



Designation: **E2838 – 11** ~~E2838 – 11~~ (Reapproved 2016)

Standard Test Method for Determination of Thiodiglycol on Wipes by Solvent Extraction Followed by Liquid Chromatography/Tandem Mass Spectrometry (LC/MS/MS)¹

This standard is issued under the fixed designation E2838; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon (ϵ) indicates an editorial change since the last revision or reapproval.

1. Scope

1.1 This procedure details the determination of thiodiglycol (TDG), also known as 2,2'-thiobis-ethanol, on wipes with 3,3'-thiodipropanol (TDP) as the surrogate. This method is based upon solvent extraction of wipes by either sonication or a pressurized fluid extraction (PFE) technique as an alternative option. The extract is filtered, concentrated and analyzed by liquid chromatography/tandem mass spectrometry (LC/MS/MS). TDG is qualitatively and quantitatively determined.

1.2 *Units*—The values stated in SI units are to be regarded as standard. No other units of measurement are included in this standard.

1.3 The Method Detection Limit (MDL)² and Reporting Range³ for TDG are listed in [Table 1](#).

1.4 *This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.*

2. Referenced Documents

2.1 *ASTM Standards*:⁴

[D653 Terminology Relating to Soil, Rock, and Contained Fluids](#)

[D1193 Specification for Reagent Water](#)

[D3694 Practices for Preparation of Sample Containers and for Preservation of Organic Constituents](#)

[D3740 Practice for Minimum Requirements for Agencies Engaged in Testing and/or Inspection of Soil and Rock as Used in Engineering Design and Construction](#)

[E2554 Practice for Estimating and Monitoring the Uncertainty of Test Results of a Test Method Using Control Chart Techniques](#)

2.2 *Other Documents*:

[EPA Publication SW-846 Test Methods for Evaluating Solid Waste, Physical/Chemical Methods](#)⁵

[The Code of Federal Regulations, 40 CFR Part 136, Appendix B](#)

3. Terminology

3.1 *Abbreviations*:

3.1.1 *mM*—millimolar, 1×10^{-3} moles/L

3.1.2 *ND*—non-detect

3.1.3 *SRM*—single reaction monitoring

¹ This test method is under the jurisdiction of ASTM Committee E54 on Homeland Security Applications and is the direct responsibility of Subcommittee E54.03 on Decontamination.

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² The MDL is determined following the Code of Federal Regulations, 40 CFR Part 136, Appendix B utilizing solvent extraction of wipes by sonication.

³ Reporting range concentrations are calculated from [Table 4](#) concentrations assuming a 10 μ L injection of the lowest and highest level calibration standards with a 2 mL final extract volume. Volume variations will change the reporting limit and ranges. The reporting limit (RL), lowest concentration of the reporting range, is calculated from the concentration of the Level 1 calibration standard as shown in [Table 4](#).

⁴ For referenced ASTM standards, visit the ASTM website, www.astm.org, or contact ASTM Customer Service at service@astm.org. For *Annual Book of ASTM Standards* volume information, refer to the standard's Document Summary page on the ASTM website.

⁵ Available from National Technical Information Service (NTIS), U.S. Department of Commerce, 5285 Port Royal Road, Springfield, VA, 22161 or at <http://www.epa.gov/epawaste/hazard/testmethods/index.htm>

TABLE 1 Method Detection Limit and Reporting Range

Analyte	CAS ^A Number	MDL (µg/wipe)	Reporting Range (µg/wipe)
Thiodiglycol	111-48-8	0.085	1-80
3,3'-Thiodipropanol (Surrogate)	10595-09-2	Not done for surrogates	1-80

^A Chemical Abstract Service (CAS), A division of the American Chemical Society, 2540 Olentangy River Road, Columbus, OH, 43202, USA.

3.1.4 *MRM*—multiple reaction monitoring

3.1.5 *VOA*—volatile organic analysis

4. Summary of Test Method

4.1 For TDG wipe analysis, samples are shipped to the lab between 0°C and 6°C. The samples are to be extracted, concentrated, and analyzed directly by LC/MS/MS within 7 days of collection. The handling, storage, preservation, and LC/MS/MS analysis are consistent between the two extraction procedures described in this test method. Only one extraction procedure is required, documenting which was performed.

4.2 TDG and TDP are identified by retention time and one SRM transition. The target analyte and surrogate are quantitated using the SRM transitions utilizing an external calibration. The final report issued for each sample lists the concentration of TDG and the TDP recovery.

