



**SLOVENSKI STANDARD**  
**SIST EN 12932:2000**

**01-november-2000**

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**Kemikalije, ki se uporabljajo za pripravo pitne vode - Kemikalije za uporabo v sili - Natrijev dikloroizocianurat, dehidriran**

Chemicals used for water intended for human consumption - Chemicals for emergency use - Sodium dichloroisocyanurate, dihydrate

Produkte zur Aufbereitung von Wasser für den menschlichen Gebrauch - Produkte für den Notfall - Natriumdichloroisocyanurat, Dihydrat

Produits chimiques utilisés pour le traitement de l'eau destinée a la consommation humaine - Produits chimiques utilisés en cas d'urgence - Dichloroisocyanurate de sodium dihydraté

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**Ta slovenski standard je istoveten z: EN 12932:2000**

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**ICS:**

13.060.20	Pitna voda	Drinking water
71.100.80	Kemikalije za čiščenje vode	Chemicals for purification of water

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**en**

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EUROPEAN STANDARD  
NORME EUROPÉENNE  
EUROPÄISCHE NORM

EN 12932

April 2000

ICS 71.100.80

English version

Chemicals used for treatment of water intended for human  
consumption - Chemicals for emergency use - Sodium  
dichloroisocyanurate, dihydrate

Produits chimiques utilisés pour le traitement de l'eau  
destinée à la consommation humaine - Produits chimiques  
utilisés en cas d'urgence - Dichloroisocyanurate de sodium  
dihydraté

Produkte zur Aufbereitung von Wasser für den  
menschlichen Gebrauch - Produkte für den Notfall -  
Natriumdichloroisocyanurat, Dihydrat

This European Standard was approved by CEN on 22 March 2000.

CEN members are bound to comply with the CEN/CENELEC Internal Regulations which stipulate the conditions for giving this European Standard the status of a national standard without any alteration. Up-to-date lists and bibliographical references concerning such national standards may be obtained on application to the Central Secretariat or to any CEN member.

This European Standard exists in three official versions (English, French, German). A version in any other language made by translation under the responsibility of a CEN member into its own language and notified to the Central Secretariat has the same status as the official versions.

CEN members are the national standards bodies of Austria, Belgium, Czech Republic, Denmark, Finland, France, Germany, Greece, Iceland, Ireland, Italy, Luxembourg, Netherlands, Norway, Portugal, Spain, Sweden, Switzerland and United Kingdom.



EUROPEAN COMMITTEE FOR STANDARDIZATION  
COMITÉ EUROPÉEN DE NORMALISATION  
EUROPÄISCHES KOMITEE FÜR NORMUNG

Central Secretariat: rue de Stassart, 36 B-1050 Brussels

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## Foreword

This European Standard has been prepared by Technical Committee CEN/TC 164 "Water supply", the secretariat of which is held by AFNOR.

This European Standard shall be given the status of a national standard, either by publication of an identical text or by endorsement, at the latest by October 2000, and conflicting national standards shall be withdrawn at the latest by October 2000.

According to the CEN/CENELEC Internal Regulations, the national standards organizations of the following countries are bound to implement this European Standard: Austria, Belgium, Czech Republic, Denmark, Finland, France, Germany, Greece, Iceland, Ireland, Italy, Luxembourg, Netherlands, Norway, Portugal, Spain, Sweden, Switzerland and the United Kingdom.

Annex A is informative. The annexes B and C are normative.

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## Introduction

In respect of potential adverse effects on the quality of water intended for human consumption, caused by the product covered by this Standard :

- 1) this Standard provides no information as to whether the product may be used without restriction in any of the Member States of the EU or EFTA ;
- 2) it should be noted that, while awaiting the adoption of verifiable European criteria, existing national regulations concerning the use and/or the characteristics of this product remain in force.

