
**Infant formula and adult
nutritionals — Determination of *trans*
and total (*cis* + *trans*) vitamin K₁
content — Normal phase HPLC**

*Formules infantiles et produits nutritionnels pour adultes —
Détermination de la teneur en vitamine K₁ trans et totale (cis +
trans) — Chromatographie liquide à haute performance (CLHP) en
phase normale*

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Published in Switzerland

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Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

The procedures used to develop this document and those intended for its further maintenance are described in the ISO/IEC Directives, Part 1. In particular, the different approval criteria needed for the different types of ISO documents should be noted. This document was drafted in accordance with the editorial rules of the ISO/IEC Directives, Part 2 (see www.iso.org/directives).

Attention is drawn to the possibility that some of the elements of this document may be the subject of patent rights. ISO shall not be held responsible for identifying any or all such patent rights. Details of any patent rights identified during the development of the document will be in the Introduction and/or on the ISO list of patent declarations received (see www.iso.org/patents).

Any trade name used in this document is information given for the convenience of users and does not constitute an endorsement.

For an explanation of the voluntary nature of standards, the meaning of ISO specific terms and expressions related to conformity assessment, as well as information about ISO's adherence to the World Trade Organization (WTO) principles in the Technical Barriers to Trade (TBT) see www.iso.org/iso/foreword.html.

This document was prepared by Technical Committee ISO/TC 34, *Food products* in collaboration with AOAC INTERNATIONAL. It is being published by ISO and separately by AOAC INTERNATIONAL. The method described in this document is equivalent to the AOAC Official Method 2015.09: *Trans vitamin K₁ in Infant, Pediatric, and Adult Nutritionals, HPLC with Fluorescence Detection*.

Any feedback or questions on this document should be directed to the user's national standards body. A complete listing of these bodies can be found at www.iso.org/members.html.

Infant formula and adult nutritionals — Determination of *trans* and total (*cis* + *trans*) vitamin K₁ content — Normal phase HPLC

1 Scope

This document specifies a method for the quantitative determination of *trans* and total (*cis* + *trans*) vitamin K₁ in infant, pediatric and adult nutritionals using normal phase (NP) high-performance liquid chromatography (HPLC) with post-column reduction and fluorescence detection. The method demonstrated good linearity over a standard range of ~2 µg/l to 80 µg/l *trans* vitamin K₁, and the limit of quantification (LOQ) was estimated to be 0,4 µg/l for standards and 0,09 µg/100 g ready to feed (RTF) for samples assuming 4 grams of sample are diluted to 10 ml.

2 Normative references

There are no normative references in this document.

3 Terms and definitions

For the purposes of this document, the following terms and definitions apply.

ISO and IEC maintain terminological databases for use in standardization at the following addresses:

- ISO Online browsing platform: available at <https://www.iso.org/obp>
- IEC Electropedia: available at <http://www.electropedia.org/>

3.1

adult nutritional

nutritionally complete, specially formulated food, consumed in liquid form, which may constitute the sole source of nourishment, made from any combination of milk, soy, rice, whey, hydrolysed protein, starch and amino acids, with and without intact protein

3.2

infant formula

breast-milk substitute specially manufactured to satisfy, by itself, the nutritional requirements of infants during the first months of life up to the introduction of appropriate complementary feeding

[SOURCE: Codex Standard 72-1981]

4 Principle

Vitamin K₁ is extracted from products with iso-octane after precipitation of proteins and release of lipids with methanol. Prepared samples are injected onto a silica HPLC column where *cis* and *trans* vitamin K₁ are separated with an iso-octane-isopropanol mobile phase. The column eluent is mixed with a dilute ethanolic solution of zinc chloride, sodium acetate, and acetic acid, and *cis* and *trans* vitamin K₁ are reduced to fluorescent derivatives in a zinc reactor column. The resulting fluorescent compounds are then detected by fluorescence at an excitation wavelength of 245 nm and an emission wavelength of 440 nm.

5 Reagents and materials

During the analysis, unless otherwise stated, use only reagents of recognized analytical grade and distilled or demineralized water or water of equivalent purity.

- 5.1 **Acetic acid**, glacial, > 99 %, American Chemical Society (ACS).
- 5.2 **Helium or nitrogen**, zero grade or equivalent helium or nitrogen.
- 5.3 **Hexane**, HPLC grade.
- 5.4 **Iso-octane (2,2,4-trimethylpentane)**, HPLC grade.
- 5.5 **Isopropanol (isopropyl alcohol)**, HPLC grade.
- 5.6 **Methanol**, HPLC grade.
- 5.7 **Phytonadione/phyloquinone (vitamin K₁)**, primary reference standard. Store per label instructions.
- 5.8 **Laboratory or distilled water**, with conductivity of 0,067 µS/cm (15 Mohm/cm).
- 5.9 **Ethanol**, 95 %, ACS.
- 5.10 **Sodium acetate**, anhydrous, ACS.
- 5.11 **Zinc**, < 150 µm, 99,995 % or equivalent.
- 5.12 **Zinc chloride**, > 97 %, ACS.

