



Designation: E720 – 11

# Standard Guide for Selection and Use of Neutron Sensors for Determining Neutron Spectra Employed in Radiation-Hardness Testing of Electronics<sup>1</sup>

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*This standard has been approved for use by agencies of the U.S. Department of Defense.*

## 1. Scope

1.1 This guide covers the selection and use of neutron-activation detector materials to be employed in neutron spectra adjustment techniques used for radiation-hardness testing of electronic semiconductor devices. Sensors are described that have been used at many radiation hardness-testing facilities, and comments are offered in table footnotes concerning the appropriateness of each reaction as judged by its cross-section accuracy, ease of use as a sensor, and by past successful application. This guide also discusses the fluence-uniformity, neutron self-shielding, and fluence-depression corrections that need to be considered in choosing the sensor thickness, the sensor covers, and the sensor locations. These considerations are relevant for the determination of neutron spectra from assemblies such as TRIGA- and Godiva-type reactors and from Californium irradiators. This guide may also be applicable to other broad energy distribution sources up to 20 MeV.

NOTE 1—For definitions on terminology used in this guide, see Terminology E170.

1.2 This guide also covers the measurement of the gamma-ray or beta-ray emission rates from the activation foils and other sensors as well as the calculation of the absolute specific activities of these foils. The principal measurement technique is high-resolution gamma-ray spectrometry. The activities are used in the determination of the energy-fluence spectrum of the neutron source. See Guide E721.

1.3 Details of measurement and analysis are covered as follows:

1.3.1 Corrections involved in measuring the sensor activities include those for finite sensor size and thickness in the calibration of the gamma-ray detector, for pulse-height analyzer deadtime and pulse-pileup losses, and for background radioactivity.

1.3.2 The primary method for detector calibration that uses secondary standard gamma-ray emitting sources is considered in this guide and in General Methods E181. In addition, an alternative method in which the sensors are activated in the known spectrum of a benchmark neutron field is discussed in Guide E1018.

1.3.3 A data analysis method is presented which accounts for the following: detector efficiency; background subtraction; irradiation, waiting, and counting times; fission yields and gamma-ray branching ratios; and self-absorption of gamma rays and neutrons in the sensors.

1.4 The values stated in SI units are to be regarded as standard. No other units of measurement are included in this standard.

1.5 *This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.*

## 2. Referenced Documents

2.1 General considerations of neutron-activation detectors discussed in Practice E261, Test Method E262, and Guides E721 and E844 are applicable to this guide. Background information for applying this guide are given in these and other relevant standards as follows:

2.2 *ASTM Standards:*<sup>2</sup>

E170 Terminology Relating to Radiation Measurements and Dosimetry

E181 Test Methods for Detector Calibration and Analysis of Radionuclides

E261 Practice for Determining Neutron Fluence, Fluence Rate, and Spectra by Radioactivation Techniques

<sup>1</sup> This guide is under the jurisdiction of ASTM Committee E10 on Nuclear Technology and Applications and is the direct responsibility of Subcommittee E10.07 on Radiation Dosimetry for Radiation Effects on Materials and Devices.

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<sup>2</sup> For referenced ASTM standards, visit the ASTM website, www.astm.org, or contact ASTM Customer Service at service@astm.org. For *Annual Book of ASTM Standards* volume information, refer to the standard's Document Summary page on the ASTM website.

- E262** Test Method for Determining Thermal Neutron Reaction Rates and Thermal Neutron Fluence Rates by Radioactivation Techniques
- E263** Test Method for Measuring Fast-Neutron Reaction Rates by Radioactivation of Iron
- E264** Test Method for Measuring Fast-Neutron Reaction Rates by Radioactivation of Nickel
- E265** Test Method for Measuring Reaction Rates and Fast-Neutron Fluences by Radioactivation of Sulfur-32
- E266** Test Method for Measuring Fast-Neutron Reaction Rates by Radioactivation of Aluminum
- E393** Test Method for Measuring Reaction Rates by Analysis of Barium-140 From Fission Dosimeters
- E496** Test Method for Measuring Neutron Fluence and Average Energy from  $^3\text{H}(d,n)^4\text{He}$  Neutron Generators by Radioactivation Techniques
- E704** Test Method for Measuring Reaction Rates by Radioactivation of Uranium-238
- E705** Test Method for Measuring Reaction Rates by Radioactivation of Neptunium-237
- E721** Guide for Determining Neutron Energy Spectra from Neutron Sensors for Radiation-Hardness Testing of Electronics
- E844** Guide for Sensor Set Design and Irradiation for Reactor Surveillance, E 706 (IIC)
- E944** Guide for Application of Neutron Spectrum Adjustment Methods in Reactor Surveillance, E 706 (IIA)
- E1018** Guide for Application of ASTM Evaluated Cross Section Data File, Matrix E706 (IIB)
- E1297** Test Method for Measuring Fast-Neutron Reaction Rates by Radioactivation of Niobium

### 3. Significance and Use

3.1 Because of the wide variety of materials being used in neutron-activation measurements, this guide is presented with the objective of bringing improved uniformity to the specific field of interest here: hardness testing of electronics primarily in critical assembly reactor environments.