5. Significance and Use

5.1 This is a performance based method, and modifications are allowed to improve performance.

5.1.1 Due to the rapid development of newer instrumentation and column chemistries changes to the analysis described in this standard are allowed as long as better or equivalent performance data result. Any modifications shall be documented and performance data generated. The user of the data generated by this Standard shall be made aware of these changes and given the performance data demonstrating better or equivalent performance.

5.2 TDG is a Schedule 2 compound under the Chemical Weapons Convention (CWC).⁶ Schedule 2 chemicals include those that are precursors to chemical weapons, chemical weapons agents or have a number of other non-military commercial uses. Schedule 2 chemicals can also be found in applications unrelated to chemical weapons. These chemicals are used as ingredients to produce insecticides, herbicides, lubricants, and some pharmaceutical products. TDG is a mustard gas precursor and a degradant as well as an ingredient in water-based inks, ballpoint pen inks, dyes, and some pesticides.

5.3 This method has been investigated for use on surface wipes. TDG is also a human metabolite resulting from sulfur mustard exposure but this method has not been investigated for such determinations.

6. Interferences

6.1 Method interferences may be caused by contaminants in solvents, reagents, glassware, and other apparatus producing discrete artifacts or elevated baselines. All of these materials shall be demonstrated to be free from interferences by analyzing laboratory reagent blanks under the same conditions as samples.

6.2 All reagents and solvents shall be of pesticide residue purity or higher to minimize interference problems.

6.3 Matrix interferences may be caused by contaminants that are co-extracted from the sample. The extent of matrix interferences can vary considerably from sample source depending on variations of the sample matrix.

7. Apparatus

7.1 *LC/MS/MS System:*

7.1.1 *Liquid Chromatography (LC) System*⁷—A LC system is required in order to analyze samples. A LC system that is capable of performing at the flows, pressures, controlled temperatures, sample volumes, and requirements of the standard shall be used.

⁶ Additional information about CWC and thiodiglycol is available on the Internet at <http://www.opcw.org> (2009).

⁷ A Waters Alliance® High Performance Liquid Chromatography (HPLC) System was used to develop this test method and generate the precision and bias data presented in Section 17. The sole source of supply of the apparatus known to the committee at this time is Waters Corporation, Milford, MA 01757. If you are aware of alternative suppliers, please provide this information to ASTM International Headquarters. Your comments will receive careful consideration at a meeting of the responsible technical committee,¹ which you may attend.

7.1.2 *Analytical Column*⁸—A column that achieves adequate resolution shall be used. The retention times and order of elution may change depending on the column used and need to be monitored. A reverse-phase analytical column with strong embedded basic ion-pairing groups was used to develop this test method.

7.1.3 *Tandem Mass Spectrometer (MS/MS) System*⁹—A MS/MS system capable of multiple reaction monitoring (MRM) analysis or a system that is capable of performing at the requirements in this standard shall be used.

7.2 *Pressurized Fluid Extraction (PFE) Device*¹⁰ (optional)—PFE devices with appropriately-sized extraction cells are available that will accommodate the wipe sample sizes used in this test method. Cells shall be made of stainless steel or other material capable of withstanding the pressure requirements (≥ 2000 psi) necessary for this procedure. A pressurized fluid extraction device shall be used that can meet the necessary requirements in this test method.

7.3 *Glass Fiber Filters*.¹¹

7.4 *Solvent Blowdown Device*, with 24- and 50-vial capacity trays and a water bath maintained at 50 to 60°C for analyte concentration from solvent volumes up to 50 mL or similar device shall be used.¹²

7.5 *Sonication Device*, capable of holding 40 mL vials.¹³

7.6 *Nitrogen Evaporation Device*, equipped with a water bath that can be maintained at 50°C for final analyte concentration (<10 mL volume) or similar shall be used.¹⁴

7.7 *Wipes*.¹⁵

7.8 *Filter Paper*.¹⁶

7.9 *Kuderna-Danish Vials (K-D)*, 10 mL.

7.10 *Amber VOA Vials*, 40 mL for sonication, or 60 mL for PFE.

7.11 *Filtration Device*:

7.11.1 *Hypodermic Syringe*—A luer-lock tip glass syringe capable of holding a syringe driven filter unit.

