

**SLOVENSKI STANDARD****SIST EN 1204:1999****01-april-1999****Kemikalije, ki se uporabljajo za pripravo pitne vode – Monokalcijev fosfat**

Chemicals used for treatment of water intended for human consumption - Monocalcium phosphate

Produkte zur Aufbereitung von Wasser für den menschlichen Gebrauch -  
Monocalciumphosphat

**iTeh STANDARD PREVIEW**

Produits chimiques utilisés pour le traitement de l'eau destinée à la consommation humaine - Bis-dihydrogénophosphate de calcium  
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**ICS:**

13.060.20	Pitna voda	Drinking water
71.100.80	Kemikalije za čiščenje vode	Chemicals for purification of water

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**EUROPEAN STANDARD  
NORME EUROPÉENNE  
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**EN 1204**

September 1997

ICS 71.100.80

Descriptors: potable water, water treatment, chemical compounds, phosphates, calcium phosphates, description, physical properties, chemical properties, impurities, toxic substances, tests, labelling, storage, information

English version

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This European Standard was approved by CEN on 26 September 1997.

CEN members are bound to comply with the CEN/CENELEC Internal Regulations which stipulate the conditions for giving this European Standard the status of a national standard without any alteration. Up-to-date lists and bibliographical references concerning such national standards may be obtained on application to the Central Secretariat or to any CEN member.

**ITEN STANDARD REVIEW**

This European Standard exists in three official versions (English, French, German). A version in any other language made by translation under the responsibility of a CEN member into its own language and notified to the Central Secretariat has the same status as the official versions.

CEN members are the national standards bodies of Austria, Belgium, Czech Republic, Denmark, Finland, France, Germany, Greece, Iceland, Ireland, Italy, Luxembourg, Netherlands, Norway, Portugal, Spain, Sweden, Switzerland and United Kingdom.

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EUROPEAN COMMITTEE FOR STANDARDIZATION  
COMITÉ EUROPÉEN DE NORMALISATION  
EUROPÄISCHES KOMITEE FÜR NORMUNG

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**Foreword**

This European Standard has been prepared by Technical Committee CEN/TC 164 "Water supply", the secretariat of which is held by AFNOR.

This European Standard shall be given the status of a national standard, either by publication of an identical text or by endorsement, at the latest by March 1998, and conflicting national standards shall be withdrawn at the latest by March 1998.

According to the CEN/CENELEC Internal Regulations, the national standards organizations of the following countries are bound to implement this European Standard: Austria, Belgium, Czech Republic, Denmark, Finland, France, Germany, Greece, Iceland, Ireland, Italy, Luxembourg, Netherlands, Norway, Portugal, Spain, Sweden, Switzerland and the United Kingdom.

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**Introduction**

In respect of potential adverse effects on the quality of water intended for human consumption, caused by the product covered by this Standard :

- 1) This Standard provides no information as to whether the product may be used without restriction in any of the Member States of the EU or EFTA ;
- 2) It should be noted that, while awaiting the adoption of verifiable European criteria, existing national regulations concerning the use and/or the characteristics of this product remain in force.

**1 Scope**

This European Standard describes the characteristics and specifies the requirements and the corresponding tests methods for monocalcium phosphate used for treatment of water intended for human consumption. It gives information on its use in water treatment.

**2 Reference normatives**

This European Standard incorporates by dated or undated reference, provisions from other publications. These normative references are cited at the appropriate places in the text and the publications are listed hereafter. For dated references, subsequent amendments to or revisions of any of these publications apply to this European standard only when incorporated in it by amendment or revision. For undated references the latest edition of the publication referred to applies.