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6 Standard and solution preparation

CAUTION — Since vitamin K₁ is light-sensitive, all standards shall be prepared, handled, and stored in the dark or under yellow shielded lighting unless otherwise stated. If the standards are transported through or into an area without yellow shielded lighting, they shall be wrapped tightly in foil. All standard solutions shall be prepared using Class A volumetric glassware.

6.1 Mixture with volume fraction of 10 % isopropanol in iso-octane

Add about 70 ml of iso-octane (5.4) to a 100 ml volumetric flask. Add 10 ml isopropanol (5.5) to the volumetric flask and dilute to volume with iso-octane. Expiration is 6 months.

6.2 Mobile phase

Add about 900 ml iso-octane (5.4) to a 1 000 ml volumetric flask. Add 3 ml to 4 ml 10 % isopropanol (6.1) to the volumetric flask and dilute to volume with iso-octane. Expiration is 6 months if stored in tightly stoppered container.

NOTE The isopropanol concentration in the mobile phase can be adjusted slightly until baseline resolution of *cis* and *trans* vitamin K₁ from other peaks present in some samples is achieved, see Figures A.2 and A.3.

6.3 Post-column electrolyte solution

Transfer 0,5 g ± 0,05 g zinc chloride (5.12) and 0,20 g ± 0,02 g sodium acetate anhydrous (5.10) to a 1 000 ml volumetric flask with ethanol (5.9). Add 150 µl ± 15 µl glacial acetic acid (5.1) and dilute to volume with ethanol. Mix solution for about 30 min or until solution is clear and all salts are dissolved. Expiration is 6 months.

6.4 Vitamin K₁ (phytonadione) stock standard solution

6.4.1 Vitamin K₁ (phytonadione) stock standard solution

Accurately weigh to 0,000 01 g about 0,055 00 g vitamin K₁ (phytonadione) (5.7) into a 250 ml volumetric flask. Dissolve standard and dilute to volume with iso-octane (5.4). Store in a refrigerator in a tightly stoppered container protected from light. Expiration is 6 months.

6.4.2 Stock standard concentration (for non USP standard)

To determine the stock standard concentration, evaporate 0,5 ml of vitamin K₁ stock standard under a stream of nitrogen and redissolve the residue in 10,0 ml of hexane. Measure the absorbance of this solution in a 1 cm cell against a reference of hexane at wavelength of 248 nm with a spectrophotometer. Calculate the vitamin K₁ stock standard concentration, ρ_{ss} , in milligrams per litre, using Formula (1):

$$\rho_{ss} = \frac{A_{248} \times 10\,000 \times 20}{419} \quad (1)$$

where

A_{248} is the absorbance of the solution at 248 nm;

419 is the $A_{1\text{cm}}^{1\%}$ value of vitamin K₁ in hexane at 248 nm;

10 000 is the conversion of $A_{1\text{cm}}^{1\%}$ to milligrams per litre;

20 is the dilution of the stock standard solution.

6.5 Vitamin K₁ (phytonadione) intermediate I standard solution

Dilute 1,0 ml vitamin K₁ stock standard (6.4.1) to 100 ml with iso-octane. Prepare from stock standard solution each time new working standards are made.

6.6 Vitamin K₁ (phytonadione) intermediate II standard solution

Dilute 10,0 ml vitamin K₁ intermediate I standard (6.5) to 50 ml with iso-octane. Prepare each time new working standards are made.

6.7 Vitamin K₁ (phytonadione) working standard solutions

Dilute 1,0 ml, 3,0 ml, 6,0 ml, 10,0 ml and 20,0 ml intermediate II standard (6.6) to 100 ml with iso-octane. Store at 2 °C to 8 °C in a refrigerator in tightly closed containers protected from light. Expiration is 3 months.

Transfer working standards to autosampler vials with Pasteur pipets or equivalent glass. Do not pour the standards from the volumetric flasks into vials.

7 Apparatus

7.1 HPLC system.

Two isocratic pumps; autosampler capable of injecting 20 µl; fluorescence detector; instrument degasser (optional), high-pressure mixing tee or T-junction, and post-column reactor column about 20 mm × 4 mm stainless steel e.g. old HPLC column with packing removed or equivalent. The system should be configured as shown in [Figure 1](#).

