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Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

The procedures used to develop this document and those intended for its further maintenance are described in the ISO/IEC Directives, Part 1. In particular, the different approval criteria needed for the different types of ISO documents should be noted. This document was drafted in accordance with the editorial rules of the ISO/IEC Directives, Part 2 (see www.iso.org/directives).

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For an explanation of the voluntary nature of standards, the meaning of ISO specific terms and expressions related to conformity assessment, as well as information about ISO's adherence to the World Trade Organization (WTO) principles in the Technical Barriers to Trade (TBT) see <u>www.iso</u> .org/iso/foreword.html.

This document was prepared by Technical Committee ISO/TC 158, *Gas analysis* in collaboration with Technical Committee ISO/TC 197, *Hydrogen technologies*.

Any feedback or questions on this document should be directed to the user's national standards body. A complete listing of these bodies can be found at <u>www.iso.org/members.html</u>.

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Introduction

The hydrogen supply infrastructure for fuel cell electric vehicles (FCVs) requires specifications and an operational protocol for maintaining the quality of the hydrogen used to fuel the vehicles. To fulfil these requirements, several documents have been written: ISO 14687 which sets forth the quality specifications of hydrogen, and ISO 19880-8 which specifies the quality assurance and control protocol for ensuring them. There was still a need for developing a standard on analytical methods to measure the level of contaminants found in the gaseous hydrogen fuel. The development and validation of these analytical protocols is necessary in order to assure the hydrogen quality required by ISO 14687 for permeating commercialized FCVs and hydrogen infrastructure in the market. This document sets criteria to validate the analytical methods used for the quality control at hydrogen distribution facilities.

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Gas analysis — Analytical methods for hydrogen fuel — Proton exchange membrane (PEM) fuel cell applications for road vehicles

1 Scope

This document specifies the validation protocol of analytical methods used for ensuring the quality of the gaseous hydrogen (H_2) at hydrogen distribution bases and hydrogen fuelling stations for road vehicles using proton exchange membrane (PEM) fuel cells. It also gives recommendations on the calculation of an uncertainty budget for the amount fraction.

This document is established mainly for analysis done in laboratories after the sampling of hydrogen either at hydrogen distribution bases or at hydrogen refuelling stations. The specific requirements for on-line monitoring are not covered by this document.

This document gives a list of suitable analytical techniques used to measure each impurity in hydrogen, according to the specification of hydrogen grade D defined by ISO 14687:—¹⁾.

Moreover, recommendations for keeping the integrity of the sample are also given in order to ensure the quality of the measurement. It also includes the requirements for reporting the analytical results.

2 Normative references ://standards.iteh.ai)

The following documents are referred to in the text in such a way that some or all of their content constitutes requirements of this document. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

ISO 14687:—, Hydrogen fuel quality — Product specification

3 Terms and definitions

No terms and definitions are listed in this document.

ISO and IEC maintain terminological databases for use in standardization at the following addresses:

- ISO Online browsing platform: available at https://www.iso.org/obp
- IEC Electropedia: available at <u>http://www.electropedia.org/</u>

4 Symbols

- *b*(%) relative bias in %
- *k*_Q multiplier used in calculating limit of quantification
- *m* number of replicate observations used during the validation of analytical method
- *n* number of replicate observations averaged when reporting results

1) Under preparation. Stage at the time of publication: ISO/DIS 14687:2018.

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R	relative recovery (apparent recovery) in %
R'	relative spike recovery in %
S	standard deviation
<i>s</i> ₀	estimated standard deviation of single results at or near zero concentration
<i>s</i> ′ ₀	standard deviation used for calculating a limit of detection (LOD) or limit of quantification (LOQ)
Sr	repeatability standard deviation
s _R	reproducibility standard deviation
u	standard uncertainty
XCRM	certified property value of a certified reference material
xLOD	value of the limit of detection
<i>x</i> _{LOQ}	value of the limit of quantification
<i>X</i> threshold	value of the specification for each impurity listed in ISO 14687:— grade D
x _{mean}	mean value (arithmetic average)
x _{ref}	reference value iTeh Standards
$x'_{\rm blank}$	mean value of the unspiked sample in a recovery experiment
x' _{mean}	mean value of spiked sample in a recovery experiment
<i>X</i> refproficiency	reference value assigned during a proficiency test
X _{spike}	added concentration in a recovery experiment

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5 Quality characteristics of the fuel

The quality specifications of hydrogen fuel dispensed to PEM fuel cells for road vehicles are listed in ISO 14687:—. The fuel quality at the dispenser nozzle applicable to the hydrogen fuel for PEM fuel cells in road vehicles shall meet the requirements of grade D of ISO 14687:—.

