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## Gel permeation chromatography (GPC) —

### Part 2: N,N-Dimethylacetamide (DMAC) as eluent

**iTeh STANDARD PREVIEW**  
*Chromatographie par perméation de gel (GPC) —*  
*Partie 2: Eluant au N,N-Diméthylacetamide (DMAC)*  
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# Contents

Page

<b>Foreword</b>	<b>v</b>
<b>1 Scope</b>	<b>1</b>
<b>2 Normative references</b>	<b>1</b>
<b>3 Terms and definitions</b>	<b>1</b>
<b>4 Principle</b>	<b>2</b>
<b>5 Apparatus</b>	<b>2</b>
5.1 Eluent supply	3
5.2 Pump	3
5.3 Injection system	3
5.4 Separation columns	3
5.5 Column temperature control	5
5.6 Detector	5
<b>6 Reagents</b>	<b>6</b>
<b>7 Calibration of the apparatus</b>	<b>6</b>
7.1 General	6
7.2 Specification for the calibration standards	6
7.3 Preparation of the calibration solutions for injection	7
7.4 Conditions for calibration runs	7
7.5 Measurement of elution volume	7
7.6 Plotting the calibration curve	7
<b>8 Sampling</b>	<b>8</b>
<b>9 Preparation for the test</b>	<b>8</b>
9.1 Preparation of the injection solution	8
9.2 Preparation of the apparatus	9
<b>10 Analytical parameters</b>	<b>9</b>
<b>11 Data acquisition and evaluation</b>	<b>10</b>
11.1 General	10
11.2 Calculation of the net chromatogram from the raw data	10
11.2.1 Determination of the baseline	10
11.2.2 Correction of the measured values and of the net chromatogram	10
11.2.3 Evaluation limits	10
11.3 Calculation of the average values	11
11.4 Calculation of the distribution curves	12
<b>12 Precision</b>	<b>12</b>
12.1 General	12
12.2 Repeatability	12
12.3 Reproducibility	13
<b>13 Test report</b>	<b>13</b>
13.1 General	13
13.2 General data on the equipment and settings	13
13.2.1 Data on the equipment used	13
13.2.2 Calibration	14
13.2.3 Evaluation	14
13.3 Special data on the sample	15
<b>Annex A (informative) Conversion of experimental parameters for variant column sizes</b>	<b>16</b>
<b>Annex B (informative) Example of a data sheet for a polymer standard</b>	<b>17</b>
<b>Annex C (informative) Explanations</b>	<b>19</b>

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## Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

The procedures used to develop this document and those intended for its further maintenance are described in the ISO/IEC Directives, Part 1. In particular, the different approval criteria needed for the different types of ISO documents should be noted. This document was drafted in accordance with the editorial rules of the ISO/IEC Directives, Part 2 (see [www.iso.org/directives](http://www.iso.org/directives)).

Attention is drawn to the possibility that some of the elements of this document may be the subject of patent rights. ISO shall not be held responsible for identifying any or all such patent rights. Details of any patent rights identified during the development of the document will be in the Introduction and/or on the ISO list of patent declarations received (see [www.iso.org/patents](http://www.iso.org/patents)).

Any trade name used in this document is information given for the convenience of users and does not constitute an endorsement.

For an explanation of the voluntary nature of standards, the meaning of ISO specific terms and expressions related to conformity assessment, as well as information about ISO's adherence to the World Trade Organization (WTO) principles in the Technical Barriers to Trade (TBT), see [www.iso.org/iso/foreword.html](http://www.iso.org/iso/foreword.html).

This document was prepared by Technical Committee ISO/TC 35, *Paints and varnishes*.

A list of all parts in the ISO 13885 series can be found on the ISO website.

Any feedback or questions on this document should be directed to the user's national standards body. A complete listing of these bodies can be found at [www.iso.org/members.html](http://www.iso.org/members.html).

