available lime; (Standards.iteh.ai)
- amendments of the methods for particle size determination;

kSIST FprEN 459-2:2020

- amendment for the description of bulk density sist/02449556-6644-4abc-95d9d4a4f71624ee/ksist-fpren-459-2-2020
- amendments for the determination of reactivity;
- amendments for the Table B.1 and Table B.2 in Annex B;
- editorial changes were made and minor mistakes corrected.

EN 459, *Building lime*, is currently composed of the following parts:

- Part 1: Definitions, specifications and conformity criteria;
- Part 2: Test methods;
- Part 3: Conformity evaluation.

The existing standards from the EN 196 series were used as a basis for the testing of physical and mechanical properties in EN 459-2. For the testing of chemical properties of building limes, test methods described in EN 12485 have been incorporated into this document.

#### Introduction

The objective in this document has been to adopt as many European standardized methods as possible, and where this has not been possible, to use other appropriate proven methods.

Unless otherwise stated, tolerance class m of ISO 2768-1 apply (Indications on drawings by "ISO 2768-m").

All dimensions are in millimetres.

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#### 1 Scope

This document describes the test methods for all building limes covered by EN 459-1.

This document specifies in Table 2 the methods used for the chemical analyses and the determination of physical properties of building limes.

This document describes primary methods of chemical analyses. The reference methods and, in certain cases, alternative methods in this document can be considered to be equivalent.

#### 2 Normative references

The following documents are referred to in the text in such a way that some or all of their content constitutes requirements of this document. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

EN 196-1:2016, Methods of testing cement — Part 1: Determination of strength

EN 196-3, Methods of testing cement — Part 3: Determination of setting times and soundness

EN 196-7, Methods of testing cement — Part 7: Methods of taking and preparing samples of cement

EN 459-1, Building lime — Part 1: Definitions, specifications and conformity criteria

EN 932-1, Test for general properties of aggregates — Part 1: Methods for sampling

EN ISO 6506-1, Metallic materials Brinell hardness test Part 1: Test method (ISO 6506-1)

EN 22768-1, General tolerances — Part In Tolerances for linear and angular dimensions without individual tolerance indications (ISO 2768 1) and ards/sist/02449556-6644-4abc-95d9-d4a4f71624ee/ksist-fpren-459-2-2020

#### 3 Terms and definitions

No terms and definitions are listed in this document.

ISO and IEC maintain terminological databases for use in standardization at the following addresses:

- IEC Electropedia: available at <a href="http://www.electropedia.org/">http://www.electropedia.org/</a>
- ISO Online browsing platform: available at <a href="http://www.iso.org/obp">http://www.iso.org/obp</a>

#### 4 Sampling

#### 4.1 General

Sampling shall be carried out as specified in 4.2 to 4.4 taking into account the need to minimize moisture and carbon dioxide absorption. Samples shall therefore be transported and stored in air-tight containers and all the handling shall be carried out as quickly as possible.

#### 4.2 Sampling of powdered material

Sampling shall be carried out in accordance with EN 196-7.

#### 4.3 Sampling of granular material

Sampling shall be carried out in accordance with EN 932-1.

#### 4.4 Sampling of lime putty and milk of lime

The spot sample size shall be  $(10 \pm 5)$  dm<sup>3</sup>.

Where lime putty or milk of lime is sampled, the increments shall be blended thoroughly.

#### 4.5 Preparation of the test portion

Before carrying out the analysis, the sample shall be reduced in mass by means of a sample divider and/or by any homogenization process (quartering, mixing, ...) to produce a representative homogeneous test sample of suitable mass for the intended determinations.

Lime putty and milk of lime shall be dried before the chemical analysis (see 6.5.4.2).

The sample preparation for the appropriate test is described in Table 1.

Table 1 — Sample preparation for the single tests

Test	Clause in this document	Type and form of the building lime	Sample preparation
Chemical analysis	6	All types of building lime	The sample of granular material shall be crushed and ground. All tests shall be performed on materials of a grain size $\leq 0.2$ mm
Grain size distribution by sieving	7.1 and 7.2 <b>iTe</b>	h STANDARD PR Quicklime (standards.iteh.	Material in the as-delivered state
Grain size distribution by air-jet sieving	7.2 https://stand	Hydrated lime hydrated dolomitic atalogs and lime hydrated hydraulic properties form 459-2-	Material in the as-delivered state
Bulk density	7.3	All types of lime with hydraulic properties	See 7.3.2
	7.4.2.1 and 7.4.2.2	Hydrated lime, lime with hydraulic properties	Material in the as-delivered state
Soundness	7.4.2.3	Hydraulic lime with an $SO_3$ content of more than 3 % and up to 7 %	Material in the as-delivered state
Soundness	7.4.3	Hydrated lime, lime putty und hydrated dolomitic lime	Material in the as-delivered state
	7.4.4	Quicklime, lime putty, dolomitic quicklime, hydrated dolomitic lime	See 7.4.4.3.1
Setting times	7.5	Lime with hydraulic properties Material in the as-delivered state	

