
**Application of ISO 5725 for the
determination of repeatability and
reproducibility of precision tests
performed in standardization work
for chemical analysis of steel**

*Application de la norme ISO 5725 pour la détermination de la
répétabilité et la reproductibilité des essais de précision réalisés en
travaux de normalisation pour l'analyse chimique de l'acier*

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ISO copyright office
Ch. de Blandonnet 8 • CP 401
CH-1214 Vernier, Geneva, Switzerland
Tel. +41 22 749 01 11
Fax +41 22 749 09 47
copyright@iso.org
www.iso.org

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Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

The procedures used to develop this document and those intended for its further maintenance are described in the ISO/IEC Directives, Part 1. In particular the different approval criteria needed for the different types of ISO documents should be noted. This document was drafted in accordance with the editorial rules of the ISO/IEC Directives, Part 2 (see www.iso.org/directives).

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The committee responsible for this document is ISO/TC 17, *Steel*, Subcommittee SC 1, *Methods of determination of chemical composition*.

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Application of ISO 5725 for the determination of repeatability and reproducibility of precision tests performed in standardization work for chemical analysis of steel

1 Scope

This document describes how to determine the repeatability and reproducibility of precision tests performed within standardization work using the chemical analysis method. Specifically, this document explains the procedure for calculating precision, using precision test data of ISO 5725-3:1994, Table D.2 for the precision test in ISO 9647:1989 as an example.

The procedure of the international test for determining precision is described in ISO 5725-2 and ISO 5725-3.

2 Normative references

There are no normative references in this document.

3 Terms and definitions

No terms and definitions are listed in this document.

ISO and IEC maintain terminological databases for use in standardization at the following addresses:

- IEC Electropedia: available at <http://www.electropedia.org/>
- ISO Online browsing platform: available at <http://www.iso.org/obp>

4 Precision test

4.1 Structure of the precision test

The structure of the precision test normally used within standardization work using chemical analysis is a 3-factor, staggered-nested structure, as shown in [Figure 1](#).

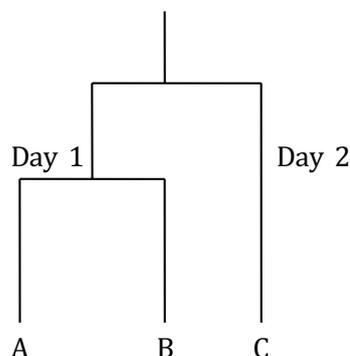


Figure 1 — Structure of the precision test

- a) Both A and B (Day 1) are obtained under repeatability where independent test results are obtained with the same method on identical items in the same laboratory by the same operator using the same equipment within short intervals of time.
- b) C (Day 2) is obtained under time-different intermediate precision conditions, except for the time factor. The measurement is performed by the same operator and, in addition, the measurements at a given level are performed using the same sample and equipment throughout.

4.2 Homogeneity of samples

For a precision test it is important to use homogeneous samples. Therefore, it is necessary to control the homogeneity of the samples selected for each precision test, if the samples are not certified reference materials, before starting a test in order to be sure that the heterogeneity level of the sample can be included in the expected precision values.

4.3 Number of laboratories and number of levels

In principle, the number of laboratories that participate in an international cooperative test is decided on the basis of the required precision. As this approach is often difficult to implement, the practical rule typically followed is the selection of 8 to 15 (or more) laboratories see ISO 5725-1:1994, 6.3.4, preferably in 5 countries. The number of levels depends on the range and scope of the method to be tested. A minimum of two levels by decade with the scheme (for example, 1–10) is required, and in the case of limited application ranges, three or more levels by decade (for example, 1–2–5–10) can be selected.

NOTE Fully-nested experiments offer higher reliability of repeatability than staggered-nested experiments. However, it will not improve the reliability of reproducibility significantly. From the standpoint of improving reliability, it is effective to increase the number of participating laboratories.

5 Representation of the experimental results

5.1 General

First, on the basis of precision test results, prepare the following tables and graphs.

5.2 Table of results and number of decimal places

Prepare a list of precision data.

In the list of data, the number of decimal places is the number required in the experiment plus one according to the convenor's requirement.

[Table 1](#) shows an example of a list of data.

The data obtained by laboratory i are indicated by y_{ij} ($j = 1, 2, 3$).

The symbol p represents the number of laboratories participating in the experiment. (It should be noted that the number changes if outliers are deleted.)

Table 1 — Original results

Lab. no.	Day 1		Day 2
	A	B	C
1	y11	y12	y13
—	—	—	—
i	yi1	yi2	yi3
—	—	—	—
p	yp1	yp2	yp3

5.3 Graphical representation of the data

5.3.1 General

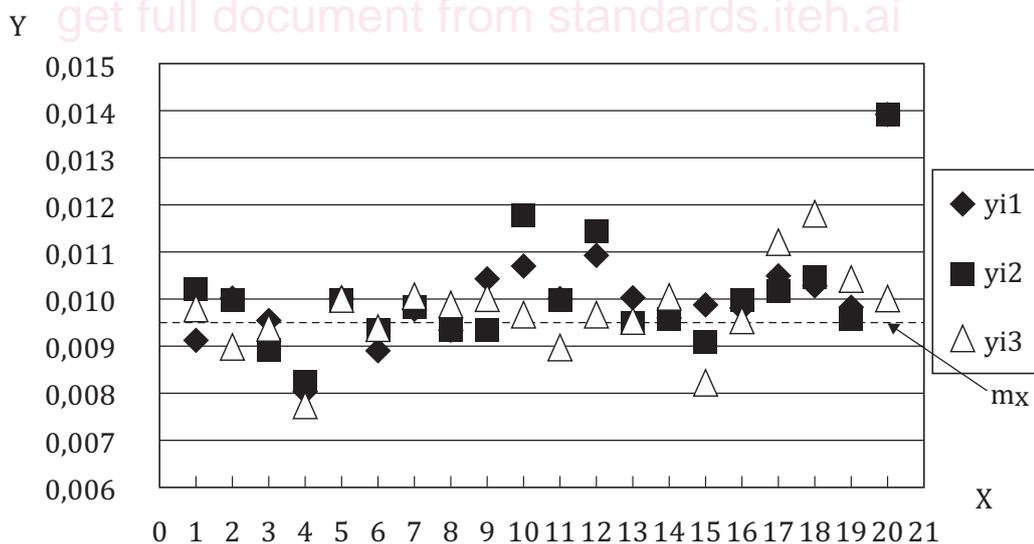
All data can be evaluated by graphical representation just to get an overview of the data population distribution. If there are laboratories which have obviously erroneous values for several levels, eliminating those laboratories as an outlier may be considered if deemed necessary.

5.3.2 Data plot

Draw a graph for each of the levels by plotting the data as follows.

- For each of the laboratories in [Table 1](#), plot Day 1 to 1 (= yi1), Day 1 to 2 (= yi2) and Day 2 (= yi3) using different symbols.
- Indicate the average m_x for each level.

An example of the graph for the original results is shown in [Figure 2](#).



Key

- X laboratory no.
- Y contents % (mass fraction)

Figure 2 — Original results