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# Gel permeation chromatography (GPC) —

## Part 2:

## N,N-Dimethylacetamide (DMAC) as eluent

**WARNING** — This document can involve hazardous materials, operations or equipment. It does not purport to address all of the safety problems, if any, associated with its use. It is the responsibility of the user to establish appropriate safety and health practices and to ensure compliance with any national regulatory conditions.

### 1 Scope

This document specifies the determination of the molar-mass distribution and the average molar mass values  $M_n$  (number average) and  $M_w$  (weight average) of polymers that are soluble in DMAC (N,N-Dimethylacetamide) by gel permeation chromatography (GPC).

**NOTE** Also known as size exclusion chromatography (SEC).

Even though the chromatograms obtained show good repeatability, it is possible that this method cannot be used with certain polymer types because of specific interactions (e.g. adsorption) within the sample/eluent/column system.

The conditions specified in this document are not applicable to the GPC analysis of polymer samples with  $M_w$  values greater than  $10^6$  g/mol and/or polymers with elution limits outside the calibration range (see 7.6 and Annex C).

This document includes no correction method (e.g. for the elimination of peak broadening). If absolute molar mass values are required, an absolute method (e.g. membrane osmometry for  $M_n$  or light scattering for  $M_w$ ) can be used.

### 2 Normative references

The following documents, in whole or in part, are normatively referenced in this document and are indispensable for its application. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

ISO 1513, *Paints and varnishes — Examination and preparation of test samples*

ISO 4618, *Paints and varnishes — Terms and definitions*

ISO 15528, *Paints, varnishes and raw materials for paints and varnishes — Sampling*

### 3 Terms and definitions

For the purposes of this document, the terms and definitions given in ISO 4618 and the following apply.

ISO and IEC maintain terminological databases for use in standardization at the following addresses:

- ISO Online browsing platform: available at <https://www.iso.org/obp>
- IEC Electropedia: available at <http://www.electropedia.org/>

3.1  
gel permeation chromatography  
GPC

separation of molecules, mainly based on exclusion effects such as differences in the size and/or shape of molecules (size exclusion chromatography) or in charge (ion exclusion chromatography)

[SOURCE: ISO 13885-1:2020, 3.1]

3.2  
system peak

signal peculiar to the *gel permeation chromatography* (3.1) using a refractive index detector

Note 1 to entry: These signals appear at the total penetration limit of the columns and are not part of the sample, but of the overall system.

[SOURCE: ISO 13885-1:2020, 3.2]

4 Principle

The dissolved (molecularly disperse) molecules of a polymer sample are fractionated on a porous column material, with separation taking place according to the size of the molecule (or, more precisely, the polymer coil size which forms in this eluent). Small molecules diffuse into the pores of the column material more frequently and are therefore retarded more than large molecules. Thus, large molecules are eluted earlier, small molecules later. Under the test conditions given, the elution volume is solely a function of the coil size of the molecule.

The polymer content of a sample is determined, the sample is then diluted with eluent to give a concentration of less than 5 g/l and an aliquot of the diluted sample is injected into the GPC system. The concentration of the molecules eluted from the column is measured in order of decreasing coil size with a concentration-sensitive detector (typically a differential refractometer). With the aid of a calibration curve that has been determined for the particular GPC system, the relative molar-mass distribution, the relative quantities  $M_n$  and  $M_w$  and the heterogeneity or polydispersity  $M_w/M_n$  are calculated from the chromatogram obtained.

5 Apparatus

The apparatus shall consist of the components shown in Figure 1, which are described below.

All the components which come into contact with the eluent or the sample solution shall be resistant and shall not exhibit adsorption or memory effects in any form.

The eluent described in Clause 6 can cause corrosion in the case of long-term use. For this reason, it is necessary that high-quality steel or titanium is used for all components and that the individual modules are connected to one another by means of capillary tubes made of high-quality steel or titanium.

