



Designation: ~~D3448-85~~ (Reapproved 2005) Designation: D3448 - 10

## Standard Test Method for Specific Aqueous Conductance of Trichlorotrifluoroethane<sup>1</sup>

This standard is issued under the fixed designation D3448; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon ( $\epsilon$ ) indicates an editorial change since the last revision or reapproval.

### 1. Scope

1.1 This test method covers the determination of aqueous conductivity due to ionizable contaminants extracted from trichlorotrifluoroethane. The value of 1.9  $\mu\text{S}/\text{cm}$  as maximum specific aqueous conductance corresponds to 0.1 ppm of chloride ion (100 ppb). It was calculated by putting the solvent specification of 0.1 ppm chloride into the equation of the subject method to find  $C_2 - C_1$  equals 19.08. The specific conductance due to contaminants from the solvent was then calculated as follows:

$$(C_2 - C_1)C_3 = (19)(0.1) = 1.9 \mu\text{S}/\text{cm} \quad (1)$$

1.2 The above is exclusive of the background specific conductance of the water used in the analysis.

1.3

1.3 The values stated in SI units are to be regarded as standard. No other units of measurement are included in this standard.

1.4 This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to consult and establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use. Specific precautionary statements are given in Section 6.

### 2. Summary of Test Method

2.1 Ionizable impurities in the solvent are extracted with water of known low conductivity. The increase in water conductivity measures the amount of these impurities present.

### 3. Significance and Use

3.1 This test method allows for the determination of ionizable contaminants (expressed in terms of ppb of HCl) in trichlorotrifluoroethane via measurement of specific aqueous conductance.

3.2 This test method can be used for establishing manufacturing and purchasing specifications.

### 4. Apparatus

4.1 Conductivity Bridge.<sup>2</sup>

4.2 Conductivity Cell,<sup>3</sup> dip-type. Cell constant of 0.1 reciprocal centimetre. Electrodes must be bright platinum.<sup>4,8-10</sup>

4.3 Polyethylene Beaker, 250-mL.

4.4 Polyethylene Bottles, wide mouth, 300-mL and 500-mL size.

4.4.1 Calibrate and mark the outside of the 300-mL polyethylene bottles to indicate 20, 120, and 220 mL (graduate) of water volume, using deionized water to determine the liquid levels.

4.5 Polyethylene Tubing, long enough to reach the bottom of the 300-mL polyethylene bottle.

### 5. Reagents

5.1 Purity of Reagents—Reagent grade chemicals shall be used in all tests. Unless otherwise indicated, it is intended that all reagents shall conform to the specifications of the Committee on Analytical Reagents of the American Chemical Society, where

<sup>1</sup> This test method is under the jurisdiction of ASTM Committee D26 on Halogenated Organic Solvents and Fire Extinguishing Agents and is the direct responsibility of Subcommittee D26.04 on Test Methods.

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<sup>2</sup> Serfass Model RCM-15B1 has been found satisfactory and can be obtained from Industrial Instrument Inc., Cedar Grove, NJ. An equivalent may be used.

<sup>3</sup> The sole source of supply of the apparatus (Serfass Model RCM-15B1) known to the committee at this time is Industrial Instrument Inc., Cedar Grove, NJ. If you are aware of alternative suppliers, please provide this information to ASTM International Headquarters. Your comments will receive careful consideration at a meeting of the responsible technical committee, which you may attend.

<sup>4</sup> Model Cel-A01 has been found satisfactory and can be obtained from Industrial Instrument Inc., Cedar Grove, NJ. An equivalent may be used.

<sup>5</sup> The sole source of supply of the apparatus (Model Cel-A01) known to the committee at this time is Industrial Instrument Inc., Cedar Grove, NJ. If you are aware of alternative suppliers, please provide this information to ASTM International Headquarters. Your comments will receive careful consideration at a meeting of the responsible technical committee, which you may attend.

such specifications are available.<sup>4</sup> Other grades may be used, provided it is first ascertained that the reagent is of sufficiently high purity to permit its use without lessening the accuracy of the determination.

5.2 *Alcoholic Hydrochloric Acid* (0.01 N)—Dilute 5.0 mL (pipet) of concentrated hydrochloric acid to 6 L with denatured 2B ethyl alcohol.

