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Standard Guide for Rapid Screening of Vegetation for Radioactive Strontium Aerial Deposition¹

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1. Scope

1.1 This guide provides a rapid procedure by which vegetation samples may be screened for surface contamination of radioactive strontium (^{89}Sr and ^{90}Sr , collectively referred to as radiostrontium) following an airborne radioactive dispersal event. It provides a conservative estimate of radiostrontium deposition that can be used by decision makers for immediate actions prior to obtaining definitive results from a fixed laboratory asset.

1.2 Insoluble forms of radiostrontium, such as the strontium (^{90}Sr) titanate (SrTiO_3) used in radio-isotope thermal-electric generators (RTGs), will not be measured by this method.

1.3 Non-SI units are used in the calculations of this guide for ease of use during the emergency phase of an event. The instrumentation used typically provides count rates in counts per minute (cpm) rather than per second (s^{-1} , the SI unit), thus activity is expressed in dpm (decays per minute) rather than Bq. Additionally, US EPA protective guidelines for surface contamination are expressed in $\text{dpm}/100\text{ cm}^2$.

1.4 *This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.*

2. Referenced Documents

2.1 *ASTM Standards:*²

D1129 Terminology Relating to Water

D1193 Specification for Reagent Water

D3648 Practices for the Measurement of Radioactivity

¹ This guide is under the jurisdiction of ASTM Committee D19 on Water and is the direct responsibility of Subcommittee D19.04 on Methods of Radiochemical Analysis.

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² For referenced ASTM standards, visit the ASTM website, www.astm.org, or contact ASTM Customer Service at service@astm.org. For *Annual Book of ASTM Standards* volume information, refer to the standard's Document Summary page on the ASTM website.

2.2 *Other Documents:*

EPA Protective Action Guidelines³

3. Terminology

3.1 *Definitions*—See Terminology D1129 for terms related to water.

3.2 *Definitions of Terms Specific to This Standard:*

3.2.1 *ROI, n*—region of interest; the span of channels, or region, in the spectrum in which the counts due to a specific radioisotope appear on a functioning, calibrated liquid scintillation spectrometry system.

3.3 *Acronyms:*

3.3.1 *RLS, n*—rapid liquid sampler

3.3.2 *SPE, n*—solid phase extraction

4. Summary of Guide

4.1 Vegetation is collected from an area equivalent to 100 cm^2 . The leafy material is shaken with $\text{pH} = 2$ water to solubilize radiostrontium deposited on the vegetation. The radiostrontium is then extracted onto a solid phase extraction (SPE) disk for counting and quantification.

4.2 Testing has shown that chemical recoveries for ^{90}Sr under these extraction conditions average 30–50 %, with similar recoveries expected for ^{89}Sr .

4.3 A counting efficiency of 80–85 % can be achieved using liquid scintillation spectrometry.

4.4 Quantification may also be accomplished using a simple gas-filled count rate meter (a “pancake probe”); however the presence of other beta-emitting radionuclides can not be discerned when using such a non-discriminatory detector.

5. Significance and Use

5.1 Strontium-90 is a major component of nuclear waste and is also a potential radioisotope for use as a weapon of mass destruction in a radiological dispersal device. It is a beta-emitting radioisotope with moderate half-life (~30 years).

³ Available from United States Environmental Protection Agency (EPA), Ariel Rios Bldg., 1200 Pennsylvania Ave., NW, Washington, DC 20460, <http://www.epa.gov>.

Strontium-89 is also a beta emitting radionuclide, but with a half-life of only ~50 days it is not usually present in significant quantities. If ingested the radiostrontium may deposit in the bone of an individual and thus can contribute a significant radiological dose to an affected person.

5.2 Following an explosion in which radioactive material was present, the potential exists for the material to become airborne. It will quickly attach to atmospheric particles and be deposited on surfaces as the plume passes. This guide provides a rapid procedure by which vegetation can be screened to determine if radiostrontium is present and to provide a conservative estimate of its deposition on vegetation.

5.3 This guide is intended to be used in a field portable lab, or if needed, can be performed completely in the field; therefore no hazardous chemicals are required to complete the analysis. However, an option for the use of acid in certain steps is documented in this guide.

5.4 This guide is not intended to be used for screening food products or animal feed following an accident or incident.

6. Interferences

6.1 Liquid scintillation (LS) counting is the preferred method of counting. Because this is a screening method, chemical decontamination from other beta-emitting isotopes is not as rigorous as found in traditional laboratory methods. Careful evaluation of the liquid scintillation spectrum may provide indications of the presence of contaminants.

