



Designation: **D7283—13 D7283 – 17**

Standard Test Method for Alpha and Beta Activity in Water By Liquid Scintillation Counting¹

This standard is issued under the fixed designation D7283; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon (ϵ) indicates an editorial change since the last revision or reapproval.

1. Scope

1.1 This test method covers the measurement of gross alpha- and beta- activity concentrations in a homogeneous water sample. It is applicable to alpha emitters with activity concentration levels above 0.11 Bq/L (3 pCi/L) and beta emitters with activity concentration levels above 0.15 Bq/L (4 pCi/L). This test method is not applicable to samples containing radionuclides that are volatile under conditions of the analysis.

1.2 This test method may also be used for the direct measurement of gross alpha- and beta- activity concentrations in homogeneous water samples with alpha emitter activity concentration levels above 1.8 Bq/L (50 pCi/L) and beta emitter activity concentration levels above 3.7 Bq/L (100 pCi/L).

1.3 This test method was tested using single-operator tests.^{2,3} A collaborative study following the U.S. EPA “Protocol for the Evaluation of Alternate Test Procedures for Analyzing Radioactive Contaminants in Drinking Water” was performed. The results of this study are on file at ASTM Headquarters.⁴

~~1.4 Standard methods under the jurisdiction of ASTM Committee D19 may be published for a limited time preliminary to the completion of full collaborative study validation. Such standards are deemed to have met all other D19 qualifying requirements but have not completed the required validation studies to fully characterize the performance of the Test Method across multiple laboratories and matrices. Preliminary publication is done to make current technology accessible to users of standards, and to solicit additional input from the user community.~~

1.4 The values stated in SI units are to be regarded as standard. ~~No other units of measurement are included in this~~ The values given in parentheses are mathematical conversions to pCi/L that are provided for information only and are not considered standard. An exception is noted in Section 14 standard.

1.5 ~~This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.~~

1.6 This international standard was developed in accordance with internationally recognized principles on standardization established in the Decision on Principles for the Development of International Standards, Guides and Recommendations issued by the World Trade Organization Technical Barriers to Trade (TBT) Committee.

2. Referenced Documents

2.1 *ASTM Standards:*⁵

D1129 Terminology Relating to Water

D1125 Test Methods for Electrical Conductivity and Resistivity of Water

D1193 Specification for Reagent Water

D1890 Test Method for Beta Particle Radioactivity of Water

¹ This test method is under the jurisdiction of ASTM Committee D19 on Water and is the direct responsibility of Subcommittee D19.04 on Methods of Radiochemical Analysis.

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² Wong, C. T., Soliman, V. M., and Perera, S. K., *Journal of Radioanalytical and Nuclear Chemistry*, Vol 264, No. 2, 2005, pp. 357–363.

³ Ruberu, S.R., Liu, Y.G., and Perera, S.K., *Health Physics*, Vol 95, No. 4, October 2008, pp. 397–406.

⁴ Supporting data have been filed at ASTM International Headquarters and may be obtained by requesting Research Report RR:D19-1195. Contact ASTM Customer Service at service@astm.org.

⁵ For referenced ASTM standards, visit the ASTM website, www.astm.org, or contact ASTM Customer Service at service@astm.org. For *Annual Book of ASTM Standards* volume information, refer to the standard’s Document Summary page on the ASTM website.

[D1943 Test Method for Alpha Particle Radioactivity of Water](#)
[D3370 Practices for Sampling Water from Closed Conduits](#)
[D3648 Practices for the Measurement of Radioactivity](#)
[D3856 Guide for Management Systems in Laboratories Engaged in Analysis of Water](#)
[D4448 Guide for Sampling Ground-Water Monitoring Wells](#)
[D5847 Practice for Writing Quality Control Specifications for Standard Test Methods for Water Analysis](#)
[D6001 Guide for Direct-Push Groundwater Sampling for Environmental Site Characterization](#)
[D7902 Terminology for Radiochemical Analyses](#)
[E177 Practice for Use of the Terms Precision and Bias in ASTM Test Methods](#)
[E691 Practice for Conducting an Interlaboratory Study to Determine the Precision of a Test Method](#)