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## 1 Scope

This European Standard is applicable to sodium dichloroisocyanurate dihydrate used for emergency treatment of water intended for human consumption. It describes the characteristics of sodium dichloroisocyanurate dihydrate and specifies the requirements and the corresponding test methods for sodium dichloroisocyanurate dihydrate. It gives information on its use in water treatment.

## 2 Normative references

This European Standard incorporates by dated or undated reference, provisions from other publications. These normative references are cited at the appropriate places in the text and the publications are listed hereafter. For dated references, subsequent amendments to or revisions of any of these publications apply to this European Standard only when incorporated in it by amendment or revision. For undated references the latest edition of the publication referred to applies.

EN ISO 3696, *Water for analytical laboratory use - Specification and test methods* (ISO 3696:1987).

EN 1483, *Water quality - Determination of mercury*.

ISO 3165, *Sampling of chemical products for industrial use - Safety in sampling*.

ISO 6206, *Chemical products for industrial use - Sampling - Vocabulary*.

ISO 8213, *Chemical products for industrial use - Sampling techniques - Solid chemical products in the form of particles varying from powders to coarse lumps*.

ISO 8288, *Water quality - Determination of cobalt, nickel, copper, zinc, cadmium and lead - Flame atomic absorption spectrometric methods*.

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ISO 9174, *Water quality - Determination of chromium - Atomic absorption spectrometric methods*.

## 3 Description

### 3.1 Identification

#### 3.1.1 Chemical name

1-sodium, 3, 5-dichloro-1,3,5-triazine-2, 4, 6-trione dihydrate.

#### 3.1.2 Synonym or common name

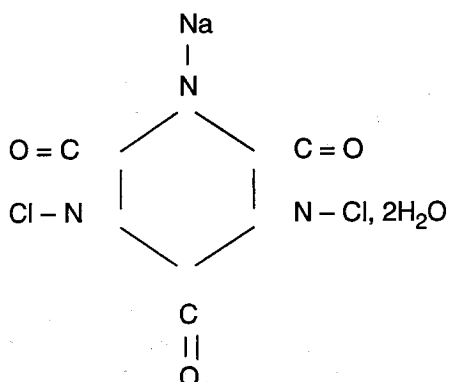
Sodium dichloroisocyanurate, dihydrate.

#### 3.1.3 Relative molecular mass

256,02.

#### 3.1.4 Empirical formula

$C_3N_3O_3Cl_2Na \cdot 2H_2O$ .

**3.1.5 Chemical formula****3.1.6 CAS Registry Number <sup>1)</sup>**

51580-86-0.

**3.1.7 EINECS reference <sup>2)</sup>**

Not listed.

**3.2 Commercial form**

The product is available in various granular forms.

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**3.3 Physical properties**

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**3.3.1 Appearance and odour**

The product is a white granular solid with chlorinous odour.

**3.3.2 Density**The bulk density is approximately 0,9 g/cm<sup>3</sup>.**3.3.3 Solubility in water**

The solubility in water is 291 g/l at 25 °C.

**3.3.4 Vapour pressure**

Not applicable.

**3.3.5 Boiling point at 100 kPa <sup>3)</sup>**

Not applicable, the product decomposes before fusion.

1) Chemical Abstracts Service Registry Number.

2) European Inventory of Existing Commercial Chemical Substances.

3) 100 kPa = 1 bar.



### 3.3.6 Melting point

Not applicable.

### 3.3.7 Specific heat

1090 ± 40 kJ/kg · K at 20 °C.

### 3.3.8 Viscosity, (dynamic)

Not applicable.

### 3.3.9 Critical temperature

Not applicable.

### 3.3.10 Critical pressure

Not applicable.

### 3.3.11 Physical hardness

Not applicable.

## 3.4 Chemical properties

The product is a strong oxidant. It liberates chlorine by hydrolysis.

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## 4 Purity criteria

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Limits have been given for impurities and toxic substances where these are likely to be present in significant quantities from the current production process and raw materials. If a change in the production process or raw materials leads to significant quantities of other impurities or by-products being present, this shall be notified to the user.

