



Designation: D5356 – 10

Standard Test Method for pH of Chrome Tanning Solutions¹

This standard is issued under the fixed designation D5356; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reappraisal. A superscript epsilon (ϵ) indicates an editorial change since the last revision or reappraisal.

1. Scope

1.1 This test method covers the determination of the pH of chrome tanning solutions.

1.2 The values stated in SI units are to be regarded as standard. No other units of measurement are included in this standard.

1.3 *This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.*

2. Referenced Documents

2.1 *ASTM Standards:*²

D3898 Test Method for Chromic Oxide in Basic Chromium Tanning Liquors

E70 Test Method for pH of Aqueous Solutions With the Glass Electrode

3. Terminology

3.1 *Definitions of Terms Specific to This Standard:*

3.1.1 *pH, of a solution*—the negative logarithm of the hydrogen ion activity.

3.1.1.1 *Discussion*—A solution of pH 7 is neutral at 24°C. Lower numbers indicate increasing acidity, higher numbers indicate increasing alkalinity.

4. Summary of Test Method

4.1 The chrome tanning liquor is diluted to a concentration of 25.0 ± 0.5 g chromic oxide equivalent per litre, allowed to come to equilibrium and the pH is then measured with the appropriate meter. For general details of a procedure, Test Method **E70** should be consulted. The provisions of the

standard shall be governing except where the text of this test method specifically deviates.

5. Significance and Use

5.1 This test method is designed to measure the pH of a chrome tanning solution which has been diluted to a specific concentration. This is considered to be a measure of the acidity or alkalinity of the solution.

5.2 This test method is suitable for quality control in the manufacture of leather.

6. Apparatus

6.1 *pH Meter*, either battery or line-operated equipped with appropriate electrodes. The meter shall have a maximum scale division of 0.1 pH unit and reproducibility of 0.05 pH unit.

NOTE 1—Sellers list reasonably priced models having more sensitivity and greater reproducibility.

7. Reagents

7.1 *Purity of Reagents*—Reagent grade chemicals shall be used in all tests. Unless otherwise indicated, it is intended that all reagents shall conform to the specifications of the Committee on Analytical Reagents of the American Chemical Society, where such specifications are available.³ Other grades may be used, provided it is first ascertained that the reagent is of sufficiently high purity to permit its use without lessening the accuracy of the determination.

7.2 *Purity of Water*—Unless otherwise indicated, references to water shall be understood to mean distilled water or water of equal purity.

7.3 *Buffer Solutions*—Commercially available buffer solutions of 2.00 ± 0.02 pH and 4.00 ± 0.02 pH shall be used for meter standardization, in accordance with the manufacturer's recommendations and the provisions set forth in Test Method **E70**.

¹ This test method is under the jurisdiction of ASTM Committee **D31** on Leather and is the direct responsibility of Subcommittee **D31.06** on Chemical Analysis

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² For referenced ASTM standards, visit the ASTM website, www.astm.org, or contact ASTM Customer Service at service@astm.org. For *Annual Book of ASTM Standards* volume information, refer to the standard's Document Summary page on the ASTM website.

³ *Reagent Chemicals, American Chemical Society Specifications*, American Chemical Society, Washington, DC. For suggestions on the testing of reagents not listed by the American Chemical Society, see *Analar Standards for Laboratory Chemicals*, BDH Ltd., Poole, Dorset, U.K., and the *United States Pharmacopeia and National Formulary*, U.S. Pharmaceutical Convention, Inc. (USPC), Rockville, MD.