NOTE 2—Some of the techniques discussed are useful for 14-MeV dosimetry. See Test Method **E496** for activation detector materials suitable for 14-MeV neutron effects testing.

NOTE 3—The materials recommended in this guide are suitable for  $^{252}\text{Cf}$  or other weak source effects testing provided the fluence is sufficient to generate countable activities.

3.2 This guide is organized into two overlapping subjects; the criteria used for sensor selection, and the procedures used to ensure the proper determination of activities for determination of neutron spectra. See Terminology **E170** and General Methods **E181**. Determination of neutron spectra with activation sensor data is discussed in Guides **E721** and **E944**.

### 4. Foil Sets

#### 4.1 Reactions Considered:

4.1.1 Neutron-induced reactions appropriate for this guide are listed in **Table 1**. The table includes most of the reactions used in this field. Those not marked with an asterisk are recommended because of their demonstrated compatibility with other reactions used in spectrum adjustment determinations. This compatibility is primarily based on experience with

the ENDF/B-VI (**1**, **2**),<sup>3</sup> and IRDF-90 (**3**) cross-sections. These recommendations may change modestly as revisions are made in the ENDF/B and IRDF dosimetry cross sections. Other reactions may be useful in particular circumstances with appropriate care. It is important that the user take full account of both the footnotes attached to each reaction and the discussions in the body of the text about individual reactions when implementing the foil-activation technique.

4.1.2 The four paired columns under the labels fast burst (**13**) and “TRIGA (**14**) Type” list the energy ranges within which 95 % of the response occurs for these two representative spectra. These limits are just a guide because the response often varies widely within each range. The response limits for an idealized fission spectrum with no  $1/E$  tail can be much different (shifted toward higher energy) for resonance reactions. For example, in a Watt fission spectrum the  $^{197}\text{Au}(n, \gamma)^{198}\text{Au}$  has a 95 % response between  $5.0 \times 10^{-2}$  and 2.7 MeV. The recommended foil mass column gives values that are designed to minimize self-absorption, self-shielding, and other corrections, provided the foils are 1.27 cm in diameter. The  $E_{\text{r}} \cong 0$  fission foils,  $^{235}\text{U}$  and  $^{239}\text{Pu}$ , have similar cross-section shapes. However, the  $^{235}\text{U}$  foil is preferred since it is less expensive and is much less of a health hazard than  $^{239}\text{Pu}$ . In addition, when measuring soft (TRIGA) spectra, the  $^{235}\text{U}$  foil is useful in determining the correction for the  $^{235}\text{U}$  impurity in the  $^{238}\text{U}$  foil (which is readily available with about 400 ppm or less  $^{235}\text{U}$  impurity).

4.1.3 Although sulfur is listed and is used widely as a monitor foil, it is the only recommended sensor requiring beta particle detection and, therefore, requires a different calibration and counting technique. The  $^{58}\text{Ni}(n,p)^{58}\text{Co}$  reaction has about the same threshold energy and, therefore, can be used instead of the  $^{32}\text{S}(n,p)^{32}\text{P}$  if it acquires sufficient activity. Many facilities use sulfur as a routine monitor because its two-week half-life allows a convenient period for counting and permits reuse of the sensor after 6 to 9 months. Automated beta counters are commercially available. Neither nickel nor sulfur should be counted for the  $(n,p)$  reaction products immediately after irradiation because for nickel the  $^{58}\text{Co}$  must build up through a metastable state, and for sulfur there are competing reactions. According to Test Method **E264** the waiting period for  $^{58}\text{Co}$  should be 4 days. For  $^{32}\text{P}$ , Test Method **E265** recommends waiting 24 h. Corrections can be made for shorter waiting periods.

4.1.4 In selecting dosimetry reactions one should consider the validation of the cross sections and associated uncertainty as demonstrated in the  $^{235}\text{U}$  thermal fission and the  $^{252}\text{Cf}$  spontaneous fission benchmark neutron fields. Ref (**15**) provides a recent comparison of the measured and calculated spectrum-averaged cross sections for these benchmark fields.

4.1.5 Some frequently used reactions have shown relatively consistent deviations of measured to calculated activity ratios in many different spectra determinations. For example, when ENDF/B-V cross sections are used in the reaction  $^{63}\text{Cu}(n, \gamma)^{64}\text{Cu}$ , the calculated activity is usually low, and an adjustment code will try to raise the spectrum in the vicinity of Cu

<sup>3</sup> The boldface numbers in parentheses refer to the list of references at the end of this guide.

