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**Corrosion of metals and alloys —  
Electrochemical test methods for  
high-temperature corrosion testing of  
metallic materials in molten salts**

*Corrosion des métaux et alliages — Méthodes d'essais  
électrochimiques de corrosion à haute température de matériaux  
métalliques dans des sels fondus*

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Published in Switzerland

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## Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

The procedures used to develop this document and those intended for its further maintenance are described in the ISO/IEC Directives, Part 1. In particular, the different approval criteria needed for the different types of ISO documents should be noted. This document was drafted in accordance with the editorial rules of the ISO/IEC Directives, Part 2 (see [www.iso.org/directives](http://www.iso.org/directives)).

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This document was prepared by Technical Committee ISO/TC 156, *Corrosion of metal and alloys*.

Any feedback or questions on this document should be directed to the user's national standards body. A complete listing of these bodies can be found at [www.iso.org/members.html](http://www.iso.org/members.html).

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## Introduction

Corrosion of metals and alloys in high-temperature molten salts is generally an electrochemical phenomenon. Therefore, the corrosion resistance and corrosion mechanism can be evaluated via a variety of electrochemical techniques. This document describes the apparatus and procedures for electrochemical measurements in high temperature molten salts. The closely related guidelines for potentiostatic and potentiodynamic polarization measurement in aqueous solutions are described in ISO 17475.

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# Corrosion of metals and alloys — Electrochemical test methods for high-temperature corrosion testing of metallic materials in molten salts

## 1 Scope

This document describes the general procedure for electrochemical measurements in high-temperature molten salts using potentiodynamic polarization measurements and electrochemical impedance spectroscopy (EIS), and describes the experimental apparatus.

## 2 Normative references

The following documents are referred to in the text in such a way that some or all of their content constitutes requirements of this document. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

ISO 8044, *Corrosion of metals and alloys — Vocabulary*

ASTM E220, *Standard method for calibration of thermocouples by comparison techniques*

ASTM E230, *Standard temperature-electromotive forces tables for standardized thermocouples*

ASTM E1350, *Standard test method for testing sheathed thermocouples prior to, during and after installation*

## 3 Terms and definitions

For the purposes of this document, the terms and definitions given in ISO 8044 apply.

ISO and IEC maintain terminology databases for use in standardization at the following addresses:

- ISO Online browsing platform: available at <https://www.iso.org/obp>
- IEC Electropedia: available at <https://www.electropedia.org/>

## 4 Apparatus

### 4.1 General

The test apparatus should consist of a heating unit, gas feeding unit, electrochemical measurement unit and electrochemical cell. A basic design of a closed, vertical, apparatus is shown in [Figure 1](#).

### 4.2 Heating unit or furnace

**4.2.1** The apparatus shall be composed of a heating unit or furnace that heats all the test pieces at a uniform temperature. The heating unit shall be equipped with a testing section capable of isolating the test pieces from outside air.

**4.2.2** The temperature distribution of the heating unit shall be characterized at the exposure temperature prior to testing to determine the width of the isothermal zone by the use of a

calibrated movable thermocouple. The heating unit or furnace shall maintain the temperature of the electrochemical cell within the permissible range given in [Table 1](#).

**4.2.3** Calibration of thermocouples shall be performed in accordance with ASTM E220, ASTM E230 and ASTM E1350. A representative thermocouple taken from a batch of wire can be calibrated. Thermocouples shall be recalibrated annually or at the beginning and the end of each test, if there is uncertainty about the thermocouple stability.

**Table 1 — Permissible tolerance of temperature of test pieces**

Temperature range, °C	≤300	300 to 600	600 to 800	800 to 1 000	1 000 to 1 200	>1 200
Temperature tolerance, °C	±2	±3	±4	±5	±7	By agreement

### 4.3 Gas feeding unit

**4.3.1** The gas supply system shall be capable of supplying test gases at a constant flow rate to the test piece chamber described in [Figure 1](#). The gas flow shall be monitored by a gas flow meter.

