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ISO 1213-2:2024

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Foreword

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The procedures used to develop this document and those intended for its further maintenance are described in the ISO/IEC Directives, Part 1. In particular, the different approval criteria needed for the different types of ISO document should be noted. This document was drafted in accordance with the editorial rules of the ISO/IEC Directives, Part 2 (see www.iso.org/directives).

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This document was prepared by Technical Committee ISO/TC 27, *Coal and coke*, Subcommittee SC 1, *Coal preparation: Terminology and performance*.

This third edition cancels and replaces the second edition (ISO 1213-2:2016), of which it constitutes a minor revision. The changes are as follows:

the title has been updated to specify coal and coke; 2:2024

— solid mineral fuels have been replaced by either coal and coke or, in some cases, just coal or just coke.

A list of all parts in the ISO 1213 series can be found on the ISO website.

Any feedback or questions on this document should be directed to the user's national standards body. A complete listing of these bodies can be found at <u>www.iso.org/members.html</u>.

Coal and coke — Vocabulary —

Part 2: Terms relating to sampling, testing and analysis

1 Scope

This document defines terms commonly employed in the sampling, testing and analysis of coal (3.39) and coke (3.42).

Alternative names are given for several terms. In some cases, the use of the alternative name is deprecated (as indicated).

2 Normative references

There are no normative references in this document.

3 Terms and definitions

ISO and IEC maintain terminological databases for use in standardization at the following addresses:

- ISO Online browsing platform: available at https://www.iso.org/obp
- IEC Electropedia: available at https://www.electropedia.org/

3.1

abrasion

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loss of material from particle surfaces of a *coal* (3.39) or *coke* (3.42) sample, or from other surfaces in contact with the particles, caused by friction between contacting surfaces

3.2

abrasion index

total mass lost by the *abrasion* (3.1) of four carbon steel blades when rotated in a specified mass of *coal* (3.39) under specified conditions

Note 1 to entry: Expressed in milligrams of metal lost per kilogram of a coal and coke sample.

3.3

abrasion value

resistance to *abrasion* (3.1) of the *coke* (3.42) after reaction with carbon dioxide in the *CRI* (3.43) test, measured as the mass fraction expressed as a percentage of a sample passing through a 0,5 mm sieve after tumbling under conditions specified

3.4

adiabatic calorimeter

calorimeter that adjusts its jacket temperature constantly to be identical to bomb temperature, thereby preventing heat losses

Note 1 to entry: The inner calorimeter chamber and the jacket exchange no energy because the water temperature in both is identical during the test. The water in the external jacket is heated or cooled to match the temperature change in the calorimeter proper.

3.5

accuracy

closeness of agreement between an observation and the "true" value

Note 1 to entry: The accuracy of a result should not be confused with its precision. A result may be precise but it is only accurate when it is free of *bias* (3.18).

3.6

adventitious ash

DEPRECATED: extraneous ash

ash arising from *mineral matter* (3.136) associated with, but not inherent in a *coal* (3.39) or *coke* (3.42) sample

3.7

air-dried basis

means of expressing an analytical result based on the condition in which a *coal* (3.39) or *coke* (3.42) sample is in equilibrium with atmospheric humidity

3.8

air-drying

process of bringing the mass fraction of the sample near to equilibrium with the atmosphere, in the area in which further reduction of the sample are to take place

Note 1 to entry: The coal or coke sample in this state is composed of absorbed moisture, mineral matter and organic matter.

3.9

anthracite

coal (3.39) of high *rank* (3.174), with a low *volatile matter* (3.239) content and a semi-metallic lustre, and which does not soften or swell when heated

3.10

apparent relative density

ratio of the mass of a *coal* (3.39) (lump sample) to the mass of an equal volume of water (at the same temperature), inclusive of any voids within the fuel subjected to the test

Note 1 to entry: The apparent relative density should not be confused with the *bulk density* (3.25).

