



Designation: ~~D 5225-98 (Reapproved 2003)~~ Designation: D 5225 - 09

Standard Test Method for Measuring Solution Viscosity of Polymers with a Differential Viscometer¹

This standard is issued under the fixed designation D 5225; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon (ϵ) indicates an editorial change since the last revision or reapproval.

1. Scope ^{*}

1.1 This test method covers the determination of the solution viscosity of polymers using a differential or the modified differential viscometer. It is applicable to all polymers that dissolve completely without chemical reaction or degradation to form solutions that are stable with time and temperature. Results of the test are usually expressed as specific viscosity, intrinsic viscosity (limiting viscosity number), inherent viscosity (logarithmic viscosity number), or relative viscosity (viscosity ratio).

1.2 Since there is more than one type of viscometer available to measure a differential pressure, follow the manufacturer's directions applicable to the equipment being used.

1.3 The solution viscosity values are comparable with those obtained using a glass capillary of Test Method D 2857. This test method differs from the glass capillary in that the solvent and the solution are compared at the same time that a test is run. With a glass capillary, each solution must be referenced back to the solvent run in the same capillary at the same temperature.

1.4 *This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.* For specific hazard statements, see Section 8.

~~NOTE 1—There is no ISO equivalent method.~~ 1—There is no known ISO equivalent to this test method.

2. Referenced Documents

2.1 *ASTM Standards:*²

D 1243 *Test Method for Dilute Solution Viscosity of Vinyl Chloride Polymers*

D 2857 *Test Method Practice for Dilute Solution Viscosity of Polymers*

E 691 *Practice for Conducting an Interlaboratory Study to Determine the Precision of a Test Method*

3. Terminology

3.1 *Definitions:*

3.1.1 *inherent viscosity*—the ratio of the natural logarithm of the relative viscosity to the concentration. The IUPAC term for inherent viscosity is *logarithmic viscosity number*.

3.1.2 *intrinsic viscosity*—limit of the reduced and inherent viscosities as the concentration of the polymeric solute approaches zero and represents the capacity of the polymer to increase viscosity. The IUPAC term for intrinsic viscosity is *limiting viscosity number*.

3.1.3 *reduced viscosity*—the ratio of the specific viscosity to the concentration. Reduced viscosity is a measure of the specific capacity of the polymer to increase the relative viscosity. The IUPAC term for reduced viscosity is *viscosity number*.

3.1.4 *relative viscosity*—the ratio of the polymer solution pressure to the pressure of the solvent.

3.1.5 *specific viscosity*—the relative viscosity minus one.

3.1.6 *viscosity constant, K*—baseline reading when solvent is present in both capillaries.

4. Summary of Test Method

4.1 *Differential Viscometer* (Fig. 1):

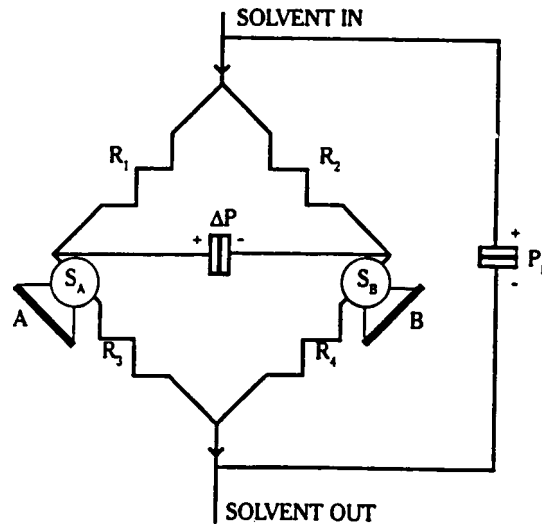
4.1.1 The viscosity measurement with the differential viscometer is based on a fluid analog of the Wheatstone Bridge. Pure solvent at constant inlet pressure P_i enters a balanced capillary network and flows through, producing a zero or baseline pressure on the differential pressure transducer. Solution is loaded into the sample reservoir A and then injected onstream by means of the

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² For referenced ASTM standards, visit the ASTM website, www.astm.org, or contact ASTM Customer Service at service@astm.org. For *Annual Book of ASTM Standards* volume information, refer to the standard's Document Summary page on the ASTM website.

*A Summary of Changes section appears at the end of this standard.



NOTE—
 R₁, R₂, R₃, R₄= Matched SS Capillary Tubing
 A, B = Solution Holdup Reservoirs
 S_A, S_B= Switching Valve
 P_i= Solvent Inlet Pressure Transducer
 ΔP = Differential Pressure Transducer

FIG. 1 Differential Viscometer

switching valve S_A. The differential pressure begins to rise until it reaches a steady state value of ΔP proportional to the specific viscosity of the solution. The differential pressure is monitored continuously on a strip chart recorder or computer, providing a baseline where ΔP is measured. The equation relating ΔP to specific viscosity is:

$$\eta_{sp} = \frac{4\Delta P}{P_i - 2\Delta P} \quad (1)$$

4.1.2 Derivation of the equation is in Annex A1.

4.2 Modified Differential Viscometer (Fig. 2):

4.2.1 The modified differential viscometer has two stainless steel capillaries connected in series with a sample loading/injection valve before the second capillary. Two differential pressure transducers, P₁ and P₂, are connected in parallel across the capillaries. A pump continuously supplies solvent flow. The ratio of the pressures P₂ and P₁ is proportional to the ratio of the viscosities of the fluid in capillary 2 to that in capillary 1.

$$\frac{P_2}{P_1} = K \frac{\eta_2}{\eta_1} = K \cdot \text{Relative Viscosity} \quad (2)$$

4.2.1.1 K, the viscosity constant, is obtained from the baseline reading when solvent is present in both capillaries, so η₂/η₁ is unity.

4.2.1.2 With the valve in LOAD position, the sample is flushed through the sample loop by the syringe pump. A baseline reading is established and recorded by the computer data acquisition system. When the valve is switched to the INJECT position, solvent flowing from capillary 1 pushes the sample into capillary 2. The differential pressure P₂ will increase due to the higher viscosity of the sample solution. The steady state value of P₂/P₁ then yields the value of relative viscosity of the sample.

$$\text{Relative Viscosity} = \frac{P_2}{K P_1} \quad (3)$$

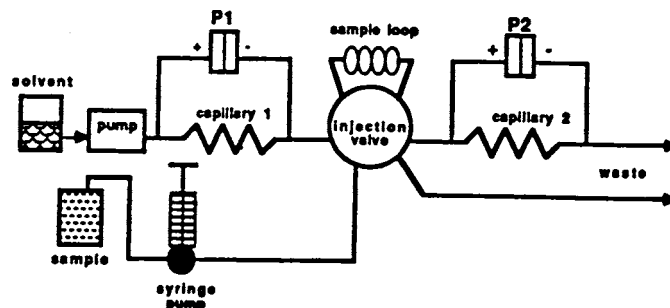


FIG. 2 Schematic of Relative Viscometer