7.2 Analytical column, silica 150 mm × 3,0 mm, 3 µm, 60 Å, or equivalent.

7.3 Analytical balance, capable of weighing to the nearest 0,000 01 g.

7.4 Beakers, glass, assorted sizes.

7.5 Centrifuge.

7.6 Centrifuge tubes and caps, 50 ml glass tubes with caps lined with polytetrafluoroethylene (PTFE).

7.7 Cylinders, graduated, glass, assorted sizes.

7.8 Gas regulator, compatible with helium or nitrogen.

7.9 Gas sparge, tubing and filtering assembly.

7.10 Magnetic stirrer and stir bar, with rack to hold centrifuge tubes.

7.11 Pipet, disposable glass, Pasteur.

7.12 Pipet, mechanical, variable volume, 0,5 ml to 5 ml and 10 µl to 100 µl.

7.13 Pipet, repeating 5 ml and 25 ml or equivalent.

7.14 Spectrophotometer, capable of measuring absorbance at 248 nm.

7.15 Volumetric flasks, glass, Class A, assorted sizes.

7.16 Volumetric pipets, glass, Class A, assorted sizes.

7.17 Vortex mixers.

7.18 Yellow lights or yellow shields, with cutoff of at least 440 nm.

8 Procedure

8.1 Sample preparation

8.1.1 Liquid samples

For ready-to-feed liquids, mix samples well or homogenize to ensure homogeneity and accurately weigh to 0,001 g, up to 4 g of sample into 50 ml centrifuge tubes. To liquids with sample masses less than 4 g,

add enough water (5.8) to the tubes so that the sample mass plus the amount of water added (g or ml) equal about 4 and mix well.

8.1.2 Powder samples

If the powder sample homogeneity is unknown, assume that it is non-homogenous and proceed as for dry-blended/non-homogenous powder samples.

8.1.3 Dry blended/non-homogenous powder samples

For dry blended/non-homogenous powder samples, accurately weigh approximately 25,0 g of powder and add 200 g of water (5.8). Record all masses. Mechanically stir or mix by hand until a homogeneous suspension is obtained. A homogenizer can be used when necessary. Accurately weigh to 0,001 g up to 4 g of homogeneous suspension into 50 ml centrifuge tubes. If less than 4 g of homogeneous suspension are weighed, add enough water to the tubes so that the sample mass plus the amount of water added (g or ml) equal about 4 and mix well.

8.1.4 Wet blended powder samples

For wet blended homogenous powder samples, accurately weigh to 0,000 1 g up to 0,5 g of powder into 50 ml centrifuge tubes. Add 4 ml of water (5.8) and mix well.

8.1.5 Extraction

Add 25 ml \pm 2,0 ml methanol (5.6) to each sample just prior to vortexing or stirring. Methanol should not be added to more than two samples consecutively without vortexing or stirring. Cap each centrifuge tube. Vortex each sample for at least 30 s at a rate that causes a vortex within the tube and then allow samples to sit undisturbed for at least 10 min, but no more than 40 min, or add a magnetic stir bar (7.10) to each centrifuge tube, cap tubes and place onto a magnetic stir plate, and stir samples for at least 10 min, but not more than 40 min, at a spin rate that causes a vortex. Begin timing after vortex forms in the tubes.

Quantitatively add 10 ml \pm 0,05 ml iso-octane (5.4) to each sample with a volumetric pipet and cap tubes. Iso-octane can be added to all samples before vortexing or stirring any of the samples. Vortex each sample for at least 45 s or stir each sample for at least 45 s at a rate that causes a vortex to form within the tubes. Begin timing after vortex forms in the tubes. Add 5 ml \pm 0,5 ml laboratory water (5.8) to each sample and cap tubes. Vortex or shake each sample for at least 20 s or stir each sample for at least 20 s at a spin rate that causes a vortex to form within the sample. Begin timing after vortex forms in the tubes.

Centrifuge the samples until a clean separation of the isooctane and aqueous-methanol layers results. The iso-octane layer should be a clear layer at the top of the centrifuge tube, and the aqueous-methanol layer should be a cloudy layer below the iso-octane layer. In some samples, there may be a small emulsion layer between the iso-octane and aqueous-methanol layers. A good separation of solvent layers can usually be achieved by centrifuging samples for approximately 10 min at 800 relative centrifugal force.

Do not add ethanol to samples to remove emulsions. Ethanol will change the final dilution volume and affect results.

Remove samples from the centrifuge and inspect each sample to verify that the iso-octane and aqueous-methanol layers are separated. With a glass pipet, carefully rinse down the upper walls of the centrifuge tube with a portion of the iso-octane layer. If the layers become mixed together, centrifuge the sample again. Pipette a portion of the clear iso-octane layer into a labelled autosampler vial and cap the vial.

