

Designation: D4735 - 09

StandardTest Method for Determination of Trace Thiophene in Refined Benzene by Gas Chromatography¹

This standard is issued under the fixed designation D4735; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon (ε) indicates an editorial change since the last revision or reapproval.

1. Scope*

- 1.1 This test method covers the determination of thiophene in refined benzene in the range from 0.80 to 1.80 mg/kg for the Flame Photometric Detector (FPD), and from 0.14 to 2.61 mg/kg for the Pulsed Flame Photometric Detector (PFPD). For the PFPD, the minimum level of quantitation (LOQ) is 0.14 mg/kg and the minimum level of detection (LOD) is 0.04 mg/kg, as described in ASTM Research Report RR:D16-1038.² The range of the test method may be extended by modifying the sample injection volume, split ratios, calibration range, or sample dilution with thiophene-free solvent.
- 1.2 This test method has been found applicable to benzene characteristic of the type described in Specifications D2359 and D4734 and may be applicable to other types or grades of benzene only after the user has demonstrated that the procedure can completely resolve thiophene from the other organic contaminants contained in the sample.
- 1.3 The following applies to all specified limits in this test method: for purposes of determining conformance to applicable specification using this test method, an observed value or a calculated value shall be rounded off "to the nearest unit" in the last right-hand digit used in expressing the specification limit in accordance with the rounding-off method of Practice
- 1.4 The values stated in SI units are to be regarded as standard.
- 1.5 This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use. For specific hazard statements, see Section 7.

2. Referenced Documents

2.1 ASTM Standards:³

D1193 Specification for Reagent Water

D1685 Test Method for Traces of Thiophene in Benzene by Spectrophotometry

D2359 Specification for Refined Benzene-535

D3437 Practice for Sampling and Handling Liquid Cyclic Products

D4057 Practice for Manual Sampling of Petroleum and Petroleum Products

D4177 Practice for Automatic Sampling of Petroleum and Petroleum Products

D4307 Practice for Preparation of Liquid Blends for Use as Analytical Standards

D4734 Specification for Refined Benzene-545

D6809 Guide for Quality Control and Quality Assurance Procedures for Aromatic Hydrocarbons and Related Materials

E29 Practice for Using Significant Digits in Test Data to Determine Conformance with Specifications

E177 Practice for Use of the Terms Precision and Bias in ASTM Test Methods 61885/astm-d4735-00

E260 Practice for Packed Column Gas Chromatography

E355 Practice for Gas Chromatography Terms and Relationships

E691 Practice for Conducting an Interlaboratory Study to Determine the Precision of a Test Method

E840 Practice for Using Flame Photometric Detectors in Gas Chromatography

E1510 Practice for Installing Fused Silica Open Tubular Capillary Columns in Gas Chromatographs

2.2 Other Document:

OSHA Regulations, 29 CFR paragraphs 1910.1000 and 1910.1200⁴

¹ This test method is under the jurisdiction of ASTM Committee D16 on Aromatic Hydrocarbons and Related Chemicals and is the direct responsibility of Subcommittee D16.04 on Instrumental Analysis.

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² Supporting data have been filed at ASTM International Headquarters and may be obtained by requesting Research Report RR:D16-1038.

³ For referenced ASTM standards, visit the ASTM website, www.astm.org, or contact ASTM Customer Service at service@astm.org. For *Annual Book of ASTM Standards* volume information, refer to the standard's Document Summary page on the ASTM website.

⁴ Available from U.S. Government Printing Office Superintendent of Documents, 732 N. Capitol St., NW, Mail Stop: SDE, Washington, DC 20401, http://www.access.gpo.gov.

TABLE 1 Thiophene in Benzene Instrumental Conditions

Column	Α	В	С	D
Tubing	6 ft 31/8 in. Ni	15 ft by 1/8 in. stainless steel	10 ft by 1/8 in. stainless steel	30 meter, Fused Silica, 0.25 (or 0.32) mm ID
Phase	TCEEP ^A	SP-1000	OV-351	Bonded Polyethylene Glycol (PEG/CW)
Concentration, weight %	7	10	10	0.5 (or 1.0) micron film thickness
Support	Chromosorb P-AW ^B	Supelcoport	Chromosorb P-AW	N/A
Mesh	100/120	60/80	80/100	N/A
Gas chromatographic conditions				
Inlet	150	170	180	200
Carrier Gas	helium	helium	helium	helium
Carrier Flow, mL/min	30	30	30	1.0-1.5
Split Ratio	N/A	N/A	N/A	50:1
Column Temperature, °C	70	90	70	50°C for 1 mi., 10°C/min to 200°C, hold for 1 min
Detector	FPD	FPD	FPD	PFPD
(optimize flows per manufacturer's instructions)				(tuned for Sulfur) BG-12 Filter 2 mm combustor
H ₂ , mL/min	140	140	140	11.5
				flow optimized for S mode
Air I, mL/min	80	80	80	12.0
				flow optimized for S mode
Air 2, mL/min	70	70	70	10.0
				flow optimized for S mode
Temperature (°C)	220	220	250	250