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EN 26595	Water quality - Determination of total arsenic - Silver diethyldithiocarbamate spectrophotometric method (ISO 6595:1982)
EN ISO 3696	Water for analytical laboratory use - Specification and test methods (ISO 3696 : 1987)
ISO 2997	Phosphoric acid for industrial use - Determination of sulphate content - Method by reduction and titrimetry
ISO 3165	Sampling of chemical products for industrial use - Safety in sampling
ISO 3357	Sodium tripolyphosphate and sodium pyrophosphate for industrial use - Determination of total phosphorus (V) oxide content - Quinoline phosphomolybdate gravimetric method
ISO 3360	Phosphoric acid and sodium phosphates for industrial use (including foodstuffs) - Determination of fluorine content - Alizarin complexone and lanthanum nitrate photometric method
ISO 5373	Condensed phosphates for industrial use (including foodstuffs) - Determination of calcium content - Flame atomic absorption spectrometric method

- ISO 5666-1 Water quality - Determination of total mercury by flameless atomic absorption spectrometry - Part 1 : Method after digestion with permanganate-peroxodisulfate
- ISO 5961 Water quality - Determination of cadmium by atomic absorption spectrometry
- ISO 6206 Chemical products for industrial use - Sampling - Vocabulary
- ISO 6703-1 Water quality - Determination of cyanide - Part 1 : Determination of total cyanide
- ISO 8213 Chemical products for industrial use - Sampling techniques - Solid chemical products in the form of particles varying from powders to coarse lumps
- ISO 8288 Water quality - Determination of cobalt, nickel, copper, zinc, cadmium and lead - Flame atomic absorption spectrometric methods
- ISO 9174 Water quality - Determination of total chromium - Atomic absorption spectrometric methods
- ISO 9965 Water quality - Determination of selenium - Atomic absorption spectrometric method (hydride technique)
- ISO 11885 Water quality - Determination of 33 elements by inductively coupled plasma atomic emission spectroscopy
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### 3 Description

#### 3.1 Identification

##### 3.1.1 Chemical name

Monocalcium phosphate.

##### 3.1.2 Synonym or common name

Calcium-bis-dihydrogen monophosphate. Acid calcium phosphate monohydrate. Calcium phosphate monobasic.

##### 3.1.3 Relative molecular mass

234,0.

##### 3.1.4 Empirical formula

$\text{Ca}(\text{H}_2\text{PO}_4)_2$ .

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### 3.1.5 Chemical formula

Ca(H2PO4)2

### 3.1.6 CAS Registry Number <sup>1)</sup>

7758-11-4.

### 3.1.7 EINECS reference <sup>2)</sup>

231-837-1.

## 3.2 Commercial form

Monocalcium phosphate is a powder.

## 3.3 Physical properties

### 3.3.1 Appearance

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### 3.3.2 Density

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Bulk density from 500 g/dm<sup>3</sup> to 800 g/dm<sup>3</sup>.

### 3.3.3 Solubility in water

Approximately 0,005 g/l at 25 °C.

### 3.3.4 Vapour pressure

Not applicable.

### 3.3.5 Boiling point at 100 kPa <sup>3)</sup>

Not applicable.

### 3.3.6 Melting point

1 000 °C.

<sup>1)</sup> Chemical Abstracts Service Registry Number

<sup>2)</sup> European Inventory of Existing Commercial Chemical Substances

<sup>3)</sup> 100 kPa = 1 bar

**3.3.7 Specific heat**

Not known.

**3.3.8 Viscosity (dynamic)**

Solid : Not applicable.

**3.3.9 Critical temperature**

Not applicable.

**3.3.10 Critical pressure**

Not applicable.

**3.3.11 Physical hardness**

Not applicable.

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**3.4 Chemical properties**

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 Suspensions of monocalcium phosphate have acidic reactions.  
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The pH value of a suspension containing 10 % (m/m) is approximately 3.

**4 Purity criteria**

Limits have been given for impurities and toxic substances where these are likely to be present in significant quantities from the current production process and raw materials. If a change in the production process or raw materials leads to significant quantities of other impurities or by-products being present, this shall be notified to the user.

**4.1 Composition of commercial product**

The product shall conform to the following requirements on a dry mass basis :

- Phosphate content expressed as P<sub>2</sub>O<sub>5</sub> : (56 ± 2,0) percent by mass (% (m/m)) ;
- Calcium content expressed as CaO : (23 ± 2,0) percent by mass (% (m/m)).