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Standard Test Method for Measuring Neutron Fluence Rates by Radioactivation of Cobalt and Silver[1]

This standard is issued under the fixed designation E481; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon (ε) indicates an editorial change since the last revision or reapproval.

1. Scope

1.1 This test method covers a suitable means of obtaining the thermal neutron fluence rate, or fluence, in well moderated nuclear reactor environments where the use of cadmium, as a thermal neutron shield as described in Method E262, is undesirable because of potential spectrum perturbations or of temperatures above the melting point of cadmium.

1.2 This test method describes a means of measuring a Westcott neutron fluence rate (Note 1) by activation of cobalt- and silver-foil monitors (See Terminology E170). The reaction $^{59}Co(n,\gamma)^{60}Co$ results in a well-defined gamma emitter having a half-life of ~~1925.5~~1925.28 days **(1)**.[2] The reaction $^{109}Ag(n,\gamma)^{110m}Ag$ results in a nuclide with a complex decay scheme which is well known and having a half-life of 249.76 days **(~~14~~(1)**. Both cobalt and silver are available either in very pure form or alloyed with other metals such as aluminum. A reference source of cobalt in aluminum alloy to serve as a neutron fluence rate monitor wire standard is available from the National Institute of Standards and Technology (NIST) as Standard Reference Material 953.[3] The competing activities from neutron activation of other isotopes are eliminated, for the most part, by waiting for the short-lived products to die out before counting. With suitable techniques, thermal neutron fluence rate in the range from 10^9 $cm^{-2} \cdot s^{-1}$ to 3×10^{15} $cm^{-2} \cdot s^{-1}$ can be measured. For this method to be applicable, the reactor must be well moderated and be well represented by a Maxwellian low-energy distribution and an $(1/E)$ epithermal distribution. These conditions are usually met in positions surrounded by hydrogenous moderator without nearby strongly absorbing materials. Otherwise the true spectrum must be calculated to obtain effective activation cross sections over all energies.

NOTE 1—Westcott fluence rate = $v_0 \int_0^\infty n(v)dv$.

1.3 The values stated in SI units are to be regarded as the standard.

1.4 *This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.*

2. Referenced Documents

2.1 *ASTM Standards:*[4]

E170 Terminology Relating to Radiation Measurements and Dosimetry

E177 Practice for Use of the Terms Precision and Bias in ASTM Test Methods

E181 Test Methods for Detector Calibration and Analysis of Radionuclides

E262 Test Method for Determining Thermal Neutron Reaction Rates and Thermal Neutron Fluence Rates by Radioactivation Techniques

3. Significance and Use

3.1 The pertinent data for these two reactions are given in Table 1. This test method uses one monitor (cobalt) with a nearly $1/v$ absorption cross-section curve and a second monitor (silver) with a large resonance peak so that its resonance integral is large compared to the thermal cross section. The equations are based on the Westcott formalism **(~~(~~2, 3, ~~4~~)** and determine a Westcott 2200 m/s neutron fluence rate nv_0 and the Westcott epithermal index parameter $r\sqrt{T/T_0}$. ~~References~~ **. References 4, 5, ~~6~~, 6, and ~~7~~6**

[1] This test method is under the jurisdiction of ASTM Committee E10 on Nuclear Technology and Applications and is the direct responsibility of Subcommittee E10.05 on Nuclear Radiation Metrology.

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[2] The boldface numbers in parentheses refer to references listed at the end of this test method.

[3] Standard Reference Material 953 is available from National Institute of Standards and Technology, U.S. Dept. of Commerce, Washington, DC 20234.

[4] For referenced ASTM standards, visit the ASTM website, www.astm.org, or contact ASTM Customer Service at service@astm.org. For *Annual Book of ASTM Standards* volume information, refer to the standard's Document Summary page on the ASTM website.