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3.11 ash ps://standards.iteh.ai/catalog/standards/iso/c8c5418c-16a7-4262-afdf-8c1cdb20b0bb/iso-1213-2-2024

residue obtained by incineration of a *coal* (3.39) or *coke* (3.42) sample under specified conditions

3.12

ash analysis

analysis of ash (3.11) for its elemental composition

Note 1 to entry: The elements usually determined are silicon, aluminium, iron, magnesium, manganese, titanium, calcium, sodium, potassium, phosphorus and sulfur, and these are usually expressed as oxides.

3.13

ash fusibility

characteristic physical state of the ash (3.11) obtained by heating under specified conditions

Note 1 to entry: Ash fusibility is determined under either *oxidizing atmosphere* (3.61) or *reducing atmosphere* (3.176) conditions.

Note 2 to entry: See also deformation temperature (3.58), sphere temperature (3.215), hemisphere temperature (3.98) and flow temperature (3.75).

3.14

ash viscosity

measure of the resistance to flow of ash (3.11) in the fused state

3.15

as received basis

as sampled basis

means of expressing an analytical result based on the condition where total moisture (3.232) is included

3.16

base/acid ratio

ratio of the mass of basic oxides (iron(III) oxide, calcium oxide, magnesium oxide, disodium oxide and dipotassium oxide) to the mass of acidic oxides (silica, aluminium oxide and titanium (IV) oxide) in ash (3.11)

Note 1 to entry: This ratio can be used in the determination of the *fouling factor* (3.81) and the slagging factor.

3.17

batch

quantity of *coal* (3.39) or *coke* (3.42) produced at one time under relatively uniform conditions

3.18

bias

systematic *error* (3.68) which leads to the average value of a series of results being persistently higher or persistently lower than those obtained using a reference sampling method

Note 1 to entry: Bias is the total systematic error as contrasted to random error. There may be one or more systematic error components contributing to the bias. A larger systematic difference from the accepted reference value is reflected by a larger bias value.

3.19

bias of scale

bias (3.18) that is constant and independent of the range of values measured

3.20

bituminous coal

general descriptive term for coal (3.39) of rank (3.174) between anthracite (3.9) and brown coal and lignite (3.24)

Note 1 to entry: The vitrinites in all coals in the bituminous range melt and form a coke when the coal is heated above 400 °C in the absence of air.

Note 2 to entry: In some countries, coals of rank immediately below that of bituminous coal are referred to as subbituminous coals.

3.21

blast furnace coke

strong, *large coke* (3.116) for use in blast furnaces

Note 1 to entry: Blast furnace coke is generally produced from blends of *bituminous coals* (3.20), which may incorporate additives.

Note 2 to entry: Blast furnace coke usually has a low reactivity to carbon dioxide.

3.22

breakage

particle size reduction (3.155) resulting from impact and/or compression

3.23

breeze

undersize after separating the smallest size of graded coke (3.91)

Note 1 to entry: Breeze is usually less than 10 mm in size.

3.24

brown coal and lignite

coals (3.39) of low rank (3.174) characterized by high inherent moisture, high volatile matter (3.239) and low calorific value

Note 1 to entry: In some countries, the terms are used to describe all low-rank coals up to *bituminous coals* (3.20). In other countries, the coals at the higher end of the range are referred to as sub-bituminous coals.

3.25

bulk density

mass of a portion of a *coal* (3.39) or *coke* (3.42) sample divided by the volume of the container which is filled by that portion under specified conditions

Note 1 to entry: Bulk density values can have range and may depend on previous handling, time and weather. The values on stockpiles can also vary from loose free fall situations to compacted filled by that portion under specified conditions.

3.26

bulk sample

sample of large mass, taken in a particular operation for a specific reason such as for *float sink analysis* (3.78)

3.27

caking of coal

property of *coal* (3.39) when heating without access of air to a plastic condition with formation of the connected non-volatile residue

3.28

caking index

measure of the caking power of a coal in terms of the mechanical strength (3.132) of the coke (3.42) obtained by carbonization, under specified conditions, of an intimate mixture of the coal (3.39) and standard anthracite (3.9)

3.29

Document Preview calorific value gross at constant volume

absolute value of the specific energy of combustion, in joules, for unit mass of a *coal* (3.39) or *coke* (3.42)sample burned in oxygen in a calorimetric bomb under the conditions specified

Note 1 to entry: The products of combustion are assumed to consist of gaseous oxygen, nitrogen, carbon dioxide and sulfur dioxide, of liquid water (in equilibrium with its vapour) saturated with carbon dioxide under the conditions of the bomb reaction, and of solid ash, all at the reference temperature.