**TABLE 1 Activation Foils**

Reaction	Fast Burst <sup>A</sup>		TRIGA Type <sup>A</sup>		$E_{\gamma}^B$ , (keV)	Gamma Emission Probability <sup>B</sup>	Fast Fission Yield, <sup>C</sup> %	$T_{1/2}^B$	Recommended Foil Mass, g <sup>D</sup>	Footnotes
	$E_L$ , MeV	$E_H$ , MeV	$E_L$ , MeV	$E_H$ , MeV						
<sup>197</sup> Au( <i>n,γ</i> ) <sup>198</sup> Au	4.00 – 6	7.20 – 4	3.80 – 6	9.20 – 6	411.8025	95.54		2.6944 days	0.06	<i>E,F,G</i>
<sup>59</sup> Co( <i>n,γ</i> ) <sup>60</sup> Co	7.60 – 6	4.50 – 4	6.90 – 7	1.43 – 4	1173.2	99.85		5.2711 years	0.06	<i>E,G</i>
					1332.5	99.98				
<sup>58</sup> Fe( <i>n,γ</i> ) <sup>59</sup> Fe	1.00 – 6	2.10 + 0	5.25 – 7	1.00 – 2	1099.245	56.59		44.495 days	0.15	<i>E,H</i>
					1291.59	43.21				
<sup>55</sup> Mn( <i>n,γ</i> ) <sup>56</sup> Mn	5.25 – 7	6.60 – 1	4.75 – 7	1.10 – 3	846.76	98.85		2.57878 h	0.05	<i>E,F</i>
					1810.726	26.9				
<sup>63</sup> Cu( <i>n,γ</i> ) <sup>64</sup> Cu	1.15 – 6	2.30 + 0	5.25 – 7	9.60 – 3	1345.77	0.4743		12.700 h	0.15	<i>E</i>
<sup>23</sup> Na( <i>n,γ</i> ) <sup>24</sup> Na	6.30 – 7	2.00 + 0	5.25 – 7	3.00 – 3	1368.626	99.993		14.9574 h	0.10	<i>E,I,J</i>
					2754.1	99.872				
<sup>45</sup> Sc( <i>n,γ</i> ) <sup>46</sup> Sc	4.25 – 7	1.00 + 0	4.00 – 7	4.75 – 4	889.27	99.983		83.788 days	0.05	<i>E</i>
					1120.537	99.986				
<sup>235</sup> U( <i>n,β</i> ) <sup>140</sup> La	9.20 – 2	4.70 + 0	6.30 – 4	3.80 + 0	1596.2	95.4	6.105	40.28 h	0.30	<i>E,K,L</i>
<sup>235</sup> U( <i>n,β</i> ) <sup>95</sup> Zr	9.20 – 2	4.70 + 0	6.30 – 4	3.80 + 0	724.2	44.27	6.363	64.03 days	0.60	<i>E,L</i>
					756.7	54.4				
<sup>239</sup> Pu( <i>n,β</i> ) <sup>140</sup> La	1.43 – 1	4.80 + 0	8.80 – 4	4.30 + 0	1596.2	95.4	5.326	40.28 h	1.00	<i>E,K,L</i>
<sup>239</sup> Pu( <i>n,β</i> ) <sup>95</sup> Zr	1.43 – 1	4.80 + 0	8.80 – 4	4.30 + 0	724.2	44.1	4.685	64.02 days	0.60	<i>E,L</i>
					756.7	54.4				
<sup>93</sup> Nb( <i>n,n</i> ) <sup>93m</sup> Nb	8.40 – 1	5.70 + 0	1.00 + 0	5.50 + 0	16.5-19.6	11.0		16.12 years		<i>M</i>
<sup>103</sup> Rh( <i>n,n</i> ) <sup>103m</sup> Rh	5.50 – 1	5.70 + 0	6.90 – 1	5.70 + 0	39.8	0.068		56.1 min		<i>M</i>
<sup>237</sup> Np( <i>n,β</i> ) <sup>140</sup> La	5.75 – 1	5.60 + 0	6.60 – 1	5.50 + 0	1596.2	95.4	5.489	40.28 h	0.60	<i>E,K,L,N</i>
<sup>237</sup> Np( <i>n,β</i> ) <sup>95</sup> Zr	5.75 – 1	5.60 + 0	6.60 – 1	5.50 + 0	724.2	44.1	5.699	64.02 days	0.60	<i>E,L</i>
					756.7	54.4				
<sup>115</sup> In( <i>n,n</i> ) <sup>115m</sup> In	1.00 + 0	6.00 + 0	1.20 + 0	5.80 + 0	336.2	45.9		4.49 h	0.12	
<sup>238</sup> U( <i>n,β</i> ) <sup>140</sup> La	1.50 + 0	6.90 + 0	1.50 + 0	6.60 + 0	1596.2	95.4	5.948	40.28 h	1.00	<i>E,K,L,O</i>
