



Designation: **D381 – 12 D381 – 12 (Reapproved 2017)**

Standard Test Method for Gum Content in Fuels by Jet Evaporation¹

This standard is issued under the fixed designation D381; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon (ϵ) indicates an editorial change since the last revision or reapproval.

This standard has been approved for use by agencies of the U.S. Department of Defense.

1. Scope^{*}

1.1 This test method covers the determination of the existent gum content of aviation fuels, and the gum content of motor gasolines or other volatile distillates in their finished form, (including those containing alcohol and ether type oxygenates and deposit control additives—see **Note 7** for additional information) at the time of test.

1.2 Provisions are made for the determination of the heptane insoluble portion of the residue of non-aviation fuels.

1.3 The values stated in SI units are to be regarded as standard. No other units of measurement are included in this standard.

1.3.1 The accepted SI unit of pressure is the Pascal (Pa); the accepted SI unit for temperature is degrees Celsius.

1.4 **WARNING**—Mercury has been designated by many regulatory agencies as a hazardous material that can cause central nervous system, kidney and liver damage. Mercury, or its vapor, may be hazardous to health and corrosive to materials. Caution should be taken when handling mercury and mercury containing products. See the applicable product Material Safety Data Sheet (MSDS) for details and EPA's website—<http://www.epa.gov/mercury/faq.htm>—for additional information. Users should be aware that selling mercury and/or mercury containing products into your state or country may be prohibited by law.

1.5 *This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.* For specific warning statements, see **6.4**, **7.4**, and **9.1**.

1.6 *This international standard was developed in accordance with internationally recognized principles on standardization established in the Decision on Principles for the Development of International Standards, Guides and Recommendations issued by the World Trade Organization Technical Barriers to Trade (TBT) Committee.*

2. Referenced Documents

2.1 *ASTM Standards:*²

D1655 Specification for Aviation Turbine Fuels

D4057 Practice for Manual Sampling of Petroleum and Petroleum Products

E1 Specification for ASTM Liquid-in-Glass Thermometers

E29 Practice for Using Significant Digits in Test Data to Determine Conformance with Specifications

2.2 *Energy Institute Standard:*³

IP Standard Methods for Analysis and Testing of Petroleum Products

IP 540 Determination of the existent gum content of aviation turbine fuel – jet evaporation method

3. Terminology

3.1 *Definitions of Terms Specific to This Standard:*

3.1.1 *existent gum, n*—the evaporation residue of aviation fuels, without any further treatment.

3.2 For non-aviation fuels, the following definitions apply.

¹ This test method is under the jurisdiction of ASTM Committee **D02** on Petroleum Products, Liquid Fuels, and Lubricants and is the direct responsibility of Subcommittee **D02.14** on ~~Stability and Cleanliness~~ Stability, Cleanliness and Compatibility of Liquid Fuels.

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² For referenced ASTM standards, visit the ASTM website, www.astm.org, or contact ASTM Customer Service at service@astm.org. For *Annual Book of ASTM Standards* volume information, refer to the standard's Document Summary page on the ASTM website.

³ Available from Energy Institute, 61 New Cavendish St., London, ~~W1G 7AR, U.K.~~ U.K., <http://www.energyinst.org>.

***A Summary of Changes section appears at the end of this standard**

3.3 *solvent washed gum content, n*—the residue remaining when the evaporation residue (see 3.4) has been washed with heptane and the washings discarded.

3.3.1 *Discussion*—

TABLE 1 Schedule of Test Conditions

Sample Type	Vaporizing Medium	Operating Temperature	
		Bath	Test Well
Aviation and motor gasoline	air	160 to 165°C	150 to 160°C
Aviation turbine fuel	steam	232 to 246°C	229 to 235°C

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Sample Type	Vaporizing Medium	Operating Temperature	
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Aviation and motor gasoline	air	160 °C to 165 °C	150 °C to 160 °C
Aviation turbine fuel	steam	232 °C to 246 °C	229 °C to 235 °C

For motor gasoline or non-aviation gasoline, solvent washed gum content was previously referred to as existent gum.

3.4 *unwashed gum content, n*—the evaporation residue of the product or component under test, without any further treatment.