7.11.1.1 A 25 or 50 mL luer-lock tip glass syringe size is recommended in this test method.

7.11.2 *Filter Units*¹⁷—A filter unit of polytetrafluoroethylene (PTFE) 0.20 μm was used for the sonication extraction and a polyvinylidene fluoride (PVDF) 0.22 μm was used for the PFE process. Either PTFE or PVDF filter units shall be used.

⁸ A SIELC- Primesep SB™ 5 μm , 100 Å particle, 150 by 2.1 mm column was used to develop this test method and generate the precision and bias data presented in Section 17. The sole source of supply of the apparatus known to the committee at this time is SIELC Technologies, Prospect Heights, IL 60070. If you are aware of alternative suppliers, please provide this information to ASTM International Headquarters. Your comments will receive careful consideration at a meeting of the responsible technical committee,¹ which you may attend.

⁹ A Waters Quattro micro™ API mass spectrometer was used to develop this test method and generate the precision and bias data presented in Section 17. The sole source of supply of the apparatus known to the committee at this time is Waters Corporation, Milford, MA 01757. If you are aware of alternative suppliers, please provide this information to ASTM International Headquarters. Your comments will receive careful consideration at a meeting of the responsible technical committee,¹ which you may attend.

¹⁰ A Dionex Accelerated Solvent Extraction (ASE® 200) system with appropriately-sized extraction cells was used to develop this test method and generate the precision and bias data presented in Section 17. The sole source of supply of the apparatus known to the committee at this time is Dionex Corporation, Sunnyvale, CA 94088. If you are aware of alternative suppliers, please provide this information to ASTM International Headquarters. Your comments will receive careful consideration at a meeting of the responsible technical committee,¹ which you may attend.

¹¹ Whatman Glass Fiber Filters 19.8 mm, Part # 047017, specially designed for the PFE system¹⁰ were used to develop this test method and generate the precision and bias data presented in Section 17. The sole source of supply of the apparatus known to the committee at this time is Dionex Corporation, Sunnyvale, CA 94088. If you are aware of alternative suppliers, please provide this information to ASTM International Headquarters. Your comments will receive careful consideration at a meeting of the responsible technical committee,¹ which you may attend.

¹² The sole source of supply of the apparatus (a TurboVap LV) known to the committee at this time is Caliper Life Sciences, Hopkinton, MA 01748. If you are aware of alternative suppliers, please provide this information to ASTM International Headquarters. Your comments will receive careful consideration at a meeting of the responsible technical committee,¹ which you may attend.

¹³ The sole source of supply of the apparatus (a Branson® Model 5510 Sonicator) known to the committee at this time is Branson Ultrasonics, Americas Headquarters, 41 Eagle Road, Danbury, CT 06810. If you are aware of alternative suppliers, please provide this information to ASTM International Headquarters. Your comments will receive careful consideration at a meeting of the responsible technical committee,¹ which you may attend.

¹⁴ The sole source of supply of the apparatus (N-Evap 24-port nitrogen evaporation device) known to the committee at this time is Organomation Associates Inc., West Berlin, MA 01503. If you are aware of alternative suppliers, please provide this information to ASTM International Headquarters. Your comments will receive careful consideration at a meeting of the responsible technical committee,¹ which you may attend.

¹⁵ Certi-Gauze™ pads, sterile, 3 by 3 in. (Part # 52639), were used to develop this test method and generate the precision and bias data presented in Section 17. The sole source of supply of the pads known to the committee at this time is Certified Safety Mfg, Kansas City, MO. If you are aware of alternative suppliers, please provide this information to ASTM International Headquarters. Your comments will receive careful consideration at a meeting of the responsible technical committee,¹ which you may attend.

¹⁶ Whatman 42 ashless, 125 mm filter paper (Catalog # 1442 125) were used to develop this test method and generate the precision and bias data presented in Section 17. The sole source of supply of the filter paper known to the committee at this time is Whatman Inc., Building 1, 800 Centennial Avenue, Piscataway, NJ 08854. If you are aware of alternative suppliers, please provide this information to ASTM International Headquarters. Your comments will receive careful consideration at a meeting of the responsible technical committee,¹ which you may attend.

¹⁷ An IC Millex®-LG Syringe Driven Filter Unit PTFE 0.20 μm (Catalog # SLLGC25NS) and Millex®-GV Syringe Driven Filter Unit PVDF 0.22 μm (Catalog # SLGV033NS) were used to develop this test method and generate the precision and bias data presented in Section 17. The sole source of supply of the filter units known to the committee at this time is Millipore Corporation, 290 Concord Road, Billerica, MA 01821. If you are aware of alternative suppliers, please provide this information to ASTM International Headquarters. Your comments will receive careful consideration at a meeting of the responsible technical committee,¹ which you may attend.