Note 2 to entry: Equipment such as Adiabatic and or Isothermal bomb calorimeters are used to determine this result.

3.30

calorific value net at constant volume

absolute value of the specific energy of combustion, in joules, for unit mass of the *coal* (3.39) or *coke* (3.42)sample burned in oxygen under conditions of constant volume and such that all the water of the reaction remains as water vapour (in a hypothetical state at 0,1 Mpa), the other products being as for the gross calorific value all at the reference temperature

Note 1 to entry: The net calorific value at constant volume is the negative value of the net specific energy of combustion.

3.31

calorific value net at constant pressure

absolute value of the specific heat (enthalpy) of combustion in joules, for unit mass of the *coal* (3.39) or *coke* (3.42) sample burned in oxygen at constant pressure under such conditions that all the water of the reaction products remains as water vapour (at 0,1 Mpa), the other products being as for the gross calorific value, all at the reference temperature

3.32 carbominerite collective term for inter growths of minerals and *macerals* (3.121)

Note 1 to entry: The various types of carbominerite with their compositions are given in Table 1.

Туре	Volume fraction % of minerals	
Carbargilite	20 to 60, clay minerals	
Carbopyrite	5 to 20, sulfides	
Carbankerite	20 to 60, carbonates	
Carbosilicite	20 to 60, quartz	
Carbopolyminerite ^a	20 to 60, various minerals	
 The term is used also for carbopolyminerite containing a maximum of % of mineral matter, provided that sulfides form a substantial part of the mineral matter. 		

Table 1 — Types and compositions of carbominerite

3.33

carbon in mineral matter

carbon in the mineral matter carbonates of a *coal* (3.39) or *coke* (3.42) sample

3.34

carboxyreactivity

rate of reaction of a *coal* (3.39) or *coke* (3.42) sample with carbon dioxide under specified conditions

3.35

channel sample

sample of raw *coal* (3.39) and associated inorganic material taken by removing a channel of even cross-section from the seam

Note 1 to entry: Where the full section of the seam is not accessible or not required, this term may refer to a sample taken either from a specifically defined portion of the seam, or from the floor to roof as mined or exposed.

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3.36 chars://standards.iteh.ai/catalog/standards/iso/c8c5418c-16a7-4262-afdf-8c1cdb20b0bb/iso-1213-2-2024

solid, partially or non-agglomerated carbonaceous material produced by the pyrolysis of a *coal* (3.39) sample

3.37

chute

inclined trough for conveying *coal* (3.39) or *coke* (3.42) to a lower level

3.38

clinkering

aggregation of particles of *ash* (3.11) after it has melted during the course of combustion of *coal* (3.39) sample or during gasification

Note 1 to entry: The aggregated particles may include small amounts of unburnt *coal* (<u>3.39</u>).

3.39

coal

combustible sedimentary rock formed from altered plant remains consolidated under superimposed strata

Note 1 to entry: The characteristics of different coals are due to differences in source plant material, in the conditions and the degree of change that the material has undergone in its geological history, and in the range of impurities present. Coals can be characterized macroscopically by their lithotype composition and microscopically by their maceral and *microlithotype* (3.123) compositions.

3.40

coalification

process by which accumulated plant matter is compacted and transformed into coal (3.39)

3.41

coefficient of variation

standard deviation (3.216), expressed as a percentage of the absolute value of the arithmetic mean

Note 1 to entry: to entry:
$$CV = \frac{s}{\overline{x}} \times 100$$

where

- *s* is the value of standard deviation;
- \overline{x} is the arithmetic mean;

Note 2 to entry: *CV* is normally denoted as *v*.

