



Designation: D6889 – 03 (Reapproved 2011)

Standard Practice for Fast Screening for Volatile Organic Compounds in Water Using Solid Phase Microextraction (SPME)¹

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1. Scope

1.1 This practice covers a procedure for the screening of trace levels of volatile organic compounds in water samples by headspace solid phase microextraction (SPME) in combination with fast gas chromatography with flame ionization detection.

1.2 The results from this screening procedure are used to estimate analyte concentrations to prevent contamination of purge and trap or headspace analytical systems.

1.3 The compounds of interest must have a greater affinity for the SPME absorbent polymer or adsorbent than the sample matrix or headspace phase in which they reside.

1.4 Not all of the analytes which can be determined by SPME are addressed in this practice. The applicability of the absorbent polymer, adsorbent or combination to extract the compound(s) of interest must be demonstrated before use.

1.5 Where used it is the responsibility of the user to validate the application of SPME to the analytes of interest.

1.6 The values stated in SI units are to be regarded as standard. No other units of measurement are included in this standard.

1.7 *This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.* For specific hazard statements, see Section 9.

2. Referenced Documents

2.1 *ASTM Standards:*²

[D1129 Terminology Relating to Water](#)

[D1193 Specification for Reagent Water](#)

¹ This practice is under the jurisdiction of ASTM Committee D19 on Water and is the direct responsibility of Subcommittee D19.06 on Methods for Analysis for Organic Substances in Water.

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² For referenced ASTM standards, visit the ASTM website, www.astm.org, or contact ASTM Customer Service at service@astm.org. For *Annual Book of ASTM Standards* volume information, refer to the standard's Document Summary page on the ASTM website.

[D3694 Practices for Preparation of Sample Containers and for Preservation of Organic Constituents](#)

[D3856 Guide for Management Systems in Laboratories Engaged in Analysis of Water](#)

[D4210 Practice for Intralaboratory Quality Control Procedures and a Discussion on Reporting Low-Level Data \(Withdrawn 2002\)](#)³

[D6520 Practice for the Solid Phase Micro Extraction \(SPME\) of Water and its Headspace for the Analysis of Volatile and Semi-Volatile Organic Compounds](#)

3. Summary of Practice

3.1 This practice employs adsorbent/gas extraction to isolate compounds of interest, see Practice [D6520](#). An aqueous sample is added to a small (2 mL) septum sealed vial. Salt is used to improve analyte recovery. After the addition of a surrogate standard and a short mixing cycle, a SPME fused silica fiber coated with a thick polymer film is then exposed to the aqueous headspace for a few seconds. The fiber is then desorbed in the heated injection port of a GC/FID or GC-MS and the resulting analytes chromatographed on a short narrow bore capillary column. The total analysis time is approximately 3 min.

3.2 The concentrations of the volatile organics in the water sample are estimated to determine whether the sample may be analyzed directly or first diluted prior to purge and trap or headspace analysis.

4. Significance and Use

4.1 This practice provides a general procedure for the solid-phase microextraction (SPME) of volatile organic compounds from the headspace of an aqueous matrix. Absorbent extraction is used as the initial step in the extraction of organic constituents for the purpose of screening and subsequently estimating the concentration of the volatile organic components found in water samples. This information may then be used to determine whether a sample may be analyzed directly by purge and trap or headspace or will require dilution prior to analysis.

³ The last approved version of this historical standard is referenced on www.astm.org.

4.2 Typical detection limits that can be achieved using SPME techniques with gas chromatography (GC) with a flame ionization detector (FID) range from milligrams per litre (mg/L) to micrograms per litre (µg/L). The detection limit, linear concentration range, and sensitivity of this test method for a specific organic compound will depend upon the aqueous matrix, the fiber phase, the sample temperature, sample volume, sample mixing, and the determinative technique employed.

4.3 Solid phase microextraction has the advantage of speed, reproducibility, simplicity, no solvent, small sample size, and automation.

4.3.1 Extraction devices vary from a manual SPME fiber holder to automated commercial devices specifically designed for SPME.