NOTE 1—Any filter unit brand may be used that meets the requirements of the test method.

8. Reagents and Materials

8.1 *Purity of Reagents*—High Performance Liquid Chromatography (HPLC) pesticide residue analysis and spectrophotometry grade chemicals shall be used in all tests. Unless indicated otherwise, it is intended that all reagents shall conform to the Committee on Analytical Reagents of the American Chemical Society.¹⁸ Other reagent grades may be used provided they are first determined to be of sufficiently high purity to permit their use without affecting the accuracy of the measurements.

8.2 *Purity of Water*—Unless otherwise indicated, references to water shall be understood to mean reagent water conforming to ASTM Type I of Specification **D653**. It shall be demonstrated that this water does not contain contaminants at concentrations sufficient to interfere with the analysis.

8.3 *Gases*—Nitrogen (purity $\geq 97\%$) and argon (purity $\geq 99.999\%$).

8.4 *Acetic Acid* ($\text{CH}_3\text{CO}_2\text{H}$, CAS# 64-19-7).

8.5 *Acetone* (CH_3COCH_3 , CAS # 67-64-1).

8.6 *Acetonitrile* (CH_3CN , CAS # 75-05-8).

8.7 *Ammonium Formate* ($\text{NH}_4\text{CO}_2\text{H}$, CAS # 540-69-2).

8.8 *Formic Acid* (HCO_2H , CAS# 64-18-6).

8.9 *Methanol* (CH_3OH , CAS # 67-56-1).

8.10 *Thiodiglycol* ($\text{S}(\text{CH}_2\text{CH}_2\text{OH})_2$, CAS # 111-48-8).

8.11 *3,3'-Thiodipropanol* ($\text{S}(\text{CH}_2\text{CH}_2\text{CH}_2\text{OH})_2$, CAS # 10595-09-2).

8.12 *Drying Agent*.¹⁹

8.13 *Sand*—Reagent Grade sand, such as Ottawa Sand.

9. Hazards

9.1 Normal laboratory safety applies to this method. Analysts shall wear safety glasses, gloves, and lab coats when working in the lab. Analysts shall review the Material Safety Data Sheets (MSDS) for all reagents used in this method and shall be fully trained to perform the tests.

10. Glassware Washing, Sampling and Preservation

10.1 *Glassware Washing*—All glassware is washed in hot tap water with a detergent and rinsed in hot water conforming to ASTM Type I of Specification **D653**. The glassware is then dried and heated in an oven at 250°C for 15 to 30 minutes. All glassware is subsequently cleaned with acetone and methanol, respectively.

10.2 *Sampling*—The wipe sample is folded and placed into a 40 mL pre-cleaned amber glass VOA vial with a PTFE-lined cap in the field. The wipe is shipped to the laboratory between 0°C and 6°C. The required surrogate and matrix spike solutions are added to the wipe in the VOA vial at the laboratory. Field blanks are needed to follow conventional sampling practices.

10.3 *Preservation*—Store samples between 0°C and 6°C from the time of collection until analysis. Analyze the sample within 7 days of collection.

11. Preparation of LC/MS/MS

11.1 LC Chromatograph Operating Conditions for the LC used to develop this test method⁷:

11.1.1 Injection volumes of all calibration standards and samples are 10 μL . The first sample analyzed after the calibration curve is a blank to ensure there is no carry-over. The gradient conditions for the liquid chromatograph are shown in **Table 2**.

11.1.2 *Temperatures*—Column, 30°C; Sample compartment, 15°C.

11.1.3 *Seal Wash*—Solvent: 50% Acetonitrile/50% Water; Time: 5 minutes.

11.1.4 *Needle Wash*—Solvent: 50% Acetonitrile/50% Water; Normal Wash, approximately a 13 second wash time.

11.1.5 *Autosampler Purge*—Three loop volumes.

¹⁸ *Reagent Chemicals, American Chemical Society Specifications*, American Chemical Society, Washington, D.C. For suggestions on the testing of reagents not listed by the American Chemical Society, see *Annual Standards for Laboratory Chemicals*, BDH Ltd., Poole, Dorset, U.K., and the *United States Pharmacopeia and National Formulators*, U.S. Pharmacopeial Convention, Inc. (USPC), Rockville, MD.

