

ISO/CD 8217:2022(E)

ISO/DTR 18588:2023(E)

ISO TC 28/SC 4/WG 6

Date: 2023-03-27 05:11

Secretariat: AFNOR

Petroleum products — Characterisation of marine fuels by viscosity-gravity constant

Produits pétroliers — Caractérisation des combustibles pour la marine par la constante viscosité-gravité

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Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

The procedures used to develop this document and those intended for its further maintenance are described in the ISO/IEC Directives, Part 1. In particular, the different approval criteria needed for the different types of ISO documents should be noted. This document was drafted in accordance with the editorial rules of the ISO/IEC Directives, Part 2 (see www.iso.org/directives).

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This document was prepared by Technical Committee ISO/TC 28, *Petroleum and related products, fuels and lubricants from natural or synthetic sources*, Subcommittee SC 4, *Classifications and specifications*.

Any feedback or questions on this document should be directed to the user's national standards body. A complete listing of these bodies can be found at www.iso.org/members.html.

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Introduction

On 1 January 2020, MARPOL Annex-VI Regulation 14.1 reduced the maximum sulfur (S) content of marine fuels when operating outside an emission control area (ECA) from 3,50 % by mass to 0,50 % by mass. This resulted in a more diverse range of fuel blend formulations being offered to the marine market and an increasing interest by the marine industry in the ~~characterisation~~ characterization of petroleum-derived fuels not including fatty acid methyl ester (FAME), in terms of their aromatic or paraffinic nature.

~~This document was prepared in cooperation with ship owners, ship operators, shipping associations, national standards bodies, classification societies, fuel testing services, engine designers, fuel treatment equipment manufacturers, marine fuel suppliers, fuel additive suppliers and the petroleum industry. This document provides an indicator ~~to~~ for defining whether a petroleum-derived residual fuel which does not including ~~include~~ FAME, as supplied to a ship, is more aromatic or paraffinic in nature, ~~thus therefore providing additional onboard handling information of marine fuels irrespective of sulfur content.~~~~

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Characterisation Petroleum products — Characterization of marine fuels by viscosity gravity constant

1 Scope

This document provides ~~general guidance~~ best practices on the calculation and use of an informative indicator for the ~~characterisation~~ characterization of fossil residual marine fuels not containing FAME as defined in ISO 8217. ~~This document~~ is based on established guidance used by the petroleum industry to characterize petroleum fractions in terms of their paraffinic ~~to~~ naphthenic or aromatic nature, ~~by their viscosity-gravity constant~~.

2 Normative references

There are no normative references in this document.

3 Terms and definitions

For the purposes of this document, the following terms and definitions apply.

ISO and IEC maintain terminology databases for use in standardization at the following addresses:

- ISO Online browsing platform: available at <https://www.iso.org/obp>
- IEC Electropedia: available at <https://www.electropedia.org/>

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3.1

fatty acid methyl ester FAME

~~fatty acid methyl ester(s)~~ derived by (trans-)esterification of fats and vegetable oils

3.2

very low sulfur fuel oil VLSFO

marine fuel with a maximum sulfur content of 0,50 % by mass

3.3

high sulfur fuel oil HSFO

marine fuel with a sulfur content exceeding 0,50 % by mass

4 Characterisation Characterization of petroleum fuels

4.1 General

The ASTM publication “Characterisation and properties of petroleum fractions” by M.R. Riazi^[1] ~~refers to~~ cites two methods for the ~~characterisation~~ characterization of petroleum fractions:

- the Bureau of Mines correlation index (BMCI);

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— the viscosity-gravity constant (VGC).

4.2 BMCI (or CI)

4.2.1 General

BMCI, often also referred to as correlation index (CI) is used to characterize petroleum fractions/crude oils. Riazi^[1] reports that values of BMCI between 0 and 15 indicate a predominantly paraffinic oil. A value of BMCI greater than 50 indicates a predominance of aromatic compounds^[2]. According to the CI scale^[3], all n-paraffins have a CI value of 0, while cyclohexane (the simplest naphthene), has a CI value of 50, and benzene has a CI value of 100. Using the CI scale, petroleum fractions can be classified as described in Table 1.

Table 1 — Classification of petroleum fractions according to CI values

Table 1 — Classification of petroleum fractions according to CI values

Nature of petroleum fraction	CI value
Paraffinic (P)	<29,8
Naphthenic (N)	<57,0
Aromatic (A)	>75,0

4.2.2 Calculation of BMCI (or CI)

BMCI (or CI) relies on terms that are not available from the routine testing of marine fuels. It can be determined from Formula (1).

$$BMCI = \frac{48\,640}{T_b} + 473,7 \times SG - 456,8$$
$$I_{BMC} = \frac{48\,640}{T_b} + 473,7 \times G_s - 456,8 \quad (1)$$

where

— SG is the liquid specific gravity at 60 °F (see Formula (2));

— T_b is the volume average boiling point, expressed in degree kelvin (K) (see Formula (3)).