4.3.2 A partial list of volatile organic compounds that can be screened by this practice is shown in **Table 1**.

5. Principles of SPME

5.1 Solid phase microextraction is an equilibrium technique where analytes are not completely extracted from the matrix. With liquid samples, the recovery is dependent on the partitioning or equilibrium of analytes among the three phases present in the sampling vial: the aqueous sample and headspace (**Eq 1**), the fiber coating and aqueous sample (**Eq 2**), and the fiber coating and the headspace (**Eq 3**):

$$K_1 = C_L/C_g \quad (1)$$

$$K_2 = C_F/C_L \quad (2)$$

$$K_3 = C_F/C_g \quad (3)$$

where:

C_L , C_G , and C_F = concentrations of the analyte in these phases.

5.1.1 Distribution of the analyte among the three phases:

$$C_0V_L = C_GV_G + C_LV_L + C_FV_F \quad (4)$$

5.1.2 Concentration of analyte in fiber:

TABLE 1 Check Standard Composition for Screening VOCs in Water

Analyte	Sample Composition, µg/L	Detection Limit, µg/L
TBA	100 000	10 000
Methyl-t-butyl ether	1000	150
cis-1,2-Dichloroethene	3000	300
1,1,1-Trichloroethane	1000	200
Benzene	400	40
1,1,1-Trichloroethane	700	120
Toluene	200	10
Tetrachloroethene	300	50
Chlorobenzene	150	10
Ethylbenzene	100	5
m-Xylene	100	5
styrene	100	5
o-Xylene	100	5
Isopropylbenzene	100	5
2-Chlorotoluene	100	5
1,2,4-Trimethylbenzene	100	5
1,4-Dichlorobenzene-d4	150	5
1,2-Dichlorobenzene	100	5
Naphthalene	100	5

$$C_F = C_0V_LK_1K_2/V_G + K_1V_L + K_1K_2V_F \quad (5)$$

6. Interferences

6.1 Reagents, glassware, septa, fiber coatings and other sample processing hardware may yield discrete artifacts or elevated baselines that can cause poor precision and accuracy. See Terminology **D1129**.

6.1.1 Plastics other than PTFE-fluorocarbon should be avoided. They are a significant source of interference and can adsorb some organics.

7. Apparatus

7.1 *SPME Holder*, manual or automated sampling.

7.1.1 *SPME Fiber Assembly*—Polydimethylsiloxane (PDMS), 30µM or equivalent fiber suitable for volatiles adsorption.

7.2 *Vials with Septa and Caps*, for manual or automated SPME. Vials for automation, 2 mL.

7.3 *Gas Chromatograph*, with flame ionization detector.

7.3.1 *GC Column*, 10 m by 0.25 mm, 1µM film Methyl Silicone, or equivalent.

7.3.2 *GC Guard Column*, 1m by 0.32 mm uncoated, or equivalent.

7.3.3 *Split/splitless Injector*, with 0.75 to 1.0 mm inside diameter insert.

7.3.4 *Optional Septum Replacement Device*.

7.3.5 *Optional SPME Autosampler*.

7.3.6 *GC Compatible Workstation*.

8. Reagents

8.1 *Purity of Water*—Unless otherwise indicated, reference to water shall be understood to mean reagent water conforming to Type II of Specification **D1193**.

8.2 Chemicals, standard materials and surrogates should be reagent or ACS grade or better. When they are not available as reagent grade, they should have an assay of 90 % or better.

8.3 *Sodium Chloride* (NaCl), reagent grade, granular.

8.4 *Surrogate Standard*, 30 mg/L, 1,4-dichlorobenzene-d₄ in methanol.

8.5 *Check Standard*—Prepare a check standard in methanol. Check standard should contain 30 mg/L 1,4-dichlorobenzene-d₄ plus VOCs that will be screened. A typical check standard will provide aqueous concentrations shown in **Table 1** when spiking 4 µL of check standard to 700 µL water sample.

9. Hazards

9.1 The toxicity and carcinogenicity of chemicals used or that could be used in this practice have not been precisely defined. Each chemical should be treated as a potential health hazard. Exposure to these chemicals should be minimized. Each laboratory is responsible for maintaining awareness of OSHA regulations regarding safe handling of chemicals used in this practice.