



Designation: D4672 – 18

Standard Test Method for Polyurethane Raw Materials: Determination of Water Content of Polyols¹

This standard is issued under the fixed designation D4672; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon (ϵ) indicates an editorial change since the last revision or reapproval.

1. Scope*

1.1 This test method measures water content of polyols and many other organic compounds.

1.2 The values stated in SI units are to be regarded as the standard. The values given in parentheses are for information only.

1.3 *This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety, health, and environmental practices and determine the applicability of regulatory limitations prior to use.*

NOTE 1—This test method is equivalent to ISO 14897.

1.4 *This international standard was developed in accordance with internationally recognized principles on standardization established in the Decision on Principles for the Development of International Standards, Guides and Recommendations issued by the World Trade Organization Technical Barriers to Trade (TBT) Committee.*

2. Referenced Documents

- 2.1 *ASTM Standards*:²
 - D1193 Specification for Reagent Water
 - D883 Terminology Relating to Plastics
 - E203 Test Method for Water Using Volumetric Karl Fischer Titration
 - E691 Practice for Conducting an Interlaboratory Study to Determine the Precision of a Test Method
- 2.2 *ISO Standards*:³
 - ISO 14897 Plastics—Polyols for use in the production of polyurethane—Determination of water content

¹ This test method is under the jurisdiction of ASTM Committee D20 on Plastics and are the direct responsibility of Subcommittee D20.22 on Cellular Materials - Plastics and Elastomers.

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² For referenced ASTM standards, visit the ASTM website, www.astm.org, or contact ASTM Customer Service at service@astm.org. For *Annual Book of ASTM Standards* volume information, refer to the standard's Document Summary page on the ASTM website.

³ Available from American National Standards Institute (ANSI), 25 W. 43rd St., 4th Floor, New York, NY 10036, http://www.ansi.org.

3. Terminology

3.1 Definitions:

3.1.1 *polyurethane, n*—a polymer prepared by the reaction of an organic diisocyanate with compounds containing hydroxyl groups.

3.1.1.1 *Discussion*—Polyurethanes, or urethanes, as they are sometimes called, can be thermosetting, thermoplastic, rigid or soft and flexible, cellular or solid. (See Terminology D883.)

4. Summary of Test Methods

4.1 This method is based essentially on volumetric or coulometric titrations that follow the reduction of iodine by sulfur dioxide in the presence of water. This reaction proceeds quantitatively when methanol or another alcohol (ROH) and pyridine (C₅H₅N) or a similar amine (R'N) are present to react with the sulfur trioxide (SO₃) and hydriodic acid (HI) produced according to the following reactions:



4.2 To determine water, Karl Fischer reagent (a solution of iodine, sulfur dioxide, imidazole, and pyridine or a pyridine substitute) is added to a solution of the sample in methanol or other alcohol until all of the water present has been consumed. The titrant is either added by buret (volumetry) or generated electrochemically in the titration cell (coulometry). Coulometric titrations eliminate the need for standardizing the reagent. Pyridine is less commonly used recently due to its toxicity. If pyridine is to be used, refer to the SDS for proper precautions.

4.3 This method provides details specific to water determinations in polyols. General guidance to the use of Karl-Fischer analysis, including a list of interferences, can be found in Test Method E203.

5. Significance and Use

5.1 This test method is suitable for quality control, as a specification test, and for research. The water content of a polyol is important since isocyanates react with water.

6. Apparatus

6.1 Several commercial Karl Fischer autotitrators are available that employ volumetric or coulometric titrations. These

*A Summary of Changes section appears at the end of this standard