



**SLOVENSKI STANDARD**  
**kSIST-TS FprCEN/TS 17759:2021**  
**01-december-2021**

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**[Not translated]**

Inorganic fertilizers - Determination of pH of a solution of ammonium nitrate fertilizers of high nitrogen content

Anorganische Düngemittel - Bestimmung des pH-Wertes in einer Lösung mit Ammoniumnitratdüngemitteln mit hohem Stickstoffgehalt

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**Ta slovenski standard je istoveten z: FprCEN/TS 17759**

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**ICS:**

65.080                      Gnojila                                      Fertilizers

**kSIST-TS FprCEN/TS 17759:2021                      en,fr,de**

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TECHNICAL SPECIFICATION  
SPÉCIFICATION TECHNIQUE  
TECHNISCHE SPEZIFIKATION

**FINAL DRAFT**  
**FprCEN/TS 17759**

November 2021

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ICS 65.080

English Version

**Inorganic fertilizers - Determination of pH of a solution of  
ammonium nitrate fertilizers of high nitrogen content**

Anorganische Düngemittel - Bestimmung des pH-  
Wertes in einer Lösung mit  
Ammoniumnitratdüngemitteln mit hohem  
Stickstoffgehalt

This draft Technical Specification is submitted to CEN members for Vote. It has been drawn up by the Technical Committee CEN/TC 260.

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Recipients of this draft are invited to submit, with their comments, notification of any relevant patent rights of which they are aware and to provide supporting documentation.

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EUROPEAN COMMITTEE FOR STANDARDIZATION  
COMITÉ EUROPÉEN DE NORMALISATION  
EUROPÄISCHES KOMITEE FÜR NORMUNG

**CEN-CENELEC Management Centre: Rue de la Science 23, B-1040 Brussels**

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<b>Contents</b>	<b>Page</b>
European foreword .....	3
<b>1</b> Scope.....	<b>4</b>
<b>2</b> Normative references.....	<b>4</b>
<b>3</b> Terms and definitions.....	<b>4</b>
<b>4</b> Principle.....	<b>4</b>
<b>5</b> Reagents.....	<b>4</b>
<b>6</b> Apparatus and equipment.....	<b>4</b>
<b>7</b> Sampling and sample preparation .....	<b>5</b>
<b>8</b> Procedure .....	<b>5</b>
<b>8.1</b> Calibration of the pH meter.....	<b>5</b>
<b>8.2</b> Determination.....	<b>5</b>
<b>9</b> Expression of the results.....	<b>5</b>
<b>10</b> Test report.....	<b>5</b>
<b>Bibliography</b> .....	<b>6</b>

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## European foreword

This document (FprCEN/TS 17759:2021) has been prepared by Technical Committee CEN/TC 260 “Fertilizers and liming materials” the secretariat of which is held by DIN.

This document is currently submitted to the Vote on TS.

This document has been prepared under a standardization request given to CEN by the European Commission and the European Free Trade Association.

**WARNING** — The use of this document can involve hazardous materials, operations and equipment. This document does not purport to address all the safety problems associated with its use. It is the responsibility of the user of this document to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.

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**FprCEN/TS 17759:2021 (E)****1 Scope**

This document specifies a method for the determination of pH of a solution of ammonium nitrate fertilizer of high nitrogen content.

**2 Normative references**

The following documents are referred to in the text in such a way that some or all of their content constitutes requirements of this document. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

EN 1482-2, *Fertilizers and liming materials - Sampling and sample preparation - Part 2: Sample preparation*

EN 12944-1, *Fertilizers and liming materials and soil improvers - Vocabulary - Part 1: General terms*

EN 12944-2, *Fertilizers and liming materials and soil improvers - Vocabulary - Part 2: Terms relating to fertilizers*

**3 Terms and definitions**

For the purposes of this document, the terms and definitions given in EN 12944-1 and EN 12944-2 apply. ISO and IEC maintain terminological databases for use in standardization at the following addresses:

- IEC Electropedia: available at <http://www.electropedia.org/>
- ISO Online browsing platform: available at <http://www.iso.org/obp>

**4 Principle**

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The measurement of the pH of an ammonium nitrate solution is carried out by means of a pH meter.

**5 Reagents**

Use only reagents of recognized analytical grade.

**5.1 Distilled or demineralized water**, free from carbon dioxide.

**5.2 Buffer solution A**, pH 6,88 at 20 °C **or buffer solution B** pH 4,00 at 20 °C.

For buffer solution A, dissolve 3,40 g ± 0,01 g of potassium dihydrogen phosphate (KH<sub>2</sub>PO<sub>4</sub>) in approximately 400 ml of water. Then dissolve 3,55 g ± 0,01 g of disodium hydrogen phosphate (Na<sub>2</sub>HPO<sub>4</sub>) in approximately 400 ml of water (5.1). Transfer the two solutions without loss into a 1000-ml graduated flask (6.2), fill up to the mark and mix.

Keep this solution in an airtight vessel.

For buffer solution B, dissolve 10,21 g ± 0,01 g of potassium hydrogen phthalate (KHC<sub>8</sub>O<sub>4</sub>H<sub>4</sub>) in water (5.1), transfer without loss into a 1000-ml graduated flask (6.2), fill up to the mark and mix.

Keep this solution in an airtight vessel.

Alternatively a commercially available pH standard solutions may be used.

**6 Apparatus and equipment**

Usual laboratory glassware and equipment and, in particular, the following.

- 6.1 **pH meter, equipped with glass and calomel electrodes or equivalent**, sensitivity 0,05 pH unit.
- 6.2 **Graduated flask**, capacity 1 000 ml.
- 6.3 **Beaker**, capacity 250 ml.
- 6.4 **Balance**, capable of weighing to the nearest 0,01 g.
- 6.5 **Airtight vessel**.

## 7 Sampling and sample preparation

Sampling is not part of the method specified in this document. A recommended sampling method is given in EN 1482-1 [1].

Sample preparation shall be carried out in accordance with EN 1482-2.

## 8 Procedure

### 8.1 Calibration of the pH meter

Calibrate the pH meter (6.1) at a temperature of  $20\text{ °C} \pm 1\text{ °C}$ , using the buffer solutions (5.2). Pass a slow stream of nitrogen onto the surface of the solution and maintain this throughout the test.

### 8.2 Determination **iTeh STANDARD PREVIEW**

Pour 100,0 ml of water onto 10 g  $\pm$  0,01 g of the sample in a 250 ml beaker (6.3). Remove the insolubles by filtering, decanting or centrifuging the liquid. Measure the pH of the clear solution at a temperature of  $20\text{ °C} \pm 1\text{ °C}$  according to the same procedure as for the calibration of the pH meter.

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### 9 Expression of the results <https://standards.iteh.ai/catalog/standards/sist/1f83d05/ksist-ts-fprcen-ts-17759-2021>

Express the result in pH units, to the nearest 0,1 unit, and state the temperature used.

## 10 Test report

The test report shall contain at least the following information:

- a) all information necessary for the complete identification of the sample;
- b) test method used with reference to this document, FprCEN/TS 00260260:2021;
- c) test result obtained expressed in pH units, to the nearest 0,1 unit, and state the temperature used.
- d) date of sampling and sampling procedure (if known);
- e) date when the analysis was finished;
- f) all operating details not specified in this document, or regarded as optional, together with details of any incidents that occurred when performing the method which might have influenced the test result(s).