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Železove rude - Določevanje žvepla - 3. del: Metoda s sežiganjem/infrardeča metoda

Iron ores - Determination of sulfur content - Part 3: Combustion/infrared method

# iTeh STANDARD PREVIEW

Minerais de fer - Dosage du souffe 2 Partie 31 Méthode par combustion et infrarouge

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# INTERNATIONAL STANDARD

ISO 4689-3

Third edition 2017-03

# Iron ores — Determination of sulfur content —

Part 3: **Combustion/infrared method** 

Minerais de fer — Dosage du soufre —

iTeh STPartie 3: Méthode par combustion et infrarouge

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ISO 4689-3:2017(E)

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#### ISO 4689-3:2017(E)

#### Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

The procedures used to develop this document and those intended for its further maintenance are described in the ISO/IEC Directives, Part 1. In particular the different approval criteria needed for the different types of ISO documents should be noted. This document was drafted in accordance with the editorial rules of the ISO/IEC Directives, Part 2 (see <a href="www.iso.org/directives">www.iso.org/directives</a>).

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For an explanation on the voluntary nature of standards, the meaning of ISO specific terms and expressions related to conformity assessment, as well as information about ISO's adherence to the World Trade Organization (WTO) principles in the Technical Barriers to Trade (TBT) see the following URL: <a href="https://www.iso.org/iso/foreword.html">www.iso.org/iso/foreword.html</a>. (standards.iteh.ai)

This document was prepared by Technical Committee ISO/TC 102, Iron ore and direct reduced iron, Subcommittee SC 2, Chemical analysis.

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This third edition cancels and replaces the second edition (ISO 4689-3:2015), of which it constitutes a minor revision, with the following changes:

- "0,001 %" has been inserted before "sulfur" in 5.1;
- in 6.7:
  - under "Combustion crucible", "Length" has been changed to "Height";
  - under "Combustion crucible", outer diameter has been changed from "10 mm" to "26 mm";
  - under "Combustion crucible", inner diameter has been changed from "26 mm" to "10 mm";
- "0,2 g" has been added after "sulfur" in 8.2;
- <u>Formula (7)</u> and the relevant descriptions in <u>9.2.4</u> have been modified to harmonize this clause across all standards for which ISO/TC 102/SC 2 is responsible.

A list of all parts in the ISO 4689 series can be found on the ISO website.

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## Introduction

ISO 4689-2 was originally published as ISO 4690:1986. Under a policy of rationalization of the numbering system used in ISO/TC 102, it has been decided to re-designate ISO 4690:1986 as ISO 4689-2. It was further decided to introduce a combustion/infrared method, numbered ISO 4689-3, i.e. this document.

When next revised, ISO 4689:1986 will be re-designated as ISO 4689-1.

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# Iron ores — Determination of sulfur content —

# Part 3:

# Combustion/infrared method

WARNING — This document may involve hazardous materials, operations, and equipment. This document does not purport to address all of the safety problems associated with its use. It is the responsibility of the user of this document to establish appropriate health and safety practices and determine the applicability of regulatory limitations prior to use.

### 1 Scope

This document specifies a combustion/infrared method, using a high-frequency induction furnace, for the determination of the sulfur content of iron ores.

This method is applicable to sulfur contents between 0,002 % (mass fraction) and 0,25 % (mass fraction) in natural iron ores, iron ore concentrates and agglomerates, including sinter products. The method is not applicable to iron ores containing more than 1,0 % (mass fraction) of combined water. The apparatus, of which the metal filter is equipped with a heating device, can be applied to iron ores containing less than 3,0 % (mass fraction) of combined water.

# 2 Normative references (standards.iteh.ai)

The following documents are referred to in the text in such a way that some or all of their content constitutes requirements of this document. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

ISO 648, Laboratory glassware — Single-volume pipettes

ISO 1042, Laboratory glassware — One-mark volumetric flasks

ISO 2596, Iron ores — Determination of hygroscopic moisture in analytical samples — Gravimetric, Karl Fischer and mass-loss methods

ISO 3082, Iron ores — Sampling and sample preparation procedures

ISO 7764, Iron ores — Preparation of predried test samples for chemical analysis

ISO Guide 35, Reference materials — General and statistical principles for certification

#### 3 Terms and definitions

No terms and definitions are listed in this document.

