

## SLOVENSKI STANDARD SIST EN 16166:2022

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Nadomešča: SIST EN 16166:2013

# Tla, obdelani biološki odpadki in blato - Določevanje adsorbiranih organsko vezanih halogenov (AOX)

Soil, treated biowaste and sludge - Determination of adsorbed organically bound halogens (AOX)

## iTeh STANDARD

Boden, behandelter Bioabfall und Schlamm - Bestimmung von adsorbierten organisch gebundenen Halogenen (AOX)

Boues, biodéchets traités, sols et sédiments Détermination des composés organiques halogénés adsorbables (AOX)

<u>SIST EN 16166:2022</u>

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### SIST EN 16166:2022

# EUROPEAN STANDARD NORME EUROPÉENNE EUROPÄISCHE NORM

## EN 16166

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Supersedes EN 16166:2012

**English Version** 

# Soil, treated biowaste and sludge - Determination of adsorbed organically bound halogens (AOX)

Boues, biodéchets traités, sols et sédiments -Détermination des composés organiques halogénés adsorbables (AOX) Boden, behandelter Bioabfall und Schlamm -Bestimmung von adsorbierten organisch gebundenen Halogenen (AOX)

This European Standard was approved by CEN on 22 November 2021.

CEN members are bound to comply with the CEN/CENELEC Internal Regulations which stipulate the conditions for giving this European Standard the status of a national standard without any alteration. Up-to-date lists and bibliographical references concerning such national standards may be obtained on application to the CEN-CENELEC Management Centre or to any CEN member.

This European Standard exists in three official versions (English, French, German). A version in any other language made by translation under the responsibility of a CEN member into its own language and notified to the CEN-CENELEC Management Centre has the same status as the official versions.

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### <u>SIST EN 16166:2022</u>

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EUROPEAN COMMITTEE FOR STANDARDIZATION COMITÉ EUROPÉEN DE NORMALISATION EUROPÄISCHES KOMITEE FÜR NORMUNG

CEN-CENELEC Management Centre: Rue de la Science 23, B-1040 Brussels

### SIST EN 16166:2022

### EN 16166:2021 (E)

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### **European foreword**

This document (EN 16166:2021) has been prepared by Technical Committee CEN/TC 444 "Environmental characterization of solid matrices", the secretariat of which is held by NEN.

This European Standard shall be given the status of a national standard, either by publication of an identical text or by endorsement, at the latest by June 2022, and conflicting national standards shall be withdrawn at the latest by June 2022.

Attention is drawn to the possibility that some of the elements of this document may be the subject of patent rights. CEN shall not be held responsible for identifying any or all such patent rights.

This document supersedes EN 16166:2012.

In comparison with the previous edition, the following technical modifications have been made:

- Calculation harmonized with EN ISO 9562;
- Initial check and daily check procedure harmonized with EN ISO 9562 and extended to include full measuring process.

Any feedback and questions on this document should be directed to the users' national standards body. A complete listing of these bodies can be found on the CEN website.

According to the CEN-CENELEC Internal Regulations, the national standards organisations of the following countries are bound to implement this European Standard: Austria, Belgium, Bulgaria, Croatia, Cyprus, Czech Republic, Denmark, Estonia, Finland, France, Germany, Greece, Hungary, Iceland, Ireland, Italy, Latvia, Lithuania, Luxembourg, Malta, Netherlands, Norway, Poland, Portugal, Republic of North Macedonia, Romania, Serbia, Slovakia, Slovenia, Spain, Sweden, Switzerland, Turkey and the United Kingdom.

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### Introduction

This document is applicable and validated for several types of matrices as indicated in Table 1 (see also Annex A for the results of the validation).

NOTE This method can also be applied to other environmental solid matrices, provided the user has verified the applicability.

Matrix	Materials used for validation
Sludge	Municipal sludge
Compost	Fresh compost
	Compost
Soil	Sludge amended soil
	Agricultural soil

Table 1 — Matrices for which this document is applicable and validated

**WARNING** — Persons using this document should be familiar with usual laboratory practice. This document does not purport to address all of the safety problems, if any, associated with its use. It is the responsibility of the user to establish appropriate safety and health practices and to ensure compliance with any national regulatory conditions.

**IMPORTANT** — It is absolutely essential that tests conducted according to this document be carried out by suitably trained staff.

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#### Scope 1

This document specifies an operationally defined method for the direct determination of organically bound halogens (chlorine, bromine and iodine) adsorbed and occluded to the sample matrix. AOX being a methodologically defined parameter, it is essential that the procedure is applied without any modification.

This document is intended for analysis of sludge, treated biowaste or soil in concentrations ranging from 5 mg/kg dry matter. The upper limit and exact concentration range covered depend on the instrumentation used for determination.