2.2 Other Standards and Publications:

[EPA 900.0 Gross Alpha and Gross Beta Radioactivity in Drinking Water, from Prescribed Procedures for Measurement of Radioactivity in Drinking Water \(EPA-600/4-80-032\)](#)⁶
[Standard Methods 7110C Coprecipitation Method for Gross Alpha Radioactivity in Drinking Water](#)⁷
[Standard Methods 8010E Table 8010: Recommended Composition for Reconstituted Fresh Water](#)⁷
[ISO 9696 Water Quality—Measurement of Gross Alpha Activity in Non-saline Water—Thick Source Method](#)⁸
[ISO 11704:2010 Water Quality—Measurement of Gross Alpha and Beta Activity Concentration in non-saline water – Liquid Scintillation Counting Method](#)⁸

3. Terminology

3.1 *Definitions*—For definitions of terms used in this test method, refer to ~~Terminology~~ [Terminologies D1129 or D7902](#). For terms not defined in this test method or in Terminology [D1129](#), reference may be made to other published glossaries.

3.2 Definitions of Terms Specific to This Standard:

3.2.1 *alpha-to-beta spillover, n*—in the measurement of radioactivity, for a given emitting source, that fraction of alpha particles that are misclassified as beta particles by the counter.

3.2.2 *alpha particle detection efficiency, n*—in the measurement of radioactivity, for a given emitting source, that fraction of alpha particles that are identified as alpha particles by the counter.

3.2.3 *beta-to-alpha spillover, n*—in the measurement of radioactivity, for a given emitting source, that fraction of beta particles that are misclassified as alpha particles by the counter.

3.2.4 *beta energy, maximum, n*—the maximum energy of the beta particle energy spectrum produced during beta decay of a given radionuclide.

⁶ Available from United States Environmental Protection Association (EPA), Ariel Rios Agency (EPA), William Jefferson Clinton Bldg., 1200 Pennsylvania Ave., NW, Washington, DC 20460, <http://www.epa.gov>.

⁷ Available from American Water Works Association (AWWA), 6666 W. Quincy Ave., Denver, CO 80235, <http://www.awwa.org>.

⁸ Available from International Organization for Standardization (ISO), 1 rue de Varembe, Case postale 56, CH-1211, Geneva 20, Switzerland, <http://www.iso.ch>. ISO Central Secretariat, BIBC II, Chemin de Blandonnet 8, CP 401, 1214 Vernier, Geneva, Switzerland, <http://www.iso.org>.

3.2.4.1 Discussion—

Since a given beta emitter may decay to several different nuclear energy levels of the progeny, more than one maximum energy may be listed for a given radionuclide.

3.2.5 *beta particle detection efficiency, n*—in the measurement of radioactivity, for a given emitting source, that fraction of beta particles that are identified as beta particles by the counter.

3.2.6 *detector background, n*—in the measurement of radioactivity, the counting rate resulting from factors other than the radioactivity of the sample and reagents used.

3.2.6.1 Discussion—

Detector background varies with the location, shielding of the detector, and the electronics; such background includes cosmic rays, contaminating radioactivity, and electronic noise.

3.2.7 *figure of merit, n*—a numerical quantity based on one or more characteristics of a system or device, representing a measure of efficiency or effectiveness; figure of merit is generally calculated as the square of the efficiency divided by the background.

3.2.8 *gross alpha, n*—in the measurement of radioactivity, a semi-quantitative estimate of the combined activity of alpha-emitting radionuclides in a test sample.

3.2.9 *gross beta, n*—in the measurement of radioactivity, a semi-quantitative estimate of the combined activity of beta-emitting radionuclides in a test sample.

3.2.10 *homogeneous water sample, n*—water in which the alpha and beta activity is uniformly dispersed throughout the volume of water sample and remains so until the measurement is completed or until the sample is evaporated or precipitating reagents are added to the sample.