<sup>238</sup> U( <i>n,β</i> ) <sup>95</sup> Zr	1.50 + 0	6.90 + 0	1.50 + 0	6.60 + 0	724.2	44.1	5.105	64.02 days	1.00	<i>E,L</i>
					756.7	54.4				
<sup>232</sup> Th( <i>n,β</i> ) <sup>140</sup> Ba	1.50 + 0	7.40 + 0	1.50 + 0	7.10 + 0	537.3	24.4	7.704	12.753 days	1.00	<i>E,K,P</i>
<sup>232</sup> Th( <i>n,β</i> ) <sup>95</sup> Zr	1.50 + 0	7.40 + 0	1.50 + 0	7.10 + 0	724.2	44.1	5.374	64.02 days	1.00	<i>E,L</i>
					756.7	54.4				
<sup>54</sup> Fe( <i>n,p</i> ) <sup>54</sup> Mn	2.30 + 0	7.70 + 0	2.30 + 0	7.40 + 0	834.838	99.9746		312.13 days	0.15	<i>E</i>
<sup>58</sup> Ni( <i>n,p</i> ) <sup>58</sup> Co	2.00 + 0	7.60 + 0	2.00 + 0	7.30 + 0	810.7	99.45		70.83 days	0.30	<i>E</i>
<sup>47</sup> Ti( <i>n,p</i> ) <sup>47</sup> Sc	1.90 + 0	7.60 + 0	1.90 + 0	7.30 + 0	159.4	68.3		3.35 days	0.15	<i>E,Q,R</i>
<sup>32</sup> S( <i>n,p</i> ) <sup>32</sup> P	2.40 + 0	7.50 + 0	2.30 + 0	7.30 + 0	1710.6	100. (beta)		14.284 days	...	<i>S</i>
<sup>64</sup> Zn( <i>n,p</i> ) <sup>64</sup> Cu	2.60 + 0	7.70 + 0	2.60 + 0	7.40 + 0	1345.7	0.4743		12.700 h	0.30	<i>E</i>
<sup>27</sup> Al( <i>n,p</i> ) <sup>27</sup> Mg	3.50 + 0	9.40 + 0	3.40 + 0	9.20 + 0	843.8	71.8		9.46 min	0.30	<i>E</i>
					1014.4	28.0				
<sup>46</sup> Ti( <i>n,p</i> ) <sup>46</sup> Sc	3.80 + 0	9.60 + 0	3.70 + 0	9.20 + 0	889.3	99.983		83.788 days	0.15	<i>E,Q</i>
					1120.5	99.986				
<sup>56</sup> Fe( <i>n,p</i> ) <sup>56</sup> Mn	5.50 + 0	1.14 + 1	5.50 + 0	1.10 + 1	846.7	98.85		2.57878 h	0.15	<i>E,T</i>
					1810.7	26.9				
<sup>24</sup> Mg( <i>n,p</i> ) <sup>24</sup> Na	6.50 + 0	1.17 + 1	6.50 + 0	1.13 + 1	1368.6	99.993		14.9574 h	0.03	<i>E,J</i>
					2754.1	99.872				
<sup>27</sup> Al( <i>n,α</i> ) <sup>24</sup> Na	6.50 + 0	1.21 + 1	6.50 + 0	1.17 + 1	1368.6	99.993		14.9574 h	0.30	<i>E,J</i>
					2754.1	99.872				
<sup>48</sup> Ti( <i>n,p</i> ) <sup>48</sup> Sc	5.90 + 0	1.24 + 1	5.90 + 0	1.20 + 1	983.5	100.1		43.7 h	0.15	<i>E</i>
					1037.5	97.56				
					1312.1	100.1				
<sup>93</sup> Nb( <i>n,2n</i> ) <sup>92m</sup> Nb	9.70 + 0	1.45 + 1	9.40 + 0	1.40 + 1	934.4	99.1		10.15 days		
<sup>127</sup> I( <i>n,2n</i> ) <sup>126</sup> I	9.70 + 0	1.47 + 1	9.70 + 0	1.43 + 1	388.6	35.6		12.93 days	0.25	<i>E</i>
					666.3	32.9				
<sup>65</sup> Cu( <i>n,2n</i> ) <sup>64</sup> Cu	1.08 + 1	1.57 + 1	1.07 + 1	1.53 + 1	1345.7	0.475		12.701 h	0.15	<i>E,M</i>
<sup>63</sup> Cu( <i>n,2n</i> ) <sup>62</sup> Cu	1.19 + 1	1.66 + 1	1.19 + 1	1.63 + 1	875.7	0.150		9.74 min	0.15	<i>E,H</i>
<sup>90</sup> Zr( <i>n,2n</i> ) <sup>89</sup> Zr	1.28 + 1	1.69 + 1	1.27 + 1	1.67 + 1	909.1	99.0		78.4 h	0.10	
<sup>58</sup> Ni( <i>n,2n</i> ) <sup>57</sup> Ni	1.32 + 1	1.71 + 1	1.31 + 1	1.69 + 1	1377.6	81.2		35.9 h	0.30	