^A Tetracyanoethylated pentaerythritol or pentrile.

3. Summary of Test Method

3.1 The thiophene concentration in refined benzene is determined at the milligram thiophene per kilogram sample level using conventional gas-liquid chromatography with a flame photometric detector (FPD) or pulsed flame photometric detector (PFPD). A reproducible volume of sample is injected. Quantitative results are obtained by the external standard technique using the measured peak area of thiophene.

4. Significance and Use ai/catalog/standards/sist/6l

- 4.1 This test method is suitable for setting specifications on benzene and for use as an internal quality control tool where benzene is either produced or used in a manufacturing process.
- 4.2 This test method was found applicable for determining thiophene in refined benzene conforming to the specifications described in Specification D2359 and may be applicable toward other grades of benzene if the user has taken the necessary precautions as described in the text.
- 4.3 This test method was developed as an alternative technique to Test Method D1685.

5. Apparatus

- 5.1 Gas Chromatograph—Any chromatograph having a flame photometric detector (FPD or PFPD) may be used which can operate at the typical conditions described in Table 1. The user is referred to Practices E260 and E355 for additional information about gas chromatography principles and procedures. An automatic sampler is recommended. The GC should have the following performance characteristics:
- 5.1.1 Column Temperature Programmer—The chromatograph shall be capable of linear programmed temperature operation over a range sufficient for the separation of the

compounds of interest. The programming shall be sufficiently reproducible to obtain retention time repeatability throughout the scope of the analysis.

- 5.1.2 Sample Inlet System—The sample inlet system shall have variable temperature control capable of operating continuously at a temperature up to the maximum column temperature employed. The sample inlet system shall allow a constant volume of sample to be injected by means of a syringe. For the PFPD a heated flash vaporizing injector designed to provide a linear sample split injection (that is, 50:1) is required for proper sample introduction. The associated carrier gas flow controls shall be of sufficient precision to provide reproducible column flows and split ratios in order to maintain analytical integrity.
- 5.2 *Column*—The column shall provide complete resolution of thiophene from benzene and any other hydrocarbon impurities because of potential quenching effects by hydrocarbons on the light emissions from the thiophene. The columns described in Table 1 have been judged satisfactory. The user is referred to Practice E1510 for assistance on installing fused silica capillary columns into the gas chromatograph.
- 5.3 *Detector*—Any flame photometric detector (FPD or PFPD) can be used, provided it has sufficient sensitivity to produce a minimum peak height twice that of the base noise for a 4-µL injection on the FPD, or a 1.0-µL injection for the PFPD of 0.5 mg/kg thiophene in benzene. The user is referred to Practice E840 for assistance in optimizing the operation and performance of the FPD.
- 5.4 Data Acquisition System—The use of an electronic integrating device or computer data system is recommended for determining the detector response. The device and software shall have the following capabilities: a) graphic presentation of

^B Chromosorb P is a registered trademark of the Manville Corp.

the chromatogram, b) digital display of chromatographic peak areas, c) identification of peaks by retention time or relative retention time, or both, d) calculation and use of response factors, and e) internal standardization, external standardization, and data presentation.

- 5.5 Microsyringe, 5 or 10-µL capacity.
- 5.6 Volumetric Flasks, 50, 100 and 500-mL capacity.
- 5.7 Separatory Funnel, 1-L capacity.