NOTE This method can also be applied to other environmental solid matrices, provided the user has verified the applicability.

#### Normative references 2

There are no normative references in this document.

#### **Terms and definitions** 3

For the purposes of this document, the following terms and definitions apply.

ISO and IEC maintain terminological databases for use in standardization at the following addresses:

- IEC Electropedia: available at https://www.electropedia.org
- ISO Online browsing platform: available at https://www.iso.org/obp/ui

# 3.1 (standards.iteh.ai) adsorbed organically bound halogens, AOX

amount of chlorine, bromine, and iodine contained as organic compounds in the sample matrix, expressed as chloride, when determined according to this document 2

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Note 1 to entry: This includes halogenated organic compounds adsorbed at the dry sample matrix as well as water soluble halogenated compounds that can be adsorbed on to activated carbon.

#### **Principle** 4

Activated carbon is added to a dried, homogenized solid sample. Inorganic halides are eluted and at the same time water soluble organic compounds are adsorbed onto the activated carbon by shaking with acidified nitrate solution.

The loaded carbon/sample mixture is combusted in an oxygen stream.

The hydrogen halides produced are absorbed followed by the determination of the halide ions, for example, by microcoulometric titration. The result is expressed as the mass fraction of chloride.

### **5** Interferences

Sparingly soluble or occluded inorganic halides are included in the determination and can, if present, give a significant positive bias. Adequate washing is essential to remove inorganic interference.

Iodide contained in the sample can be adsorbed to the active carbon and not be washed off, thus leading to positive bias.

Organic bromine and iodine compounds can, during combustion, lead to the formation of elemental bromine or iodine respectively or to the formation of halogen oxides. The determination of these AOX fractions can be incomplete, thus leading to negative bias.

For samples with significant amounts of volatile, purgeable organic halogen compounds negative bias can occur. In this case, the sample is to be suspended in water and purged with oxygen at 60 °C for 30 min. The purged gases are analysed by the analytical instrument. The resulting portion of purgeable organic halogens is added to the result of the regular determination.

### **6** Reagents

Use only reagents of recognized analytical grade. Water should be of high purity.

The AOX contribution from water, reagents and gases should be significantly lower than the lowest AOX content to be determined. The overall AOX content of water, chemicals, and gases shall be checked by measuring the total blank (see 9.5).

### **6.1 Activated carbon**, grain size approximately 10 μm to 50 μm.

For information for the storage of activated carbon, see Annex B.

The blank value of the activated carbon, expressed as chloride, shall be less than 15  $\mu g$  per gram of activated carbon.

- **6.2** Nitric acid, HNO<sub>3</sub>,  $\rho$  = 1,4 g/ml, 65 % solution.
- **6.3 Hydrochloric acid**, *c*(HCl) = 0,010 mol/l.
- 6.4 Sulfuric acid,  $H_2SO_4$ ,  $\rho = 1,84$  ig/mth STANDARD
- **6.5** Gases for combustion, e.g. oxygen  $(O_2)$ , or a mixture of oxygen and an inert gas.
- 6.6 Sodium nitrate, NaNO<sub>3</sub>, for the preparation of stock solution.
- **6.7** Nitrate stock solution, acidified, *c*(NaNO<sub>3</sub>) = 0,2 mol/l.

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Dissolve 17 g of sodium nitrate (6.6) in water in a 1 c000 ml volumetric flash, add 15 ml of nitric acid (6.2), and make up to volume with water 4b6-87d8-e6f528e3e3ec/sist-en-16166-2022

If stored in a brown glass bottle, the solution is stable for three months.

### **6.8** Nitrate washing solution, $c(NaNO_3) = 0,01 \text{ mol/l}$ .

Pipette 50 ml of the nitrate stock solution (6.7) in a 1 000 ml volumetric flask, and make up to volume with water.

If stored in a brown glass bottle, the solution is stable for three months.

### **6.9 4-chlorophenol, stock solution**, $\rho_{Cl}(AOX) = 200 \text{ mg/l}.$

Dissolve 72,5 mg of 4-chlorophenol ( $C_6H_5ClO$ ) in water in a 100 ml volumetric flask and make up to volume with water.

If stored at  $(5 \pm 3)$  °C the solution is stable for one month.

**6.10 2-Chlorobenzoic acid**, stock solution  $\rho_{Cl}(AOX) = 250 \text{ mg/l}$ .

Dissolve 110,4 mg 2-chlorobenzoic acid in a 100 ml volumetric flask in water and make up to volume with water.

The dissolution of 2-chlorobenzoic acid is very slow. It is recommended to prepare this solution the day before using it. This stock solution may be stored for one month at 2 °C to 8 °C in a glass bottle.

