INTERNATIONAL STANDARD

ISO 19694-4

First edition 2023-03

Stationary source emissions — Determination of greenhouse gas emissions in energy-intensive industries —

Part 4:

Teh ST Aluminium industry

Emissions de sources fixes — Détermination des émissions de gaz à effet de serre dans les industries énergo-intensives —

Partie 4: Industrie de l'aluminium

ISO 19694-4:2023

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Published in Switzerland

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Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

The procedures used to develop this document and those intended for its further maintenance are described in the ISO/IEC Directives, Part 1. In particular, the different approval criteria needed for the different types of ISO documents should be noted. This document was drafted in accordance with the editorial rules of the ISO/IEC Directives, Part 2 (see www.iso.org/directives).

Attention is drawn to the possibility that some of the elements of this document may be the subject of patent rights. ISO shall not be held responsible for identifying any or all such patent rights. Details of any patent rights identified during the development of the document will be in the Introduction and/or on the ISO list of patent declarations received (see www.iso.org/patents).

Any trade name used in this document is information given for the convenience of users and does not constitute an endorsement.

For an explanation of the voluntary nature of standards, the meaning of ISO specific terms and expressions related to conformity assessment, as well as information about ISO's adherence to the World Trade Organization (WTO) principles in the Technical Barriers to Trade (TBT), see www.iso.org/iso/foreword.html.

This document was prepared by Technical Committee ISO/TC 146, *Air quality*, Subcommittee SC 1, *Stationary source emissions*.

A list of all parts in the ISO 19694 series can be found on the ISO website. 8-b3e2-4940-b955-

Any feedback or questions on this document should be directed to the user's national standards body. A complete listing of these bodies can be found at www.iso.org/members.html.

Introduction

This document serves the following purposes:

- to measure, test and quantify GHG emissions from the aluminium industry;
- to assess the level of GHG emissions performance of production processes over time at production sites;
- to establish and provide reliable, accurate and quality information for reporting and verification purposes.

This document can be used to measure, report and compare the GHG emissions of an aluminium production facility. Data for individual facilities, sites or works can be combined to measure, report and compare GHG emissions for a company, corporation or group.

Direct fuel-based emissions are not included; for calculation of this part of the GHG emissions, see ISO 19694-1.

This document deals with sector-specific aspects for the determination of greenhouse gas (GHG) emissions from aluminium production and is based on documents mentioned under tier 3 of Section 4.4.2.4 of the 2006 IPCC guidelines [6].

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Stationary source emissions — Determination of greenhouse gas emissions in energy-intensive industries —

Part 4:

Aluminium industry

1 Scope

This document specifies a harmonized method for calculating the emissions of greenhouse gases from the electrolysis section of primary aluminium smelters and aluminium anode baking plants. This document also specifies key performance indicators for the purpose of benchmarking of aluminium and boundaries.

2 Normative references

The following documents are referred to in the text in such a way that some or all of their content constitutes requirements of this document. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

ISO 19694-1, Stationary source emissions — Determination of greenhouse gas emissions in energy-intensive industries — Part 1: General aspects

3 Terms and definitions

For the purposes of this document, the terms and definitions given in ISO 19694-1 and the following apply.

ISO and IEC maintain terminology databases for use in standardization at the following addresses:

- ISO Online browsing platform: available at https://www.iso.org/obp
- IEC Electropedia: available at https://www.electropedia.org/

3.1

aluminium electrolysis

section of an aluminium primary smelter where aluminium is converted from aluminium oxide to aluminium metal in electrolysis cells

3.2

anode baking plant

production of carbon anodes for use in aluminium prebake electrolysis cells

3.3

PFC gas

gas emitted from aluminium electrolysis (3.1) consisting of CF₄ and C₂F₆

3.4

grid specific CO₂ factor

 ${\rm CO_2}$ factor (t ${\rm CO_2/MWh}$) associated with the electricity delivered to a specific aluminium smelter from their supplier

Note 1 to entry: The unit for grid specific ${\rm CO_2}$ factor is t ${\rm CO_2/MWh.}$

4 Symbols and abbreviated terms

4.1 Abbreviated terms

AE Anode effect

CWPB Centre-worked prebake

DAE Direct anode emissions

DEE Direct electrolysis emissions

GHG Green house gas

HSS Horizontal stud Søderberg

IPCC Intergovernmental Panel on Climate Change

PFC Perfluorocarbon

PFPB Point feeder prebake

SWPB Side-worked prebake

TIE Electrolysis electricity consumption

VSS Vertical stud Søderberg

WBCSD World Business Council for Sustainable Development

WRI World Resources Institute

4.2 Symbols and chemical formulae de Ob4b7a/iso-19694-4-2023

4.2.1 Symbols

 $A_{\rm EM}$ Anode effect minutes per cell-day (equals to frequency multiplied by average duration)

 $A_{\rm EO}$ Anode effect overvoltage

 $A_{\rm NC}$ Net anode consumption

 $A_{\rm sha}$ Ash content in baked anodes

A_{shp} Ash content in pitch, % mass fraction

 $A_{\rm shpc}$ Ash content in packing coke, % mass fraction

*B*_A Baked anode production

 $B_{\Delta W}$ Baked anode mass

 $B_{\rm C}$ typical binder content in paste, % mass fraction

 $C_{\rm RA}$ Carbon content of baked anodes

 $C_{\rm Butt}$ Carbon content of anode butts

 $C_{\rm E}$ Current efficiency for aluminium production

 $C_{\rm D}$ Carbon in skimmed dust from anode from Søderberg cells, tonnes carbon per tonne aluminium

