



Designation: ~~D5770~~—~~11~~ **D5770 – 11 (Reapproved 2017)**

Standard Test Method for Semiquantitative Micro Determination of Acid Number of Lubricating Oils During Oxidation Testing¹

This standard is issued under the fixed designation D5770; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon (ϵ) indicates an editorial change since the last revision or reapproval.

1. Scope*~~Scope~~

1.1 This test method is a semiquantitative micro method intended for monitoring the changes in acidic constituents occurring in lubricating oils during oxidation testing, when the acid number of such oils falls within the range from ~~0.020.02 mg~~ to ~~1.0 mg~~ 1.0 mg of potassium hydroxide per gram of sample. It is applicable to such oils as turbine oils, hydraulic oils, and other circulating oils.

NOTE 1—This test method is a micro version of Test Method **D974** and it produces results similar to that method.

1.2 This test method is designed for use where sample size is limited. It shall not be used as a replacement for higher precision methods such as Test Methods **D974** or **D664**. It shall not be used to monitor oils in-service.

1.3 The values stated in SI units are to be regarded as the standard. ~~The values given in parentheses are for information only.~~

1.3.1 *Exception*—The values given in parentheses are for information only.

1.4 *This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.*

1.5 *This international standard was developed in accordance with internationally recognized principles on standardization established in the Decision on Principles for the Development of International Standards, Guides and Recommendations issued by the World Trade Organization Technical Barriers to Trade (TBT) Committee.*

2. Referenced Documents

2.1 *ASTM Standards:*²

D664 Test Method for Acid Number of Petroleum Products by Potentiometric Titration

D943 Test Method for Oxidation Characteristics of Inhibited Mineral Oils

D974 Test Method for Acid and Base Number by Color-Indicator Titration

D3339 Test Method for Acid Number of Petroleum Products by Semi-Micro Color Indicator Titration

D4871 Guide for Universal Oxidation/Thermal Stability Test Apparatus

3. Terminology

3.1 *Definitions:*

3.1.1 *acid number, n*—the quantity of a specified base, expressed in milligrams of potassium hydroxide per gram of sample, required to titrate a sample in a specified solvent to a specified endpoint using a specified detection system.

¹ This test method is under the jurisdiction of ASTM Committee **D02** on Petroleum Products—~~Products, Liquid Fuels, and Lubricants~~ and is the direct responsibility of Subcommittee **D02.06** on Analysis of Liquid Fuels and Lubricants.

Current edition approved ~~May 15, 2011~~ May 1, 2017. Published ~~July 2011~~ June 2017. Originally approved in 1995. Last previous edition approved in ~~2007~~ 2011 as **D5770-02(2007):D5770-11**. DOI: ~~10.1520/D5770-11~~ 10.1520/D5770-11R17.

² For referenced ASTM standards, visit the ASTM website, www.astm.org, or contact ASTM Customer Service at service@astm.org. For *Annual Book of ASTM Standards* volume information, refer to the standard's Document Summary page on the ASTM website.

3.1.1.1 *Discussion*—

In this test method, the acid number is calculated from the number of drops required to produce a change in solution color from

*A Summary of Changes section appears at the end of this standard

blue-green to orange, compared to the number of drops required to produce an identical color change using a reference standard. Because this is a direct comparison method, the acid number value can be reported in milligrams of potassium hydroxide per gram of sample.

4. Summary of Test Method

4.1 A ~~2.0 mL~~ 2.0 mL portion of the titration solution is titrated with a sample using a dropping pipet. The number of drops of sample required to turn the blue-green titration solution to a persistent orange color is noted.

4.2 A second ~~2.0 mL~~ 2.0 mL portion of the titration solution is titrated with an acid number reference solution of known acid number. The number of drops of the reference solution required to turn the blue-green titration solution to a persistent orange color is noted.

4.3 The estimated acid number of the sample is calculated using the acid number of the reference solution and the ratio of the number of drops of the reference solution required to effect the color change to the number of drops of the sample required for the same change.