Test	Clause in this document	Type and form of the building lime	Sample preparation
Reactivity	7.6	Quicklime, dolomitic quicklime	Method 1:  The test shall be performed in the as delivered state.  If material contains particles coarser than 5 mm, the fraction above 5 mm shall be crushed to get 100 % of the material ≤ 5 mm.  Method 2:  The test shall be performed on materials of a grain size of ≥ 95 % passing 0,2 mm sieve.  The method used for the test shall be declared in the test report.
Yield	7.7	Quicklime	See 7.7.2
Mortar tests	7.8 to 7.10	Hydrated lime, hydrated dolomitic lime, lime with hydraulic properties	Material in the as-delivered state
Compressive strength	7.11 iTeh ST	Lime D Awith D hydraulic properties	Material in the as-delivered state

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#### 5 General requirements for testing

#### 5.1 Methods of chemical analysis

The methods to be used for the chemical analysis of building lime products and the principle of each method are listed in Table 2.

Table 2 — Methods of chemical analysis

Determination	Method	Principle
Calcium oxide	6.3	Complexometric titration
		XRF melting tablet
		XRF powder tablet
Magnesium oxide	6.3	Complexometric titration
		XRF melting tablet
		XRF powder tablet
		ICP-OES
		AAS-Flame technique
Sulfate, expressed as SO	6.4 AND	Gravimetry XRF melting tablet
(st	andaı	<b>ds iteh ai)</b> XRF powder tablet
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Free water	6.5	Gravimetry, drying oven
		Gravimetry, moisture balance
Carbon dioxide	6.6	Volumetry
	6.7	Gravimetry
		Combustion at 900 °C minimum temperature and IR detection
Loss on ignition at 1 050 °C	6.8	Gravimetry
Available lime	6.9	Acidimetric titration

#### 5.2 Number of tests

Analysis of a building lime may require the determination of a number of its chemical properties. For each determination one or more tests shall be carried out in which the number of measurements to be taken shall be as specified in the relevant clause of this document.

Where the analysis is one of a series subject to statistical control, determination of each chemical property by a single test shall be the minimum required.

Where the analysis is not part of a series subject to statistical control, the number of tests for determination of each chemical property shall be two (see also 5.3).

#### 5.3 Repeatability and reproducibility

Repeatability — Precision under repeatability conditions where independent test results are obtained with the same method on identical test items (material) in the same laboratory by the same operator using the same equipment within short intervals of time.

Reproducibility — Precision under reproducibility conditions where test results are obtained with the same method on identical test items (material) in different laboratories with different operators using different equipment.

Repeatability and reproducibility in this document (see Annex B) are expressed as repeatability standard deviation(s) and reproducibility standard deviation(s) in e.g. absolute percent, grams, etc., according to the property tested.

#### 5.4 Expression of masses, volumes, factors and results

Express masses in grams to the nearest 0,001 g and volumes from burettes in millilitres to the nearest 0,05 ml.

Express the factors of solutions, given by the mean of three measurements, to three decimal places.

Express the results generally to one decimal place.

If the result is below 1, then express the results, generally to two decimal places. If the test is in duplicate, express the result as a mean value. If the two test results differ by more than twice the standard deviation of repeatability, repeat the test and take the mean of the two closest test results.

The results of all individual tests shall be recorded. PREVIEW

# 5.5 Blank determinations (standards.iteh.ai)

Carry out a blank determination without a sample where relevant, following the same procedure and using the same amounts of reagents: Correct the results obtained for the analytical determination accordingly.

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#### 5.6 Reagents

Use only reagents of analytical quality. References to water mean distilled or deionised water having an electrical conductivity  $\leq 0.5$  mS/m.

Unless otherwise stated percent means percent by mass.

For the concentrated liquids used to make up the reagents in this document, the densities ( $\rho$ ) are given in grams per millilitre at 20 °C. The degree of dilution is always given as a volumetric sum, for example: dilute hydrochloric acid 1 + 2 means that 1 volume of concentrated hydrochloric acid is to be mixed with 2 volumes of water.

The concentrations of reference and standard volumetric solutions are specified as amount-of-substance concentrations, c (mol/l),

#### 5.7 Evaluation of test results

#### 5.7.1 General

The chemical requirements for building limes are specified in the relevant tables in EN 459-1.

#### 5.7.2 Test results for quicklime

For quicklime the specified values correspond to the finished product.