C_{SM} Emissions of cyclohexane soluble matter, kilograms per tonnes aluminium

 $E_{ ext{CF}_4}$ Emissions of tetrafluoromethane, kg $ext{CF}_4$ per year

 $E_{C_2F_6}$ Emissions of hexafluoroethane, kg C_2F_6 per year

 $E_{\rm CO_2}$ CO₂ emissions, tonnes per year

 $E_{\rm FPC}$ Emission factor of packing coke, tCO₂/t of packing coke

 $\frac{F_{\text{C}_2\text{F}_6}}{\text{CF}_4}$ Mass fraction of $\frac{\text{C}_2\text{F}_6}{\text{CF}_4}$

 $G_{\rm A}$ Mass of loaded green anodes, $G_{\rm A} = \left(\frac{G_{\rm AW}}{B_{\rm AW}}\right) B_A$

 G_{AW} Green anode mass

 G_{WP} Global warming potential; use latest G_{WP} data from IPCC

 $H_{\rm w}$ Hydrogen content in green anode

 $H_{\rm p}$ Hydrogen content in pitch, % mass fraction

 $M_{\rm BA}$ Total mass of baked anodes

 $M_{\rm Butt}$ Total mass of anode butts 10 and Sitch 21

 $M_{\rm P}$ Total metal production, tonnes aluminium per year

 N_{AC} Net anode consumption, tonnes per tonnes aluminium 168-63-24940-6955

 O_{FPC} Oxidation factor of packing coke (typically 1 for this stream)

 $O_{\rm VC}$ Overvoltage coefficient for CF_4

*P*_C Paste consumption, tonnes per tonnes aluminium

 P_{CC} Packing coke consumed per tonnes of baked anode

 $P_{\rm CW}$ Packing coke mass

 R_{CF_A} Emission rates of CF_4 , kg per tonne of aluminium produced

 $R_{C_2F_6}$ Emission rates of C_2F_6 , kg per tonne of aluminium produced

 S_a Sulfur content in baked anodes

 S_c Sulfur content in calcined coke, % mass fraction

 S_p Sulfur content in pitch, % mass fraction

 $S_{\rm pc}$ Sulfur content in packing coke, % mass fraction

 S_{CF_4} Slope coefficient for CF_4 , kg CF_4 per tonne aluminium per anode effect minute per cell day

 $W_{\rm T}$ Waste tar collected

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4.2.2 Chemical formulae

Al Aluminium

 Al_2O_3 Aluminium oxide (alumina)

C Carbon

CF₄ Tetrafluoromethane

C₂F₆ Hexafluoroethane

CO Carbon monoxide

CO₂ Carbon dioxide

NaAlF₆ Sodium aluminium hexafluoride (cryolite)

NaF Sodium fluoride

5 Calculation methods — General remarks

5.1 General

This document shall be used in conjunction with ISO 19694-1 which contains generic, overall requirements, definitions and rules applicable to the determination of GHG emissions for all energy-intensive sectors, provides common methodological issues and specifies the details for applying the rules. The application of this document to the sector-specific standards ensures accuracy, precision and reproducibility of the results.

5.2 Calculation methods for process GHG emissions from primary aluminium production 56aede0b4b7a/iso-19694-4-2023

<u>Figure 1</u> gives sources of process emissions and references to where in the standard calculation methods are specified.

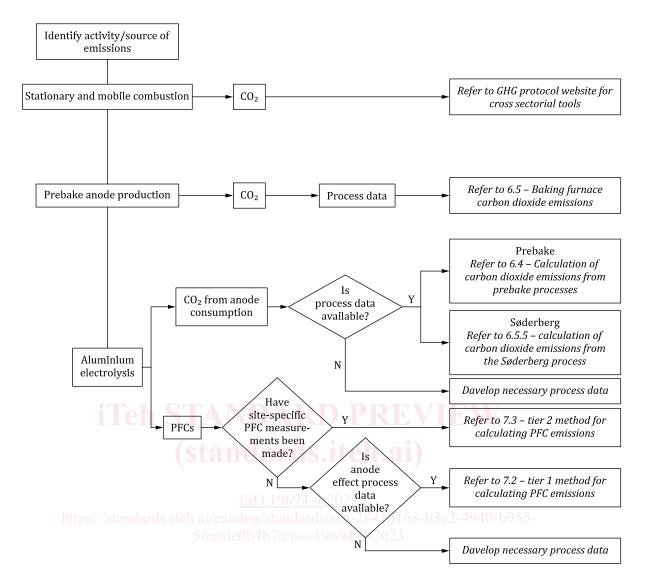


Figure 1 — Decision tree for process carbon dioxide and perfluorocarbon emissions from primary aluminium production

Process CO_2 emissions in state-of-the-art aluminium smelters comprise around 90 % of total direct CO_2 equivalent emissions, with the balance of emissions consisting of CO_2 from fossil fuel combustion and PFC emissions. Guidance on CO_2 emissions from fuel combustion is not included in this document. Methodology for calculating CO_2 emissions from the combustion of fuel in anode baking furnaces is described elsewhere [6],[7], while methodology for calculating process CO_2 emissions is given in Clause 7.

5.3 Sources of greenhouse gases

5.3.1 Electrolysis

Most of the CO_2 emissions result from the electrolytic reaction of the carbon anode with alumina as given in Formula (1):

$$2Al_2O_3 + 3C \rightarrow 4Al + 3CO_2 \tag{1}$$

Carbon dioxide is also emitted during the electrolysis reaction as the carbon anode reacts with other sources of oxygen, primarily from the air. Carbon dioxide is also formed as a result of the Boudouard reaction where CO_2 reacts with the carbon anode forming carbon monoxide, which is then oxidized to