3.2.11 *reagent background, n*—in the measurement of radioactivity of water samples, the counting rate observed when a sample is replaced by mock sample salts or by reagent chemicals used for chemical separations that contain no analyte.

3.2.11.1 Discussion—

Reagent background varies with the reagent chemicals and analytical methods used and may vary with reagents from different manufacturers and from different processing lots.

4. Summary of Test Method

4.1 The test sample is reduced by evaporation, transferred to a scintillation vial and mixed with a suitable liquid scintillation cocktail. Gross alpha- and beta- activity concentrations are measured simultaneously by liquid scintillation using alpha/beta discrimination. By optimizing the alpha/beta discriminator, a high efficiency of alpha- and beta- particle detection can be achieved with acceptable misclassification of beta particles into the alpha multi-channel analyzer (MCA) and alpha particles into the beta MCA. The alpha- and beta- particle efficiency and spillover calibrations of the liquid scintillation system are determined by using known activities of established reference nuclides in test sources having cocktail-solvent ratios comparable to that of the test samples. Some commonly employed reference standards include ^{241}Am , ^{239}Pu , ^{230}Th , natural isotopic abundance uranium (^{234}U , ^{235}U , and ^{238}U), for gross alpha, and $^{90}\text{Sr}/^{90}\text{Y}$, and $^{137}\text{Cs}/^{137\text{m}}\text{Ba}$ for gross beta. Results are reported in activity units equivalent along with the reference radionuclide (for example, Bq/L gross alpha equiv. ^{241}Am).

4.2 If the measurement quality objectives (MQOs) do not require a low detection limit, an aliquant of the sample may be mixed directly with a suitable liquid scintillation cocktail for analysis.

5. Significance and Use

5.1 This test method is intended for the measurement of gross alpha- and beta-activity concentrations in the analyses of environmental and drinking waters. For samples submitted to satisfy regulatory or permit requirements, the submitter should assure that this or any other method used is acceptable to the regulator or permit issuer.

5.2 This test method is also applicable to the direct analysis of gross alpha- and beta-activity concentrations in water when low detection limits are not required. Direct analysis provides a rapid method for determination of gross alpha- and beta-activity concentrations when low detection limits are not required.

5.3 This test method is not capable of discriminating among alpha emitting radionuclides or among beta emitting radionuclides. Those intending to identify and quantify specific radionuclides should use test methods specific to the radionuclides of interest.

5.4 This test method may not be cited as a method for the determination of gross alpha- or beta-activity concentrations in a solid/soil matrix or the acid digestate of the same. The use of this test method for such applications brings the potential for serious bias and incomparability of results dependent on the matrix constituents, manner of sample preparation or treatment, or both.

6. Interferences

6.1 The counting efficiencies and spillover for both the alpha and beta components are dependent on the energy of the alpha- or beta-emitter chosen to determine the calibration coefficient. Biases may occur if the energies of the alpha- or beta-particle emitting nuclides in the test sample differs significantly from those used to determine the respective counting efficiencies. Best results are obtained when the radionuclide composition of the sample is known and the calibration radionuclide is selected to match as closely as possible the energy of the sample radionuclide.

6.2 The use of $^{137}\text{Cs}/^{137\text{m}}\text{Ba}$ as a calibration standard for samples containing radionuclides other than ^{137}Cs may introduce a low bias in the analytical results unless there is a correction for conversion electron emissions. The conversion electrons from the $^{137\text{m}}\text{Ba}$ progeny are detected by liquid scintillation yielding greater than 100 % detection efficiency for the $^{137}\text{Cs}/^{137\text{m}}\text{Ba}$ calibration standard.

6.3 When using uranium as a calibration standard the isotopic abundance of each of the isotopes (^{234}U , ^{235}U , and ^{238}U) must be known to accurately determine the standard activity concentration. Many uranium standards used for mass measurements are depleted uranium. Natural isotopic abundance uranium and depleted uranium standards contain short-lived decay progeny (^{234}Th , $^{234\text{m}}\text{Pa}$) which interfere with the spillover calibration unless they are removed immediately prior to calibration.