SG

I_{BMC} is the Bureau of Mines correlation index;

— G_s is the liquid specific gravity at 15.6 °C (60 °F) [see Formula (2)];

— T_b is the volume average boiling point, expressed in degree kelvin (K) [see Formula (3)].

G_s can be calculated from the fuel's density as follows:

$$SG = \frac{\left(\frac{\rho_{15}}{1\,000} - K \right)}{H} \quad G_s = \frac{\left(\frac{\rho_{15}}{1\,000} - K \right)}{H} \quad (2)$$

where

— ρ_{15} is the density at 15 °C, expressed in kilogram per cubic metre (kg/m³);

— H and K are variables the values of which are to be selected from Table 2.

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Table 2 — Values of H and K versus SG

where ρ_{15} is the density at 15 °C, expressed in kilogram per cubic metre (kg/m³); K value is the density at 15 °C, expressed in kilogram per cubic metre (kg/m³); H and K are variables the values of which are to be selected from Table 2.

Table 2 — Values of H and K versus G_s

G _s value		H value	K value
min.	max.		
≥0,79	<0,81	0,998 342	0,000 964
≥0,81	<0,83	0,998 492	0,000 842
≥0,83	<0,849 999 9	0,998 641	0,000 718
≥0,849 999 9	<0,875 000 1	0,998 761	0,000 616
≥0,875 000 1	<0,900 000 1	0,998 881	0,000 511
≥0,900 000 1	<1	0,999 011	0,000 394
≥1	<1,1	0,998 861	0,000 544

The volume average boiling point T_b can be calculated from Formulae Formula (3) and (4):

$$T_b = 9,3369 \times e^{(1,6514 \times 10^{-4} \times M + 1,4103 \times SG - 7,5152 \times 10^{-4} \times M \times SG)} \times M^{0,5369} \times SG^{-0,7276} \quad (3)$$

$$M = 223,56 \times v_{38}^{(-1,2435 + 1,1228 \times G_s)} \times v_{99}^{(3,4758 - 3,038 \times G_s)} \times SG^{-0,6665} \quad (4)$$

where

- T_b is the volume average boiling point, expressed in degree kelvin (K);
- M is a variable;
- v_{38} is the viscosity at 38 °C, expressed in square millimeter per second (mm²/s);
- v_{99} is the viscosity at 99 °C, expressed in square millimeter per second (mm²/s);
- T_b is the volume average boiling point, expressed in degree kelvin (K);
- G_s is the liquid specific gravity at 15,6 °C (60 °F) [see Formula (2)];
- e is the Euler number;
- M is a variable, the value of which can be calculated from Formula (4).

$$M = 223,56 \times v_{38}^{(-1,2435 + 1,1228 \times G_s)} \times v_{99}^{(3,4758 - 3,038 \times G_s)} \times G_s^{-0,6665} \quad (4)$$

where

- v_{38} is the viscosity at 38 °C, expressed in square millimetre per second (mm²/s);

ν_{99} is the viscosity at 99 °C, expressed in square millimetre per second (mm²/s).

Hence, the conversion of viscosity at 50 °C, ν_{50} , to viscosity at 38 °C and 99 °C is required essential. Formulae (5) and (6) can be used to convert viscosity at 50 °C to viscosity at 38 °C and 99 °C as follows:

$$\nu_{38} = e^{\left(\frac{\ln(\ln(\nu_{50}+0,7))+3,55 \times \ln\left(\frac{273,15+50}{273,15+38}\right)}{0,7} \right)}$$

$$\nu_{38} = e^{\left(\frac{\ln[\ln(\nu_{50}+0,7)]+3,55 \times \ln\left(\frac{273,15+50}{273,15+38}\right)}{0,7} \right)} \quad (5)$$

$$\nu_{99} = e^{\left(\frac{\ln(\ln(\nu_{50}+0,7))+3,55 \times \ln\left(\frac{273,15+50}{273,15+99}\right)}{0,7} \right)}$$

$$\nu_{99} = e^{\left(\frac{\ln[\ln(\nu_{50}+0,7)]+3,55 \times \ln\left(\frac{273,15+50}{273,15+99}\right)}{0,7} \right)} \quad (6)$$

where

ν_{50} is the viscosity at 50 °C, expressed in square millimetre per second (mm²/s).

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