<sup>A</sup> Energy limits inside of which 95 % of the detector response occurs for each reaction (see Practice E261 and Refs (4,5)). The foils are assumed to have Cd covers as described in Footnote E.

<sup>B</sup> Data taken from Refs (6-8). Ref (8) takes precedent, but it only addresses reactions used in detector calibration. In other cases, Ref (6) provides the half-life and Ref (7) provides the gamma yields. Many gamma-ray energies rounded to nearest 0.1 keV. For uncertainties on values, see references.

<sup>C</sup> Fission yields can be found in Ref (9).

<sup>D</sup> Choice of mass is based on assumed foil diameter of 1.27 cm.

<sup>E</sup> Cd covers 0.5 to 1-mm thicknesses. Pairs of bare and Cd-covered foils are advantageous for resonance reactions.

<sup>F</sup> Use <sup>59</sup>Co instead of <sup>197</sup>Au and <sup>55</sup>Mn for very long irradiations.

<sup>G</sup> Use dilute aluminum-gold alloy (<0.2 % Au) when possible.

<sup>H</sup> Do not count the 0.511 line.

<sup>I</sup> Use in the form of NaCl.

<sup>J</sup> The 1986 edition of Ref (10) has a typographical error for the half-life of <sup>24</sup>Na. The correct number can be found in previous editions. The correct number can also be found in Ref (6).

<sup>K</sup> This is the 40.27-h daughter of 12.75-day <sup>140</sup>Ba. Wait 5 days for maximum decay rate (see Test Method E393).

<sup>L</sup>  $E_{\gamma} = 0.01$  MeV shielded with <sup>10</sup>B sphere. (Use of <sup>10</sup>B shield is important for soft (TRIGA) spectra where  $\Phi(E < 0.01$  MeV) will otherwise dominate).

<sup>M</sup> Precautions must be taken in counting because of the low gamma-ray energy. See Test Method E1297 for details of <sup>93m</sup>Nb use. For <sup>103m</sup>Rh, X-rays are typically counted rather than listed gamma ray. See Ref (7).

<sup>N</sup> If a <sup>10</sup>B sphere is used for the <sup>239</sup>Pu foil, then a <sup>10</sup>B sphere should also be used for the <sup>237</sup>Np foil so that correction for <sup>239</sup>Pu impurity in the <sup>237</sup>Np foil can be made.

<sup>O</sup> If a <sup>10</sup>B sphere is used for the <sup>235</sup>U foil, then a <sup>10</sup>B sphere should also be used for the <sup>238</sup>U foil so that correction for <sup>235</sup>U impurity in the <sup>238</sup>U foil can be made.

<sup>P</sup> Radioactivity of <sup>232</sup>Th interferes with the <sup>140</sup>La line.

<sup>Q</sup> At high energies (>10 MeV), account for (*n,np*) contributions from higher atomic number *Ti* isotopes.

<sup>R</sup> See Refs (11) and (12).

<sup>S</sup> Requires β counting techniques, see Test Method E265.

<sup>T</sup> Maximum Mn impurity = 0.001 %, Cd covered. Do not use <sup>56</sup>Fe foil for long irradiations.

\* Not recommended for use at this time either because of large uncertainties or because of conflicts with other reactions during spectrum adjustment procedures.

resonances. In fact, however, this consistent behavior indicates that the tabulated cross-section values in some important energy region are too small. The analyst must then choose one of the following alternatives: (1) leave out reactions which have demonstrated consistent deviations; (2) seek better cross-section sets; (3) assign wide error bars or low statistical weight to these reactions. It is recommended that the first option be chosen because a sufficient number of well-established cross sections do exist to satisfactorily determine fast reactor spectra. Furthermore, if the cross section for a particular reaction is not well established, and it is assigned too large a weight in the spectrum adjustment procedure, the final spectrum can be severely distorted. Other suspect reactions are noted in Table 1 with an asterisk.

NOTE 4—Some of the reactions not recommended at this time (on the basis of inconsistencies among recommended cross sections) may be upgraded when more recent evaluations are applied to a wide range of neutron spectra.

## 4.2 Foil Impurities:

4.2.1 Foil impurities are especially serious for a moderated source (TRIGA reactor) when an impurity leads to the same reaction product by way of thermal-neutron capture. Some examples of these foils, with impurities in parentheses, are <sup>238</sup>U (<sup>235</sup>U), <sup>27</sup>Al (<sup>23</sup>Na), <sup>56</sup>Fe (<sup>55</sup>Mn), and <sup>24</sup>Mn (<sup>23</sup>Na).

4.2.2 For a soft spectrum, such as the TRIGA J-tube spectrum [boral (boron-aluminum alloy) shielded], the number of fissions in the <sup>235</sup>U foil (Cd covered) is about 100 times the number occurring in the <sup>238</sup>U foil; therefore, the <sup>238</sup>U must have an impurity level of <sup>235</sup>U of no more than about 200 ppm for an uncertainty of 2 % or less in determining accurately the <sup>238</sup>U activity. Higher impurity levels of <sup>235</sup>U can be tolerated for Godiva-type reactors where the fluence below 10 keV is much lower, or with TRIGA-type reactors if the <sup>235</sup>U foil data are used for correcting the <sup>238</sup>U activity.

4.2.3 When the <sup>56</sup>Fe foil (Cd covered) is used in a TRIGA spectrum, it should have no more than 10 ppm <sup>55</sup>Mn impurity to keep the contribution from the <sup>55</sup>Mn(*n,γ*)<sup>56</sup>Mn reaction to less than 2 %. Similarly, the <sup>55</sup>Mn impurity should be no more than 100 ppm when using the <sup>56</sup>Fe foil at 50 cm from a Godiva-type reactor (which is approximately 2 m above the concrete floor) in order to achieve the same level of accuracy. Data from a <sup>55</sup>Mn foil (Cd covered) can be used to correct the <sup>56</sup>Fe data if the impurity correction is ≤20 % of the total (*n,p*) activation, and the percent of manganese in iron is accurately known.

