



Designation: **C1517–09 C1517 – 16**

Standard Test Method for Determination of Metallic Impurities in Uranium Metal or Compounds by DC-Arc Emission Spectroscopy¹

This standard is issued under the fixed designation C1517; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon (ϵ) indicates an editorial change since the last revision or reapproval.

1. Scope

1.1 This test method describes the steps necessary for the preparation and determination of impurity metals in uranium metal and uranium compounds by DC arc emission spectroscopy.

1.2 The method is valid for those materials that can be dissolved in acid ~~and/or~~ converted to an oxide in a muffle furnace ~~furnace, or both~~ (see Practice C1347).

1.3 This method uses the carrier distillation technique to selectively carry the impurities into the arc, leaving the uranium oxide in the electrode. If it is necessary to determine the carrier ~~metal~~ (usually metal (usually a silver or strontium, or gallium compound) as an impurity, another technique must be chosen for that element.

1.4 *Units*—The values stated in SI units are to be regarded as standard. The values given in parentheses are for information only.

1.5 *This standard may involve hazardous materials, operations and equipment. This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.*

2. Referenced Documents

2.1 *ASTM Standards:*²

C753 Specification for Nuclear-Grade, Sinterable Uranium Dioxide Powder

C761 Test Methods for Chemical, Mass Spectrometric, Spectrochemical, Nuclear, and Radiochemical Analysis of Uranium Hexafluoride

C776 Specification for Sintered Uranium Dioxide Pellets

C788 Specification for Nuclear-Grade Uranyl Nitrate Solution or Crystals

C859 Terminology Relating to Nuclear Materials

C967 Specification for Uranium Ore Concentrate

C1347 Practice for Preparation and Dissolution of Uranium Materials for Analysis

E130 Practice for Designation of Shapes and Sizes of Graphite Electrodes (Withdrawn 2013)³

E135 Terminology Relating to Analytical Chemistry for Metals, Ores, and Related Materials

3. Terminology

3.1 ~~See definitions and terms~~ Except as otherwise defined herein, definitions of terms are as given in Terminologies C859 and E135.

4. Summary of Test Method

4.1 Uranium metal, solutions and compounds are converted to uranium oxide (U_3O_8) in a muffle furnace. A weighed amount of the oxide is mixed with an appropriate spectrographic carrier and loaded into a graphite electrode. The electrode is excited in a DC arc and the light is dispersed by a spectrograph or spectrometer. The resulting spectrum is measured electronically using a

¹ This test method is under the jurisdiction of ASTM Committee C26 on Nuclear Fuel Cycle and is the direct responsibility of Subcommittee C26.05 on Methods of Test. Current edition approved June 1, 2009 April 1, 2016. Published July 2009 May 2016. Originally approved in 2002. Last previous edition approved in 2002 2009 as C1517–02-C1517 – 09. DOI: 10.1520/C1517-09.10.1520/C1517-16.

² For referenced ASTM standards, visit the ASTM website, www.astm.org, or contact ASTM Customer Service at service@astm.org. For *Annual Book of ASTM Standards* volume information, refer to the standard's Document Summary page on the ASTM website.

³ The last approved version of this historical standard is referenced on www.astm.org.

CCD, CID₂ or CMOS camera or photographed on photographic plates or film sensitive to the proper regions. The line intensities are compared directly to standard plates or to calibration curves derived from the arced standards.

5. Significance and Use

5.1 This test method is applicable to uranium metal, uranium oxides and compounds soluble in nitric or sulfuric acid, and uranium solutions which can be converted to uranium oxide (U₃O₈) in a muffle furnace. It may be used to determine the impurities in uranium compounds as listed in Specifications **C753**, **C776**, **C788**, and **C967**.

6. Apparatus

6.1 *Spectrograph or Spectrometer*—A spectrograph with sufficient resolving power and linear dispersion to separate the analytical lines from other lines in the spectrum of the sample in the spectral region of ~~230.0~~230 to ~~855.0~~855 nm is required. Some spectrographs may be able to access wavelengths lower than 230 nm that may allow the determination of other analytes. Instruments with a reciprocal linear dispersion in the first order of 0.5 nm/mm or less are satisfactory. ~~A direct-reading spectrometer of comparable quality may be substituted for equipment listed, in which case the directions given by the manufacturer should be substituted for those in this procedure.~~ The spectrometer should include a CCD, CID, or CMOS camera for electronic measurement of the resulting spectrum.

6.2 *Excitation Source*—Use an arc power source capable of providing a dc arc of up to 14-A dc, depending on the carrier used and electrode design.

6.3 *Excitation Stand*—Conventional type with adjustable water-cooled electrode holders (may be fitted with automatic sample changers if desired).

~~6.4 *Photographic Processing Equipment*—Use developing, fixing, washing and drying equipment.~~

~~6.5 *Microphotometer*, having a precision of at least $\pm 1\%$ for transmittances.~~

6.4 *Mixer*, for dry materials.

6.5 *Platinum Crucible*.

6.6 *Venting Tool*, (see Fig. 8, Test Methods **C761**); — 01).

6.7 *Muffle Furnace*, 1000°C capability.

7. Reagents and Materials

7.1 *Purity of Materials*—Reagent grade chemicals shall be used in all tests. Unless otherwise indicated, it is intended that all reagents conform to the specifications of the Committee of Analytical Reagents of the American Chemical Society⁴ where such specifications are available. Other grades may be used provided it is first ascertained that the reagent is of sufficiently high purity to permit its use without lessening the accuracy of the determination.

7.2 *Electrodes*—The anode and counter electrodes should be of the S-2, S 16 and C-1 types as given in Practice **E130** (or equivalent).