4. Summary of Test Method

4.1 When testing either aviation or motor gasoline, a ~~5050 mL~~ 0.5 mL ~~± 0.5 mL~~ quantity of fuel is evaporated under controlled conditions of temperature and flow of air (see Table 1). When testing aviation turbine fuel, a ~~5050 mL~~ 0.5 mL ~~± 0.5 mL~~ quantity of fuel is evaporated under controlled conditions of temperature and flow of steam (see Table 1). For aviation gasoline and aviation turbine fuel, the resulting residue is weighed and reported as milligrams per ~~100 mL~~ 100 mL. For motor gasoline, the residue is weighed before and after extracting with heptane and the results reported as milligrams per ~~100 mL~~ 100 mL.

NOTE 1—Specification D1655 allows the existent gum of aviation turbine fuels to be determined by either Test Method D381 or IP 540, with Test Method D381 identified as the referee test method. Test Method D381 specifically requires the use of steam as the evaporating medium for aviation turbine fuels, whereas IP 540 allows either air or steam as the evaporating medium for aviation turbine fuels.

5. Significance and Use

5.1 The true significance of this test method for determining gum in motor gasoline is not firmly established. It has been proved that high gum can cause induction-system deposits and sticking of intake valves, and in most instances, it can be assumed that low gum will ensure absence of induction-system difficulties. The user should, however, realize that the test method is not of itself correlative to induction-system deposits. The primary purpose of the test method, as applied to motor gasoline, is the measurement of the oxidation products formed in the sample prior to or during the comparatively mild conditions of the test procedure. Since many motor gasolines are purposely blended with nonvolatile oils or additives, the heptane extraction step is necessary to remove these from the evaporation residue so that the deleterious material, gum, may be determined. With respect to aviation turbine fuels, large quantities of gum are indicative of contamination of fuel by higher boiling oils or particulate matter and generally reflect poor handling practices in distribution downstream of the refinery.

6. Apparatus

6.1 *Balance*, capable of weighing test specimens to the nearest ~~0.1 mg~~ 0.1 mg.

6.2 *Beakers*, of ~~100 mL~~ 100 mL capacity, as illustrated in Fig. 1. Arrange the beakers in sets, the number in each set depending upon the number of beaker wells in the evaporating bath. Mark each beaker in the set, including the tare beaker, with an identifying number or letter.

6.3 *Cooling Vessel*—A tightly covered vessel, such as a desiccator without desiccant, for cooling the beakers before weighing.

NOTE 2—The use of a desiccant could lead to erroneous results.

6.4 *Evaporation Bath* (**Warning—Warning!**—If a liquid-filled evaporation bath is used, care must be taken that the flash point of the liquid used is at least ~~30°C~~ 30 °C higher than the highest bath temperature expected.) Either a solid metal block bath or a liquid bath, electrically heated, and constructed in accordance with the general principles shown in Fig. 1 may be used. (Although all dimensions are given in SI units, older baths conforming to Test Method D381 – 94, or earlier, are suitable.) The bath should have wells and jets for two or more beakers. The rate of flow from each outlet when fitted with the conical adapters with ~~500 μm~~ 500 μm to ~~600 micron~~ 600 μm copper or stainless steel screens should not differ from ~~1000 mL~~ 1000 mL ~~mL/s~~ s by more than 15 %. A liquid

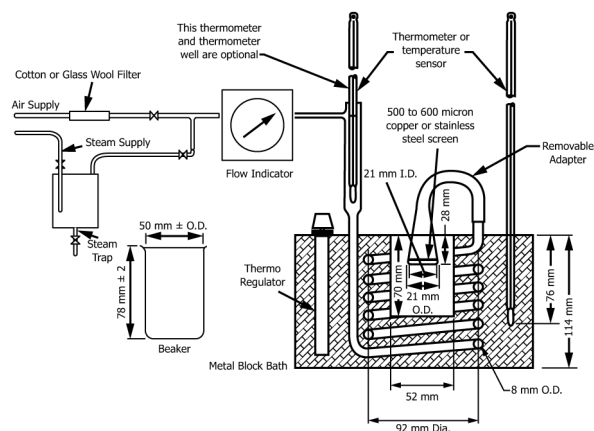


FIG. 1 Apparatus for Determining Gum Content by Jet Evaporation

bath, if used, shall be filled to within ~~25 mm~~ 25 mm of the top with a suitable liquid. Temperature shall be maintained by means of thermostatic controls or by refluxing liquids of suitable composition.

