



Designation: **E2209 – 02 (Reapproved 2006)^{ε2} E2209 – 13**

Standard Test Method for Analysis of High Manganese Steel Using ~~by Spark Atomic~~ Emission Spectrometry¹

This standard is issued under the fixed designation E2209; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon (ϵ) indicates an editorial change since the last revision or reapproval.

^{ε1} NOTE—Updated Section 2 Referenced Documents in December 2006.

^{ε2} NOTE—Added research report footnote to Section 16 editorially in April 2009.

1. Scope

1.1 This test method provides for the analysis of high manganese steel by spark atomic emission spectrometry ~~using the point-to-plane technique~~ for the following elements in the ~~concentration~~ ranges shown:

Elements Elements	Concentration Range, % Composition Range, %
Aluminum (Al)	0.02 to 0.15
Carbon (C)	0.3 to 1.4
Chromium (Cr)	0.25 to 2.00
Manganese (Mn)	8.0 to 16.2
Molybdenum (Mo)	0.03 to 2.0
Nickel (Ni)	0.05 to 4.0
Phosphorus (P)	0.025 to 0.06
Silicon (Si)	0.25 to 1.5

NOTE 1—The ranges represent the actual levels at which this method was tested.² These ~~concentration~~ composition ranges can be extended to higher ~~concentrations~~ compositions by the use of suitable reference materials. Sulfur is not included because differences in results between laboratories exceeded acceptable limits at all analyte levels.

1.2 *This test method may involve hazardous materials, operations, and equipment. This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.*

2. Referenced Documents

2.1 *ASTM Standards:*³

A128/A128M [Specification for Steel Castings, Austenitic Manganese](#)

[E135 Terminology Relating to Analytical Chemistry for Metals, Ores, and Related Materials](#)

¹ This test method is under the jurisdiction of ASTM Committee E01 on Analytical Chemistry for Metals, Ores, and Related Materials and is the direct responsibility of Subcommittee E01.01 on Iron, Steel, and Ferroalloys.

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² Supporting data have been filed at ASTM International Headquarters and may be obtained by requesting Research Report RR:E01-1035.

³ For referenced ASTM standards, visit the ASTM website, www.astm.org, or contact ASTM Customer Service at service@astm.org. For *Annual Book of ASTM Standards* volume information, refer to the standard's Document Summary page on the ASTM website.

~~E158 Practice for Fundamental Calculations to Convert Intensities into Concentrations in Optical Emission Spectrochemical Analysis (Withdrawn 2004)⁴~~

~~E172 Practice for Describing and Specifying the Excitation Source in Emission Spectrochemical Analysis (Withdrawn 2001)⁴~~

E305 Practice for Establishing and Controlling Atomic Emission Spectrochemical Analytical Curves

E353 Test Methods for Chemical Analysis of Stainless, Heat-Resisting, Maraging, and Other Similar Chromium-Nickel-Iron Alloys

E406 Practice for Using Controlled Atmospheres in Spectrochemical Analysis

~~E876 Practice for Use of Statistics in the Evaluation of Spectrometric Data (Withdrawn 2003)⁴~~

E1019 Test Methods for Determination of Carbon, Sulfur, Nitrogen, and Oxygen in Steel, Iron, Nickel, and Cobalt Alloys by Various Combustion and Fusion Techniques

E1059 Practice for Designating Shapes and Sizes of Nongraphite Counter Electrodes (Withdrawn 2013)⁴

E1329 Practice for Verification and Use of Control Charts in Spectrochemical Analysis

E1601 Practice for Conducting an Interlaboratory Study to Evaluate the Performance of an Analytical Method

E1806 Practice for Sampling Steel and Iron for Determination of Chemical Composition

2.2 Other Document:

~~ASTM Manual on Presentation of Data and Control Chart Analysis, ASTM MNL 7A, seventh revision, 2002; MNL 7~~ [ASTM Manual on Presentation of Data and Control Chart Analysis, 8th Edition, 2010.](https://standards.iteh.ai/)

3. Terminology

3.1 For definition of terms used in this method, refer to Terminology **E135**.

4. Summary of Test Method

4.1 A controlled discharge is produced between the flat surface of the specimen and the counter electrode. The radiant energies of selected analytical lines are converted into electrical energies by ~~photo-multiplier~~ photomultiplier tubes and stored on capacitors. This discharge is terminated after a fixed ~~exposure~~ integration time. At the end of the exposure period, the charge on each capacitor is ~~measured~~, measured and converted to ~~concentration~~ mass fraction percent.

5. Significance and Use

5.1 The chemical composition of high manganese steel alloys must be determined accurately to ensure the desired metallurgical properties. This procedure is suitable for manufacturing control and inspection testing.

6. Interferences

6.1 Interferences may vary with spectrometer design and excitation characteristics. Direct spectral interferences may be present on one or more of the wavelengths listed in a method. Frequently, these interferences may be determined and proper corrections made by the use of various reference materials. The composition of the sample being analyzed should match closely the composition of one or more of the reference materials used to prepare and control the calibration curve that is employed. Alternatively, mathematical corrections may be used to solve for interelement effects (refer to ~~Practice effects~~, **E158**). Various mathematical correction procedures are commonly utilized. Any of these are acceptable that will achieve analytical accuracy equivalent to that provided by this method.

7. Apparatus

7.1 Sample Preparation Equipment:

7.1.1 *Sample Mold*, to produce chilled cast samples approximately 38 mm (1½ in.) in diameter that are homogeneous, free of voids or porosity in the region to be excited, and representative of the material to be analyzed. Refer to Practice **E1806** for steel sampling procedures.

7.1.2 *Immersion Sampler*, to take a sample from the bath or from the metal stream when pouring can be used. The sample should produce a sample of the same dimensions as listed in **7.1.1**.

7.1.3 *Surface Grinder or Sander With Abrasive Belts or Disk*, capable of providing a flat uniform surface on the reference materials and specimens. The following table shows the various methods of sample preparation used in the Inter-Laboratory Study (ILS):

Type of Grinding Preparation	Belt and/or Disk
Grinding Medium	Aluminum Oxide, Zirconium Oxide
Grit of Grinding Medium	36 to 180

NOTE 2—Silicon carbide grinding medium may be used but it was not utilized by the laboratories in the Inter-Laboratory Study (ILS).

⁴ The last approved version of this historical standard is referenced on www.astm.org.