ISO and IEC maintain terminological databases for use in standardization at the following addresses:

- IEC Electropedia: available at <a href="http://www.electropedia.org/">http://www.electropedia.org/</a>
- ISO Online browsing platform: available at <a href="http://www.iso.org/obp">http://www.iso.org/obp</a>

### 4 Principle

The sample is mixed with a flux containing iron, tin and tungsten and heated in a high-frequency induction furnace, using oxygen as a supporting fuel and carrier gas.

The evolved sulfur dioxide is carried by oxygen into the cell of an infrared detector, where the absorption scale is read.

## 5 Reagents

During analysis, use only reagents of recognized analytical grade, and only distilled water or water of equivalent purity.

- **5.1 Metallic iron,** particle size 0,2 mm to 1,2 mm containing less than 0,001 % sulfur.
- **5.2 Magnesium perchlorate,** Mg(ClO<sub>4</sub>)<sub>2</sub>, anhydrous, free flowing, particle size 0,5 mm to 2 mm.
- **5.3 Iron(III) oxide,** containing less than 0,003 % sulfur.
- **5.4 Metallic tin,** particle size 0,2 mm to 1,2 mm, containing less than 0,001 % sulfur.
- **5.5 Metallic tungsten**, particle size 0,2 mm to 1,2 mm containing less than 0,001 % sulfur.
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- 5.6 Sodium hydroxide coated silica, particle size 0,5 mm to 2 mm. (standards.iteh.ai)
- **5.7 Sodium hydroxide**, particle size 0,5 mm to 2 mm.

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**5.8 Standard sulfur solution**, prepared as follows. 6d422d0216dd/sist-iso-4689-3-2017

Dry 8 g of potassium sulfate [purity > 99,99 % (mass fraction)] to constant mass at 105 °C to 110 °C and cool in a desiccator. Weigh the dried masses of potassium sulfate specified in <u>Table 1</u> and dissolve each portion in water. Cool, transfer each portion to separate volumetric flasks and dilute to volume with water.

Table 1 —	Sulfur standard	solution	(calibration	series)

Solution no.	Mass of potassium sulfate	Concentration of sulfur mg/ml
1	0,543 5	1,00
2	1,087 0	2,00
3	1,902 2	3,50
4	2,717 4	5,00

#### 6 Apparatus

Ordinary laboratory apparatus, including one-mark pipettes and one-mark volumetric flasks complying with the specifications of ISO 648 and ISO 1042, respectively, as well as the following. See Figure 1.

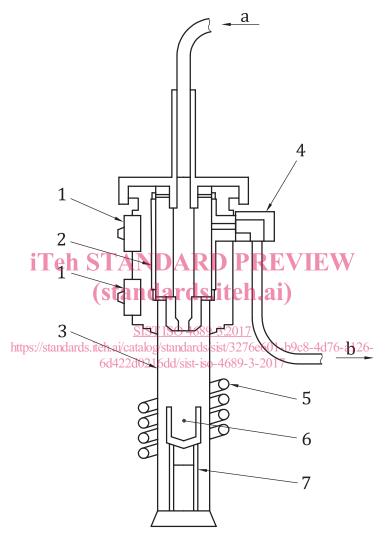
- 6.1 Oxygen supply.
- **6.2 Absorption tower,** containing soda-asbestos (5.6) or sodium hydroxide (5.7).
- **6.3 Drying tower,** containing magnesium perchlorate (5.2).

**6.4 High-frequency induction furnace**, having a heating induction coil (height 30 mm to 55 mm, 4 to 5 rolls), capable of heating to a temperature of 1 200 °C.

An example of a combustion furnace, with combustion gas-refining section, is given in Figure 1.

### **6.5** Combustion tube, quartz.

### 6.6 Refractory combustion crucible.



#### Key

- 1 metal filter
- 2 heating device
- 3 quartz combustion tube
- 4 dust trap
- 5 high-frequency induction coil
- 6 crucible
- 7 holder
- a Oxygen.
- b Combustion gas.

Figure 1 — Example of combustion furnace with combustion gas-refining section