### 4.1 Composition of commercial product

The product shall contain at least 55 percent by mass (% (*m/m*)) of available chlorine.

### 4.2 Impurities and main by-products

The sodium chloride content shall be less than 0,05 percent by mass (% (*m/m*)) of the product.

### 4.3 Toxic substances

NOTE For the purpose of this standard, "toxic substances" are those defined in the EU Directive 80/778/EEC of 15 July, 1980 (see [1]).

The content of toxic substances shall conform to the requirements specified in Table 1.

**Table 1 - Toxic substances**

Element		Limit mg/kg of product	
		Type 1	Type 2
Arsenic (As)	max.	10	10
Cadmium (Cd)	max.	1	1
Chromium (Cr)	max.	6	10
Mercury (Hg)	max.	0,02	0,02
Nickel (Ni)	max.	3	3
Lead (Pb)	max.	4	4
Antimony (Sb)	max.	5	5
Selenium (Se)	max.	1	1

NOTE Cyanide which does not exist in a strong oxidizing medium such as sodium dichloroisocyanurate, dihydrate is not a relevant toxic substance. Pesticides and polycyclic aromatic hydrocarbons are not by-products of the manufacturing process.

## 5 Test methods

### 5.1 Sampling

Observe the general recommendations of ISO 3165 and take account of ISO 6206. Prepare the laboratory sample(s) required by the relevant procedure described in ISO 8213.

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### 5.2 Analysis

#### 5.2.1 Determination of available chlorine (main product)

##### 5.2.1.1 Principle

The available chlorine is determined by measuring active chlorine in the sample. The oxidizing chlorine reacts with potassium iodide releasing iodine which is then titrated with sodium thiosulfate standard volumetric solution in the presence of starch indicator solution.

##### 5.2.1.2 Reagents

All reagents shall be of a recognized analytical grade and the water used shall conform to grade 3 in accordance with EN ISO 3696.

5.2.1.2.1 Potassium iodide crystals (KI).

5.2.1.2.2 Glacial acetic acid.

5.2.1.2.3 Sodium thiosulfate standard volumetric solution,  $c(\text{Na}_2\text{S}_2\text{O}_3 \cdot 5\text{H}_2\text{O}) = 0,1 \text{ mol/l}$ .

Dissolve 24,8 g of  $\text{Na}_2\text{S}_2\text{O}_3 \cdot 5\text{H}_2\text{O}$  with water. Add 0,5 ml of chloroform as preservative, dilute to volume with water in a 1 000 ml one-mark volumetric flask and mix thoroughly.

To standardize : weigh, to the nearest 0,1 mg ( $m$ ), (160 ± 10) mg of primary standard potassium dichromate into a tared glass beaker. Place the contents of the beaker in a 500 ml stoppered conical flask, add 100 ml of water and (2 ± 0,5) g of potassium iodide and stir to dissolve. Add (15 ± 1) ml of hydrochloric acid solution (diluted 1 + 1 by volume), swirl and allow to stand for 5 min. Titrate with the sodium thiosulfate solution until the solution is pale yellow. Add (5 ± 1) ml of starch solution (5.2.1.2.4) and titrate to the end point, i.e. to the disappearance of the blue-black colour. Record the volume ( $V$ ) used.

The concentration,  $c$ , of the sodium thiosulfate standard volumetric solution ( $\text{Na}_2\text{S}_2\text{O}_3 \cdot 5\text{H}_2\text{O}$ ) expressed in moles per litre is given by the following equation :

$$c = \frac{m}{V \times 49,0317}$$

where

$m$  is the mass, in milligrams, of potassium dichromate ( $\text{K}_2\text{Cr}_2\text{O}_7$ ) weighed ;

$V$  is the volume, in millilitres, of the sodium thiosulfate standard volumetric solution used ;

49,0317 is the molar mass, in grams per mole, of potassium dichromate ( $\frac{1}{6} \text{K}_2\text{Cr}_2\text{O}_7$ )

#### 5.2.1.2.4 Starch solution, 1 % ( $m/m$ ).

Make a slurry with (1 ± 0,1) g starch and (5 ± 1) ml water. Add (90 ± 5) ml boiling water to the slurry. Stir to dissolve it and cool the solution. This solution needs refrigeration to avoid the decomposition of the starch which results in a vague end point. Keep the solution cool and use it within one week.