6.4 Radon is a noble gas, and therefore easily emanates from most matrices. If the radon progeny of the uranium (^{222}Rn), thorium (^{220}Rn), and actinium (^{219}Rn) series emanate from the sample test source prior to counting, radioactive equilibrium is

disrupted. EPA 900.0 recognizes this disruption by suggesting a delay of 72 h before the prepared sample is counted for gross alpha. Other published methods such as Standard Methods 7110C provide for a shorter delay of 3 h. Thus the activity of samples containing ^{226}Ra will increase significantly with time during the first several weeks after preparation. This delay will result in overestimation of the activity of samples relative to their true ^{226}Ra concentration. This test method advises that any such delay period used by the laboratory be based on the ~~measurement quality objectives~~ MQOs inherent in the intended data use (see 11.7).

6.5 Radionuclides may be present in the sample in disequilibrium with their parent radionuclides. Many factors, including differential solubility of radionuclides from the matrix in which the parent radionuclide occurs can cause this disequilibrium. Where these radionuclides have a half-life on the order of a few days or shorter, the time elapsed between sampling and the beginning of sample counting will tend to bias the final result low. In those ~~cases the measurement quality objectives cases,~~ the MQOs inherent in the intended data may dictate the maximum time between sample collection and the beginning of sample counting. The laboratory should be aware of such requirements and be prepared to comply with them.

6.6 Radionuclides incorporated in volatile compounds are lost during the conduct of this test method. These include tritium in HTO or ^{14}C in the carbon dioxide formed during the addition of acid. The pertechnetate ion ($\text{FeO}(\text{}^{99}\text{TcO}_4\text{}^{2-})$) is an example of a radionuclide which may be lost through semi-volatility. The ~~Measurement Quality Objectives~~ MQOs should address the potential loss of such radionuclides and provide direction for their quantification by specific methods.

6.7 When counting gross alpha- and beta-activity by a liquid scintillation counter using alpha/beta discrimination, some pulses resulting from alpha particles are misclassified as beta particles and some pulses resulting from beta particles are misclassified as alpha particles. The “spillover” characteristics are determined during the calibration of the specific instrument being used.

6.8 Quenching of the photon output in the liquid scintillation cocktail reduces detection efficiency and introduces additional uncertainty in spillover corrections. Quenching is caused by molecular species in the sample and cocktail mixture that reduce the intermolecular transfer of energy or absorb emitted visible and UV photons prior to detection. This test method describes the use of an external standard source to compensate for the effects of quenching.

6.9 The presence of solid particles in the scintillation cocktail may lead to erroneous results. This test method requires complete dissolution of the sample prior to addition of the scintillation cocktail.

6.10 The sample aliquant/scintillation cocktail mixture ratio should be within the cocktail manufacturer’s recommendations to insure a homogeneous mixture. If the sample aliquant/scintillation cocktail mixture forms two phases, repeat the analysis with a different sample aliquant/scintillation cocktail mixture/ratio.

6.11 The exterior of the vials must be free of dirt, markings, and fingerprints.

6.12 ‘Dark adapting’ of scintillator solutions is dependent upon the ~~fluor,~~ the instrument fluor used in the scintillation cocktail, the instrument, and the lighting conditions of the count room. Evaluation of these parameters for the adaptation to the ‘dark’ conditions is necessary for counting optimization. [ASTM D7283-17](#)

6.13 Samples and standards should be counted with the same instrument operating parameters including temperature. For refrigerated instruments, time should be allowed for the samples to cool to the operating temperature of the instrument. Be aware of the potential for phase separation when cooling prepared samples.

7. Apparatus

7.1 ~~Liquid Scintillation~~ scintillation vials, approximately 20 mL, of low-potassium glass are recommended.

7.2 ~~Electric hot plate.~~ Hot plate, heating block, drying oven or other appropriate device to evaporate the samples.