¹⁹ Varian – Chem Tube – Hydromatrix®, 1 kg (Part # 198003) was used to develop this test method and generate the precision and bias data presented in Section 17 by recommendation of the PFE manufacturer. The sole source of supply of the drying agent known to the committee at this time is Agilent Technologies, United States, 5301 Stevens Creek Blvd, Santa Clara, CA 95051. If you are aware of alternative suppliers, please provide this information to ASTM International Headquarters. Your comments will receive careful consideration at a meeting of the responsible technical committee,¹ which you may attend. (Note: Some drying agents have been shown to clog PFE transfer lines.)



TABLE 2 Gradient Conditions for Liquid Chromatography

Time (min)	Flow ($\mu\text{L}/\text{min}$)	Percent CH_3CN	Percent Water	Percent 500 mM Ammonium Formate/2% Formic Acid
0	300	0	95	5
2	300	0	95	5
3	300	50	45	5
6	300	90	5	5
10	300	90	5	5
12	300	0	95	5
16	300	0	95	5

11.1.6 Specific instrument manufacturer wash and purge specifications shall be followed in order to eliminate sample carry-over in the analysis.

11.2 Mass Spectrometer Parameters⁹:

11.2.1 To acquire the maximum number of data points per SRM channel while maintaining adequate sensitivity, the tune parameters shall be optimized according to the instrument. Each peak requires at least 10 scans per peak for adequate quantitation. This standard contains one target compound and one surrogate which are in different SRM experiment windows in order to optimize the number of scans and sensitivity. Variable parameters regarding retention times, SRM transitions, and cone and collision energies are shown in Table 3 for the mass spectrometer used to develop this test method. Other mass spectrometer parameters used in the development of this method are listed below:

The instrument is set in the Electrospray (+) positive source setting.

Capillary Voltage: 3.5 kV

Cone: Variable depending on analyte (Table 3)

Extractor: 2 V

RF Lens: 0.2 V

Source Temperature: 120°C

Desolvation Temperature: 300°C

Desolvation Gas Flow: 500 L/hr

Cone Gas Flow: 25 L/hr

Low Mass Resolution 1: 14.5

High Mass Resolution 1: 14.5

Ion Energy 1: 0.5 V

Entrance Energy: -1 V

Collision Energy: Variable depending on analyte (Table 3)

Exit Energy: 2 V

Low Mass Resolution 2: 14.5

High Mass resolution 2: 14.5

Ion Energy 2: 0.5 V

Multiplier: 650 V

Gas Cell Pirani Gauge: 0.33 Pa

Inter-Channel Delay: 0.02 s

Inter-Scan Delay: 0.1 s

Repeats: 1

Span: 0 Daltons

Dwell: 0.1 s

12. Calibration and Standardization

12.1 The mass spectrometer shall be calibrated per manufacturer specifications before analysis. In order to obtain valid and accurate analytical values within the confidence limits, the following procedures shall be followed when performing the test method.

12.2 *Calibration and Standardization*—To calibrate the instrument, analyze eight calibration standards containing the eight concentration levels of TDG and TDP in water prior to analysis as shown in Table 4. A calibration stock standard solution is prepared from standard materials or purchased as certified solutions. Aliquots of Level 8 are then diluted with water to prepare the desired calibration levels in 2 mL amber glass LC vials. The calibration vials shall be used within 24 hours to ensure optimum

TABLE 3 Retention Times, SRM Transitions, and Analyte-Specific Mass Spectrometer Parameters

Analyte	SRM Mass Transition (m/z) (Parent > Product)	Retention Time (min)	Cone Voltage (Volts)	Collision Energy (eV)
Thiodiglycol	123.1 > 104.9	2.75	18	5
3,3'-Thiodipropanol	151.2 > 133.1	5.75	19	8

TABLE 4 Concentrations of Calibration Standards (µg/L)

Analyte/Surrogate	LV 1	LV 2	LV 3	LV 4	LV 5	LV 6	LV 7	LV 8
Thiodiglycol	500	1000	2000	4000	8000	16 000	32 000	40 000
3,3'-Thiodipropanol	500	1000	2000	4000	8000	16 000	32 000	40 000

results. Stock calibration standards are routinely replaced every six months if not previously discarded for quality control failure. The analyst is responsible for recording initial component weights carefully when working with pure materials and correctly carrying the weights through the dilution calculations. Calibration standards are not filtered.