NOTE Commercial indicators for iodine titration exist and can be used in place of the specified starch solution provided that their efficiency has been previously tested. [SIST EN 12932:2000](https://standards.iteh.ai/catalog/standards/sist/feaa42b6-b186-4dd9-8044-10ce04e3ca1e/sist-en-12932-2000)

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#### 5.2.1.3 Apparatus

Ordinary laboratory apparatus and glassware.

#### 5.2.1.4 Procedure

##### 5.2.1.4.1 Test portion

Weigh, to the nearest 0,1 mg, 0,25 g of the laboratory sample ( $m_0$ ) into a tared stoppered weighing bottle.

##### 5.2.1.4.2 Determination

Transfer the test portion to a 200 ml volumetric flask.

Add 10 ml of water and 10 ml of glacial acetic acid (5.2.1.2.2), stir for 5 min and then add 100 ml of water and 2 g of potassium iodide (5.2.1.2.1), mix to dissolve and wait 10 min. Titrate with the sodium thiosulfate standard volumetric solution (5.2.1.2.3) to a light yellow colour. Add 5 ml of the starch solution (5.2.1.2.4) and continue titration to the disappearance of the blue black colour. Record the volume ( $V_1$ ) of the sodium thiosulfate standard volumetric solution used.

**5.2.1.5 Expression of results**

The chlorine ( $\text{Cl}_2$ ) content,  $C_1$ , expressed as a percentage by mass (%) ( $m/m$ ), is given by the following equation :

$$C_1 = \frac{V_1 \times 0,003545 \times C}{m_0} \times 100$$

where

$V_1$  is the volume, in millilitres, of the sodium thiosulfate standard volumetric solution (5.2.1.2.3) used for the titration ;

$m_0$  is the mass, in grams, of the test portion (5.2.1.4.1) ;

0,003545 is the mass, in grams, of chlorine ( $\text{Cl}_2$ ) corresponding to 1,00 ml of sodium thiosulfate solution,  $c(\text{Na}_2\text{S}_2\text{O}_3 \cdot 5\text{H}_2\text{O}) = 0,1 \text{ mol/l}$  ;

$c$  is the actual concentration, expressed in moles per litre, of the sodium thiosulfate standard volumetric solution (5.2.1.2.3).

**5.2.1.6 Accuracy**

The result is accurate to within  $\pm 0,1 \%$ .

**5.2.2 Impurities**

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**5.2.2.1 Sodium chloride**

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**5.2.2.1.1 General**

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This method applies to products with sodium chloride content in the range of 0 % ( $m/m$ ) to 0,05 % ( $m/m$ ).

**5.2.2.1.2 Principle**

The sodium dichloroisocyanurate dihydrate is acidified and then digested to remove all traces of available chlorine. Sodium is analysed in solution by atomic absorption spectrometry.

**5.2.2.1.3 Reagents**

All reagents shall be of a recognized analytical grade and the water used shall conform to grade 3 in accordance with EN ISO 3696.

**5.2.2.1.3.1** Sulfuric acid concentrated, ( $\text{H}_2\text{SO}_4$ ) 95 % ( $m/m$ ) minimum.

**5.2.2.1.3.2** Sodium standard stock solution  $c(\text{Na}) = 1\,000 \mu\text{g/ml}$ .

Weigh, 2,542 1 g of sodium chloride that has been previously dried for 1 h at 110 °C. Dissolve with water and dilute to volume with water in a 1 000 ml one mark volumetric flask. Mix well.