5. Significance and Use

5.1 This test method provides a semiquantitative estimate of the acid number of lubricating oils obtained from laboratory oxidation tests using smaller amounts of sample than Test Methods **D974**, **D664**, or **D3339**. It has specific application in Test Method **D943** and in Test Method **D4871**. This test method, therefore, provides a means of monitoring the relative oxidation of lubricating oils by measuring changes in acid number, at different time intervals and under various oxidizing test conditions.

5.2 Since this test method is semiquantitative, each laboratory shall develop its own criteria for each oxidation test method for determining when to switch from this semiquantitative test method to a more precise test method for acid number.

6. Apparatus

6.1 *Glass Beakers*, ~~10 mL~~ 10 mL capacity, or glass vials, ~~4 dram~~ or 6 dram.

6.2 *Glass Syringe*, ~~2.5 mL~~ 2 mL to 5 mL capacity. A ~~5 mL~~ 5 mL interchangeable syringe with a ~~20 gage~~, ~~12 cm~~ 20 gage, 12 cm needle is satisfactory.

6.3 *Dropping Pipet*, glass with rubber bulb, capable of delivering ~~35~~ 35 drops to 40 drops of oil.

7. Reagents

7.1 *Purity of Reagents*—Reagent grade chemicals shall be used in all tests. Unless indicated otherwise, it is intended that all reagents shall conform to the specifications of the Committee on Analytical Reagents of the American Chemical Society, where such specifications are available.³ Other grades can be used provided it is first ascertained that the reagent is of sufficiently high purity to permit its use without lessening the accuracy of the determination.

7.2 *Acid Number Reference Solution*—Any stable, oil-soluble acid which will produce an acid number of approximately ~~0.5~~ 0.5 mg KOH/g KOH is acceptable. This solution shall be standardized using Test Method **D974** or **D664**. A solution of 0.20 % by mass dodecenylsuccinic anhydride in HVI 250 base oil or in an ISO 10-22 (60-100 SUS) oil has proven satisfactory.

7.3 *Alcoholic Potassium Hydroxide Standard Solution (0.1 M)*—Prepare approximately ~~0.1 M~~ 0.1 M solution by dissolving KOH in propanol-2-ol. Standardize against pure potassium acid phthalate in about ~~100 mL~~ 100 mL of water, using phenolphthalein to detect the end point. (Commercial grades of this reagent have been found to be satisfactory.) (**Warning**—Corrosive.) (**Warning**—Poisonous if ingested, alkaline, causes irritation producing dermatitis.)

7.4 *Dodecylsuccinic Anhydride*—(**Warning**—Irritating to eyes and skin.)

7.5 *p-Naphtholbenzein Indicator Solution*—The p-naphtholbenzein must meet the specifications in Appendix X1 of Test Method **D974**. Prepare a solution containing ~~10 g~~ 10 g of p-naphtholbenzein/L in a 50:50 mixture of toluene:propanol-2-ol. (**Warning**—Flammable. Vapor harmful.)

NOTE 2—In a 1992 study, only Kodak, Fisher, and Baker (Mallinkrot) were found to meet the specifications. However, the Kodak brand is no longer available. The Fisher Reagent Solution was the only commercially available solution to meet the specifications.

7.6 *Propanol-2-ol (Isopropyl Alcohol)*—(**Warning**—Flammable.) (**Warning**—It has been reported that, unless inhibited against it, peroxides can form in propanol-2-ol and, as the storage vessel or equipment such as a dispensing bottle become empty and approach dryness, an explosive mixture can occur.)

³ *Reagent Chemicals, American Chemical Society Specifications*, American Chemical Society, Washington, DC. For suggestions on the testing of reagents not listed by the American Chemical Society, see *Analar Standards for Laboratory Chemicals*, BDH Ltd., Poole, Dorset, U.K., and the *United States Pharmacopeia and National Formulary*, U.S. Pharmacopeial Convention, Inc. (USPC), Rockville, MD.