6 Requirements for analytical method validation and fit for purpose

6.1 General

Method validation of fit for purpose is basically the process of defining the application requirements, and confirming that the method under consideration has capabilities consistent with what the application requires. The method performance characteristics (Table 1) that are associated with method validation shall be evaluated as described in Eurachem guide ^[2]. Using the information produced during the validation, the laboratory shall determine if the method is suitable for the analysis of impurities in H₂. Criteria are defined for establishing the fit for purpose of the developed method. If the method doesn't fulfil these criteria, it shall not be used for the analysis of impurities in H₂; another method shall be used.

A laboratory may adopt a validated procedure which has been published as a standard, or buy a complete measuring system to be used for a specific application from a commercial manufacturer. In both of these cases, basic validation work has already been carried out but the laboratory shall confirm its ability to apply the method. This means that some experimental work shall be done to demonstrate

that the method works in the end-user's laboratory. The laboratory shall also determine if the method is suitable for the analysis of impurities in H_2 .

6.2 Characteristics for analytical methods

6.2.1 List of main characteristics

The list of the main characteristics commonly evaluated during method validation is given in <u>Table 1</u>.

Performance characteristic	Clause/Subclause
Selectivity	<u>6.2.2</u>
Limit of detection (LOD) and limit of quantification (LOQ)	<u>6.2.3</u>
Working range	<u>6.2.4</u>
Trueness	<u>6.2.5</u>
bias, recovery	
Precision	<u>6.2.6</u>
repeatability, intermediate precision and reproducibility	
Measurement uncertainty	<u>6.2.7</u>
Ruggedness (robustness)	6.2.8

Table 1 — Overview of characteristics for analytical methods

6.2.2 Selectivity

6.2.2.1 Definition and estimation

Analytical selectivity relates to "the extent to which the method can be used to determine particular analytes in mixtures or matrices without interferences from other components of similar behaviour"^[3].

The selectivity of a method is usually investigated by studying its ability to measure the analyte of interest in samples to which specific interferences have been deliberately introduced (those thought likely to be present in samples).

6.2.2.2 Fit for purpose for H₂ analysis

The selectivity is acceptable if:

- the effect of interferences is examined by the analysis of test samples containing various suspected interferents in the presence of analytes of interest;
- the presence of suspected interferents doesn't inhibit detection or quantification of the analytes;
- the presence of suspected interferents doesn't increase the uncertainty of the measurements.

If detection or quantification is inhibited by interferences, further method development is required or another method shall be used.

In the case of impurities in hydrogen, the interferences with the other impurities specified in ISO 14687:— have to be investigated. If other impurities not specified could have an effect on the measurement, they shall also be studied.

6.2.3 Limit of detection and limit of quantification

6.2.3.1 Definition and calculation

It is necessary to distinguish between the instrument detection limit and the method detection limit (x_{LOD}) . The instrument detection limit can be based on the analysis of a sample, often a blank, presented directly to the instrument (i.e. omitting any sample preparation steps), or on the signal to noise ratio in, for example, a chromatogram.

To obtain the method detection limit x_{LOD} , the calculation shall be based on the analysis of samples according to the whole measurement procedure (including sample preparation or preconcentration) and calculated with the same formula as for the test samples.

The determination limit or limit of quantification (x_{LOQ}) is the lowest amount of analyte in a sample that can be quantitatively determined with a stated acceptable precision and accuracy, under stated experimental conditions. Depending on the degree of rigour needed or the level of risk that can be tolerated, the x_{LOQ} is typically 4 to 20 times the standard deviation of a low level sample or blank measurement^[2].

The suitable samples for estimating x_{LOD} and x_{LOQ} should preferably be:

- a) blank samples, i.e. matrices containing no detectable analyte;
- b) test samples with concentrations of analyte close to expected x_{LOD} .

Blank samples work well for methods where a measurable signal is obtained for a blank, such as spectro-photometry and atomic spectroscopy. However, for techniques such as chromatography, which rely on detecting a peak above the noise, samples with concentration levels close to or above the x_{LOD} are required. These can be prepared by spiking a blank sample.

The standard deviation s_0 is normally obtained under repeatable conditions. It is important that this standard deviation is representative of the precision obtained for typical test samples, and that sufficient replicate measurements are made to give a reliable estimate. The number of replicates (m) should be at least 6 to give a representative value; 10 replicates are often recommended in validation protocols. The standard deviation s_0 is obtained by the following Formula (1).

$$s_0 = \sqrt{\sum (x_i - x_{\text{mean}})^2 / (m - 1)}$$
(1)

where

x_i is the measurement i;

x_{mean} is the average value;

m is the number of replicates used during validation of the analytical method.

In many measurement procedures, the mean of n replicates is reported in the routine use of the method, where each replicate is obtained by following the entire measurement procedure. In this case,