NOTE 1—Exact shapes and dimensions of the electrodes are not as critical as given in Practice **E130**; however, dimensions of the electrodes used should be consistent and it is essential that the same dimension electrodes be used for standards and samples.

~~7.3 *Photographic Processing Solutions*—Prepare solutions as noted in available literature (1,2).~~

~~7.4 *Photographic Film/Plates*—Use photo emulsion SA-1 and 1-N or equivalent.~~

~~7.5 *Powder Paper*.~~

7.3 *Nitric Acid (HNO₃)*—concentrated (70 %), electronic grade, or equivalent.

7.4 *Sulfuric Acid (H₂SO₄)*—concentrated, electronic grade, or equivalent.

7.5 *Spectrographic Carrier*—The following spectrographic carriers have been used successfully for DC Arc analysis:

7.5.1 *Silver Chloride-Lithium Fluoride*, 11:1 w/w ratio.

7.5.2 *Silver Chloride-Silver Fluoride*, 4:1 w/w ratio.

7.5.3 *Gallium Oxide*, 99.99 % or better.

7.5.4 *Silver Chloride-Strontium Fluoride*, 16.4 mol % SrF₂ in AgCl.

7.5.5 *Gallium Oxide-Lithium Fluoride*, 11:1 w/w ratio.

7.6 *Mixing Vial*, plastic, 12.7 mm (1/2 in.) by 25.4 mm (1 in.) with cap and 10 mm (3/8 in.) plastic mixing bead.

⁴ *Reagent Chemicals, American Chemical Society Specifications*, American Chemical Society, Washington, DC. For suggestions on the testing of reagents not listed by the American Chemical Society, see *Analar Standards for Laboratory Chemicals*, BDH Ltd., Poole, Dorset, U.K., and the *United States Pharmacopeia and National Formulary*, U.S. Pharmacopeial Convention, Inc. (USPC), Rockville, MD.

TABLE 1 Carrier—Sample Combinations

Carrier Material	Carrier Wt, (mg)	Oxide Wt, (mg)	Electrode Charge, (mg)	Mixing Time, (s)
AgCl/AgF (4:1)	50	250	50	180
AgCl/LiF (11:1)	30	270	100	60
AgCl/SrF ₂ (~6:1)	50	450	100	30
Ga ₂ O ₃ /LiF (11:1)	20	380	100	25
Ga ₂ O ₃ ^A	7	343	100	60

^A For the determination of Ag and Li only.

7.7 *Standard Uranium Oxide (U₃O₈) Diluent*—Use NBL CRM 129^{4,3} (or its replacement or equivalent) of known impurity level as a diluent.

8. Precautions

8.1 Consult manufacturer's Material Safety Data Sheets (MSDS)(SDS) for chemical incompatibilities, specific hazards, or spill cleanup for any hazardous materials used in this method.

8.2 All mixing and weighing operations involving uranium oxides should be carried out in properly functioning hoods or exhaust boxes; gloveboxes.

9. Standardization and Calibration

9.1 Standards:

9.1.1 Standards may be synthesized by adding the impurity elements to purified U₃O₈ (NBL CRM 129^{4,3}, or equivalent) and homogenizing. Impurities in powder form, preferably as oxides, may be blended in U₃O₈; impurities in solution may be added to U₃O₈ and the mixture dried, blended and reignited, or the impurities and uranium may be combined in solution and reconverted to U₃O₈. The individual elements should grade in such a ratio as to facilitate visual comparisons covering the desired analytical range for each.

9.1.2 The compounds used to make U₃O₈ impurity standards should be of the highest purity available.

9.1.3 Alternatively, commercially available uranium impurity standards, such as NBL CRM 123⁵ and 124⁵ series standards, may be used. (Other standards may be available; the user should determine quality and/or applicability or applicability, or both, prior to use.) These may be supplemented by synthetic standards to extend calibration ranges, if necessary.

9.1.4 For each standard used, prepare in the same ratio of uranium oxide to carrier as for samples (see Table 1 for further details).

9.1.5 Charge the electrode and arc at the same conditions as determined to be optimum for the instrument in use.

9.2 Calibration Curves:

9.2.1 ~~If a microphotometer is used, determine emulsion calibration curves and analytical curves as described in available literature (1,2).~~

9.2.1 ~~If In using a direct reading spectrometer is used, spectrometer, calibrate according to manufacturer's instruction.~~

9.2.3 ~~For visual comparison, the darkness/width of each standard and impurity element line will establish the individual analyte concentrations.~~

10. Procedure

10.1 Sample Preparation

10.1.1 Liquid Samples:

10.1.1.1 Transfer into a platinum crucible a sufficient amount of liquid to yield not more than 2 g uranium oxide.

10.1.1.2 Place the crucible on a hot plate and evaporate to dryness. Do not allow to boil.

NOTE 2—The addition of H₂SO₄ may be necessary for solutions of uranyl chloride to effect complete conversion to oxide.

10.1.1.3 Continue the preparation at 10.1.2.2.

10.1.2 Miscellaneous Uranium Oxides, Solids or Compounds:

10.1.2.1 Transfer not more than 2 g of the uranium compound to a platinum crucible.

10.1.2.2 Place in a cool muffle and set the muffle to 900 ± 50°C.

⁵ Available from the US Department of Energy, New Brunswick Laboratory, ~~D350~~, Building 350, 9800 South Cass Avenue, Argonne, IL 60439, ATTN: Reference Material Sales. <http://www.nbl.doe.gov> <http://science.energy.gov/nbl/>.