



Designation: ~~D3792–05 (Reapproved 2009)~~ **D3792 – 05 (Reapproved 2015)**

Standard Test Method for Water Content of Coatings by Direct Injection Into a Gas Chromatograph¹

This standard is issued under the fixed designation D3792; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon (ϵ) indicates an editorial change since the last revision or reapproval.

This standard has been approved for use by agencies of the U.S. Department of Defense.

1. Scope

1.1 This test method is for the determination of the total water content of waterborne paints. It has been evaluated for latex systems (styrene-butadiene, poly(vinylacetate)-acrylic, acrylic), epoxy acrylic resin systems and acrylic systems. The established working range of this test method is from 15 to 90 %. There is no reason to believe that it will not work outside of this range.

1.2 The values stated in SI units are to be regarded as the standard. The values given in parentheses are for information only.

1.3 *This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.*

2. Referenced Documents

2.1 *ASTM Standards:*²

D1193 Specification for Reagent Water

E180 Practice for Determining the Precision of ASTM Methods for Analysis and Testing of Industrial and Specialty Chemicals (Withdrawn 2009)³

3. Summary of Test Method

3.1 A suitable aliquot of whole paint is internally standardized, diluted with dimethylformamide, and then injected into a gas chromatographic column containing a porous polymer packing that separates water from other volatile components.

4. Significance and Use

4.1 In order to calculate volatile organic content (VOC) in waterborne paints, it is necessary to know the water content. This gas chromatographic test method provides a relatively simple and direct way to determine water content.

5. Apparatus

5.1 *Gas Chromatography*—Any gas-liquid chromatographic instrument equipped with a thermoconductivity detector may be used. Temperature programming capability is preferable, but isothermal operations may be adequate. See **Table 1**.

5.2 *Column*—The column should be at least 1.22 m (4 ft) of 3.2-mm (1/8-in.) outside diameter tubing of stainless steel, or other suitable material, lined with a TFE-fluorocarbon coating packed with 60/80 mesh (180 to 250 μm) porous polymer packing material.⁴ A longer 1.83 m (6-ft) column can be used to improve resolution.

5.3 *Integrator*—Any electronic integrator that can accurately quantify a gas chromatographic peak is acceptable. Alternatively, a recording potentiometer with a full-scale deflection of 1-10mV, full scale response time of 2 s or less and sufficient sensitivity and stability to meet the requirements of **5.1**.

¹ This test method is under the jurisdiction of ASTM Committee **D01** on Paint and Related Coatings, Materials, and Applications and is the direct responsibility of Subcommittee **D01.21** on Chemical Analysis of Paints and Paint Materials.

Current edition approved ~~June 1, 2009~~ June 1, 2015. Published ~~June 2009~~ June 2015. Originally approved in 1979. Last previous edition approved in 2005/2009 as ~~D3792–05~~ D3792 – 05 (2009). DOI: ~~10.1520/D3792-05R09~~ 10.1520/D3792-05R15.

² For referenced ASTM standards, visit the ASTM website, www.astm.org, or contact ASTM Customer Service at service@astm.org. For *Annual Book of ASTM Standards* volume information, refer to the ~~standard's~~ standard's Document Summary page on the ASTM website.

³ The last approved version of this historical standard is referenced on www.astm.org.

⁴ HayeSep R (silanized), was used in the round robin. Any other porous polymer packing or other column giving equivalent or superior performance may be used. These products are available from most gas chromatography suppliers and distributors.

TABLE 1 Suggested Instrument Conditions

Detector	thermal conductivity	
Column	1.22 m × 3.2 mm TFE-fluorocarbon coating coated stainless steel	
Packing	60/80 mesh porous polymer	
Temperatures, °C		
Sample inlet		240
Detector		250
Column		
Initial		140
Final		240
Program rate	40/min	
Carrier gas	helium	
Flow rate, ml/min	22.0-22.5	
Detector current	150 mA	
Specimen size	1-2 µl	

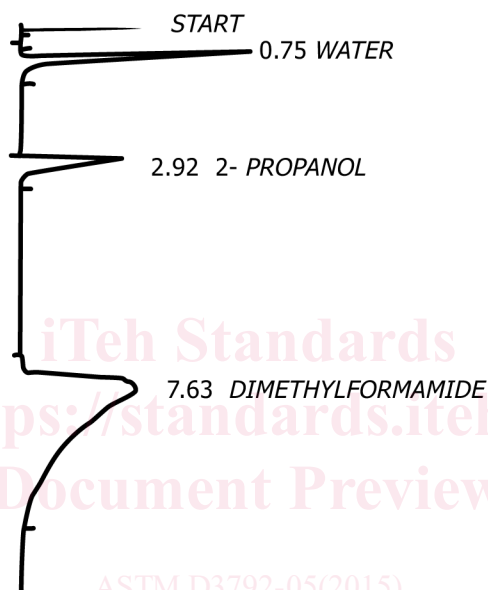


FIG. 1 Typical Chromatogram

5.4 *Liquid Charging Devices*—Micro syringes of 5–10-µL capacity with a precision of ±0.01 µL. Automatic injection of samples improves the precision of this test method.

6. Column Conditioning

6.1 *Procedure*—Install the packed column in the gas chromatographic unit leaving the exit end disconnected from the detector. This will prevent any contamination of the detector with the column bleed.

6.1.1 Set the carrier gas flow rate at 20 to 30 mL/min if a 3.2-mm (1/8-in.) outside diameter column is used. Purge the column 5 or 10 min before heating.

6.1.2 Heat the column from room temperature to 200°C at 5°C/min and hold this temperature for at least 12 h (overnight). At the end of this time, heat the column at 5°C/min to 250°C (the maximum temperature for this packing) and hold for several hours. Cool the column to room temperature and connect to the detector.

6.1.3 Reheat the column to 250°C at 5°C/min to observe if there is column bleed. Optimum conditioning of this column may take several cycles of the heating program before a flat recorder baseline is achieved.

6.2 Before each calibration and series of determinations (or daily) condition the column at 200°C for 1 h with carrier gas flow and to eliminate any residual volatile compounds retained on the column.

7. Reagents and Materials

7.1 *Purity of Reagents*—Reagent grade chemicals shall be used in all tests. Unless otherwise indicated, it is intended that all reagents shall conform to the specifications of the Committee on Analytical Reagents of the American Chemical Society, where