## 4.3 Influence of Nuclear Data on Foil Selection:

4.3.1 Since the total number of interactions is deduced from an absolute specific activity determination, that activity should be determined with good accuracy (of the order of 5 %), and the foils selected should have gamma-ray yields known to the same or better accuracy. Some of the factors involved in

determining these yields include conversion-electron production, branching ratio to a given energy level, and fission yield.

4.3.2 The 1596-keV gamma-ray transition from <sup>140</sup>La produced by <sup>232</sup>Th fission is not usually useful because of interference from <sup>232</sup>Th radioactivity. This often has led to the use of the 537-keV transition from the <sup>140</sup>Ba precursor of <sup>140</sup>La, having a gamma-transition probability of 0.244 per <sup>140</sup>Ba decay. The use of <sup>140</sup>Ba generally requires the chemical separation of this isotope from the rest of the fission products so that the 537-keV line can be seen above competing lines. See Test Method E393.

4.3.3 The choice of element, and hence the gamma-ray transition, directly influences the accuracy of determining the specific activity induced by neutron irradiation. It also influences the final choice of foil thickness, in that the selection of an element resulting in a low-energy gamma ray may lead to a large self-absorption correction. For example, the <sup>232</sup>Th foil of Table 1 has a maximum attenuation of 22 %, or an average correction of about 11 %, for the 537-keV transition. This represents an upper limit for the thickness of that foil. Therefore, the self-attenuation of gamma rays, as well as the neutron self-shielding discussed later, will influence the foil selection.

NOTE 5—For other considerations in the selection of specific foils, see Guide E844, Practice E261, and Test Methods E262, E263, E264, E265, E266, E704, and E705.

## 5. Apparatus

5.1 The gamma-ray detector should be a germanium-type detector (either Ge(Li) or intrinsic) with an energy resolution of 2.5 keV or better (full-width at half-maximum (FWHM) at 1173 keV). Associated equipment would include a multichannel pulse-height analyzer and a precision pulse generator with calibrated pulse-height and pulse-rate inputs into the detection system.

5.2 Foil and source holders should be used to provide precise positioning of a gamma-ray standard source and of each activated foil with respect to the detector. Required precision is about 0.2 mm or better in distance from the window of the detector or in lateral alignment.

5.3 National standard sources that are traceable to NIST (or their equivalent) should be used for calibration of the detection system.

## 6. Precautions

6.1 *Scattering Problems*—A sensor with a strong resonance absorption, such as a thick <sup>235</sup>U foil, should not be placed in front of a 1/*v* detector, and thick foils with covers should not be stacked because accurate corrections for the resultant scattering are difficult to determine. With an isotropic neutron-fluence,

$\Phi_o$ , incident on stacked foils, the reduction in the fluence rate caused by scattering at a given foil can be estimated by using the following equation:

$$\Phi = \Phi_o e^{-\sum_i \sigma_i X_i}$$

where  $\Phi$  is the attenuated fluence,  $\sum_i$  is a summation-over- $i$  symbol,  $\sigma_i$  is the total macroscopic scattering cross section in  $\text{cm}^{-1}$ , and  $X_i$  is the thickness of the  $i^{\text{th}}$  foil in centimetres. The summation is up to the foil of interest, located at its appropriate depth (distance from source) in the foil stack. For best results, the reduction in fluence rate should be less than 10 % for the foil located at the maximum depth.

**6.2 Foil Self-Shielding**—For the thicknesses of the foils recommended, the correction for self-shielding is appreciable only for pure gold foils (with its highly absorbing resonance at about 5 eV), the correction for a 0.025-mm thick foil being about a factor of two for epicadmium neutrons (that is, neutrons with energies greater than 0.5 eV) (16).

NOTE 6—Dilute aluminum-gold alloys are available and do not generally require self-shielding corrections.

**6.3 Fluence Nonuniformity**—If all the foils cannot be located in a region of uniform fluence rate (as determined by symmetry considerations), they can be located at different positions (and, hence, with different fluence rates) as long as the neutron energy spectrum is constant. If the fluence varies by more than 3 % from point to point, fluence monitors should be used with each foil. Around a Godiva-type reactor, sulfur foils can serve as monitors near the individual foils. Where space is more limited, then nickel [ $^{58}\text{Ni}(n,p)^{58}\text{Co}$ ], iron [ $^{54}\text{Fe}(n,p)^{54}\text{Mn}$ ], or even aluminum [ $^{27}\text{Al}(n,\alpha)^{24}\text{Na}$ ] should be considered for monitors. (See Practice E261 for other relevant considerations.) Often a better solution is obtained by mounting all foils on a rotating disk or ring to ensure that they receive the same fluence.