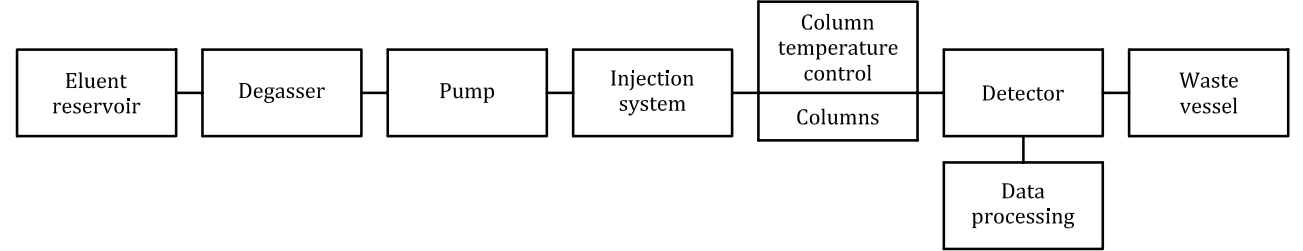


Figure 1 — Block diagram of a GPC apparatus



### 5.1 Eluent supply

The eluent reservoir shall adequately protect the eluent against external influences such as the atmosphere and light, if necessary by means of a blanket of inert gas above the liquid level.

The eluent reservoir shall contain a sufficient quantity of the eluent to bring the apparatus to equilibrium and to carry out several repeat analyses.

The eluent shall be degassed either before it is introduced into the reservoir, or by use of a device fitted between the reservoir and the pump, to prevent malfunctions of the pump or the formation of bubbles in the detector. The method of degassing used (e.g. bubble trap, online purging with helium or vacuum degassing) is open to choice.

### 5.2 Pump

The pump shall ensure that the eluent flow through the separation column is as smooth and pulse-free as possible. The flow rate shall be 1 ml/min (see [Annex A](#)). To fulfil these requirements, the pump shall operate at optimum efficiency at this flow rate.

The flow rate of the pump used shall have a variation of max. 0,1 %.

### 5.3 Injection system

The injection system serves to introduce a given amount of the sample solution into the eluent stream in a rapid and smooth fashion. This introduction may be carried out either manually or automatically.

If the introduction is carried out manually, ensure that the sample loop is filled completely with solvent before loading with the sample.

Memory effects from previous sample solutions in the injection system shall be avoided by design measures or by adequate flushing.

### 5.4 Separation columns

The apparatus shall have one or more columns connected in series and packed with spherical porous material, the diameter of the pores corresponding to the size of the polymer molecules being analysed.

The packing material typically consists of a styrene/divinylbenzene copolymer (S/DVB), produced by a special polymerization process, which swells only slightly in the eluent and therefore cannot deform under the pressure developed at the flow rate of 1 ml/min that is usually set.

In addition to these macroporous spherical S/DVB particles, packing materials based on other organic monomers or on silicon dioxide (silica) are also used. The criterion for their use is that no adsorptive interaction shall occur between their surface and the polymer molecules in the sample. Furthermore, the sample being analysed shall not be changed, either chemically or structurally, within the chromatographic system.

Certain polymers interact with the surface of the packing material (e.g. by adsorption), and other effects can sometimes interfere with the GPC separation mechanism. Details of such effects and notes on possible remedies are discussed in [Annex C](#). If it is intended to compare analyses of such polymers by different laboratories, the laboratories shall agree on details of the test conditions that are not covered by this document.