TABLE 1 Recommended Constants

Symbol	Parameter	Cobalt (^{60}Co)		Silver (^{110m}Ag)	
		Value[A]	Reference	Value[A]	Reference
~~$t_{1/2}$~~	~~Half-life~~	~~1925.5 (5) days~~	~~(1)~~	~~249.76 (4) days~~	~~(14)~~
$t_{1/2}$	Half-life	1925.28 (14) days	**(1)**	249.76 (4) days	**(1)**
~~A~~	~~Abundance of parent isotope~~	~~100 % (^{59}Co)~~	~~(14)~~	~~48.161 (8) % (^{109}Ag)~~	~~(14)~~
A	Abundance of parent isotope	100 % (^{59}Co)	**(1)**	48.161 (8) % (^{109}Ag)	**(1)**
~~Δ~~	~~Mass excess of residual isotope (scaled to Δ [^{12}C] = 0) (1 amu = 931.494MeV)[B]~~	~~−61.644 MeV~~	~~(14)~~	~~−87.340 MeV~~	~~(14)~~
Δ	Mass excess of residual isotope (scaled to Δ [^{12}C] = 0) (1 amu = 931.494MeV)[B]	−61.64904 MeV	**(1)**	−87.3424 MeV	**(1)**
~~σ_a~~	~~Absorption 2200 m/s cross section for target ^{59}Co and ^{109}Ag~~	~~37.233 b ± 0.16 %~~	~~C,D~~	~~91.0 b ± 1 %~~	~~(16)~~
σ_a	Absorption 2200 m/s cross section for target ^{59}Co and ^{109}Ag	37.233 b ± 0.16 %	C,D	91.0 b ± 1 %	**(7)**
~~σ_0~~	~~2200 m/s cross section for formation of ^{60}Co and ^{110m}Ag~~	~~37.233 b ± 0.16 %~~	~~C,D~~	~~4.7 b ±4 %~~	~~(16)~~
σ_0	2200 m/s cross section for formation of ^{60}Co and ^{110m}Ag	37.233 b ± 0.16 %	C,D	4.12 b ± 2.54 %	**(8)**
~~S_0~~	~~Correction factor which describes the departure of the cross section from the 1/v law in the epithermal region~~	~~1.69 [$^{59}Co(n,\gamma)^{60}Co$]~~	~~(7)~~	~~18.53 [$^{109}Ag(n,\gamma)^{110m}Ag$] 17.10 [$^{109}Ag(n,\gamma)^{110m+110g}Ag$]~~	~~(7)~~
S_0	Correction factor which describes the departure of the cross section from the 1/v law in the epithermal region	1.80 [$^{59}Co(n,\gamma)^{60}Co$]	E	18.13 ± 4 % [$^{109}Ag(n,\gamma)^{110m}Ag$] 17.76 [$^{109}Ag(n,\gamma)^{110m+110g}Ag$]	**(8)**
~~I_0~~	~~Resonance Integral~~	~~75.421 b ± 0.77 % [$^{59}Co(n,\gamma)^{60}Co$]~~	~~(15),E~~	~~66 b [$^{109}Ag(n,\gamma)^{110m}Ag$]~~	~~(16)~~
I_0	Resonance Integral	75.421 b ± 0.77 % [$^{59}Co(n,\gamma)^{60}Co$]	**(9)**[F]	67.9 b ± 4.5 % [$^{109}Ag(n,\gamma)^{110m}Ag$]	**(8)**
~~σ_2~~	~~Effective absorption cross section for product nuclide (reactor spectrum)~~	~~2 b~~	~~(11)~~	~~82 b~~	~~(13)~~
σ_2	Effective absorption cross section for product nuclide (reactor spectrum)	2 b	**(10)**	82 b	**(11)**
~~G_{th}~~	~~Thermal neutron self-shielding factor~~	~~Table 3~~	~~(12)~~	~~≅ 1 – 4/3 RΣ_B~~	~~(5)~~
G_{th}	Thermal neutron self-shielding factor	Table 3	**(12)**	≅ 1 – 4/3 RΣ_a	**(4)**
~~G'_{res}~~	~~Resonance neutron self-shielding factor~~	~~Table 3~~	~~(12)~~	~~Fig. 1[F]~~	~~(5)~~
G'_{res}	Resonance neutron self-shielding factor	Table 3	**(12)**	Fig. 1[G]	
~~g~~	~~Correction factor which describes the departure of the cross section from 1/v law in thermal region~~	~~1.0~~	~~(3)~~	~~See Table 4~~	~~(3)~~
g	Correction factor which describes the departure of the cross section from 1/v law in thermal region	1.0	**(2)**	See Table 4	**(2)**