**6.4 Fluence Rate Depression**—At low energies, fluence rate depression can be significant for bare thermal-neutron detectors near cadmium-covered foils if both are embedded in a moderator. This is because the cover on one foil can shadow adjacent bare foils. At high energies, depression can be significant for foils irradiated under the same conditions if the moderator contains reactor fuel. However, this is ordinarily not a problem, since in the sizable irradiation volumes normally used for radiation damage studies, the cadmium covers (as well as the foils) generally subtend a negligibly small solid angle at the point of any surrounding moderator or fuel. Fluence rate depression is usually insignificant for irradiation in a Godiva reactor glory hole.

## 7. General Handling Procedures

**7.1 Foil Encapsulation**—Fission foils should be encapsulated in sealed containers to avoid oxidation, loss of material, and for health-safety requirements. If a  $^{239}\text{Pu}$  foil is used (instead of the much safer  $^{235}\text{U}$  foil), it will require special encapsulation and periodic monitoring to check for leakage of the material. Copper encapsulation has been found satisfactory for  $^{235}\text{U}$ ,  $^{238}\text{U}$ ,  $^{237}\text{Np}$ , and  $^{232}\text{Th}$  foils. The thickness of the copper capsule should be about 0.1 to 0.25 mm at the flat surfaces and soldered at the periphery.

### 7.2 Foil Covers:

**7.2.1** As noted in Table 1, cadmium covers of 0.5 to 1-mm thickness are prescribed for all fission foils and 1/v detectors. Cadmium covers also should be used for finite-threshold foils with trace impurities that yield the same reaction product by means of thermal-neutron capture. Examples are foils such as  $^{238}\text{U}$ ,  $^{56}\text{Fe}$ ,  $^{58}\text{Ni}$ , and  $^{27}\text{Al}$  with impurities of  $^{235}\text{U}$ ,  $^{55}\text{Mn}$ ,  $^{59}\text{Co}$ , and  $^{23}\text{Na}$ , respectively. Depending on the concentration, such impurities can lead to large correction factors. For large correction factors (that is, greater than 5 %), cadmium-covered foils made of the impurity materials should be irradiated. Then, corrections can be made with good accuracy if the impurity concentration in the primary threshold foil is accurately known. If the impurity concentration is not known, a thermal-neutron activation analysis of the foil can provide data for the necessary correction. Cadmium covers may not be required for foils irradiated in the empty “glory hole” of a fast-pulse reactor, a cavity in which little or no moderator material is normally present (that is, less than 0.5  $\text{g}/\text{cm}^2$ ).

**7.2.2** Covers of  $^{10}\text{B}$  for fission foils are useful when measuring a soft TRIGA spectrum. However, if a boral shield that provides good 4- $\pi$  geometry surrounds the irradiation cavity, and if a negligible amount of moderator is contained within the shield, then the  $^{10}\text{B}$  covers may not be required. The effect of the boral shielding should be accounted for properly when the neutron spectrum is adjusted with a proper computer code. More is said about boron cover corrections in 7.2.4.

NOTE 7—Spectra adjustment codes are discussed in Guides E721 and E944.

**7.2.3** If no  $^{10}\text{B}$  covers are used for the foils, and if the TRIGA irradiation cavity is only partially shielded by boral, then it will be difficult to determine the neutron spectrum from  $10^{-2}$  MeV down to about  $3 \times 10^{-7}$  MeV. If the TRIGA irradiation cavity has only partial boral shielding, it is important that all the fission foils, all the 1/v foils, and the foils with important 1/v impurities be placed in a boral “box” or a  $^{10}\text{B}$  cover. For best results, a  $^{10}\text{B}$  cover of 1 to 1.8  $\text{g}/\text{cm}^2$  of (93 %)  $^{10}\text{B}$  should be used. In this way, the fraction of activations arising from neutrons in the energy range from  $3 \times 10^{-7}$  MeV to  $10^{-2}$  MeV will be reduced greatly. The effect of the cover thickness can be accounted for by a spectrum adjustment code provided that the effective attenuation cross section that accounts for scattering in the cover is available. See 7.2.5.

**7.2.4** For a Godiva-type reactor,  $^{10}\text{B}$  covers may not be required, and cadmium covers may be sufficient for irradiation distances of less than 1 m from the reactor when the reactor is located a few metres above the concrete floor. Cadmium covers also may be used in the glory hole where the number of low-energy neutrons is insignificant. If  $^{10}\text{B}$  covers are used, activities may require correction for scattering by the  $^{10}\text{B}$ . The correction can be determined either experimentally with pure finite-threshold fission foils ( $^{237}\text{Np}$  or  $^{232}\text{Th}$ ) that contain negligible zero-threshold impurities, or with a neutron transport calculation that takes into account the thickness of the material (17).