For good repeatability of test results, it is necessary to adhere to the minimum requirements specified below with regard to peak broadening (expressed in terms of a number of theoretical plates) and separation efficiency.

a) Number of theoretical plates

The number of theoretical plates,  $N$ , shall be determined, for the apparatus used per metre of column used, from the peak width at half height (see [Figure 2](#)). Inject up to 20 µl of a solution of methyl benzoate (concentration min. 5 g/l) on to the column (see [Annex A](#)) and evaluate the chromatogram obtained under the same conditions as are used for analysing polymers, using [Formula \(1\)](#):

$$N = 5,54 \times \left( \frac{V_e}{W_{1/2}} \right)^2 \times \frac{100}{L} \quad (1)$$

where

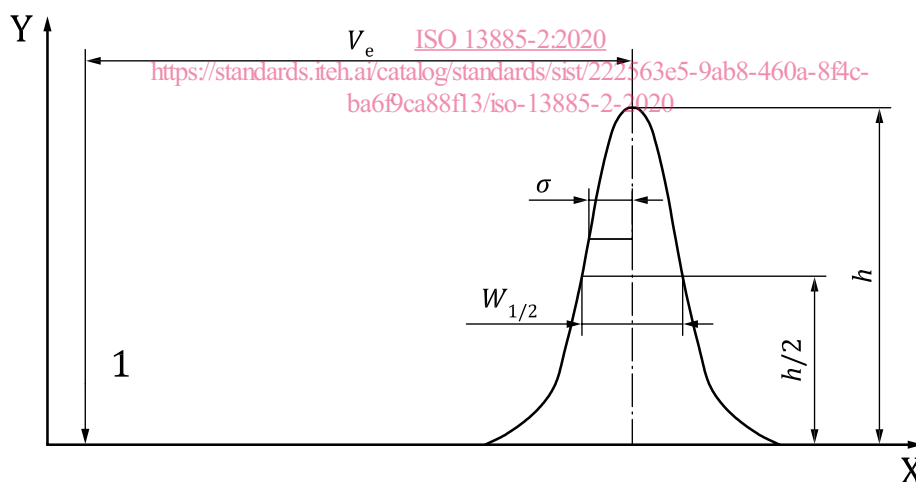
$V_e$  is the elution volume at the peak maximum;

$W_{1/2}$  is the peak width at half height (see [Figure 2](#)); the same units shall be used for  $V_e$  and  $W_{1/2}$ ;

$L$  is the length of the column (column combination), in centimetres.

Express as the result the number of theoretical plates per metre of column length. To conform to the requirements of this document, a column combination shall have at least 15 000 theoretical plates per metre.

NOTE See [Annex C](#) for tailing and fronting (asymmetry) of the peak used to calculate the plate count.



**Key**

- X elution volume
- Y peak intensity
- 1 injection
- $V_e$  elution volume measured at the peak maximum
- $W_{1/2}$  peak width at half height of the peak
- $h$  maximum peak height
- $\sigma$  standard deviation

**Figure 2 — Determination of the number of theoretical plates by the half-height method**

## b) Separation efficiency

To ensure adequate resolution, the  $\log_{10} M$  versus elution volume,  $V_e$ , calibration curve for the column combination used shall not exceed a specified gradient. For the purposes of this document, the relation given in [Formula \(2\)](#) shall apply in the area of the peak maximum for the polymer sample under investigation.

$$\frac{V_{e,M_x} - V_{e,(10 \times M_x)}}{A_c} > 6,0 \quad (2)$$

where

- $V_{e,M_x}$  is the elution volume for polymethyl methacrylate of molar mass  $M_x$ , in cubic centimetres;
- $V_{e,(10 \times M_x)}$  is the elution volume for 10 times that molar mass, in cubic centimetres;
- $A_c$  is the cross-sectional area of the column, in square centimetres;
- $M_x$  is the molar mass for polymethyl methacrylate; it shall be selected in such a way that the peak maximum for the polymer sample under investigation lies approximately half-way between these two elution volumes.

## 5.5 Column temperature control

Carry out the test at 80 °C. The temperature of the column shall not change by more than 1 °C during the analysis (see [Annex C](#)). The calibration and sample tests shall be carried out at the same temperature.

NOTE This specification of 80 °C is described in more detail in [Annex C](#).

## 5.6 Detector

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Use a differential refractometer detector. The cell volume shall not exceed 0,010 ml.