6.5 *Flow Indicator*, as illustrated in Fig. 1, such as a flowmeter, capable of metering a flow of air or steam equivalent to ~~1000~~ 1000 mL \cdot mL/s/s for each outlet.

NOTE 3—Alternatively, a pressure gage may be used to meter the flow of air or steam equivalent to ~~1000~~ 1000 mL/s \pm ~~150~~ 150 mL \cdot mL/s/s for each outlet.

6.6 *Sintered Glass Filtering Funnel*, coarse porosity, ~~150 mL~~ 150 mL capacity.

6.7 *Steam*—Supply by suitable means capable of delivering to the bath inlet the required amount of steam at ~~232~~ 232 °C to ~~246~~ 246 °C.

6.8 *Temperature Sensor*, liquid-in-glass thermometer conforming to the requirements in the specification(s) for ASTM 3C/IP73C, found in Specification E1, or another temperature measuring device or system, or both, of at least equivalent accuracy and precision over a temperature range from ~~5~~ 5 °C to ~~400~~ 400 °C.

6.9 *Graduated Cylinders*, with spout, capable of measuring ~~50~~ 50 mL \pm ~~0.5~~ 0.5 mL.

6.10 *Handling Equipment*, forceps (stainless steel, spade ended) or tongs (stainless steel) for use in handling the beakers and conical jets, as required by this test method.

7. Materials

7.1 *Air*—Supply of filtered air at a pressure not more than ~~35 kPa~~ 35 kPa.

7.2 *Gum Solvent*—A mixture of equal volumes of toluene and acetone.

7.3 *Heptane*—Minimum purity of 99.7 %.

7.4 *Steam*—Supply of steam free of oily residue and at a pressure not less than ~~35 kPa~~ 35 kPa. (**Warning—Warning If**—If a steam superheater is used, there may be exposed hot surfaces on the steam superheater. Avoid contact with exposed skin by use of protective equipment as required.)

8. Assembly of Air-Jet Apparatus

8.1 Assemble the air-jet apparatus as shown in Fig. 1. With the apparatus at room temperature, adjust the air flow to give a rate of ~~600~~ 600 mL/s \pm ~~90~~ 90 mL \cdot mL/s/s for the outlet under test. Check the remaining outlets for uniform air flow.

NOTE 4—A rate of ~~600~~ 600 mL/s \pm ~~90~~ 90 mL \cdot mL/s/s from each outlet, at room temperature and atmospheric pressure, will ensure delivery of ~~1000~~ 1000 mL/s \pm ~~150~~ 150 mL \cdot mL/s/s at the temperature of ~~155~~ 155 °C \pm ~~5~~ 5 °C for each outlet. It is recommended to follow the manufacturers' instructions to verify total flow/s (~~600~~ 600 mL \cdot mL/s/s air flow \times number of outlets = total flow/s) and uniformity from each outlet.

8.2 Apply heat to the evaporation bath (see 6.4) until the temperature of the bath is between ~~160~~ 160 °C and ~~165~~ 165 °C. Introduce air into the apparatus at a rate indicated on the flow indicator (see 6.5) from the exercise carried out in 8.1. Measure the temperature in each well with the temperature sensor (see 6.8) placed with the bulb or sensor tip resting on the bottom of the beaker in the well. Do not use any well having a recorded temperature outside the range from ~~150~~ 150 °C to ~~160~~ 160 °C.

9. Assembly of Steam-Jet Apparatus

9.1 Assemble the steam-jet apparatus as shown in Fig. 1. ~~(Warning—Warning)~~The sample and solvent vapors evaporated during the performance of this test procedure can be extremely flammable or combustible and hazardous from the inhalation standpoint. The evaporation bath must be provided with an effective exhaust hood to control such vapors and reduce the risk of thermal explosion.)

9.2 To place the apparatus in operation, apply heat to the bath. When the temperature reaches ~~232°C~~232 °C, slowly introduce dry steam into the system until a rate of ~~1000~~1000 mL/s \pm ~~150~~150 mL mL/s/s for each outlet is reached (see 10.2). Regulate the temperature of the bath to a range from ~~232~~232 °C to ~~246~~246 °C to provide a well temperature of ~~232~~232 °C \pm ~~3~~3 °C. Measure the temperature with the temperature sensor, placed resting on the bottom of a beaker in one of the bath wells with the conical adapter in place. Any well having a temperature that differs by more than ~~3~~3 °C from ~~232~~232 °C is not suitable for standard tests.