8.2 HPLC analysis

8.2.1 Instrumental operating conditions

HPLC analytical column mobile phase flow rate:	0,4 ml/min;
Post-column flow rate:	0,4 ml/min;
Injection volume:	20 µl;
Run time:	20 min;
Fluorescence excitation and emission:	245 nm and 440 nm, respectively.

8.2.2 Instrument start-up

The system should be configured as shown in [Figure 1](#).

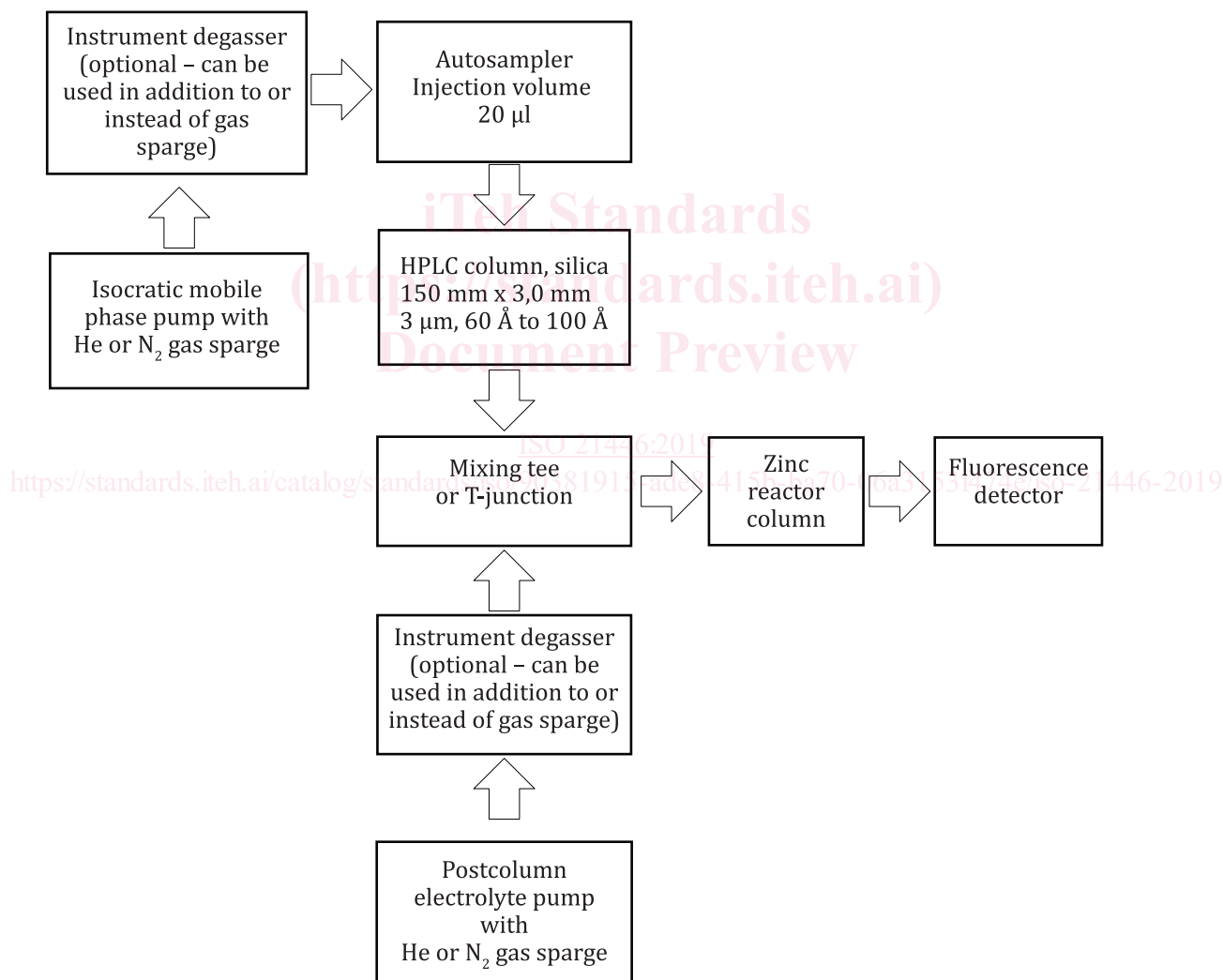


Figure 1 — Vitamin K system configuration

If necessary, remove used zinc and repack the post-column reactor column with fresh zinc ([5.11](#)). The zinc reactor column should be repacked whenever the S/N in the lowest standard is too low to accurately integrate the vitamin K₁ peak, linearity requirements ($r^2 \geq 0,999$) cannot be met unless the highest standard is excluded from the curve, peak responses from injections of the same standard drop