6.2 Lead and radium are known to also be retained by the SPE disk under these extraction conditions and do occur naturally as part of fallout deposition, although at low concentrations compared to the radiostrontium surface contamination of concern. Careful evaluation of the liquid scintillation spectrum may provide indications of the presence of alpha peaks (Ra isotopes) or other beta continuums (Pb or other isotopes).

6.3 Yttrium-90, the daughter of ^{90}Sr and also a beta emitter, will be partially retained by the SPE disk using this method. If permissible, and with appropriate personal protective equipment and engineering controls, the SPE disk can be washed with 15–20 mL of 2–3 M nitric acid. Note the time the wash finished going through the SPE disk as the start time for ^{90}Y ingrowth.

6.4 Particles containing ^{90}Sr (such as from an RTG explosion), as opposed to particles with radiostrontium adsorbed on the surface, may not be sufficiently attacked by the weak acid solution and so will not be detected by this procedure. However, particles containing ^{90}Sr would only be expected to be encountered very close to the site of the initial explosion.

7. Instrumentation

NOTE 1—See Practices **D3648** for a description of these detector systems.

7.1 *Liquid Scintillation Spectrometer*—Commercial systems are available that are reasonably portable, but may require manual sample changing. The system should have a stable background in the counting region of interest and reproducible detection efficiency for ^{90}Sr on the SPE disk. The spectrometer

should allow the option of looking at the entire counting spectrum so that evaluation of other interferences may be completed. Automatic discrimination of alpha and beta particles is desirable but not required.

7.2 *Gas Filled Count Rate Meter*—A non-discriminating beta/gamma radiation detector, often called a “pancake probe,” consisting of a gas-filled detector coupled to an electronics package that provides count rate information.

8. Reagents and Materials

8.1 *Purity of Reagents*—Reagent grade chemicals shall be used in all tests. Unless otherwise indicated, it is intended that all reagents shall conform to the specifications of the Committee on Analytical Reagents of the American Chemical Society, where such specifications are available.⁴

8.2 *Purity of Water*—Unless otherwise indicated, references to water shall be understood to mean Type I water as defined in Specification **D1193**.

8.3 *Solid Phase Extraction (SPE) Disk*—47 mm in diameter, impregnated with strontium selective resin,⁵ supplied as either loose SPE disks or in the Rapid Liquid Sampler (RLS) housing.

8.4 *pH = 2 Water*—Water acidified to a pH of 2 with nitric acid. Concentrated nitric acid may be added dropwise, mixing and monitoring the pH of the water until it begins to approach 2–4, then carefully add a more dilute nitric solution (1–2 M) until a pH of 2 is obtained. If the pH is overshoot, add a dilute Na_2CO_3 , NaHCO_3 or NaOH solution to a final pH of 2.

8.5 *Nitric Acid*—(HNO_3) concentrated and diluted solution.

8.6 *Liquid Scintillation (LS) Cocktail*—Any commercial cocktail that provides acceptable counting efficiency and backgrounds in the ROI for $^{89+90}\text{Sr}$, many are deemed environmentally friendly thus do not require disposal as a mixed waste.

8.7 *Strontium-90 Calibration Solution*—A solution whose activity is traceable to a national standards body, such as NIST or NPL, in dilute nitric acid. Alternatively, ^{89}Sr may be used but the known activity must be decay corrected to the time of use.

9. Apparatus

9.1 Wide mouth plastic bottle, 1–2 L preferred or 1–2 qt resealable plastic bags (that is, bags with a zipper-locking seal).

9.2 Device for measuring or delivering approximately 50-mL aliquots of acidified water.

⁴ *Reagent Chemicals, American Chemical Society Specifications*, American Chemical Society, Washington, DC. For suggestions on the testing of reagents not listed by the American Chemical Society, see *Analar Standards for Laboratory Chemicals*, BDH Ltd., Poole, Dorset, U.K., and the *United States Pharmacopeia and National Formulary*, U.S. Pharmacopeial Convention, Inc. (USPC), Rockville, MD.

⁵ The sole source of supply of the apparatus known to the committee at this time is trademark Strontium Empore solid phase extraction disk or equivalent, 3M Company, St. Paul, MN, <http://www.3m.com>. If you are aware of alternative suppliers, please provide this information to ASTM International Headquarters. Your comments will receive careful consideration at a meeting of the responsible technical committee,¹ which you may attend.