7.2.5 The attenuation by a boron cover of the neutron fluence is not adequately treated by many of the spectrum adjustment codes (18). Some versions of the spectrum adjustment code, SAND II (19), for example, uses a simple exponential attenuation function versus energy, and because most irradiations are conducted in wide-beam or isotropic configurations, scattered neutrons are not in general lost from the beam. As a result, the absorption cross section of the boron should generally be used to determine the attenuation. However, in many configurations (such as narrow-beam geometry or down scattering of the neutrons to lower energy) the scattering portion of the cross section can remove additional neutrons and the true effective removal cross section value will fall somewhere in between the total and the absorption cross section. This is especially noticeable if the response of the foil is concentrated above the 10-keV limit where the  $^{10}\text{B}$  absorption ceases to dominate the cross section. Thus, for high-threshold fission foils such as  $^{238}\text{U}$  and  $^{237}\text{Np}$  or a normal threshold foil such as nickel, the additional scattering will result in additional attenuation. For example, some experiments and calculations indicate that these corrections are of the order of 10 % for a 1.65-g/cm<sup>2</sup>  $^{10}\text{B}$  cover and a thin 12.7-mm diameter fission foil (20). Other work indicates that these scattering corrections may be somewhat larger (21). Strictly speaking, a calculation of the transport in the full-experiment geometry through the boron cover should be performed for each geometry (18). Measurements with a high-threshold foil,  $^{58}\text{Ni}(n,p)^{58}\text{Co}$ , have shown a transmission factor of 0.9 in a Godiva-type exposure geometry (15). This compares with a calculated value (for which only the boron capture cross section is used) of 0.96.

NOTE 8—A monitor foil such as nickel used both inside and outside a boron ball can be used to normalize the boron-covered-fission-foil exposure to that of the rest of the foil set in case positioning errors are likely to be significant. The nickel ratio is not very sensitive to spectrum shape. The procedure is to multiply the fission foil activities by a factor that ensures that the ratio of nickel activities inside and outside the boron ball is about 0.9.

7.2.6 Another advantage of using covers (B, Cd) on broad energy-response foils is that it restricts that response and permits improved definition of the spectrum during the adjustment process. If both bare and Cd-covered resonance materials (such as Au and Na) are exposed, much better definition of the shape of the spectrum in the epithermal and thermal region can be obtained.

NOTE 9—Some versions of spectrum adjustment codes handle covers through the use of auxiliary codes that apply an energy-dependent-cover correction factor to the dosimetry cross section.

7.2.7 If the spectrum is to be well defined, then the foil set must contain a large fraction of the reactions from Table 1 and possess response functions spread as uniformly over energy as is possible. This is necessary to ensure that the spectrum adjustment codes can arrive at sufficiently restricted solutions. With broad response functions the calculated fluence at one energy can influence the calculated spectrum values at distant energies. If at all possible include  $^{237}\text{Np}$ , and  $^{239}\text{Pu}$  or  $^{235}\text{U}$  to provide sensitivity between 10 keV and 1 MeV where few

other reactions have significant response. Silicon devices are also sensitive in this energy region and can be used as spectrum sensors (22).

## 8. Certification of Foil Purity

8.1 The foil purity analysis results should be recorded permanently so that appropriate impurity corrections can be made. The acceptable uncertainty in the results mainly dictates what impurity concentrations are acceptable. It also depends on the nature and source of the neutron spectrum being measured (see 4.2). If, for example, the percentage impurity of  $^{235}\text{U}$  in a foil of  $^{238}\text{U}$  is known to be 400 ppm to an accuracy of 10 %, a separate  $^{235}\text{U}$  foil can be irradiated in the same way as the primary foil to determine a proper correction factor. In this case, the impurity effect can be reduced to 10 % of its stated value (40 ppm)  $^{235}\text{U}$  in  $^{238}\text{U}$  by applying the correction factor. In determining the activity of a  $^{238}\text{U}$  foil irradiated with a TRIGA spectrum to an uncertainty of 2 % or less, up to 2000 ppm of  $^{235}\text{U}$  impurity could be tolerated (see 4.2).

## 9. Determination of Activities

9.1 A suitable set of sensors is placed in the neutron field under study. After irradiation, the specific activities of the sensor are determined by counting the gamma-ray emissions from each foil and applying appropriate corrections.

NOTE 10—Other energy response functions appropriate for spectrum adjustment procedures measured by detection of other effects, such as emulsion tracks or even displacement damage, can also be used successfully. See Guide E944, Section 4.1.

9.2 The measured specific activities of the activation foils are related to the incident neutron energy-fluence spectrum by the following equation:

$$R_j = \int_0^\infty \sigma_j(E) \Phi(E) dE \quad 1 \leq j \leq n \quad (1)$$

where:

$R_j$  = measured specific activity of an activated foil isotope  $j$ ,  
 $\sigma_j(E)$  = neutron cross section at energy  $E$  for isotope  $j$ ,  
 $\Phi(E)$  = incident neutron fluence differential in energy, and  
 $n$  = number of reactions.

9.3 The differential neutron energy-fluence spectrum  $\Phi(E)$  is calculated by means of a computer code that utilizes the specific activity data from the activation foil set. A number of these codes have been developed for this purpose and are available from the Oak Ridge National Laboratory Radiation Safety Information Computation Center (23).

## 10. Detector Calibration Procedures

10.1 Follow the general considerations in General Methods E181 and Test Method E265 on energy and efficiency calibration of the detector.

10.2 The germanium detector is usually operated at low temperatures (near the boiling point of liquid nitrogen). This requires the detector to be in a cryostat under vacuum. Normally, a thin window separates the detector's face from the outside environment. In such an enclosure, the exact position