6. Reagents and Materials

- 6.1 Purity of Reagents—Reagent grade chemicals shall be used in all tests. Unless otherwise indicated, it is intended that all reagents shall conform to the specifications of the Committee on Analytical Reagents of the American Chemical Society, where such specifications are available. Other grades may be used, provided it is first ascertained that the reagent is of sufficiently high purity to permit its use without lessening the accuracy of the determination.
- 6.2 *Purity of Water*—Unless otherwise indicated, reference to water shall be understood to mean reagent water conforming to Type IV of Specification D1193.
- 6.3 *Carrier Gas*, nitrogen, helium, or hydrogen, chromatographic grade, or shall have a purity of 99.999 % (V/V) or better.
- 6.4 *Hydrogen*, zero grade, or shall have a purity of 99.999 % (V/V) or better. (**Warning**—Hydrogen is an extremely flammable gas under high pressure.)
- 6.5 *Compressed Air,* hydrocarbon-free, or shall have a purity of 99.999 % (V/V) or better. (**Warning**—Compressed air and oxygen are gases under high pressure and they support combustion.)
- 6.6 Cadmium Chloride Solution (20 g/L)—Dissolve 20 g of anhydrous cadmium chloride CdCl₂ into 200 mL of water and dilute to 1 L.
- 6.7 Isatin Solution—Add 0.5 g of isatin to 200 mL of chloroform. Heat under a fume hood to a temperature just below the boiling point of chloroform (61°C) and maintain for 5 min with stirring. Filter the hot solution through hardened rapid-filter paper into a 250-mL volumetric flask and dilute to volume.
- 6.8 Benzene, Thiophene-Free—Wash 700 mL of benzene in a 1000-mL separatory funnel to which has been added 5 mL of isatin solution, with successive 100-mL portions of concentrated sulfuric acid until the H₂SO₄ layer is light yellow or colorless. Wash the benzene with 100 mL of water, then twice with 100 mL of cadmium chloride solution (CdCl₂). Finally, wash with another 100-mL portion of water and filter the benzene through medium filter paper into a storage bottle, stopper the bottle tightly and save for future use. Commercial

sources of thiophene-free benzene are available and can be used as an alternative to this cleanup procedure.

- 6.9 Sulfuric Acid—Concentrated H₂SO₄.
- 6.10 Thiophene.
- 6.11 Stock solutions of thiophene in benzene are commercially available and can be used for preparation of calibration standards.

7. Hazards

7.1 Benzene is considered a hazardous material. Consult current OSHA regulations and supplier's Material Safety Data Sheets, and local regulations for all materials used in this method.

8. Sampling and Handling

8.1 Sampling of benzene should follow safe rules in order to adhere to all safety precautions as outlined in the latest OSHA regulations. Refer to Practices D3437, D4057, and D4177 for proper sampling and handling of benzene.

9. Preparation of the Apparatus

- 9.1 The chromatographic separation of trace level sulfur compounds can be complicated by absorption of the sulfur compounds by the gas chromatographic system. Therefore, care should be taken to properly free the system of active sites where absorption or reactions could take place.
- 9.2 Follow the manufacturer's instructions for mounting and conditioning the column into the gas chromatograph and adjusting the instrument to conditions described in Table 1. Allow the instrument and detector sufficient time to reach equilibrium.

10. Calibration Curve 8efb8a5/astm-d4735-09

- 10.1 Prepare a 500-mL stock solution of thiophene in benzene at approximately the 100 mg/kg level by adding 0.04 g (38.0 $\mu L)$ of thiophene to 435 g (500 mL) of thiophene-free benzene.
- 10.2 Calculate the thiophene content of the stock solution according to the following equation:

Thiophene, mg/kg =
$$(A \times 10^{3})/B$$

where:

A = weight of thiophene, mg

B = weight of benzene, g

- 10.3 Prepare five calibration blends ranging from 0.00 to 5.0 mg/kg of thiophene in benzene by diluting the appropriate volume of stock solution into a known volume of thiophene-free benzene. The user is referred to Practice D4307 for assistance preparation of liquid blends for use as analytical standards.
- 10.4 For example, an 87.0 mg/kg stock solution was prepared by dissolving 0.0378 g thiophene into 435 g of benzene. Aliquots of 0.00, 0.75, 1.0, 2.0, and 5.0 mL of stock solution were dissolved in 100 mL of thiophene-free benzene to produce 0.00, 0.65, 0.87, 1.75, and 4.35 mg/kg, respectively.

⁵ Reagent Chemicals, American Chemical Society Specifications, American Chemical Society, Washington, DC. For suggestions on the testing of reagents not listed by the American Chemical Society, see Analar Standards for Laboratory Chemicals, BDH Ltd., Poole, Dorset, U.K., and the United States Pharmacopeia and National Formulary, U.S. Pharmacopeial Convention, Inc. (USPC), Rockville, MD.