### 7 Apparatus

### 7.1 Apparatus for combustion and detection

7.1.1 Combustion apparatus, consisting of furnace, heated to at least 950 °C, equipped with a combustion tube made of quartz glass or ceramic. Both vertically and horizontally arranged combustion tubes are suitable.

7.1.2 Combustion vessel, e.g. ceramic or quartz container, suitable to be inserted into the combustion tube.

7.1.3 Measuring device for determining halide concentrations, a microcoulometer, capable of determining 1 µg absolute amount of chloride with a repeatability variation coefficient of less than 10 %.

**7.1.4** Absorber, filled with sulfuric acid (6.4), to dry the gas stream and designed so that the acid does not backflush into the furnace.

Syringe, to pipette volumes of 1 µl to 10 µl of hydrochloric acid (6.3), 4-chlorophenol or 7.1.5 4-chlorobenzoic acid solutions (6.9 and 6.10).

### 7.2 Equipment for adsorption

**7.2.1** Filtration apparatus, e.g. with a funnel capacity of 0,15 l and a filter diameter of 25 mm.

7.2.2 **Low-halide polycarbonate membrane filter**, to fit the filtration apparatus (7.2.1), with a pore size of 0,45 µm, or any equivalent filtration material, such as a dedicated quartz or ceramic filter for AOX determination. (standards.iteh.ai)

Conical flask (Erlenmeyer flask) of 25 ml capacity with ground glass stopper or 12 ml to 20 ml 7.2.3 screw cap vial with polytetrafluoroethene (PTFE) lined cap.

https://standards.iteh.ai/catalog/standards/sist/b27d6fc6-Mechanical shaker device for the flasks, specified in 7,23, e.g. with a carrier plate. 7.2.4

### 7.3 Equipment for sample preparation

### 7.3.1 Porcelain evaporating dish.

**7.3.2** Oven with forced ventilation or natural ventilation through adjustable vents adjustable to (105 ± 5) °C.

**Desiccator**, provided with a suitable desiccant. 7.3.3

7.3.4 Analytical mill or porcelain mortar.

7.3.5 Precision balance.

7.3.6 pH measuring device.

### 8 Sample storage and pretreatment

### 8.1 Sampling and storage

For the sampling and storage of sludge samples, refer to existing standard, e.g. [2] and [3].

Samples shall be stored in suitable containers with an appropriate closure material such as PTFE. Samples to be frozen may be stored in aluminium containers pre-cleaned by heating to 450 °C for minimum 4 h or by rinsing with a non-chlorinated solvent.

Samples shall be kept at a temperature of  $(5 \pm 3)$  °C and in the dark. The sample pretreatment for sludge samples should take place within 24 h of sampling. Alternatively, samples can be frozen  $(-18 \pm 3)$  °C) directly after sampling and kept frozen for a maximum of one month before sample pretreatment.

### 8.2 Sample pretreatment

Pretreat the samples according to EN 16179 [8], if not otherwise specified.

Cool the dried test sample in a desiccator, comminute and homogenize in an analytical mill or porcelain mortar (7.3.4) to a particle size of no more than 0,1 mm.

Store the ground material in a desiccator or a tightly closed glass container.

The homogenised wet sample portion may alternatively be freeze-dried, which in some cases makes homogenizing of the dried sample easier.

PREVIEW

### 9 Procedure

### 9.1 General

The test portion taken for analysis shall have an AOX value within the working range of the instrument.

NOTE A typical working range is between 1 µg to 30 µg chlorine (absolute titrated amount).

### 9.2 Adsorption and inbrganic halide temovatalog/standards/sist/b27d6fc6-

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Ensure that the dried, ground test sample is homogenized by stirring or shaking before taking the test portion. Prepare the test portion for combustion as follows:

- Transfer a homogenized test portion of 10 mg to 100 mg depending on the expected AOX content to a conical flask or screw cap vial (7.2.3).
- Add approx. 20 mg activated carbon (6.1) and 10 ml nitrate stock solution (6.7) to the sample. The
  amount of activated carbon shall be the same for all test samples, blank determinations, etc.
- Shake for 1 h using a mechanical shaker (7.2.4).
- Filter the suspension through a filtration apparatus (7.2.1) using nitrate washing solution (6.8). Wash
  the filter cake with small amounts of nitrate washing solution. Use a total volume of 25 ml to ensure
  complete transfer.
- Transfer the moist filter and the filter cake to the combustion apparatus according to manufacturer's instructions and proceed according to 9.3.

The laboratory should ensure that the nitrate washing procedure is sufficient to remove all inorganic halides typically found in the types of samples routinely analysed. If the concentration of AOX is found to decrease with additional washing stages, then the number of nitrate washes should be increased accordingly, without increasing the volume of washing solution.