[A] The numbers in parenthesis following given values is the uncertainty in the last digit(s) of the value; 0.729 (8) means 0.729 ± 0.008, 70.8(1) means 70.8 ± 0.1.
[B] Isotopic masses may be calculated as A + 931.49432/Δ, where A is the atomic mass number.
[C] A 2200 m/s cross section (E = 0.0253 eV, T = 20°C) was taken from the sources indicated in Ref~~erence~~ **(~~15~~9)**.
[D] Cross section uncertainty data is taken from Ref~~erence~~ **(~~16~~7)** , the cross section comes from the other reference.
[E] ~~Cross sea~~l~~ction~~ ~~uncert~~l~~ainty comes from convariancedata provided in the cross section so urce. The other reference indicates the source of the cross sectio~~ng Eq 5.
[F] Cross section uncertainty comes from convariance data provided in the cross section source. The other reference indicates the source of the cross section.
[G] In Fig. 1, $\Theta = 4E_r kT/A\Gamma^2 = 0.2$ corresponds to the value for ^{109}Ag for T = 293 K, $\Sigma_r = N_0\sigma_{r,\,max} - \sigma_{r,\,max}$ = 29999 barn at 5.19 eV **(13)** .

contain a general discussion of the two-reaction test method. In this test method, the absolute activities of both cobalt and silver monitors are determined. This differs from the test method in the references wherein only one absolute activity is determined.

3.2 The advantages of this test method are the elimination of three difficulties associated with the use of cadmium: (*1*) the perturbation of the field by the cadmium; (*2*) the inexact cadmium cut-off energy; (*3*) the low melting temperature of cadmium. In addition, the reactivity changes accompanying the rapid insertion and removal of cadmium may prohibit the use of the cadmium-ratio method. However, the self-shielding corrections remain important unless the concentrations of cobalt and silver are small. Studies indicate that the accuracy of the two-reaction method for determination of thermal neutron fluence is comparable to the cadmium-ratio method (**14**).

3.3 The long half-lives of the two monitors permit the determination of fluence for long-term monitoring.

4. Apparatus

4.1 *NaI(Tl) or Germanium Gamma-Ray Spectrometer (using a multichannel analyzer)*—For the NaI(Tl) technique and the germanium technique, see Method E181.

4.2 *Precision Balance*.

4.3 *Digital Computer*.

5. Materials and Manufacture

5.1 The two monitors required for this test method are cobalt and silver. Although these two materials are available

commercially in very pure form, they have been used ~~(8)~~(15) alloyed with aluminum (≤1 % cobalt and ≤1 % silver) to minimize the self-shielding effect and to permit insertion into a high thermal-neutron fluence rate (>10^{15} cm $^{-2}s^{-1}$) facility ~~(7, 9)~~(6, 16). Typical alloys contain 0.1 % silver or cobalt in aluminum) see 6.1 and 8.1).

5.2 The uncertainties and nonuniformity of alloy concentrations must be established by one or more different test methods. These might include chemical and activation analysis, or spectrometry. The purity of the aluminum matrix should also be established.