12.2.1 Inject each standard and obtain its chromatogram. An external calibration is used monitoring the SRM transition of each analyte. Calibration software is utilized to conduct the quantitation of the target analyte and surrogate. The SRM transition of each analyte is used for quantitation and confirmation. Confirmation occurs by isolating the parent ion, fragmenting it to the product ion, and relating it to the retention time in the calibration standard.

12.2.2 The calibration software manual shall be consulted to use the software correctly. The quantitation method is set as an external calibration using the peak areas in ppb or ppm units as long as the analyst is consistent. Concentrations may be calculated using the data system software to generate linear regression or quadratic calibration curves. The calibration curves may be either linear or quadratic depending on your instrument. Forcing the calibration curve through the origin is not recommended. Each calibration point used to generate the curve shall have a calculated percent deviation less than 30% from the generated curve.

12.2.3 Linear calibration may be used if the coefficient of determination, r^2 , is >0.98 for the analyte. The point of origin is excluded, and a fit weighting of $1/X$ is used in order to give more emphasis to the lower concentrations. If one of the calibration standards other than the high or low point causes the r^2 of the curve to be <0.98 , this point shall be re-injected or a new calibration curve shall be regenerated. If the low or high (or both) point is excluded, minimally a five point curve is acceptable, but the reporting range shall be modified to reflect this change.

12.2.4 Quadratic calibration may be used if the coefficient of determination, r^2 , is >0.99 for the analyte. The point of origin is excluded, and a fit weighting of $1/X$ is used in order to give more emphasis to the lower concentrations. If one of the calibration standards, other than the high or low, causes the curve to be <0.99 , this point shall be re-injected or a new calibration curve shall be regenerated. If the low and/or high point is excluded, a six point curve is acceptable using a quadratic fit. An initial eight point curve over the calibration range is suggested in the event that the low or high point shall be excluded to obtain a coefficient of determination >0.99 . In this event, the reporting range shall be modified to reflect this change.

12.2.5 The retention time window of the SRM transitions shall be within 5% of the retention time of the analyte in a midpoint calibration standard. If this is not the case, re-analyze the calibration curve to determine if there was a shift in retention time during the analysis, and re-inject the sample. If the retention time is still incorrect, refer to the analyte as an unknown.

12.2.6 A midpoint calibration check standard shall be analyzed at the end of each batch of 20 samples or within 24 hours after the initial calibration curve was generated. This end calibration check shall be the same calibration standard that was used to generate the initial curve. The results from the end calibration check standard shall have a percent deviation less than 30% from the calculated concentration for the target analyte and surrogate. If the results are not within these criteria, the problem shall be corrected, and either all samples in the batch shall be re-analyzed against a new calibration curve or the affected results shall be qualified with an indication that they do not fall within the performance criteria of the test method. If the analyst inspects the vial containing the end calibration check standard and notices that the sample evaporated affecting the concentration, a new end calibration check standard shall be made and analyzed. If this new end calibration check standard has a percent deviation less than 30% from the calculated concentration for the target analyte and surrogate, the results shall be reported unqualified if all other quality control parameters are acceptable.

12.3 If a laboratory has not performed the test before or if there has been a major change in the measurement system, for example, new analyst or new instrument, perform a precision and bias study to demonstrate laboratory capability and verify that all technicians are adequately trained and follow all relevant safety procedures.

12.3.1 Analyze at least four replicates of a wipe sample containing TDG and TDP between Levels 3-6 of the calibration range in the final extract concentration. Each replicate shall be taken through the complete analytical test method.

12.3.2 Calculate the mean (average) percent recovery and relative standard deviation (RSD) of the four values and compare to the ranges of the quality control (QC) acceptance criteria for the Initial Demonstration of Performance in [Table 5](#).

12.3.3 This study shall be repeated until the single operator precision and mean recovery are within the limits in [Table 5](#).

TABLE 5 Quality Control Acceptance Criteria

Analyte/Surrogate	Test Conc. (µg/wipe)	Initial Demonstration of Performance			Lab Control Sample	
		Recovery (%)		Precision	Recovery (%)	
		Lower Limit	Upper Limit	Maximum % RSD	Lower Limit	Upper Limit
Thiodiglycol	16	30	130	40	30	130
3,3'-Thiodipropanol	16	30	130	40	30	130