#### 5.7.3 Test results for all other types

For all other types (hydrated lime, lime putty, milk of lime, lime with hydraulic properties) the values are based on the product after subtraction of its free water and bound water content. The values obtained by application of procedures described in this document for total calcium oxide and magnesium oxide (6.3), magnesium oxide (6.3), sulfate (6.4) and carbon dioxide (6.6) or (6.7) are for the products without subtraction of the free water and bound water content. To compare these values with EN 459-1 (see clauses for calcium lime or dolomitic lime), they shall first be corrected by multiplication by factor F. Factor F shall be determined in the following way:

Determine the carbon dioxide content as described in 6.6 or 6.7 and the loss on ignition as described in 6.8. The loss on ignition is the sum of the free water, bound water and carbon dioxide, provided that the sample does not contain any highly volatile compounds or oxidizable constituents. Calculate the total (free + bound) water content  $W_T$  as a mass fraction in % of the sample using:

$$W_{\rm T} = {\rm loss \ on \ ignition \ in \% --- \ carbon \ dioxide \ content \ in \%}$$
 (1)

Calculate the factor *F* from the following equation:

$$F = 100 / (100 - W_{\rm T}) \tag{2}$$

#### 5.7.4 Test results for available lime

The values for available lime, obtained by the application of the procedure described in 6.9 correspond to either available CaO for quicklime or available Ca(OH)<sub>2</sub> for all other types (hydrated lime, lime putty milk of lime, lime with hydraulic properties)andards.iteh.ai)

### 6 Chemical analysis

kSIST FprEN 459-2:2020

6.1 Fusion with lithium tetraborate 444f71624ee/ksist-fpren-459-2-2020

#### 6.1.1 General

Fusion with lithium tetraborate is used to dissolve building lime in order to determine appropriate parameters (see Table 2).

#### 6.1.2 Principle

After fusing the sparingly soluble oxide constituents of the sample with lithium tetraborate, the fusion cake is dissolved in hydrochloric acid and transferred to a suitable volumetric flask.

#### 6.1.3 Reagents

- **6.1.3.1** Lithium tetraborate,  $Li_2B_4O_7$ .
- **6.1.3.2** Hydrochloric acid,  $\rho = 1,16$  g/ml.
- **6.1.3.3** Hydrochloric acid, diluted, (1 + 5).

#### 6.1.4 Apparatus

- **6.1.4.1** Ordinary laboratory apparatus and the following:
- **6.1.4.2** Platinum-gold crucible.
- **6.1.4.3** Hot plate.

#### 6.1.5 Procedure

Weigh  $(0.25 \pm 0.02)$  g to the nearest 0.1 mg  $(m_1)$  of the sample into a platinum-gold crucible and add 1.5 g of lithium tetraborate (6.1.3.1). Roast at  $(1\ 000 \pm 25)$  °C until the melt is clear. Remove from the heat, cover the crucible with a watch glass and stand the crucible in water for a few seconds, remove and allow it to stand until it has cooled completely.

Transfer the cold melt to a beaker using approximately 50 ml of hydrochloric acid (6.1.3.1) and dissolve while stirring and heating at 100 °C. Then transfer the solution to a 250 ml volumetric flask with water and, after cooling to room temperature make up to the mark with water. This solution  $(V_1)$  is ready for further chemical analyses.

#### 6.2 Extraction with hydrochloric acid

#### 6.2.1 General

Extraction with hydrochloric acid is used to dissolve building lime in order to determine appropriate parameters (see Table 2).

#### 6.2.2 Principle

The sample is boiled with hydrochloric acid and the solution filtered. The pH value is adjusted to 6 to 7 to precipitate the iron (III) and aluminium oxides. After refiltering, the filtrate is transferred to a suitable volumetric flask.

### 6.2.3 Reagents iTeh STANDARD PREVIEW

- **6.2.3.1** Hydrochloric acid,  $\rho$  (HOI) = 1.16g/mlto 1,19g/ml.
- **6.2.3.2** Hydrogen peroxide solution  $c(H_2O_2) = 30\%$

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- **6.2.3.3** Hydrogen peroxide solution diluted, 1 = 19459-2-2020
- **6.2.3.4** Ammonium hydroxide solution, c (NH<sub>4</sub>OH) = 25 %.
- **6.2.3.5** Ammonium hydroxide solution, diluted, 1 + 9.
- **6.2.3.6** Ammonium chloride, NH<sub>4</sub>Cl.

#### 6.2.4 Apparatus

- **6.2.4.1** Ordinary laboratory equipment.
- **6.2.4.2** Hot plate.
- **6.2.4.3** Analytical balance accurate to 1 mg.
- **6.2.4.4** Magnetic stirrer and magnetic rod, inert e.g. PTFE covered.
- **6.2.4.5** pH-meter with glass electrode, capable of measuring to an accuracy of 0,05.