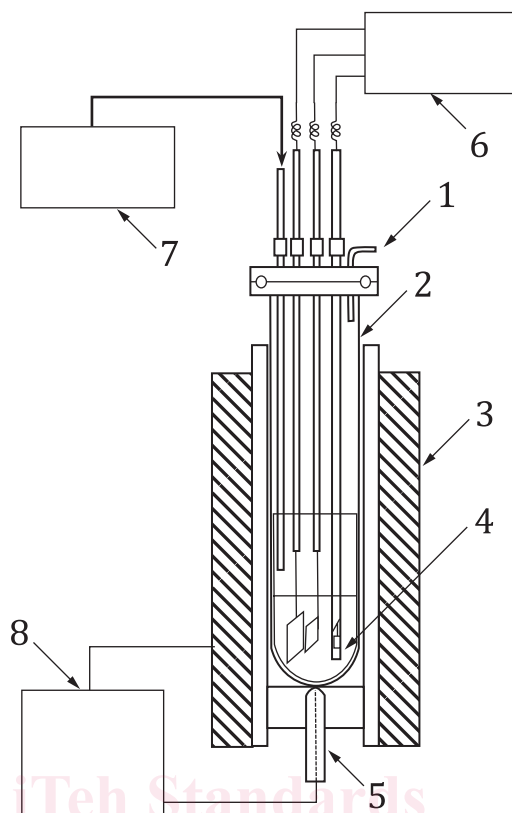
**4.3.2** When a humidifying regulator is used, it shall be capable of adjusting to the desired humidity. The gas line between the humidifying regulator and test chamber shall be kept above the dew point in order to avoid condensation. In the case that the gas is humidified, the water vapour content shall be measured. This can be achieved by, for example, the use of a hygrometer before the test chamber or by measuring the amount of water after condensation of the exhaust gases or by measuring the water consumption of the humidifier over the course of the test.

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**Key**

- 1 gas exhaust
- 2 test piece chamber
- 3 heating unit or furnace
- 4 electrochemical cell
- 5 thermocouple
- 6 electrochemical measurement unit
- 7 gas feeding unit
- 8 power/temperature control device

**Figure 1 — Schematic illustration of the entire electrochemical measurement unit for a high-temperature molten salt**

#### 4.4 Electrochemical measurement unit

**4.4.1** The potentiostat should be capable of controlling the electrode potential in the test to within  $\pm 1$  mV of a preset value. A scanning potentiostat is used for potentiodynamic measurements. For such measurements, the potentiostat shall be capable of automatically scanning the potential at a constant rate between preset potentials. For the EIS measurement, a potentiostat equipped with an electrochemical impedance measurement unit is used.

**4.4.2** The electrode potential-measuring instrument should have a high input impedance in the order of  $10^{11} \Omega$  to  $10^{14} \Omega$ , to minimize current drawn from the system during measurement. The sensitivity and accuracy of the instrument should be sufficient to detect a change of 1,0 mV.

**4.4.3** An appropriate current-measuring instrument with maximum error of 0,5 % is used.

## 4.5 Electrochemical cell

**4.5.1** A three-electrode system consisting of a working electrode (the metal to be polarized), a reference electrode for measuring the electrode potential and a counter (auxiliary) electrode shall be used. The test cell should incorporate inlet and outlet gas ports and a port for insertion of a temperature-measuring device. For the EIS measurement, a two-electrode system consisting of two identical working electrodes may be used instead. Examples of the three-electrode and two-electrode cells are shown in [Figure 2 a\)](#) and b), respectively.

**4.5.2** The test chamber shall not be composed of materials that react significantly with the gas environment or the corrosive substance during the test. If a reaction is unavoidable, the reaction shall be small enough to neglect a change in composition of the atmosphere. The crucibles used to hold the molten salt should not react with salt; for example, a high-purity alumina crucible shall be used for a sulfate or chloride bath.

**4.5.3** The working electrode is prepared from the test material of interest, usually in the form of a rod or sheet. The working electrode surface, except the test area, shall be protected by an insulating coating that does not react with the test salt and gas. The area immersed below the molten salt/gas interface shall be considered as the effective electrode surface area, if an appropriate insulating coating is not identified. In order to minimize the influence of the electrochemical reaction that occurs at the three-phase gas/electrode/molten salt interface, a flag-shaped working electrode is recommended. The flag-shaped electrode can be manufactured by punching or wire electric discharge machining. An example of the flag-shaped electrode is shown in [Figure 3](#). The level of the molten salt should not drop below the upper edge of the 'flag' section of the sample. The final finishing of the surface of the test pieces shall be performed with abrasives with a mean particle diameter of approximately 15 µm. This can be achieved by the use of abrasives according to [Table 2](#). If another surface finish is required by the parties involved, the surface finish condition shall be agreed and described. After ultrasonic degreasing in an appropriate solvent, such as isopropanol or ethanol, the test pieces shall be dried in hot air.

**Table 2 — Designation and mean diameter of particles for suitable coated abrasives**

Standard	Designation	Mean diameter µm
FEPA <sup>a</sup> 43-1 and FEPA 43-2 ISO 6344 series	P1200	15,3 ± 1,0
JIS R 6001-1	#1000	15,6 ± 1,0
ANSI B74.12	600	16,0
<sup>a</sup> Federation of European Producers of Abrasives		

**4.5.4** The reference electrode may be inserted directly into the molten salt. The type of reference electrode used will depend on the application, e.g. molten salt and temperature. An example of the reference electrode for molten chlorides, molten sulfates and molten carbonates is given in [Annex B](#).

**4.5.5** The counter electrode should be sufficiently corrosion resistant in the test environment. If the reaction at the counter electrode is likely to influence the measurement, the counter electrode shall be separated in a compartment. Pt, Au or carbon may be used as a counter electrode.