~~7.3 Glassware.~~

~~7.4 Transfer pipettes.~~

7.3 ~~Liquid scintillation counting system,~~ spectrometry system (Liquid Scintillation Counter, LSC), coincidence-type with alpha/beta discrimination. A guard detector or other background reduction electronics or software may be incorporated to reduce the instrument background.

8. Reagents and Materials

8.1 *Purity of Reagents*—Reagent grade chemicals shall be used in all tests. Unless otherwise indicated, it is intended that all reagents shall conform to the specifications of the Committee on Analytical Reagents of the American Chemical Society, where such specifications are available.⁹ Other grades may be used, provided that the reagent is of sufficiently high purity to permit its use without increasing the background of the measurement. Some reagents, even those of high purity, may contain naturally-occurring radionuclides, such as isotopes of uranium, radium, actinium, thorium, rare earths and potassium compounds

⁹ *Reagent Chemicals, American Society Specifications*, American Chemical Society, Washington, DC. For suggestions on the testing of reagents not listed by the American Chemical Society, see *Analar Standards for Laboratory Chemicals*, BDH Ltd., Poole, Dorset, U.K., and the United States *Pharmacopoeia and National Formulary*, U.S. Pharmaceutical Convention, Inc. (USPC), Rockville, MD.

and/or artificially produced radionuclides. Consequently, when such reagents are used in the analysis of low-radioactivity samples, the activity of the reagents shall be determined under analytical conditions that are identical to those used for the sample. The activity contributed by the reagents may be considered to be a component of the background and supplied as a correction when calculating the test sample result. This increased background reduces the sensitivity of the measurement.

8.2 *Purity of Water*—Unless otherwise indicated, reference to water shall be understood to mean reagent water conforming to Specification **D1193**, Type III.

8.3 *Alpha-Emitting Radioactive Standard Solution* (~ 2000 Bq/mL in $1M$ HNO_3)—traceable to a national standards laboratory (such as the National Institute of Standards and Technology, NIST, in the ~~USA~~-United States; or the National Physics Laboratory, NPL, in the ~~UK~~-United Kingdom). For gross alpha calibration the following radionuclides have found general usage: ^{241}Am , ^{239}Pu , ^{230}Th , and natural isotopic abundance uranium (^{234}U , ^{235}U , and ^{238}U).

8.4 *Beta-Emitting Radioactive Standard Solution* (~ 2000 Bq/mL in $1M$ HNO_3)—traceable to a national standards laboratory (such as NIST or NPL). For gross beta calibration the following radionuclides have found general usage: $^{90}Sr/^{90}Y$ and $^{137}Cs/^{137m}Ba$.

8.5 *Liquid scintillation cocktail*—Commercially prepared LSC cocktail or equivalent.

8.6 *Nitric Acid* (sp gr 1.42)—Concentrated nitric acid.

8.7 *Nitric Acid* (2M)—Mix 128 mL 16M HNO_3 (concentrated) with water and dilute to 1 L.

8.8 *Nitric Acid* (0.1M)—Mix 6.4 mL 16M HNO_3 (concentrated) with water and dilute to 1 L.

8.9 *Nitromethane* (sp gr 1.14)—Other quenching agents may also be used. Adjust the amount of quenching agent added to the calibration standards to produce a calibration curve covering the typical range of quench found in samples.

9. Sampling

9.1 A representative sample must be collected from the water source and should be large enough so that adequate aliquants can be taken to obtain the required sensitivity. See Practices **D3370** and Guides **D4448** and **D6001** for guidance on sampling.

9.2 Although the container material does not impact the analyte stability, the container choice should generally be plastic instead of glass to minimize losses due to breakage during transportation and handling.

9.3 Unless contrary to the ~~measurement quality objectives~~ MOOs (for example, radiocarbon or radioiodine analysis is to be performed on the same sample or dissolved gross alpha and beta activity concentrations are sought), it is recommended that the sample be preserved at the time of collection by adding enough 2M nitric acid to the sample to bring it to pH 2 or less (5 to 10 mL of 2M nitric per litre of sample is usually sufficient). Tightly cap the container and shake well to mix. Confirm the pH with a pH-indicating strip or paper.