10. Calibration and Standardization

10.1 Air Flow:

10.1.1 Verify or calibrate the air flow to ensure all outlets meet the ~~600~~600 mL/s \pm ~~90~~90 mL mL/s/s air flow requirement as measured at room temperature and atmospheric pressure. Refer to the instrument manufacturer instructions for specific guidance on performing the air flow calibration procedure. Note the setting of the flow indicator device for use with air and use this setting for subsequent tests.

10.1.1.1 One way to calibrate the air flow is to use a calibrated flow indicator device, such as a flowmeter, separate from the device specified in 6.5, to check the air flow rate at each outlet directly at room temperature and atmospheric pressure. To obtain accurate results, ensure that the back pressure of the flowmeter is less than ~~1~~1 kPa.

10.1.1.2 Alternatively, another way to calibrate the air flow is to measure and adjust as appropriate the total air flow rate (mL/s) supplied to the outlets. The total air flow rate equals the expected air flow rate at each outlet times the number of outlet positions (for example, instrument has 5 positions and a total air flow rate measurement of ~~3000~~3000 mL mL/s/s, indicating an expected air flow rate of ~~600~~600 mL mL/s/s at each outlet). Once verifying the total flow supplied to the outlets is at the appropriate rate, perform uniformity checks by comparing the relative air flow rates at each outlet position versus the requirements in 10.1.1.

10.2 Steam Flow:

10.2.1 Verify or calibrate the steam flow to ensure all outlets meet the ~~1000~~1000 mL/s \pm ~~150~~150 mL mL/s/s steam flow requirement. Refer to the instrument manufacturer instructions for specific guidance on performing the steam flow calibration procedure. Note the setting of the flow indicator device for use with steam and use this setting for subsequent tests.

10.2.1.1 One way to calibrate the steam flow, is to attach a copper tube to a steam outlet and extend the tube into a ~~2-L~~2 L graduated cylinder that has been filled with crushed ice and water that has been previously weighed. Exhaust the steam into the cylinder for approximately ~~60 s~~60 s. Adjust the position of the cylinder so that the end of the copper tube is immersed in the water to a depth of less than ~~50 mm~~50 mm to prevent excessive back pressure. After the appropriate time has elapsed, remove the copper tube from the cylinder and weigh the cylinder. The gain in mass represents the amount of steam condensed. Calculate the steam rate (mL/s) as follows:

$$R = (M - m)1000/kt \quad (1)$$

where:

R = steam rate (mL/s),

M = mass of graduated cylinder with condensed steam, g,

m = mass of graduated cylinder and ice, g,

k = ~~mass of 1000 mL of steam at 232°C at atmospheric pressure = 0.434 g, and~~

k = mass of 1000 mL of steam at 232 °C at atmospheric pressure = 0.434 g, and

t = condensing time, s.

11. Procedure

11.1 Wash the beakers, including the tare, with the gum solvent until free of gum. Rinse thoroughly with water and immerse in a mildly alkaline or neutral pH laboratory detergent cleaning solution.

11.1.1 The type of detergent and conditions for its use need to be established in each laboratory. The criterion for satisfactory cleaning shall be a matching of the quality of that obtained with chromic acid cleaning solutions on used beakers (fresh chromic acid, ~~6-h~~6 h soaking period, rinsing with distilled water and drying). For this comparison visual appearance and mass loss on heating the glassware under test conditions may be used. Detergent cleaning avoids the potential hazards and inconveniences related to handling corrosive chromic acid solutions. The latter remains as the reference cleaning practice and as such may function as an alternate to the preferred procedure-cleaning with detergent solutions.

11.1.2 Remove the beakers from the cleaning solution by means of stainless steel forceps or tongs (see 6.10) and handle only with forceps or tongs thereafter. Wash the beakers thoroughly, first with tap water and then with distilled water, and dry in an oven at ~~150~~150 °C for at least ~~1-h~~1 h. Cool the beakers for at least ~~2-h~~2 h in the cooling vessel placed in the vicinity of the balance.