3.42

coke

solid, agglomerated carbonaceous residue produced by the pyrolysis of *coal* (3.39) in the absence of air

3.43

coke reactivity index

CRI test

mass fraction loss, expressed as a percentage of *coke* (3.42) after reaction with carbon dioxide and carbon monoxide under specified conditions

3.44

coke strength after reaction

CSR

strength of *coke* (3.42) after reaction with carbon dioxide and carbon monoxide in the *CRI* (3.43) test, measured as mass fractions expressed as a percentage retained on either a 10,0 mm or a 9,5 mm sieve after tumbling under specified conditions

3.45

combustible matter

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theoretical state of *coal* ($\underline{3.39}$) or *coke* ($\underline{3.42}$) without moisture and *mineral matter* ($\underline{3.136}$) other than *pyritic sulfur* ($\underline{3.170}$) and sulfidic sulfur [SO 1213-2:2024]

3.46 standards.iteh.ai/catalog/standards/iso/c8c5418c-16a7-4262-afdf-8c1cdb20b0bb/iso-1213-2-2024

combustible sulfur

sulfur which reacts with oxygen when a *coal* (3.39) or *coke* (3.42) sample is burnt under specified controlled conditions

Note 1 to entry: Most of the reacted sulfur reports as SO_2 in the chimney gas, but under certain conditions, some of the sulfur is captured by alkaline minerals in the ash

3.47

common sample

sample collected for more than one intended use

3.48

complete seam profile sample for each bench

collective designation of the coal samples taken separately from each coal bench and band of the tested seam or a part of it which is a section of a thick seam

3.49

constant mass division

method of increment or *sample division* (3.194) in which the portions retained from individual *increments* (3.106), *partial samples* (3.153) or *gross samples* (3.94) are of uniform mass

3.50

continuous sampling

taking of a sample from each consecutive *sub-lot* (3.221) so that *increments* (3.106) are taken at uniform intervals whenever the *coal* (3.39) or *coke* (3.42) is handled at the point of sampling

3.51

correlation coefficient

measure of the degree of correlation between the members of paired sets

3.52

core sample

cylindrical sample of the whole or part of a *coal* (<u>3.39</u>) seam obtained from drilling using a coring barrel

Note 1 to entry: The diameter of the core may vary from 50 mm to 1000 mm depending on the reason for which the sample is required. However, 50 mm to 200 mm is the most common core diameter range.

3.53

crucible swelling number

CSN

number which defines, by reference to a series of standard profiles, the size and shape of the residue obtained when a specified mass of *coal* (3.39) is heated in a covered crucible under specified conditions

Note 1 to entry: ASTM Standards use the term free swelling index (FSI) for this test.

3.54

crush (verb)

action of reducing the particle size of a sample to produce particles at the required *nominal top size* (3.144) required

Note 1 to entry: See also grind (3.93).

3.55

cut coke

screened *coke* (3.42) from which the oversize has been reduced by mechanical means and rescreened

3.56

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cutter mechanical sampling device which extracts *increments* (<u>3.106</u>)^{4262-afdf-8c1cdb20b0bb/iso-1213-2-2024}

3.57 deformation interval softening interval

interval between the *deformation temperature* (3.58) and the *hemisphere temperature* (3.98)

3.58

deformation temperature

temperature at which deformation of a test piece prepared from *ash* (3.11), by a specified procedure, occurs

Note 1 to entry: When using cylindrical (or cubicoidal) test pieces, a change of the surface and the rounding of the edges at the rim or corner.

Note 2 to entry: When using pyramidal test pieces, the rounding of the tip of the test piece. Shrinkage or distortion of the test piece, or rounding of cracks and fins, are not criteria for deformation and should be ignored if the tip and edges remain sharp.

3.59 dial divisions per minute ddpm

measure of stirrer rotation rate, in the Gieseler Plastometer method

Note 1 to entry: There are 100 dial divisions for each full 360° rotation of the stirrer. The *fluidity* (3.77) result is expressed as total dial division turned by the stirrer in a one-minute time period