9.4 If the dissolved gross alpha and beta activity concentrations are sought, the sample must be passed through a 0.45 micron filter prior to acid preservation of the sample. Drinking water samples are not normally filtered prior to analysis unless the turbidity is >5 NTU-Nephelometric Turbidity Units (NTUs). The use of suction or pressure will speed the filtration process.

9.5 If samples are collected without preservation, they should be delivered to the laboratory as quickly as practicable, but no later than 5 days following collection. Upon receipt at the lab, the unpreserved samples may be filtered, as required, and then acid preserved. Once preserved at the lab, the samples should be held for a minimum of 16 h prior to initiation of sample preparation.

9.6 Sample analysis should be completed within 180 days from the time of sample collection. Samples may be held for up to one year to allow for compositing of quarterly samples.

10. Calibration and Standardization

10.1 For gross alpha and gross beta measurement, the scintillation counter must be calibrated to determine the alpha particle detection efficiency in the alpha region of interest (ROI), the alpha particle detection efficiency in the beta ROI, the beta particle detection efficiency in the beta ROI, and the beta particle detection efficiency in the alpha ROI. For gross alpha calibration the following radionuclides have found general usage: ^{241}Am , ^{239}Pu , ^{230}Th , and natural isotopic abundance uranium (^{234}U , ^{235}U , and ^{238}U). For gross beta calibration the following radionuclides have found general usage: $^{90}Sr/^{90}Y$ and $^{137}Cs/^{137m}Ba$. The laboratory must ensure that the client is aware of the radionuclides used for the alpha and beta calibrations because the intercomparability of results from other laboratories will be impacted if they used different calibration nuclides.

NOTE 1—When using uranium as a calibration standard the isotopic abundance of each of the isotopes (^{234}U , ^{235}U , and ^{238}U) must be known to accurately determine the standard activity concentration. Many uranium standards used for mass measurements are depleted uranium. Natural isotopic abundance uranium and depleted uranium standards contain short-lived decay progeny (^{234}Th , ^{234m}Pa) which interfere with the spillover calibration unless they are removed immediately prior to calibration.

10.2 Place the instrument into operation according to the manufacturer's instructions. The instrument should be set to acquire counts in the alpha/beta counting mode with discrimination of alpha and beta events.

10.3 The regions of interest (ROIs) for alpha and beta counting in the alpha and beta MCAs should be set to optimize the figure of merit (E^2/B) while ensuring that all radionuclides of interest are included in each respective region. For an unquenched sample an ROI from 400 keV to 700 keV would include most alpha emitting radionuclides of concern. Based on the quench characteristics of the samples, the ROI should be adjusted to include the radionuclides of concern, while minimizing the background count rate. For the sample to cocktail mixture used in this test method (5 mL sample plus 15 mL of scintillation cocktail) an ROI from 50 keV to 400 keV generally includes the alpha-emitting radionuclides of concern. For beta-emitting radionuclides an ROI from 0 to 2000 keV is generally used. A low energy window setting of 2 keV is generally sufficient to eliminate luminescence and low energy noise.

10.4 The optimum setting for discrimination between alpha- and beta- particles is the setting where there is equal and minimum spill of alpha pulses into the beta MCA and beta pulses into the alpha MCA. This occurs at the crossover point of the alpha-to-beta spillover and beta-to-alpha spillover curves. However, when only the alpha emitter is of interest, a discriminator setting greater than the instrument determined cross-over point may be used, to minimize misclassification of beta events into the alpha MCA at the expense of reduced alpha detection efficiency. Similarly when only the beta emitter is of interest, a discriminator setting below the optimum may be used. This minimizes the misclassification of alpha events into the beta MCA at the expense of reduced beta detection efficiency.

10.4.1 Prepare a scintillation vial containing approximately 200 Bq of an alpha-emitting radionuclide such as ^{241}Am , ^{239}Pu , or ^{230}Th in 5 mL of 0.1M HNO_3 plus 15 mL of scintillation cocktail.