NOTE For the restriction to a single detector type, see [Annex C](#).

If copolymer samples or polymer blends are analysed, ensure that all the components give a similar response factor (ratio of detector signal to concentration of analyte in the eluate or, in the case of the differential refractometer, specific refractive index increment  $dn/dc$ ), i.e. the relationship of the response factors  $k_i$  or  $k_j$  for components  $i$  or  $j$  respectively is as follows:

$$0,2 \leq \frac{k_i}{k_j} \leq 5 \quad (3)$$

If the ratio of the response factors does not fall within this range in the analysis of a set of samples, a different detector or suitable combination of detectors may be used. If it is intended to compare the results obtained by different laboratories for such a set of samples, the type of detector shall be agreed upon. If a different detector is used, the reasons for using it shall be stated in the test report. See [Annex C](#).

The detector response obtained using the injection amounts specified in this document shall, at the lowest setting for electronic damping, exhibit a noise level of less than 1 % of the maximum height of the polymer peak. As the noise level is influenced by variations in pressure, temperature and flow rate, particularly in the differential refractometer, suitable measures are to be taken to maintain a constant temperature and to damp out pulses.

The signals from the detector are recorded by means of an electronic data system (see [Clause 11](#) for details).

## 6 Reagents

A solution of 5 g/l water-free lithium bromide LiBr in N,N-Dimethylacetamide (DMAC) [analytical quality, water content < 0,3 % (mass fraction)] is used as the eluent. DMAC is hygroscopic; the elution volume can depend on the water content of the eluent for individual polymers for conditions that are otherwise the same. The water content of the eluent shall not differ by more than 0,1 % for all analyses with an apparatus that are to be compared with one another — including calibration analyses.

Discard the eluent used to condition the columns or to perform the analyses, and do not return it to the eluent reservoir.

## 7 Calibration of the apparatus

### 7.1 General

The method is not an absolute one and requires calibration with commercially available unbranched polymethyl methacrylate standards that have been characterized by independent absolute methods. The results for samples of polymers with different chemical structures are therefore only comparable within groups of samples of the same type.

Calibrate the GPC apparatus with a series of unbranched polymethyl methacrylate samples of narrow molar mass distribution (see [Annex C](#)) and whose molar masses have been determined by independent, absolute methods. The result is a calibration curve for the evaluation of GPC analyses of unbranched polymethyl methacrylate samples. If this calibration curve is used to analyse samples of other compositions, containing molecules with other structures, the results shall be expressed as the “polymethyl methacrylate molar mass equivalent”

### 7.2 Specification for the calibration standards

NOTE The requirements for polymethyl methacrylate calibration standards are identical with the characteristics required in ISO 13885-1 for polystyrene standards.

The molar-mass distribution of the standards shall be narrower than the limits given below as a function of the molar mass at the peak maximum,  $M_p$ :

$M_p < 1\,000$ g/mol	$M_w/M_n \leq 1,3$
$1\,000$ g/mol $\leq M_p < 2\,000$ g/mol	$M_w/M_n \leq 1,2$
$2\,000$ g/mol $\leq M_p < 400\,000$ g/mol	$M_w/M_n \leq 1,1$
$400\,000$ g/mol $\leq M_p$	$M_w/M_n \leq 1,2$

The empirical peak-asymmetry factor for each chromatogram, calculated from the peak widths A and B at half height before and after the perpendicular through the peak maximum, shall lie in the range

$$\frac{A}{B} = 1,00 \pm 0,15 \quad (4)$$

The widths A and B shall be determined from electronically acquired data on peaks defined by at least 60 measuring points.

The following minimum requirements shall be fulfilled in the characterization of each individual polymethyl methacrylate standard used for calibration:

- at least one average molar mass value,  $M_n$ ,  $M_w$  or  $M_z$ , shall be determined by an absolute method;
- at least one method shall be used to determine the molar-mass distribution;