5.3 Whenever possible, the alloys should be tested for interfering impurities by neutron activation.

5.4 The method of encapsulating the monitors for irradiation depends upon the characteristics of the facility in which the measurements are to be made. The monitors have essentially the same chemical characteristics as pure aluminum; therefore, an environment in which aluminum would not be adversely affected would be generally satisfactory for the alloys. However, the low mechanical strength of the monitors requires in many instances that it be encapsulated or shielded from physical disturbances by some type of container. Aluminum cans or tubing are satisfactory for many cases of interest, but for hostile environments, stainless steel or vanadium may be preferable. Perturbation due to the presence of the container must be accounted for, especially in the case of stainless steel. The container should be constructed in such a manner that it will not create a significant flux perturbation and that it may be opened easily, especially if the monitors must be removed remotely.

6. Procedure

6.1 Decide on the size and shape of the monitors to be irradiated, taking into consideration the size and shape of the irradiation space. The mass and exposure time are parameters which can be varied to obtain a desired disintegration rate for a given neutron fluence rate level. To facilitate the convergence of the two activity equations for the fluence rate and the epithermal index in Section 7, the concentration of the alloys should be chosen so that the ratio of the disintegration rates is on the order of one.

6.2 Weigh the samples to a precision of ±1.0 % (1S %) as defined in Practice E177.

6.3 Irradiate the samples for the predetermined time period. Record the power level and any changes in power during the irradiation, the time at the beginning and end of the irradiation, and the relative position of the monitors in the irradiation facility.

6.4 A waiting period is necessary between termination of the exposure and start of counting when using Co-Al and Ag-Al monitors. This allows the 0.62356 days ~~(1)~~(17) half-life ^{24}Na which is formed by fast-neutron reactions on ^{27}Al or by thermal-neutron captures by ^{23}Na impurities to decay below levels at which its radiations may cause interferences. It is sometimes advisable to count the samples periodically and follow the decay of the portions of the activities due to the ^{24}Na. The length of the waiting period can be reduced by the use of a germanium detector.

6.5 With the gamma-ray spectrometer, analyze the silver sample for ^{110m}Ag and the cobalt sample for ^{60}Co. Obtain the net count rate in each full-energy gamma-ray peak of interest, that is, 657.7623 keV or 884.684 keV for ^{110m}Ag: 1332.501 keV for ^{60}Co (see Method E181). See Table 2 for gamma radiations of ^{110m}Ag.

7. Calculation

7.1 Calculate the activities of ^{110m}Ag and ^{60}Co in disintegrations per second.

TABLE 2 Gamma Radiations of ^{110m}Ag (~~2~~17,18)

Energy of Gamma[A] (keV)	Relative[B,A] Emission Probability (%)
~~1.~~	~~657.7622 (21)~~	~~110.0 (4)~~
1.	657.7600 (11)	100
~~2.~~	~~884.685 (3)~~	~~76.8 (3)~~
2.	884.6781 (13)	77.1 (3)
~~3.~~	~~937.493 (4)~~	~~36.31 (12)~~
3.	937.485 (3)	36.3 (6)
~~4.~~	~~1384.300 (4)~~	~~25.66 (8)~~
4.	1384.2931 (20)	26.4 (8)
~~5.~~	~~763.944 (3)~~	~~23.55 (9)~~
5.	763.9424 (17)	23.98 (21)
~~6.~~	~~706.682 (3)~~	~~17.37 (10)~~
6.	706.6760 (15)	17.31 (5)
~~7.~~	~~1505.040 (5)~~	~~13.78 (5)~~
7.	1505.0280 (20)	14.42 (19)
~~8.~~	~~667.6227 (24)~~	~~10.94 (8)~~
8.	677.6217 (12)	11.20 (2)
~~9.~~	~~818.031 (4)~~	~~7.76 (4)~~
9.	818.0244 (18)	7.78 (8)
~~10.~~	~~687.015 (3)~~	~~6.80 (6)~~
10.	687.0091 (18)	6.83 (5)
~~11.~~	~~744.277 (3)~~	~~5.00 (3)~~
11.	744.2753 (18)	5.06 (9)
~~12.~~	~~1562.302 (5)~~	~~1.087 (7)~~
12.	1562.294 (18)	1.319 (17)

[A] The number of parentheses following some given values is the uncertainty in the last digit(s) of the value: 0.729 (8) means 0.729 ± 0.008, 80.8 (1) means 70.8 ± 0.1.

