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Standard Terminology Relating to Gaseous Fuels¹

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1. Scope

1.1 This standard defines the terms used in standards that are the responsibility of Committee D03 on Gaseous Fuels. These terms are used in:

- 1.1.1 The sampling of gaseous fuels,
- 1.1.2 The analysis of gaseous fuels for composition and various other physical properties, and
- 1.1.3 Other practices related to the processing, transmission, and distribution of gaseous fuels.

2. Referenced Documents

- 2.1 *ISO Standard:*
[ISO NP 14532 Natural Gas—Terminology](#)²

3. Terminology

acid gas—natural gas containing high concentrations of hydrogen sulfide or carbon dioxide, or both, which is acidic when in contact with water or water vapor.

associated gas—natural gas, also known as gas-cap gas or dome gas, that overlies and is in immediate contact, but not in solution, with crude oil in a reservoir.

at-line instrument—instrument requiring operator interaction to sample gas directly from the pipeline.

base conditions—temperature and pressure conditions at which natural gas volumes are determined for purposes of custody transfer. In natural gas measurement the properties of interest are temperature, pressure, and composition. Assuming ideal gas properties, for simplicity, tables of pure compounds can be prepared for use in calculating gas properties for any composition at “base conditions.” These “base conditions” are chosen near ambient.

Btu—British thermal unit, the amount of energy required to raise the temperature of one pound of water one degree Fahrenheit. One Btu_{IT}(International Table) is equal to 1055.056 J.

calorimeter—a device to measure the evolved heat resulting from the combustion of a material.

compressed natural gas (CNG)—natural gas that is typically pressurized to 3600 psi. CNG is primarily used as a vehicular fuel.

compressibility—the property of a material that permits it to decrease in volume when subjected to an increase in pressure.

compressibility factor (z)—a factor calculated by taking the ratio of the actual volume of a given mass of gas at a specified temperature and pressure to its volume calculated from the ideal gas law at the same conditions.

dew point—the temperature at any given pressure at which liquid initially condenses from a gas or vapor. It is specifically applied to the temperature at which water vapor starts to condense from a gas mixture (**water dew point**), or at which hydrocarbons start to condense (**hydrocarbon dew point**).

dissolved gas—natural gas held in solution in reservoir liquids at the prevailing temperature and pressure of the reservoir.

dry gas—natural gas containing little or no water vapor.

¹ This terminology is under the jurisdiction of ASTM Committee D03 on Gaseous Fuels and is the direct responsibility of Subcommittee D03.92 on Terminology Classification and Specifications.

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² Available from American National Standards Institute (ANSI), 25 W. 43rd St., 4th Floor, New York, NY 10036, <http://www.ansi.org>.