5.3 *Alumina Gel*.<sup>5</sup>

5.4 *Deionized Trichloromonofluoromethane*, in 50-lb 25-kg cylinder. This deionized solvent may be selected pure material or it may be purified by multiple passes through an alumina gel contactor (see Fig. 1). Check the acceptability of this deionized solvent as described in 7.4.

NOTE 1—The use of deionized trichloromonofluoromethane will serve as a check on the cleanliness. The deionized standard should be used as required on a periodic basis when samples are being evaluated. Standards should be checked whenever high values are obtained or there is doubt of the results.

5.5 *Deionized Water*—Distilled water passed through two mixed resin ion exchangers connected in series and having a conductance of no greater than 0.5  $\mu$ S. Check the acceptability of the deionized water in accordance with 7.3.

5.6 *Ethyl Alcohol*.

NOTE 2—Specially denatured ethyl alcohol conforming to Formula SD-2B of the U.S. Bureau of Internal Revenue may be substituted for ethyl alcohol (95 %).

5.7 *Hydrochloric Acid*, concentrated.

<sup>4</sup>Catalog No. H-4040, Harshaw Scientific Co., Cleveland, OH, has been found satisfactory. An equivalent may be used.

<sup>4</sup> *Reagent Chemicals, American Chemical Society Specifications*, American Chemical Society, Washington, DC. For suggestions on the testing of reagents not listed by the American Chemical Society, see *Analar Standards for Laboratory Chemicals*, BDH Ltd., Poole, Dorset, U.K., and the *United States Pharmacopeia and National Formulary*, U.S. Pharmacopeial Convention, Inc. (USPC), Rockville, MD.

<sup>5</sup> *Reagent Chemicals, American Chemical Society Specifications, American Chemical Society, Washington, DC. For suggestions on the testing of reagents not listed by the American Chemical Society, see Analar Standards for Laboratory Chemicals, BDH Ltd., Poole, Dorset, U.K., and the United States Pharmacopeia and National Formulary, U.S. Pharmacopeial Convention, Inc. (USPC), Rockville, MD.*

<sup>5</sup> The sole source of supply of the apparatus (Grade F-3) known to the committee at this time is Aluminum Company of America. If you are aware of alternative suppliers, please provide this information to ASTM International Headquarters. Your comments will receive careful consideration at a meeting of the responsible technical committee, which you may attend.

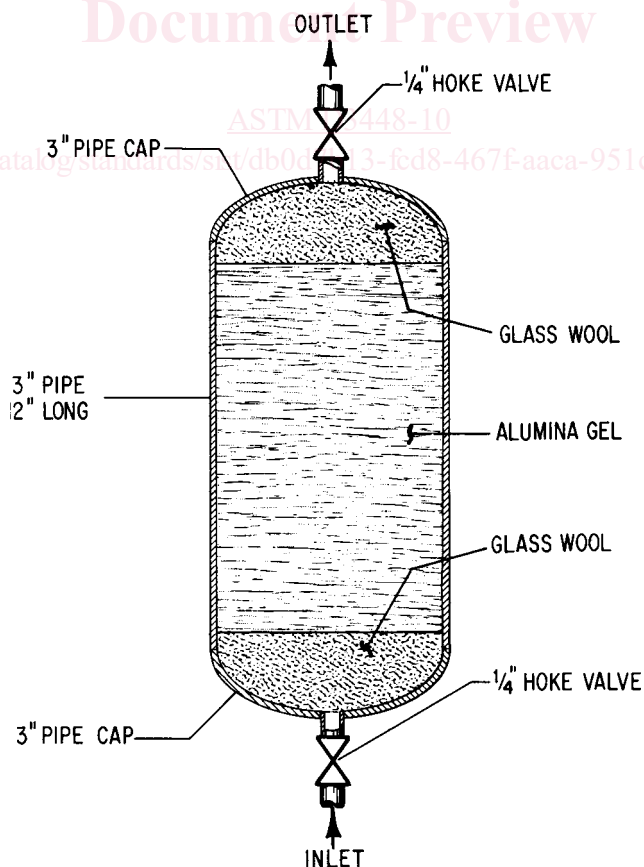


FIG. 1 Alumina Gel Contractor