[B] For absolute intensity multiply ~~e~~ emission probabilities by 0.943 ± 0.004.

7.2 A Westcott 2200 m/s neutron fluence rate, nv_0, or ϕ_w and the Westcott epithermal index parameter, $r\sqrt{T/T_0}$ are related to the measured activities of the silver and cobalt monitors by the following equation:

~~(1) $A = N0\lambda BFG\sigma 1\phi wti$~~

~~(1) $A = N0\lambda BFG\sigma 1\phi wti$~~

where:

A = measured activity at the end of the exposure time, disintegrations/s,
N_0 = number of target atoms of ^{59}Co or ^{109} Ag at start of irradiation,
~~λ_2~~λ = disintegration constant of product nuclide, s^{-1},
B = Self-absorption factor of the decay gamma ray in the monitor material,
F = burnup and decay correction factor,
G = self-shielding factor (see Eq 4, Table 3 and Fig. 1).
$\hat{\sigma}_1$ = Westcott's effective absorption cross section for production of the product nuclide, cm^2,
ϕ_w (or nv_0) = a 2200 m/s neutron fluence rate in which n is the neutron density (including both thermal and epithermal neutrons) and t_i is 2200 m/s, and
t_i = exposure time.

7.3 The self-absorption factor, if not known for the gamma rays being measured, can be approximated by the following equation:

(2) $B \simeq 1 - (4/3)(\mu aR)$

where:

μ_a = linear absorption coefficient in monitor, cm^{-1}, and
R = radius of monitor wire, cm.

7.4 The burnup and decay correction factor is given by:

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where:

$\hat{\sigma}_a$ = Westcott's effective absorption cross section for target nuclide, cm^2, and
$\hat{\sigma}_2$ = Westcott's effective absorption cross section for the product nuclide, cm^2.

7.5 The self-shielding factor is given by:

where:

g = correction factor which describes the departure of the cross section from the $1/v$ law in the thermal region (see Table 4 ~~for silver "g" factors),~~ for silver "g" factors),
G_{th} = thermal neutron self-shielding factor,
G'_{res} = resonance neutron self-shielding factor,
r = a measure of the proportion of epithermal neutrons in the reactor spectrum,
T = neutron temperature, K,
T_0 = 293.6 K, and
S_0 = correction factor which describes the departure of the cross section from the $1/v$ law in the epithermal region.

~~Although the~~

7.6 Although the Ag ^{109}(n,λ)Ag ^{110m} S_0 ~~values in~~ value in Table 1 ~~are measured values,~~ *is a measured value, S_0 can be calculated by the following equation:*

TABLE 3 Self-Shielding Factors for Cobalt Wires (12)

Wire Diameter in. (mm)	Cobalt Content, (mass %)	G'_{res}(132 eV)	G_{th}
0.050 (1.27)	0.104	1.00	1.00
0.050 (1.27)	0.976	0.95 ± 0.04	0.99 ± 0.01
0.001 (0.03)	100	0.81 ± 0.03	0.99 ± 0.02
0.005 (0.13)	100	0.52 ± 0.02	0.97 ± 0.01
0.010 (0.25)	100	0.42 ± 0.02	0.94 ± 0.01
0.015 (0.38)	100	0.38 ± 0.01	0.92 ± 0.02
0.020 (0.51)	100	0.34 ± 0.01	0.90 ± 0.02
0.025 (0.64)	100	0.32 ± 0.01	0.88 ± 0.03