10.4.2 Prepare a second scintillation vial containing approximately 200 Bq of a beta- emitting radionuclide, such as $^{90}\text{Sr}/^{90}\text{Y}$ or $^{137}\text{Cs}/^{137\text{m}}\text{Ba}$ in 5 mL of 0.1M HNO_3 plus 15 mL of scintillation cocktail.

10.4.3 Count the vials for an amount of time required to obtain a relative standard uncertainty of 1 % or less in the count (generally at least 10 000 net counts) at varying discriminator settings. Plot the alpha-to-beta spillover and beta-to-alpha spillover versus discriminator setting. The point at which the two curves intersect is the crossover point.

10.5 For each instrument the alpha particle detection efficiency in the alpha ROI, the alpha particle detection efficiency in the beta ROI, the beta particle detection efficiency in the beta ROI and the beta particle detection efficiency in the alpha ROI, at varying levels of quench are determined using the optimized ROIs and discriminator setting. The alpha and beta radionuclide standards used should be traceable to a national standards laboratory (such as NIST or NPL).

10.5.1 For the alpha particle detection efficiency in the alpha ROI and the alpha particle detection efficiency in the beta ROI, a minimum of five alpha calibration standards are prepared containing varying amounts of quenching agent. For each calibration standard, aliquot 5.00 mL of 0.1M nitric acid into a scintillation vial. Spike each of the vials with approximately 200 Bq of the alpha calibration standard. Add 15 mL of scintillation cocktail and 0 μL to 50 μL of nitromethane to each of the vials to create a series of quench standards.

10.5.2 For the beta particle detection efficiency in the beta ROI and the beta particle detection efficiency in the alpha ROI, a minimum of five beta calibration standards are prepared containing varying amounts of quenching agent. For each calibration standard, aliquot 5.00 mL of 0.1M nitric acid into a scintillation vial. Spike each of the vials with approximately 200 Bq of the beta calibration standard. Add 15 mL of scintillation cocktail and 0 μL to 50 μL of nitromethane to each of the vials to create a series of quench standards.

10.5.3 Prepare a background subtraction sample by aliquoting 5.00 mL of 0.1 M nitric acid into a scintillation vial and adding 15 mL of scintillation cocktail.

10.5.4 Count both sets of calibration standards and the background subtraction sample using the optimized ROIs and discriminator settings to obtain a relative standard uncertainty of 1 % or less in each count (10 000 counts).

11. Procedure

NOTE 2—To ensure sample integrity, step 11.1, if required, and step 11.2 should be done promptly after sample receipt at the laboratory if these actions were not performed at the time of sample collection. Samples may be kept at room temperature between receipt and analysis.

11.1 If filtration of the sample is required by the ~~measurement quality objectives—MQOs~~ and the sample was not filtered at the time of collection, pass a sample aliquant sufficient for analysis through a 0.45 μm pore membrane filter. Drinking waters normally are not filtered prior to analysis (unless turbidity is >5 NTU).

11.2 If the sample was not preserved at the time of collection, add enough 2M nitric acid to bring the sample to pH 2 or less (5 to 10 mL of 2M nitric per litre of sample is usually sufficient). Tightly cap the container and shake well to mix. Confirm the pH with pH-indicating strip or paper. Hold the acidified sample for at least 16 h before starting the analysis. Samples should be analyzed within 180 days after sample collection, or within one year when compositing quarterly samples.

11.3 When steps 11.4 and 11.5 are not performed within four days of step 11.2, recheck the pH of the sample just prior to analysis to ensure that the pH is 2 or less. If the pH is not 2 or less, add sufficient 2M nitric acid to accomplish this, and then hold the sample for at least 16 h prior to the next step.

11.4 Transfer to a clean beaker a measured aliquant of the water that contains no more than 400 mg of residue mass. The amount of dissolved solids in a given sample can be approximated by evaporating a small amount (for example, 5 mL) of the sample in